

GHS SAFETY DATA SHEET

GNB Industrial Power (A division of Exide Technologies 3950 Sussex Avenue I Aurora, IL 60504-7932 FOR FURTHER INFORMATION (Primary Contact: I Secondary Contact: I Joe Bolea (423) 989-6377 Fred Ganster (610) 921-4052 II. HAZARD Signal V Category: GH Health: STOT RE 2 Acute Tox. 4 Repr. 1A Skin Corr. 1A Flam. Gas 1 Aquatic Chronic 1 Aquatic Chronic 1 Aquatic Chronic 1 Aquatic Chronic 1 Aquatic Chronic 1 Aquatic Chronic 1	(703) 527-3 24-hour Em Ask for Env IDENTIFICATIO Vord: Danger S Codes I 302 H 314 C 332 H 360 M 373 M 220 E 410 V 260 I	(Calcium) UN2794 LY/ Electric Storage Battery C (800) 424-9300 887 – Collect ergency Response Contact vironmental Coordinator		
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Skin Corr. IA Flam. Gas 1 Aquatic Chronic 1 Aquatic Acute 1	301/330/331 I	F SWALLOWED: Inse mouth. Do NOT induce		
Aquatic Chronic 1		omiting.		
Aquetic Acute 1		F ON SKIN (or hair): Remove/Take off		
Aquetic Acute 1		mmediately all contaminated clothing. Rinse skin		
P.		vith water/shower.		
		F INHALED: Remove victim to fresh air and keep		
n'	a 305/351/338 I	t rest in a position comfortable for breathing. F IN EYES: Rinse cautiously with water for sever		
r.		ninutes. Remove contact lenses, if present and eas		
		o do. Continue rinsing.		
P		Immediately call a POISON CENTER or		
	d	loctor/physician.		
Pź		Keep away from heat/sparks/open flames/hot		
		urfaces. No smoking		
		Do not breathe dust/fume/gas/mist/vapors/spray		
		Wash thoroughly after handling.		
		Wear protective gloves/protective clothing/eye protection/face protection.		
Handling: PA		Store in well-ventilated area		
		Store locked up.		
		Collect spillage		
P	273 A	Avoid release to the environment		
	501 I	Dispose of contents/container in accordance with		
		ocal/regional/national/international regulation.		
WARNING: Batteries subjected to abusive charging at excessively hay create a surrounding atmosphere of the offensive strong inorg				

heat. Corrosive	to metals. Strong oxidizers, h	ydrogen peroxide, a	cids.		
	III. CO	MPOSITION/INFC	ORMATION ON	N INGREDIENTS	
Ingredient		CAS Number	% by Wt.		
Inorganic comp	ounds of:				
Lead		7439-92-1	52.4		
Lead Dioxid		1309-60-0	20.8		
	ous Ingredients	N/A	8.2		
Electrolyte (sulfuric acid)7664-93-919-44					
Note: Components are for a fully charged lead acid design. Inorganic lead and electrolyte (water and sulfuric acid solution) are the primary components of every battery manufactured by Exide Technologies or its subsidiaries. Other ingredients may be present dependent upon battery type. Polypropylene is the principal case material of automotive and commercial batteries. Electrolyte in this product is non-spill and completely absorbed within a solid matrix.					
IV. FIRST AID MEASURES					
Take proper p	recautions to ensure you own	health and safety l	before attemptin	ng to rescue a victim and provide first aid.	
Inhalation:	Electrolyte: Remove to fres				
Skin Contact:	including shoes, and do not v leather.	wear again until clear	ned. If acid is sp	utes; remove contaminated clothing completely, lashed on shoes, remove and discard if they contain compounds are not readily absorbed through the skin.	
Eye Contact:	-			mounts of water for at least 15 minutes; consult	
Ingestion: Electrolyte: Give large quantities of water; do not induce vomiting; consult physician. Lead compounds: Consult physician immediately.					
		V. FIRE FIGH	HTING MEASU	IRES	
Flash Point:	Not Applicable				
Flammable Lir		drogen gas in air) ; U	JEL = 74.2%		
Extinguishing media: CO ₂ ; foam; dry chemical					
Fire Fighting Procedures: Use positive pressure, self-contained breathing apparatus. Beware of acid splatter during water application and wear acid-resistant clothing, gloves, face and eye protection. If batteries are on charge, shut off power to the charging equipment, but, note that strings of series connected batteries may still pose risk of electric shock even when charging equipment is shut down.					
Hazardous Combustion Products: In operation, or when on charge, batteries generate and release flammable hydrogen and oxygen gases (hydrogen is highly flammable and oxygen supports combustion). They must always be assumed to contain this gas which, if ignited by burning cigarette, naked flame or spark, may cause battery explosion with dispersion of casing fragments and corrosive liquid electrolyte. Carefully follow manufacturer's instructions for installation and service. Keep away all sources of gas ignition and do not allow metallic articles to simultaneously contact the negative and positive terminals of a battery.					
VI. ACCIDENTAL RELEASE MEASURES					
Remove combustible materials and all sources of ignition. Stop flow of material and contain spill by diking with soda ash, etc. Carefully neutralize spill with soda ash, etc. Make certain mixture is neutral then collect residue and place in a drum or other suitable container with a label specifying "contains hazardous waste" or (if uncertain call distributor regarding proper labeling procedures). Dispose of as hazardous waste. If battery is leaking, place battery in a heavy duty plastic bag. Wear acid resistant boots, face shield, chemical splash goggles and acid resistant gloves. <i>Do not allow discharge of acid to sewer</i> . Acid must be managed in accordance with approved local, state, and federal requirements. Consult state environmental agency and/or federal EPA.					
VII. HANDLING AND STORAGE					
Handling: Single batteries pose no risk of electric shock but there may be increasing risk of electric shock from strings of connected batteries exceeding three 12-volt units. Batteries are non-spillable - potential for exposure to contents only during recycling or if outer casing is cracked or damaged.					
Storage: Store batteries under roof in cool, dry, well-ventilated areas that are separated from incompatible materials and from activities which may create flames, sparks, or heat. Keep away from metallic objects that could bridge the terminals on a battery and create a dangerous short-circuit.					

