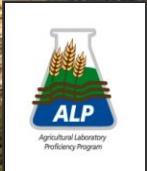


# FRST Lime Project Overview

Robert O. Miller  
*ALP Technical Director*  
*Fort Collins, CO*

August 29, 2022

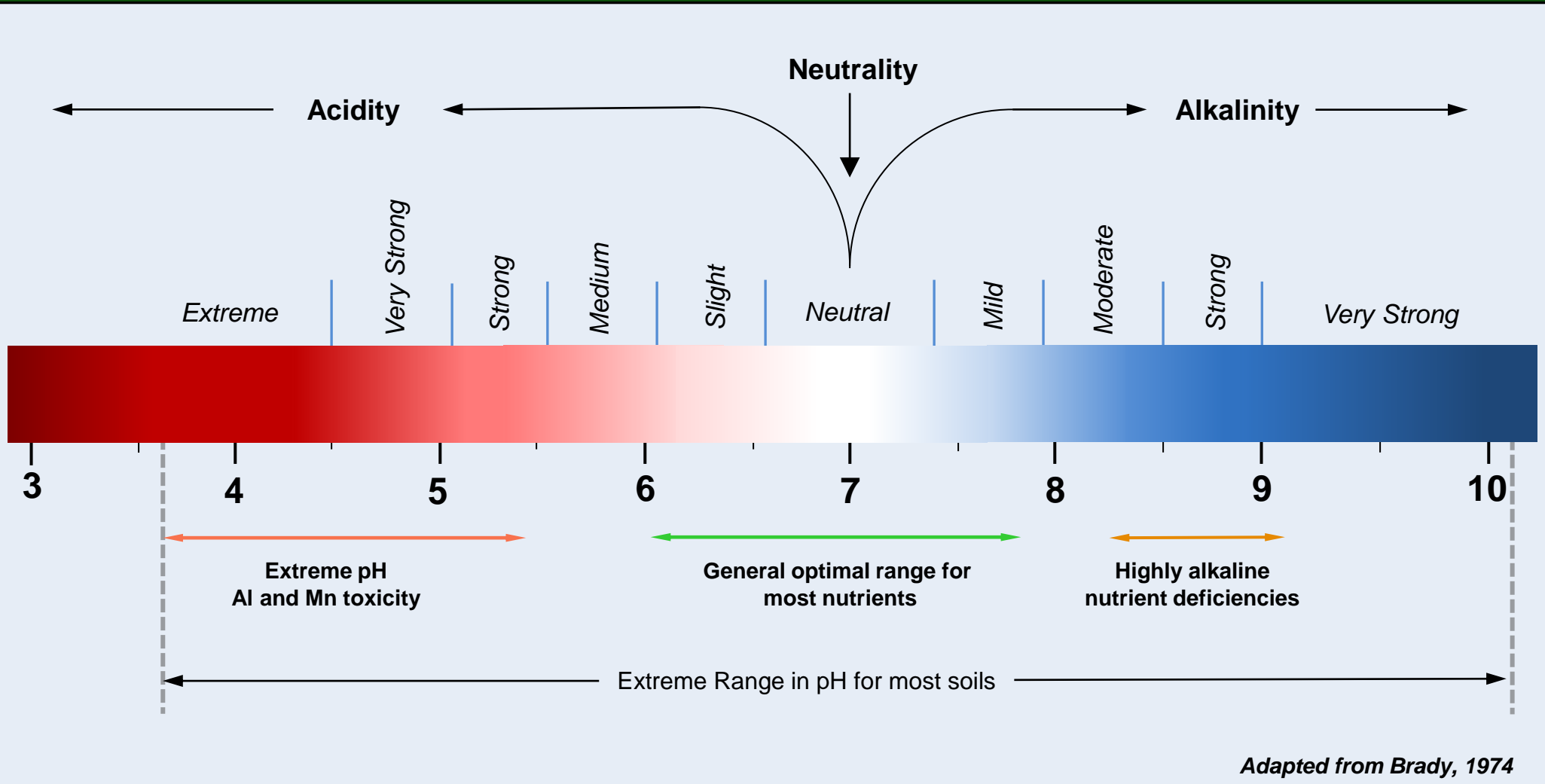


# *Background*

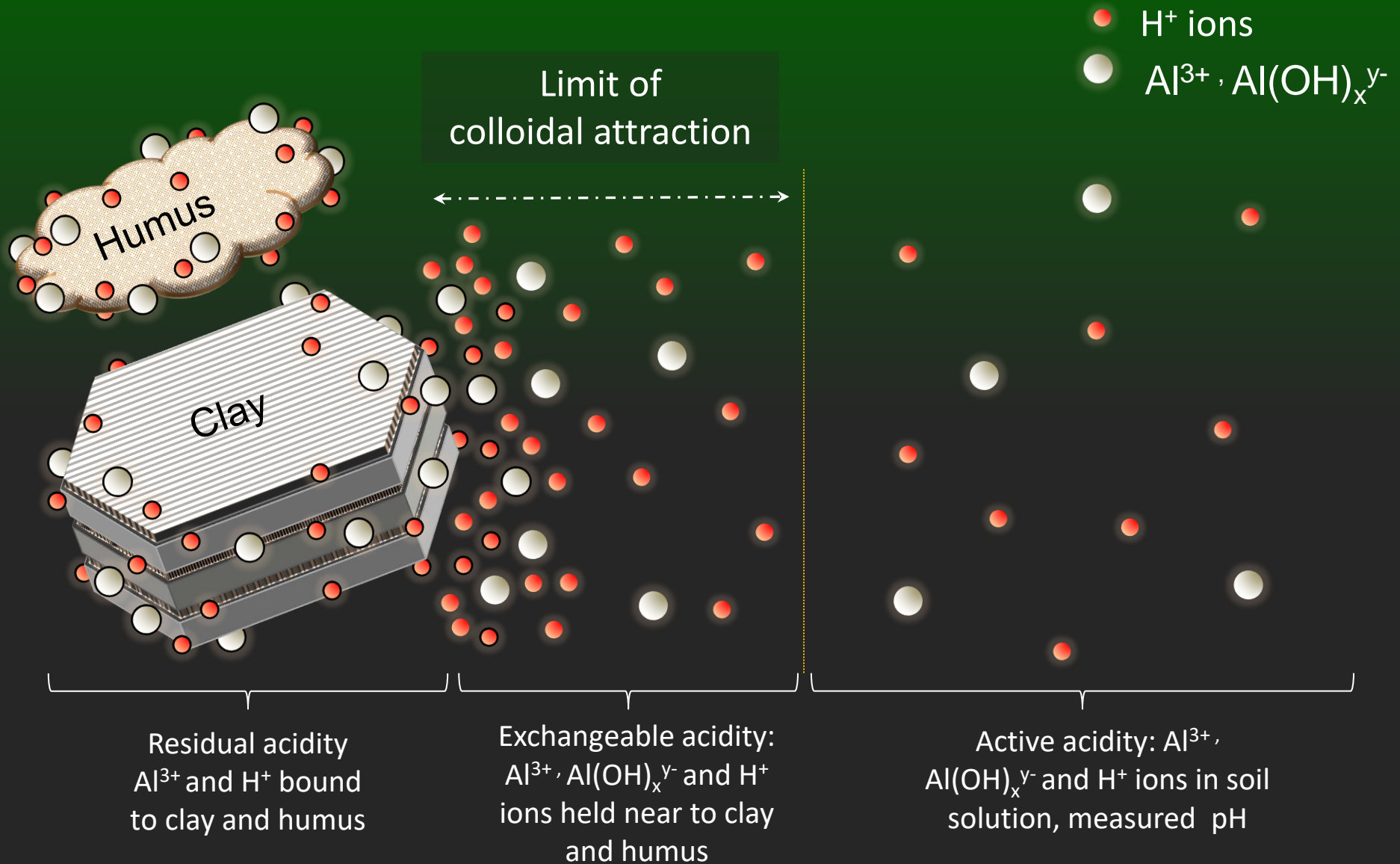
## Overview of pH, US distributions and impact of acid soils



