(#158) Dear Management:

October 3, 2019

Today, let's re-visit our 6th grade chemistry class... and the periodic table.  
  
Let's start with an assumption - you are a cable, made of metal...  
  
But what type of metal you ask???  Well that is the test...  
  
A few fun facts first:  
-----"Coinage" metals (such as silver) are not reactive with water.  
-----"Alkali" metals (such as lithium) are VERY reactive (they actually explode when exposed to water).  
----- All metals conduct electricity.  
  
So on to the test...  
  
As a cable of metal, we're gonna use you to connect the outside generator, to your office building.  
  
Now there's a rain storm coming... and the electricity must keep flowing, in order to keep the production machines running.  
  
So, if electricity is the leadership communication of your company... and rain is conflict (internal and/or external)...  
  
What type of "leader metal" are you made of... when the rain storm hits?  
  
Are you gonna be silver and keep conducting electricity smoothly and effectively?  Or are you lithium and just explode with the first few rain drops?  
  
You decide for yourself... but I'm gonna be silver.  
  
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