Charging:

There is a possible risk of electric shock from charging equipment and from strings of series connected batteries, whether or not being charged. Shut-off power to chargers whenever not in use and before detachment of any circuit connections. Batteries being charged will generate and release flammable hydrogen gas. Charging space should be ventilated. Keep battery vent caps in position. Prohibit smoking and avoid creation of flames and sparks nearby. Wear face and eye protection when near batteries being charged.

VIII. EXPOSURE CONTROLS AND PERSONAL PROTECTION

	Occupational Exposure Limits (mg/m ³)					
Ingredient:	US	US	US	Quebec	Ontario	EU
	OSHA	ACGIH	NIOSH	PEV	OEL	OEL
Inorganic forms of:						
Lead	0.05	0.05	0.05	0.05	0.05	0.15(a)
Lead Dioxide	0.05(b)	0.05(b)	0.05(b)	0.05(b)	0.05(b)	0.15(a,b)
Electrolyte (sulfuric acid/water solution)	1	0.2	1	1	0.2	0.05(c)

NOTES:

(a) as inhalable aerosol

(b) as inorganic lead

(c) thoracic fraction

Engineering Controls (Ventilation):

Store and handle in well-ventilated area. If mechanical ventilation is used, components must be acid-resistant. Handle batteries cautiously. Make certain vent caps are on securely. If battery case is damaged, avoid bodily contact with internal components. Wear protective clothing, eye and face protection, when charging or handling batteries. Follow all manufacturers' recommendations when stacking or palletizing. Do not allow metallic materials to simultaneously contact both the positive and negative terminals of the batteries. Use a battery carrier to lift a battery or place hands at opposite corners to avoid spilling acid through the vents. Avoid contact with internal components of the batteries.

Hygiene Practices:

Wash hands thoroughly before eating, drinking or smoking after handling batteries.

Respiratory Protection (NIOSH/MSHA approved):

None required under normal conditions. When concentrations of sulfuric acid mist are known to exceed PEL, use NIOSH or MSHA-approved respiratory protection.

Skin Protection:

None required under normal conditions. If battery case is damaged, use rubber or plastic acid-resistant gloves with elbow-length gauntlet, acid-resistant apron, clothing, and boots.

Eye Protection:

None required under normal conditions. If battery case is damaged, chemical goggles or face shield.

Other Protection:

In areas where water and sulfuric acid solutions are handled in concentrations greater than 1%, emergency eyewash stations and showers should be provided, with unlimited water supply.