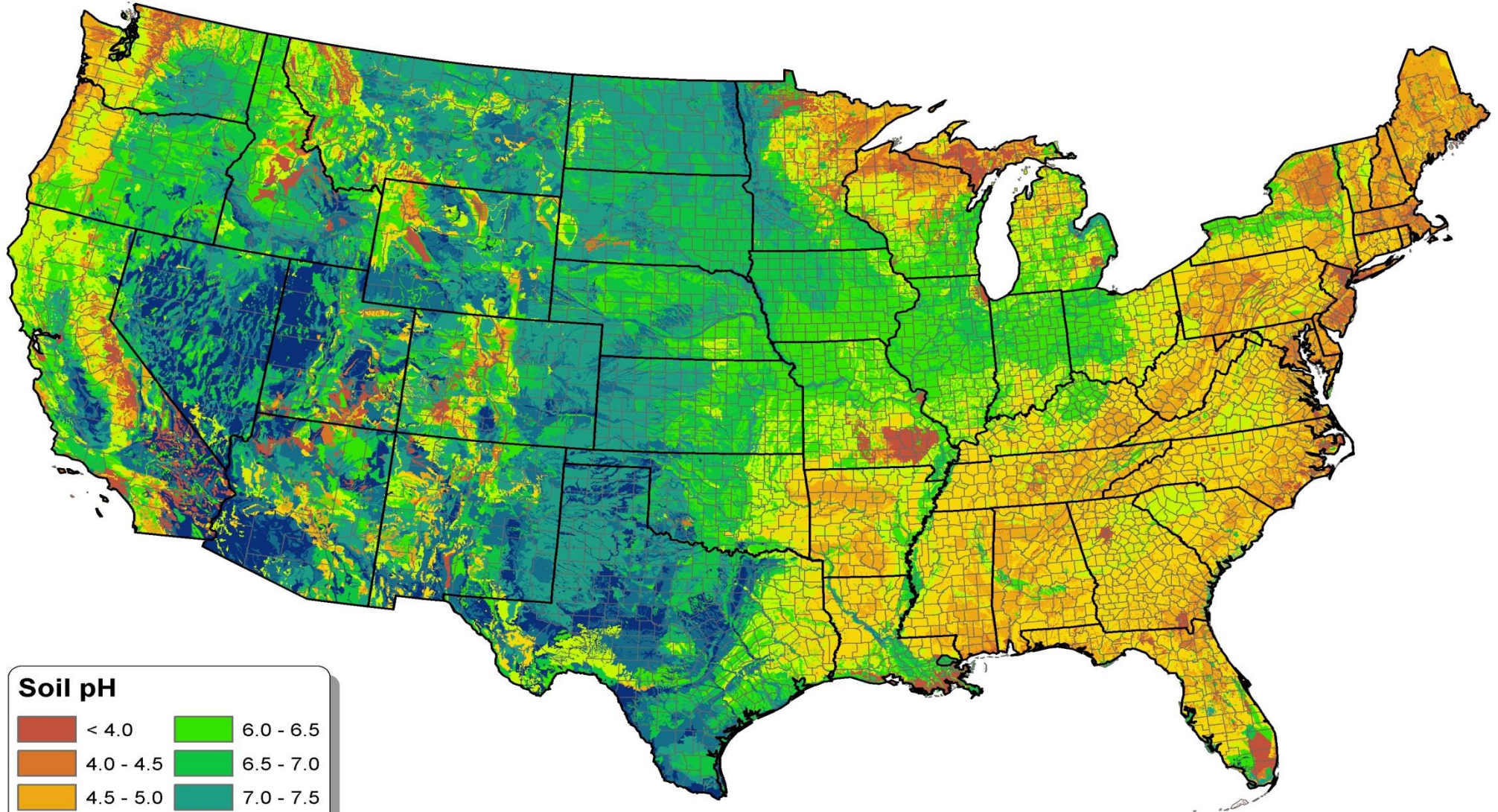
# Soil pH: Measure of acidity or alkalinity



