

LESSON PLAN OF CHEMICAL ENGINEERING DEPARTMENT

Discipline: Chemical	Semester:3rd	Name of Faculty: Siba Prasad Achary
Subject: Industrial Stoichiometry (TH-4)	No of Days per week class allotted:4	
Week	ClassDay	TheoryTopics
1 st	1 st	CHAPTER-1:UNITSANDDIMENSIONS Introduction
	2 nd	Basicandderivedunitsusedinprocessindustry.
	3 rd	Unitsofphysicalandchemicalproperties
	4 th	Relationbetweenunitsanddimension
2 nd	1 st	Unitconversionandsolvenumerical
	2 nd	Conceptsofunitoperationandunitprocess
	3 rd	Applicationofvariousgraphs in process calculation
	4 th	Solvenumerical
3 rd	1 st	CHAPTER-2:MOLECONCEPT Atomicnumber,atomicweightofelements
	2 nd	Mol.Wt.,moleunit,molefraction(orpercent)andmassfraction(orpercent),
	3 rd	Relationbetweenmoleandmassfraction
	4 th	Moleconceptwithrespecttochemicalequation.
4 th	1 st	Principleofatomconservation.
	2 nd	Molecalculationfromreaction
	3 rd	Methodsofexpressingcompositionofmixturesandsolutions
	4 th	Solverelatednumericals
5 th	1 st	CHAPTER-3:STOICHIOMETRY Introduction
	2 nd	Conceptoflimitingreactant,Atomicweight,
	3 rd	ConceptofMolecularweightandempiricalformula
	4 th	Solvednumericalbasedonlimitingreactant,mass-mass andmass volumebasis
6 th	1 st	ConceptsofEq.weight,valenceofmolecule
	2 nd	Solverelatednumerical
	3 rd	Conceptsofpreparationofsolution
	4 th	Weightandvolumepercentofsolutions
7 th	1 st	BasicsofNormality,molarityandmolality
	2 nd	Numericalonsolutionpreparation
	3 rd	Solverelatednumerical
	4 th	CHAPTER-4:GASESANDGASEOUSMIXTURES

		Introduction
8 th	1 st	Define gases, different gaseous mixture
	2 nd	Derivation of Ideal gas equation
	3 rd	Derive average molecular weight and Values of R
	4 th	Derivation of density of gas mixture
9 th	1 st	Solve related numerical
	2 nd	Composition by vol% and by weight% related to average molecular weight of gas mixture
	3 rd	Solve the examples and exercises related to Avg. mol wt. and Ideal gas equation.
	4 th	Concepts of Pressure, partial pressure and various laws related to PVT behavior.
10 th	1 st	Concepts of State Raoult's law and Henry's law
	2 nd	CHAPTER-5: MATERIAL BALANCE WITHOUT CHEMICAL REACTION Introduction
	3 rd	Basics of chemical equation and stoichiometry
	4 th	Concepts of law of conservation of mass and material balance over the reaction.
11 th	1 st	Material balance problems without chemical reactions of unit operations
	2 nd	Material balance of Evaporation and solvent numerical
	3 rd	Material balance of mixing and solvent numerical
	4 th	Material balance of crystallization
12 th	1 st	Material balance over distillation and solvent numerical
	2 nd	Material balance over drying and solvent related numerical
	3 rd	Material balance humidification and solvent related numerical
	4 th	Material balance over filtration
13 th	1 st	Material balance over absorption, extraction
	2 nd	Solvent numerical
	3 rd	CHAPTER-6: MATERIAL BALANCE WITH CHEMICAL REACTION Introduction
	4 th	Concepts of Limiting reactant, Excess reactant
14 th	1 st	Concepts of Conversion, Selectivity, Yield.
	2 nd	Basic concepts involved in material balance calculations.
	3 rd	Material balance over combustion
	4 th	Material balance over chemical reaction calculation
15 th	1 st	Concepts of heat of combustion and heat of formation.
	2 nd	Concept of recycle and bypass, purge
	3 rd	Excess air and theoretical air
	4 th	Numerical based on combustion, Excess air and theoretical air