	IX. PHYSICAL AND CHEMIC.	AL PROPERTIES - ELECTROLYTE	
Boiling Point@760 mm Hg	219 to 237° F	Specific Gravity @ 77°F (H ₂ O=1)	1.1394 to 1.3028
Melting Point	Not Applicable	Vapor Pressure (mm Hg)	13.5 to 20.8
% Solubility in Water	100	рН	Greater than 1
Evaporation Rate	Less Than 1	Vapor Density (AIR=1)	Greater than 1
(Butyl acetate=1)		Viscosity	Not applicable
Appearance and Odor Threshold	Sulfuric Acid: A clear liquid with a sharp, penetrating, pungent odor.	% Volatiles by Volume @70°F	Not Applicable
	A battery is a manufactured article; no apparent odor.		
Octanol Water	Not Applicable		
Partition			
Coefficient (Kow)			
Note: The properties	above reflect 20-40% Sulfuric acid		

X. STABILITY & REACTIVITY DATA

Stability: Stable

Conditions to Avoid: Prolonged overcharging and overheating current; sparks and other sources of ignition.

Incompatibilities: (materials to avoid)

<u>Electrolyte</u>: Contact with combustibles and organic materials may cause fire and explosion. Also reacts violently with strong reducing agents, most metals, carbides, chlorates, nitrates, picrate, sulfur trioxide gas, strong oxidizers, and water. Contact with metals may produce toxic sulfur dioxide fumes and may release flammable hydrogen gas. No further concern for mechanical impact.

Lead compounds: Avoid contact with strong acids, bases, halides, halogenates, potassium nitrate, permanganate, peroxides, nascent hydrogen, potassium, carbides, sulfides phosphorus, sulfur and reducing agents.

Hazardous Decomposition Products:

Electrolyte: Sulfur trioxide, carbon monoxide, sulfuric acid mist, sulfur dioxide, hydrogen sulfide, hydrogen.

<u>Lead compounds</u>: Temperatures above the melting point are likely to produce toxic metal fume, vapor, or dust; contact with strong acid or base or presence of nascent hydrogen may generate highly toxic arsine gas.

Hazardous Polymerization: Will Not Occur

XI. TOXICOLOGICAL DATA

Routes of Entry:

<u>Electrolyte</u>: Harmful by all routes of entry. Under normal conditions of use, sulfuric acid vapors and mist are not generated. Sulfuric acid vapors and mist may be generated when product is overheated, oxidized, or otherwise processed or damaged.

Lead compounds: Under normal conditions of use, lead dust, vapors, and fumes are not generated. Hazardous exposure can occur only when product is heated above the melting point, oxidized or otherwise processed or damaged to create dust, vapor, or fume.

Acute Toxicity:

Inhalation LD ₅₀ :	<u>Electrolyte</u> : LC ₅₀ rat: 375 mg/m ³ ; LC ₅₀ : guinea pig: 510 mg/m ³
	Elemental Lead: Acute Toxicity Point Estimate = 4500 ppmV (based on lead bullion)
Oral LD ₅₀ :	Electrolyte: rat: 2140 mg/kg
	Elemental lead: Acute Toxicity Estimate (ATE) = 500 mg/kg body weight (based on lead bullion)

Inhalation:

<u>Electrolyte</u>: Breathing of sulfuric acid vapors or mists may cause severe respiratory irritation. <u>Lead compounds</u>: Inhalation of lead dust or fumes may cause irritation of upper respiratory tract and lungs.

Ingestion:

<u>Electrolyte</u>: May cause severe irritation of mouth, throat, esophagus, and stomach. <u>Lead compounds</u>: Acute ingestion may cause abdominal pain, nausea, vomiting, diarrhea, and severe cramping. This may lead rapidly to systemic toxicity. Acute ingestion should be treated by physician.

Skin Contact:

<u>Electrolyte</u>: Severe irritation, burns, and ulceration. Sulfuric acid is not readily absorbed through the skin and is not a dermal sensitizier.

Lead compounds: Not readily absorbed through the skin and is not a dermal sensitizer.

Eye Contact:

<u>Electrolyte</u>: Severe irritation, burns, cornea damage, blindness. <u>Lead compounds</u>: May cause eye irritation.

Synergistic Products:

Electrolyte: No known synergistic products

<u>Lead compounds</u>: Synergistic effects have been noted with heavy metals (arsenic, cadmium, mercury), N-nitroso-N-(hydroxyethyl)ethylamine, N-(4-fluoro-4-biphenyl)acetamide, 2-(nitrosoethylamine)ethanol, and benzo[a]pyrene.