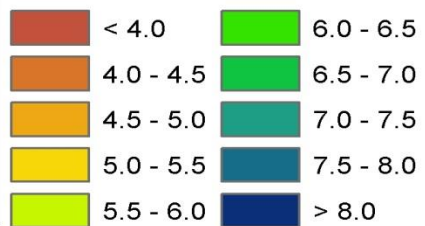
# Soil Acidity: Active vs Reserve



# US Soil pH distribution



## Soil pH

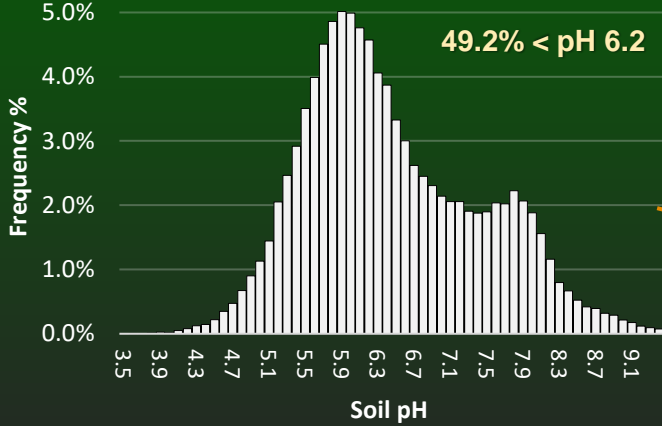


<https://forages.oregonstate.edu/sites/forages.oregonstate.edu/files/ph.jpg>

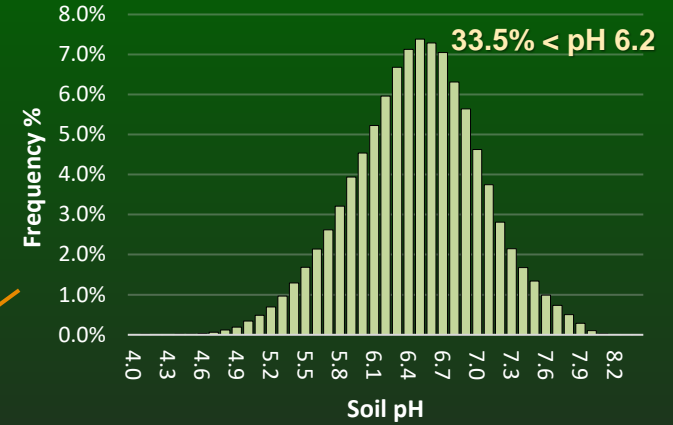
No referenced on source of map info.

