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SwissBiogas.com presents: SBGx desulphurisation and gas booster additive

v1.32b



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Desulphurisation, why?

- Protects the CHP unit
- Protects the engines
- Raises the efficiency of the whole plant
- Lowers the operating costs
- Increases the operational safety



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Current desulphurisation methods:

- No reactor intervention
- Air injection
 Addition of
 - Ferric chloride, FeCl₃
 - Iron oxide-hydroxide, FeO(OH)
 - Iron oxide, Fe_xO_y



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Air injection, why not:

- Impairs fermentation and methane generation (Hinge 2014)
- Atmospheric nitrogen is unnecessarily added to the biogas (Ammonia)
- Excessive introduction of oxygen leads to formation of corrosive sulphurous acid
- % O₂ higher than 0.9% before CHP has a negative effect on efficiency of CHP
- Sulphur deposits formed in the fermenter gas compartment break off from time to time and drop into the substrate
- Risk of explosion
- Increased corrosion: S + O₂ + H₂O → H₂SO₄
- The air flow should be controlled and adapted regularly:
 - a) Less gas production but same air injection results in too much O₂ and dilution
 - b) Higher gas production with same air injection results in too less O₂ with weak desulphurisation effect



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Ferric chloride, why not:

- Reduces volume of produced biogas (12% 32%)
 (Dentel 1982, Johnson 2003, Smith 2008, 2009, Dauknys 2011)
- Forms HCI which penetrates the bacterial membrane
- Classified as dangerous substance
- Considered a corrosive substance
- Releases chlorides during desulphurisation → More corrosive than H₂S

CORROSIVE



IRRITANT





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Iron oxide-hydroxide, why not:

Iron oxide-hydroxide binds metals.

These materials are known to bind a wide range of other compounds from water, including trace metals, arsenic, selenium, silicate, and organics. Metals such as manganese, cobalt, nickel, and zinc are known to bind to iron oxide hydroxide in simulated seawater solutions. It has also been claimed that the binding of copper and zinc by natural iron oxide-hydroxide sediments exerts a powerful control on the concentration of copper and zinc in polluted rivers and estuaries. Although not studied in seawater, it has also been observed that phosphate binding by iron oxide-hydroxide actually increases its binding of copper, cadmium, and nickel in freshwater.

Reference: http://www.reefkeeping.com/issues/2004-11/rhf/



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Iron oxide, why yes:

- Eliminates struvite formation
- Lowers H₂S very effectively
- Increases methane production
- Prevents corrosion of plant and CHP
- Reduces HRT



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What happens (inside the reactor):

$$2Fe^{3+} + H_2S \rightarrow 2Fe^{2+} + S + 2H^+$$
 (1)

$$2Fe^{2+} + (1/2) O_2 + H_2O \rightarrow 2Fe^{3+} + 2OH^{-} (2)$$





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Comparison of desulphurisation methods, SBGx by SwissBiogas.com against others:

	SBGx	Iron oxide	Iron oxide-hydroxide	Iron chloride	Air injection
Investment into					
Storage and Handling	low	low	low	high	none
Dosing Equipment	none / low	none / low	medium	medium	medium
Risk of / to					
Exposure / Personnel	low	low	low	high	none
Explosion	low	low	low	low	high
Corrosion	low	low	low	high, HCl	high, H ₂ SO ₄
Incompatibility	low	low	low	high	high
Gas Impurities	low	low	low	low	high
Reaction Products	none	none		HCI	H ₂ SO ₄
Characteristics					
Reactive Content	> 60%	30% - 60%	10% - 15%	10% - 14%	none
Digestion Speed / Volume	high	low	low	high	low
Deposit Effect	high	high	medium	none	none
Methanogen Growth	increased	normal	normal	negative	negative
Gas Yield over Normal	higher	normal	normal	negative	negative
Trace Element Addition	not necessary	required	required	required	required
Shelf Life	> 12 months	> 12 months	< 12 months	< 12 months	none
Price per chem. Reaction	medium	high	medium	high	none



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The components of SBGx as base before individual adjustments to customers' requirements