Additional Information:

Medical Conditions Generally Aggravated by Exposure:

Overexposure to sulfuric acid mist may cause lung damage and aggravate pulmonary conditions. Contact of electrolyte (water and sulfuric acid solution) with skin may aggravate skin diseases such as eczema and contact dermatitis. Contact of electrolyte (water and sulfuric acid solution) with eyes may damage cornea and/or cause blindness. Lead and its compounds can aggravate some forms of kidney, liver, and neurologic diseases.

Additional Health Data: All heavy metals, including the hazardous ingredients in this product, are taken into the body primarily by inhalation and ingestion. Most inhalation problems can be avoided by adequate precautions such as ventilation and respiratory protection covered in Section VIII. Follow good personal hygiene to avoid inhalation and ingestion: wash hands, face, neck and arms thoroughly before eating, smoking or leaving the work site. Keep contaminated clothing out of non-contaminated areas, or wear cover clothing when in such areas. Restrict the use and presence of food, tobacco and cosmetics to non-contaminated areas. Work clothes and work equipment used in contaminated areas must remain in designated areas and never taken home nor laundered with personal non-contaminated clothing. This product is intended for industrial use only and should be isolated from children and their environment. **XII. ECOLOGICAL INFORMATION** Environmental Fate: lead is very persistent in soil and sediments. No data on environmental degradation. Mobility of metallic lead between ecological compartments is slow. Bioaccumulation of lead occurs in aquatic and terrestrial animals and plants but little bioaccumulation occurs through the food chain. Most studies include lead compounds and not elemental lead. Environmental Toxicity: Aquatic Toxicity: 24-hr LC₅₀, freshwater fish (Brachydanio rerio): 82 mg/L Sulfuric acid: 96 hr-LOEC, freshwater fish (Cyprinus carpio): 22 mg/L 48 hr LC₅₀ (modeled for aquatic invertebrates): <1 mg/L, based on lead bullion Lead: XIII. DISPOSAL INFORMATION US Sulfuric Acid: Neutralize as described above for a spill, collect residue and place in a container labeled as containing hazardous waste. Dispose of as a hazardous waste. If uncertain about labeling procedures, call your local battery distributor or listed contact. DO NOT FLUSH LEAD CONTAMINATED ACID TO SEWER. Send to secondary lead smelter for recycling following applicable federal, state, and local regulations. Spent batteries XIV. TRANSPORT INFORMATION **GROUND – US-DOT/CAN-TDG/EU-ADR/APEC-ADR:** Batteries, Wet, Filled with Acid UN 2794, 8, PG III Label: "Corrosive" AIRCRAFT - ICAO-IATA: Batteries, Wet, Filled with Acid UN 2794.8 Label: "Corrosive" Reference IATA packing instructions 870 **VESSEL – IMO-IMDG:** Batteries, Wet, Filled with Acid UN 2794, 8 Label: "Corrosive" Reference IMDG packing instructions P801 **Additional Information:** - Batteries must be kept upright at all times and packaged as required to prevent short circuits. Transport may require packaging and paperwork, including the Nature and Quantity of goods, per applicable origin/destination/customs points as-shipped. XV. REGULATORY INFORMATION **United States: EPA SARA Title III** Section 302 EPCRA Extremely Hazardous Substances (EHS): Sulfuric acid is a listed "Extremely Hazardous Substance" under EPCRA, with a Threshold Planning Quantity (TPQ) of 1,000 lbs. EPCRA Section 302 notification is required if 500 lbs or more of sulfuric acid is present at one site (40 CFR 370.10). An average automotive/commercial battery contains approximately 5 lbs of sulfuric acid. Contact your Exide representative for additional information. Section 304 CERCLA Hazardous Substances: Reportable Quantity (RQ) for spilled 100% sulfuric acid under CERCLA (Superfund) and EPCRA (Emergency Planning and Community Right to Know Act) is 1,000 lbs. State and local reportable quantities for spilled sulfuric acid may vary. Section 311/312 Hazard Categorization:

EPCRA Section 312 Tier Two reporting is required for non-automotive batteries if sulfuric acid is present in quantities

of 500 lbs or more and/or if lead is present in quantities of 10,000 lbs or more.

Section 313 EPCRA Toxic Substances:

Supplier Notification: This product contains a toxic chemical or chemicals subject to the reporting requirements of section 313 of (Title) III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR Part 372.

Chemical	CAS	Percent by Weight
Lead (Pb)	7439-92-1	52.4
Electrolyte: Sulfuric Acid (H ₂ SO ₄)	7664-93-9	19-44
Lead Dioxide (PbO ₂)	1309-60-0	20.8

If you distribute this product to other manufacturers in SIC Codes 20 through 39, this information must be provided with the first shipment of each calendar year. **Note:** The Section 313 supplier notification requirement does not apply to batteries that are "consumer products".