# Regional soil pH distributions

Soil Test, WA, 2018



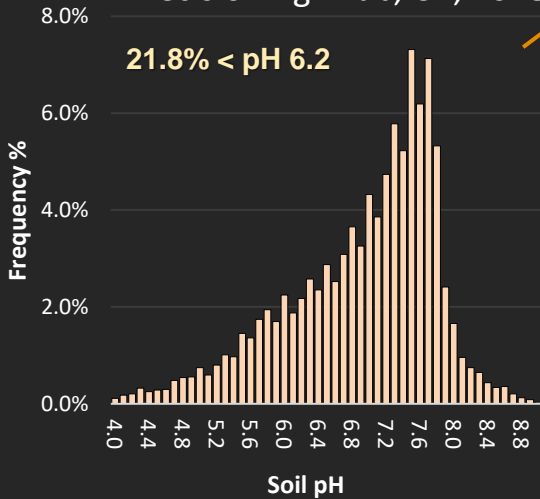
Sure-Tech Lab, IN, 2018



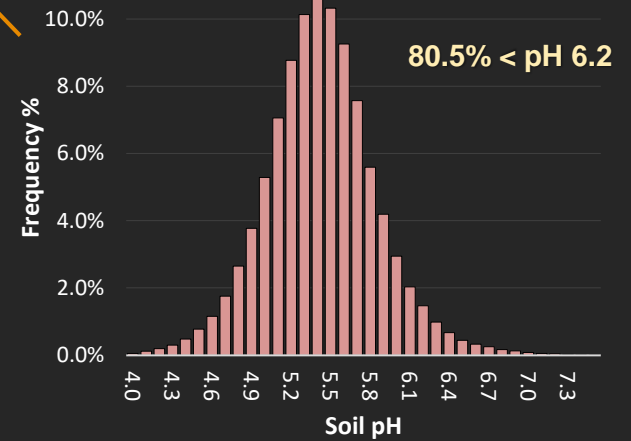
Soil  
pH 1:1 H<sub>2</sub>O



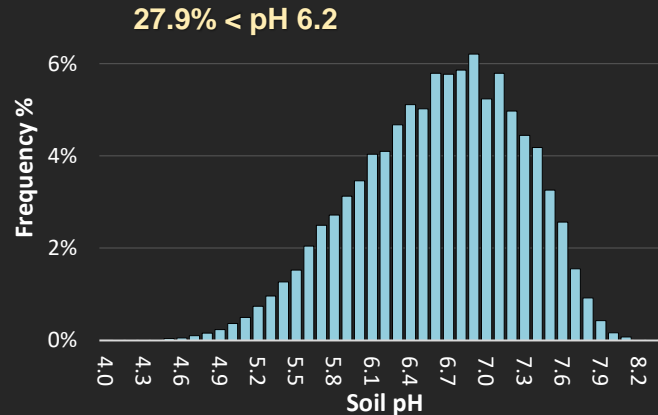
Precision Agri Lab, CA, 2018



Soil pH NCDA, NC, 2018



Rock River Lab, WI, 2018

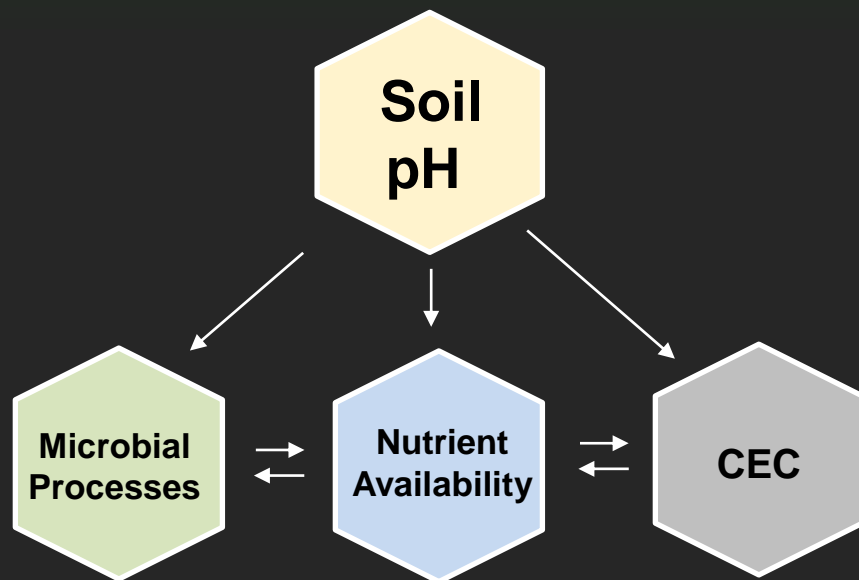


# Impact of soil pH

Nutrient availability / toxicity  
Cation Exchange Capacity (CEC)  
Base Saturation  
Nitrification  
Microbial processes

Nutrient	pH									
	4	4.5	5	5.5	6	6.5	7	7.5	8	8.5
Nitrogen										
Phosphorus										
Potassium										
Sulfur										
Calcium										
Magnesium										
Iron										
Manganese										
Boron										
Copper/Zinc										
Molybdenum										

**Note:** Darker shading indicates greater availability.



Low pH suppresses plant vigor  
and may result in Al and Mn  
toxicity for specific crops

# Overview soil pH and liming



Acid soils are extensive with medium to strongly acid soils constituting 20 – 80% of soils tested in the US, all of which would result in a lime recommendation ( $\text{pH} < 6.2$ ), estimate 3 million lab samples annually. Soil acidity impacts fertility, Al toxicity, CEC and crop productivity.

Agriculture used an estimated 11 M metric tons of ag lime in 2020, valued at \$660 million. VRT lime application has become a standard of the industry, optimizing spatial placement and minimizing cost.

Lime recommendations are a state specific and often based on calibration research conducted more than 40 years ago using past cultural practices.

