

1.17922.0001

## MQuant® Percectic Acid Test

### 1. Method

Peracetic acid reacts with a phenol derivative to form a violet dye. The concentration of peracetic acid is measured **semiquantitatively** by visual comparison of the reaction zone of the test strip with the fields of a color scale.

### 2. Measuring range and number of determinations

Measuring range / color-scale graduation	Number of determinations
500 - 1000 - 1500 - 2000 mg/l peracetic acid	100

### 3. Applications

This test is suited for the selective determination of the peracetic acid concentration in disinfectant solutions, also in cases in which hydrogen peroxide is present.

### 4. Influence of foreign substances

This was checked individually in solutions with 1000 and 0 mg/l peracetic acid. The determination is not yet interfered with up to the concentrations of foreign substances given in the table. Cumulative effects were not checked; such effects can, however, not be excluded.

Concentrations of foreign substances in mg/l or °e	
NO <sub>3</sub> <sup>-</sup>	1000
Free chlorine (hypochlorite)	50
Formaldehyde	1000
Total hardness	37.5 °e
H <sub>2</sub> O <sub>2</sub>	10 000

### 5. Reagents and auxiliaries

**The test strips are stable up to the date stated on the pack when stored closed at +2 to +8 °C.**

#### Package contents:

Tube containing 100 test strips

#### Other reagents:

MQuant® Universal indicator strips pH 0 - 14,  
Cat. No. 1.09535  
Sodium hydroxide solution 1 mol/l Titripur®,  
Cat. No. 1.09137  
Hydrochloric acid 1 mol/l Titripur®,  
Cat. No. 1.09057

### 6. Preparation

- Samples containing more than 2000 mg/l peracetic acid must be diluted with distilled water.
- **The pH must be within the range 2 - 10.** Adjust, if necessary, with sodium hydroxide solution or hydrochloric acid.

### 7. Procedure

Immerse the reaction zone of the test strip in the pretreated sample (**15 - 25 °C**) for **2 sec.** Shake off excess liquid from the strip and **after 30 sec** determine with which color field on the label the color of the reaction zone coincides most exactly. Read off the corresponding result in mg/l peracetic acid.

#### Notes on the measurement:

- The color of the reaction zone may continue to change after the specified reaction time has elapsed. This must not be considered in the measurement.
- If the color of the reaction zone is equal to or more intense than the darkest color on the scale, repeat the measurement using **fresh**, diluted samples until a value of less than 2000 mg/l peracetic acid is obtained. Concerning the result of the analysis, the dilution (see also section 6) must be taken into account:

Result of analysis = measurement value x dilution factor

### 8. Note

**Reclose the tube containing the test strips immediately after use.**

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EMD Millipore Corporation, 400 Summit Drive  
Burlington MA 01803, USA, Tel. +1-978-715-4321

Sigma-Aldrich Canada Co. or Millipore (Canada) Ltd.  
2149 Winston Park, Dr. Oakville, Ontario, L6H 6J8  
Phone: +1 800-565-1400

www.sigmaldrich.com/mquant

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