Note: The industry-leading high content of reactive Fe_xO_y

SBGx	[%]
FeO + Fe ₂ O ₃	> 70
MnO	< 1.0
С	< 0.5
Cr_2O_3	< 0.5
CaO	< 0.5
K ₂ O	< 0.5
Al_2O_3	< 0.5
SiO ₂	< 0.5
ZnO	< 0.5
Na ₂ O	< 0.2
CuO	< 0.2
S	< 0.1
TiO ₂	< 0.1
MgO	< 0.1
NiO	< 0.1
P ₂ O ₅	< 0.1
MoO ₃	< 0.05
V_2O_5	< 0.01
BaO	< 0.01
CoO	< 0.01
PbO	< 0.01
CdO	< 0.01
SnO ₂	< 0.01
WO ₃	< 0.01
CI	< 0.01
SeO ₂	< 0.01

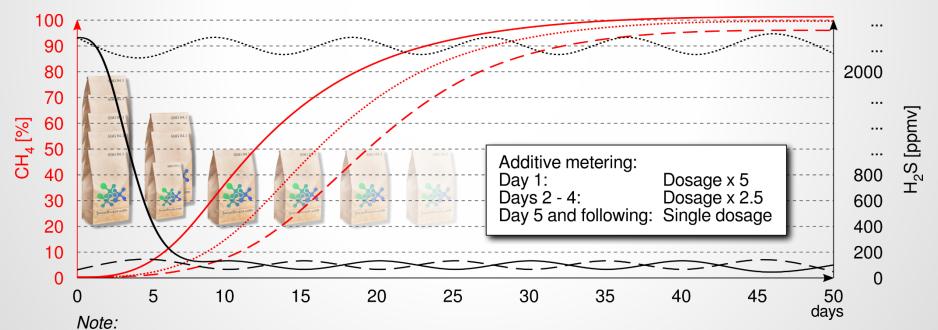


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Sample application of SBGx in practice

Comparison of CH₄ and H₂S developments

Legend	CH_4	H_2S
 SBGx treated, solid 		
 Untreated, dotted 		
 FeCl, treated dashed 		



100% CH_4 = Total reclaimed CH_4 by untreated process within 50 days



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How we pack and ship:

Delivered in powder form, in

- 20 kg bags, or as per
- individual requirements

Please contact your agent for availability.



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Where we test:







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2017 Test results by specific waste category: Kitchen Waste

Additive: EG 1117 (old naming convention)

Duration: 51 days

Increase of methane production overall:

Average: 13.5%High: 19.3%

Increase rate of methane production per unit VS:

Average: 16.3%High: 23.2%

Speeding up of fermentation rate; Peak of daily methane production:

Average w/o additive: Day 13, 183.1 ml/day

Average w/ additive: Day 10, 200.5 ml/day

High: Day 10, 206.7 ml/day

Remark: All results above achieved within 43 days!

Orig. report (2018, Q1) available upon request

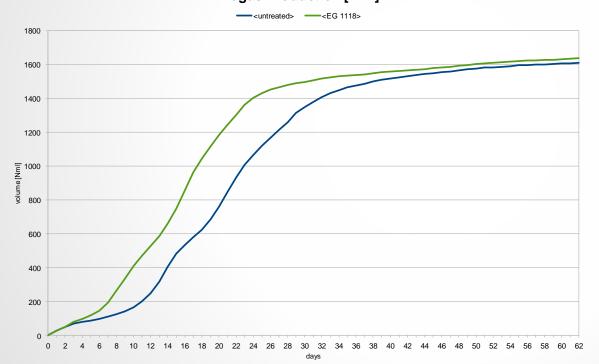




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2018 Effectivity test by specific waste category: Corn Straw





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The maximum duration of the test periods was 79 days. As some tests were terminated between day 62 and 79, only the first 62 days are analysed.

During the first 30 days, the tests showed an increased effectivity in biogas production of the substrate treated with the SBG additive EG 1118 (green, old naming convention).

Several tests were performed and the graphs show the average ("<...>") biogas production.