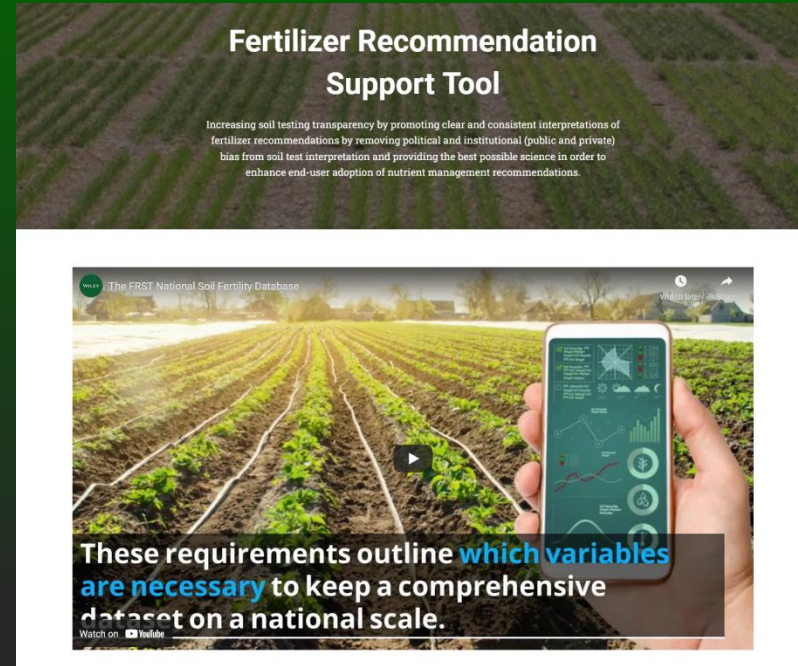


# FRST Project

The FRST (Fertilizer Recommendation Tool) was initiated in 2016 to develop promote clear and consistent soil testing interpretations of fertilizer recommendations for nutrient management, and act as a catalyst for innovation.

The FRST project team is comprised of university and government researchers across the US. Initially FRST has focused on P and K soil test calibration and recommendations. Additional projects have focused on soil sampling and a SOP for conducting P and K calibration research.

In January 2022 established the FRST Lime Project.



[soiltestfrst.org](https://soiltestfrst.org)

# FRST Lime Project Team

Brian Arnall	Oklahoma State University	Nathan Nelson	Kansas State University
Steve Culman	Ohio State University	Deanna Osmond	North Carolina St University
Luke Gatiboni	North Carolina St University	Rasel Parvej	Louisiana State University
David Hardy	NC Dept of Ag	Tim Pilkowski	USDA-NRCS
Joseph Heckman	Rutgers University	Manbir Rakkar	Montana State University
Bryan Hopkins	Brigham Young University	Ed Rayburn	West Virginia University
Sindhu Jagadamma	University of TN	Dorivar Ruiz Diaz	Kansas State University
Clain Jones	Montana State University	Kurt Schroeder	University of Idaho
John Jones	University of Wisconsin	Amy Shober	University of Delaware
Brian Kalmbach	University of Maryland	Frank Sikora	University of Kentucky
Quirine Ketterings	Cornell University	Nathan Slaton	University of Arkansas
Jay Lessl	University of Georgia	Jared Spackman	University of Idaho
Andrew Margenot	University of Illinois	Haiying Tao	University of Connecticut
Robert Miller	Colorado State University	Gurpal Toor	University of Maryland
Amber Moore	Oregon State University	Matt Yost	Utah State University
Stephanie Murphy	Rutgers University	Hailin Zhang	Oklahoma State University
Rao Mylavarapu	University of Florida		

# FRST Lime Project: objectives

Review the basis of: soil acidity, soil pH methods, BpH methods, lime rates and lime factors.



Lime recommendation survey, across the US.

Generate new lime calibration data and establish new regional or and/or national recommendations.



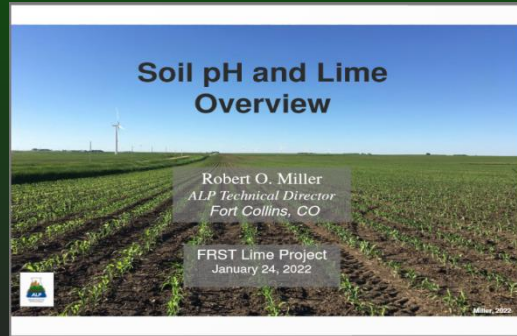
Develop a forecast model for predicting long term soil pH

Engage the laboratory industry and agronomist practioners in adopting and validating new lime recommendations.



