

Tut_1:PROBLEM:1

- **1). A large slab of aluminium, 25 cm thick, is poured in a sand mould with no superheat.**
- **a) How long will it take the slab to solidify? (Assume negligible resistance to heat flow at the mould-metal interface and within the solidified metal.)**
- **b) Show schematically cooling curves for a thermocouple 6.25 cm from the surface and one at the centre of the casting. (Remember, the solidified portion of the casting remains very close to T_M until the casting is completely solid.)**

T:1,P:1:SOLUTION

$$S = \frac{2}{\sqrt{\pi}} \left[\frac{T_M - T_o}{\rho_s H} \right] \sqrt{K_m \rho_m c_m} \sqrt{t} \quad t = \left[S \frac{\sqrt{\pi}}{2} \left(\frac{\rho_s H}{T_M - T_o} \right) \frac{1}{\sqrt{K_m \rho_m c_m}} \right]^2$$

$$S = \text{thickness} / 2 = 12.5 \text{ mm}$$

$$T_M = 660 \text{ }^\circ\text{C}, T_o = 20 \text{ }^\circ\text{C},$$

$$\rho_s = 2.7 \text{ g/cc}$$

$$H = 95 \text{ cal/g}$$

$$t = \left[12.5 \frac{\sqrt{\pi}}{2} \left(\frac{2.7 \times 95}{640} \right) \frac{1}{\sqrt{14.5 \times 10^{-4} \times 1.5 \times 0.27}} \right]^2$$

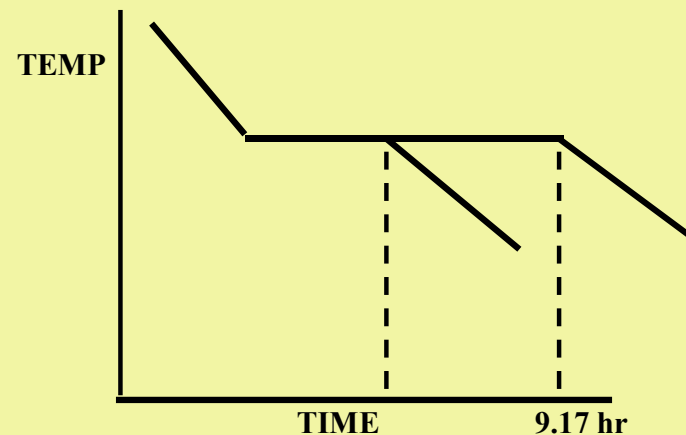
$$= 33031 \text{ seconds}$$

$$= 9.17 \text{ hours}$$

MOULD CONSTANTS FOR SAND:

$$K_m = 14.5 \times 10^{-4} \text{ cal/cm }^\circ\text{C s}$$

$$\rho_m = 1.5 \text{ g/cc}, c_m = 0.27 \text{ cal/g }^\circ\text{C}$$



2. An approximate method of calculation of solidification time of a sand casting poured with superheat is to add the heat content associated with the superheat to the heat of fusion.

How long would it take the slab casting of Problem 1 to solidify if it were poured with 100 C superheat?

TS:1,P:2:SOLUTION

100 °C SUPERHEAT

$$H = 95 + (100 \times 0.26) = 121$$

$$t = (12.5)^2 \frac{\pi}{4} \left[\frac{2.7 \times 121}{640} \right] \frac{1}{14.5 \times 10^{-4} \times 0.27 \times 1.5}$$

$$t = 54453 \text{ s} \quad \rightarrow \text{ANSWER}$$

$$t = 15 \text{ hours}$$

3. Aluminium is splat cooled on a copper substrate with a mould-metal interface heat-transfer coefficient of $1 \text{ cal/cm}^2 \text{ C s}$.

(a) Assuming heat transfer is interface-controlled, how thin must the sample be to achieve a cooling rate (in the liquid at just above the melting point) of 10^6 C/s ?

(b) For the thickness calculated above, is it reasonable to assume interface-controlled heat transfer?

TS:1,P:3:SOLUTION

$$Q/A = h (T_M - T_o)$$

THE RATE AT WHICH THE SPLAT LOSES HEAT IS GIVEN BY:

$$Q/A = \rho C (dT/dt) d$$

d = THICKNESS OF THE SPLAT

$$H = 1 \text{ cal/m}^2 \cdot \text{C} \cdot \text{s} ; T_M = 660 \text{ C} ; T_o = 20 \text{ C}$$

$$\rho = 2.7 ; c = 0.26 ; dT/dt = 10^6 \text{ C/s}$$

$$d = \frac{1 \times 640}{2.7 \times 0.26 \times 10^6} = 9.11 \times 10^{-4} \text{ cm}$$

= Answer

4. One of the advantages claimed for magnesium die-casting compared with aluminium is a shorter solidification time.

Show that this is true for processes where heat transfer is controlled at the mould-metal interface.