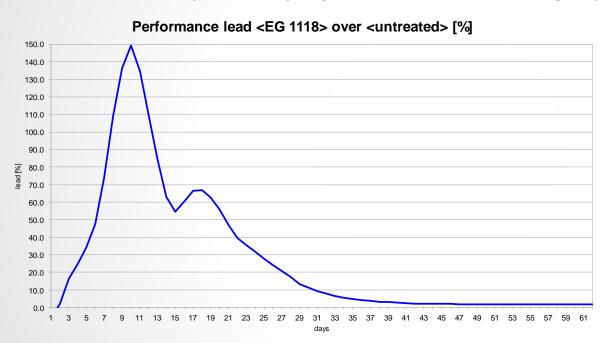
Total produced biogas with the untreated substrate after 62 days: 1609.7 Nml





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2018 Effectivity test by specific waste category: Corn Straw



If a short retention time in a reactor is key, the substrate treated with the EG 1118 additive has a significant advantage during the first 30 days. (see above)

Even at day 30, the EG 1118 additive treated substrate still has a lead of 11.1% in biogas production over the untreated substrate.

- untreated: 1347.1 Nml (83.7% of 1609.7 Nml)
- EG 1118: 1497.2 Nml (93.0% of 1609.7 Nml)

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The graph shows the advantage [%] of the additive treated substrate over the untreated substrate.

If the total produced biogas with the untreated substrate after 62 days (1609.7 Nml) is defined as 100%, 90% was reached after:

- 34 days with the untreated substrate (1448.0 Nml)
- 26 days with the EG 1118 additive treated substrate (1449.6 Nml); Performance lead: 24.3%





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2018 Test at a German biogas plant (EG 1118, old naming convent.)

	Input					Output				***************************************
Duration	Liq. Pig Man.	Grain	Maize	Recirculate	Total	CH₄	CO ₂	O_2	H₂S	Energy
	[kg]	[kg]	[kg]	[kg]	[kg]	[Vol%]	[Vol%]	[Vol%]	[ppm]	[kWh]
5.9-16.9	18'639.00	551.00	26'330.00	77'377.00	122'897.00	51.00	42.00	0.90	108.00	12'307.00
12 days prio	(40.95%)	(1.21%)	(57.84%)							***************************************
27.10-7.11	14'894.00	495.00	24'708.00	75'854.00	115'951.00	50.00	48.00	1.10	77.00	12'999.00
last 12 days	(37.14%)	(1.23%)	(61.62%)							***************************************
Change [%]	-20.09	-10.16	-6.16	-1.97	<i>-5.65</i>	-1.96	14.29	22.22	-28.70	5.62

Methane potential calculation, based on "Biogas yield by substrate", page 815 Production from selected plants and by-products:

Substrate	Biogas	CH₄ ratio	CH₄
	[m³/t]	[%]	[m³/t]
Pig Slurry:	500	65	325.0
Grain Decoction:	560	66	369.6
Maize Silage:	580	52	301.6

Source: THE EFFECT OF SUBSTRATE ON THE AMOUNT AND COMPOSITION OF BIOGAS IN AGRICULTURAL BIOGAS PLANT Nr III/2/2015, POLSKA AKADEMIA NAUK, Oddział w Krakowie, s. 809–818 Komisja Technicznej Infrastruktury Wsi
By: Joanna Kazimierowicz, Bialystok University of Technology

Link: http://dx.medra.org/10.14597/infraeco.2015.3.2.065

Comparison of the reactor's methane potential based on the substrate input:

Duration	Pig Slurry	Grain Dec.	Maize	Recirculate	Total CH₄
	[m³]	[m³]	[m³]	[m³]	[m³]
5.9-16.9	6'057.68	203.65	7'941.13	24'141.99	38'344.44
27.10-7.11	4'840.55	182.95	7'451.93	23'600.56	36'075.99
Change [%]	-20.09	-10.16	-6.16	-2.24	-5.92

Assum	otion:
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The recirculate has the same mix ratio as the freshly added substrate.

 $\rightarrow \qquad \text{If a potential of } 38,344.44 \text{ m}^3 \text{ of CH}_4 \text{ result in measured} \\ 12,307.00 \text{ kWh of energy, then a potential of} \\ 36,075.99 \text{ m}^3 \text{ of CH}_4 \text{ should mathematically result in} \\ 11,578.92 \text{ kWh (-,5.92\%) of energy, without additives.} \\$

Duration	Energy w	o additives	Energy w/ additives	
		[kWh]		[kWh]
5.9-16.9		12'307.00		
27.10-7.11	calculated:	11'578.92	measured:	12'999.00
Change [%]		-5.92		

Effect of the SwissBiogas.com additives:

Increase from 11,578.92 kWh to 12,999.00 kWh:



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