- TSCA: Each ingredient chemical listed in Section III of this SDS is also listed on the TSCA Registry.
- OSHA: Considered hazardous under Hazard Communication Act (29CFR1910.1200)
- **RCRA:** Spent lead-acid batteries are not regulated as hazardous waste when recycled. Spilled sulfuric acid is a characteristic hazardous waste; EPA hazardous waste number D002 (corrosivity).
- CAA: Exide Technologies supports preventative actions concerning ozone depletion in the atmosphere due to emissions of CFC's and other ozone depleting chemicals (ODC's), defined by the USEPA as Class I substances. Pursuant to Section 611 of the Clean Air Act Amendments (CAAA) of 1990, finalized on January 19, 1993, Exide established a policy to eliminate the use of Class I ODC's prior to the May 15, 1993 deadline.

NFPA Hazard Rating for sulfuric acid:

AFTA Hazaru Kaung Ior	summer act	u.
Flammability (Red)	=	0
Health (Blue)	=	3
Reactivity (Yellow)	=	2

	rity (Yellow) =	2					
US State Notifications & Warnings:	Identification		Notifications/Warning				
California	California Proposition 65		"WARNING: This product contains			of	
			California to cause cancer, or birth de			1	
			Battery posts, terminals, and related accessories contain lead and lead compounds, chemicals known to the State of California to cause cancer and reproductive harm. Batteries also contain other chemicals known to the State of California to cause cancer. Wash hands after handling.				
			The following chemicals identified to	exist in the finish	ned product as dist	tributed	
			into commerce are known to the State				
			or to cause reproductive harm:		,		
			 Strong inorganic acid mists inclu Lead and lead compounds; CAS 			-44% wt	
	Consumer Product Volatile		This product is not regulated as a consumer product for purposes of CARB/OTC				
	Organic Compound Emis	sions	/OC Regulations, as sold for the intended purpose and into the ndustrial/commercial supply chain.				
Country/Organization Iden		Identi	fication	Notifications/W	Varning		
Canada		listed	emical substances in this product are on the CEPA DSL/NDSL or are ot from list requirements.	accordance with Controlled Prod SDS contains al by the Controlle Refer to the Cor	s been classified in a the hazard criteri ucts Regulations a l the information of Products Regula ntrolled Products	ia of the and the required	
				Regulations for requirements	product labeling		
		NPRI	and Ontario Regulation 127/01	This product con chemicals subje	ntains the followin ct to the reporting Canada NPRI and <u>CAS #</u>	5	
				compounds Sulfuric acid	7439-92-1 7664-93-9	73.2 19-44	
		Toxic	Substances List	Lead			

EU	European Inventory of Existing Commercial Chemical Substances (EINECS):	All ingredients remaining in the finished product as distributed into commerce are exempt from, or included on, the European Inventory of Existing Commercial Chemical Substances.				
XVI. OTHER INFORMATION						
DATE ISSUED: September 11, 2013						
OTHER INFORMATION: SOURCES OF INFORMATION:	Regulations (CP Distribution into Import/Export o International Ag Monographs on Overall Evaluati Monographs Vo Ontario Ministry	Quebec to follow Canadian Controlled Product R) 24(1) and 24(2). the EU to follow applicable Directives to the Use, f the product as-sold. ency for Research on Cancer (1987), IARC the Evaluation of Carcinogenic Risks to Humans: ons of Carcinogenicity: An updating of IARC lumes 1-42, Supplement 7, Lyon, France. of Labor Regulation 654/86. Regulations osure to Chemical or Biological Agents.				
PREPARED BY: GNB INDUSTRIAL POWER A DIVISION OF EXIDE TECHNOLOGIES 3950 SUSSEX AVENUE AURORA, IL 60504-7932						
VENDEE AND THIRD PERSONS ASSUME THE RISK OF INJURY PROXIMATELY CAUSED BY THE MATERIAL IF REASONABLE SAFETY PROCEDURES ARE NOT FOLLOWED AS PROVIDED FOR IN THE DATA SHEET, AND VENDOR SHALL NOT BE LIABLE FOR INJURY TO VENDEE OR THIRD PERSONS PROXIMATELY CAUSED BY ABNORMAL USE OF THE MATERIAL EVEN IF REASONABLE PROCEDURES ARE FOLLOWED.						
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ANY I	PHOTOCOPY MUST BE OF THIS ENTIRE	DOCUMENT				