IN INTERFACE CONTROLLED SOLIDIFICATION:

$$t_f = \frac{\rho_s H}{h [T_M - T_o]} \frac{V}{A}$$

**COMPARING RELATIVE SOLIDIFICATION TIMES,
WE HAVE THE RATIO:**

$$\frac{(t_f)_{Mg}}{(t_f)_{Al}} = \frac{[\rho_s H]_{Mg} [T_M - T_o]_{Al}}{[T_M - T_o]_{Mg} [\rho_s H]_{Al}} = \frac{1.7 \times 89}{630} \times \frac{640}{2.7 \times 95}$$

$$= 0.59$$

$$(t_f)_{Mg} = 0.59 \times (t_f)_{Al}$$

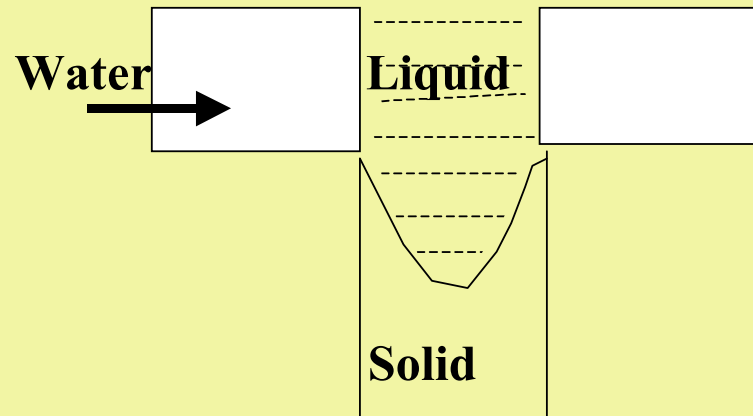
**THUS MAGNESIUM DIE
CASTINGS SOLIDIFY IN 60 % OF
THE TIME FOR ALUMINIUM
CASTINGS.**

5. You are to help design a vertical continuous-casting machine to make 8-in thick low-carbon steel slabs.

Casting speed to be aimed for is 100 in/min. What is the minimum height casting tower you must plan to build?

Note: Heat flow in the longitudinal direction can be neglected. You may use the data for pure iron.

TS:1,P:5:SOLUTION



THUS WE NEED TO HAVE A MINIMUM HEIGHT OF THE TOWER EQUAL TO 52 ft.

THICKNESS REQUIRED TO SOLIDIFY = 4 IN

FOR A WATER COOLED CHILL

$$S = 1.6 \times \sqrt{t}$$

**FOR $S = 4$ in, $t = 6.25$ minutes
IN 6.25 MINUTES, THE SLAB
WOULD TRAVEL**

$$6.2 \times 100 = 625 \text{ in} = 52 \text{ ft}$$

6. Give possible reasons for the good mould-filling characteristics which are exhibited by pure metals and eutectic alloys during the casting of small sections.

Discuss them with regard to the interface morphology.

(A) .The superior mould-filling characteristics of pure metals and eutectics during casting, especially of small sections, can be attributed to the fact that both materials exhibit smooth interface morphologies when solidifying in a positive temperature gradient at the wall of a small channel. Consequently, the liquid can flow easily through these channels.

(B).In the case of alloys, which have a solidification range, dendritic interfaces form and rapidly obstruct the channels. As a result, small sections may not be fed with the liquid metal, leading either to widespread porosity or complete absence of part of the casting.

A plate casting of dimension 200 mm × 15 mm × 25 mm was separately cast in two moulds having the properties as shown in Table.

- (i) Determine the solidification times.
- (ii) Rank the dendritic arm spacing.
- (iii) Rank the mechanical properties variation amongst the two castings.

Property	Sand I	Sand II
Density, ρ_s (kg/m^3)	1100	3400
Specific heat, c_s ($\text{Joule kg}^{-1} \text{K}^{-1}$)	930	600
Thermal conductivity, k ($\text{Watts m}^{-1} \text{K}^{-1}$)	0.71	1.7

TS:2, P:1

(i) RATIO OF SOLIDIFICATION TIMES =

$$= (\text{solidification time})_{\text{sand II}} / (\text{solidification time})_{\text{sand I}}$$

$$= (k_m \times \rho_m \times c_m)_{\text{sand I}} / (k_m \rho_m c_m)_{\text{sand II}}$$

$$(1100 \times 930 \times 0.71)_{\text{sand I}} / (3400 \times 600 \times 1.7)_{\text{sand II}}$$

$$716100 / 3468000 = 0.21$$

(ii) DENDRITE ARM SPACING (DAS) IS PROPORTIONAL TO \sqrt{t}

$$(\text{DAS})_{\text{sand II}} / (\text{DAS})_{\text{sand I}} =$$

$$[(\text{solidification time})_{\text{sand II}} / (\text{solidification time})_{\text{sand I}}]^{1/2}$$

$$= (0.21)^{0.5} = 0.447$$

(iii) MECHANICAL PROPERTIES (ULTIMATE TENSILE STRENGTH AND % ELONGATION) ARE INVERSELY PROPORTIONAL TO DAS.

**(MECHANICAL PROPERTIES OF THE CASTING CAST IN SAND II)/
(MECHANICAL PROPERTIES OF THE CASTING CAST IN SAND I)**

$$= (1/DAS)_{\text{sand II}} / (1/DAS)_{\text{sand I}}$$

$$= 1/0.447 = 2.24$$

CASTING POURED IN SAND II SOLDIFIES FASTER, SMALLER DAS AND HIGHER MECHANICAL PROPERTIES THAN CASTING POURED IN SAND I