# FRST Lime Project: presentations

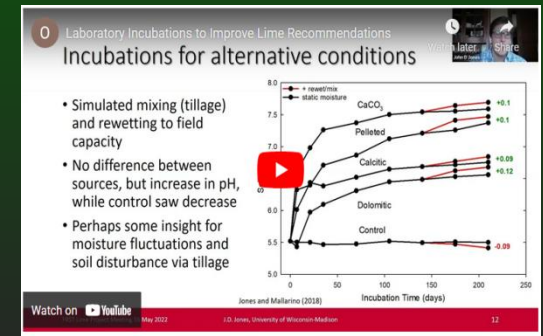
## pH overview



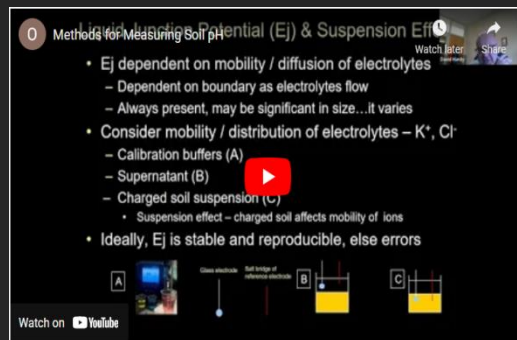
## BpH method



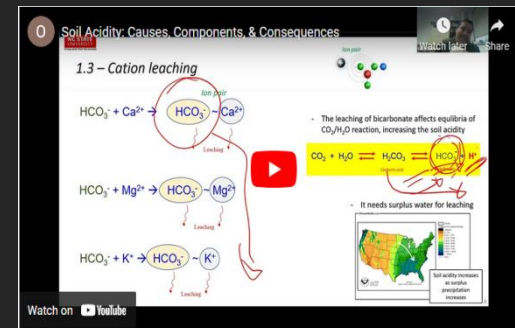
## Lime incubation



## pH method



## Soil acidity



[soiltestfrst.org/lime/](http://soiltestfrst.org/lime/)

# Current lime recommendations

- BpH lime recommendation table
- Algorithm base on  $pH_{H_2O}$  and BpH
  - i. Soil depth adjustment
  - ii. Target final pH
  - iii. Adjustment for mineral vs organic soils
  - iv. Lime neutralizing value (NV) adjustment

## Iowa State University

Buffer pH	Depth of Soil to be Neutralized								
	2 inches			3 inches			6 inches		
	Target Soil pH								
	pH 6.0	pH 6.5	pH 6.9	pH 6.0	pH 6.5	pH 6.9	pH 6.0	pH 6.5	pH 6.9
	Amount of Calcium Carbonate to Apply (pounds/acre) ‡								
7.0	0	0	400	0	0	600	0	0	1,100
6.9	0	0	600	0	0	1,000	0	0	1,900
6.8	0	200	900	0	300	1,400	0	600	2,700
6.7	0	400	1,200	0	700	1,800	0	1,300	3,500
6.6	0	700	1,500	0	1,100	2,200	0	2,100	4,400
6.5	100	900	1,700	100	1,400	2,600	200	2,800	5,200
6.4	300	1,200	2,000	400	1,800	3,000	800	3,500	6,000
6.3	500	1,400	2,300	700	2,100	3,400	1400	4,200	6,800

## University of Wisconsin

Target pH	Lime requirement formula <sup>a</sup> (tons/a 60–69 lime to apply <sup>b</sup> )
5.2	36.1 – (3.29 x BpH) – (2.67 x WpH)
5.4	48.2 – (4.84 x BpH) – (3.03 x WpH)
5.6	51.0 – (5.40 x BpH) – (2.67 x WpH)
5.8	57.2 – (5.55 x BpH) – (3.50 x WpH)
6.0	72.7 – (7.59 x BpH) – (3.78 x WpH)
6.3	103 – (12.6 x BpH) – (3.18 x WpH)
6.5	134 – (17.2 x BpH) – (2.73 x WpH)
6.6	152 – (20.3 x BpH) – (2.17 x WpH)
6.8	195 – (28.4 x BpH) + (0.144 x WpH)

<sup>a</sup> Abbreviations: BpH = buffer pH, WpH = water pH.

Buffer Method – SMP/Sikora. Nutrient application guidelines for field, vegetable, and fruit crops in Wisconsin (A2809).

## University of Delaware

Water pH	Adams-Evans Buffer pH						
	7.60	7.55	7.50	7.45	7.40	7.35	7.30
5.9	0.00	0.00	0.00	0.50	0.50	0.50	0.50
5.8	0.50	0.50	0.50	0.50	0.50	0.50	0.50
5.7	0.50	0.50	0.75	0.75	0.75	1.00	1.00
5.6	0.75	0.75	0.75	1.00	1.00	1.00	1.25
5.5	0.75	0.75	1.00	1.00	1.25	1.25	1.25
5.4	1.00	1.00	1.00	1.25	1.25	1.50	1.50
5.3	1.00	1.25	1.25	1.50	1.50	1.50	1.75
5.2	1.00	1.25	1.50	1.50	1.75	1.75	2.00
5.1	1.25	1.25	1.50	1.75	1.75	2.00	2.00
5.0	1.25	1.50	1.75	1.75	2.00	2.00	2.25

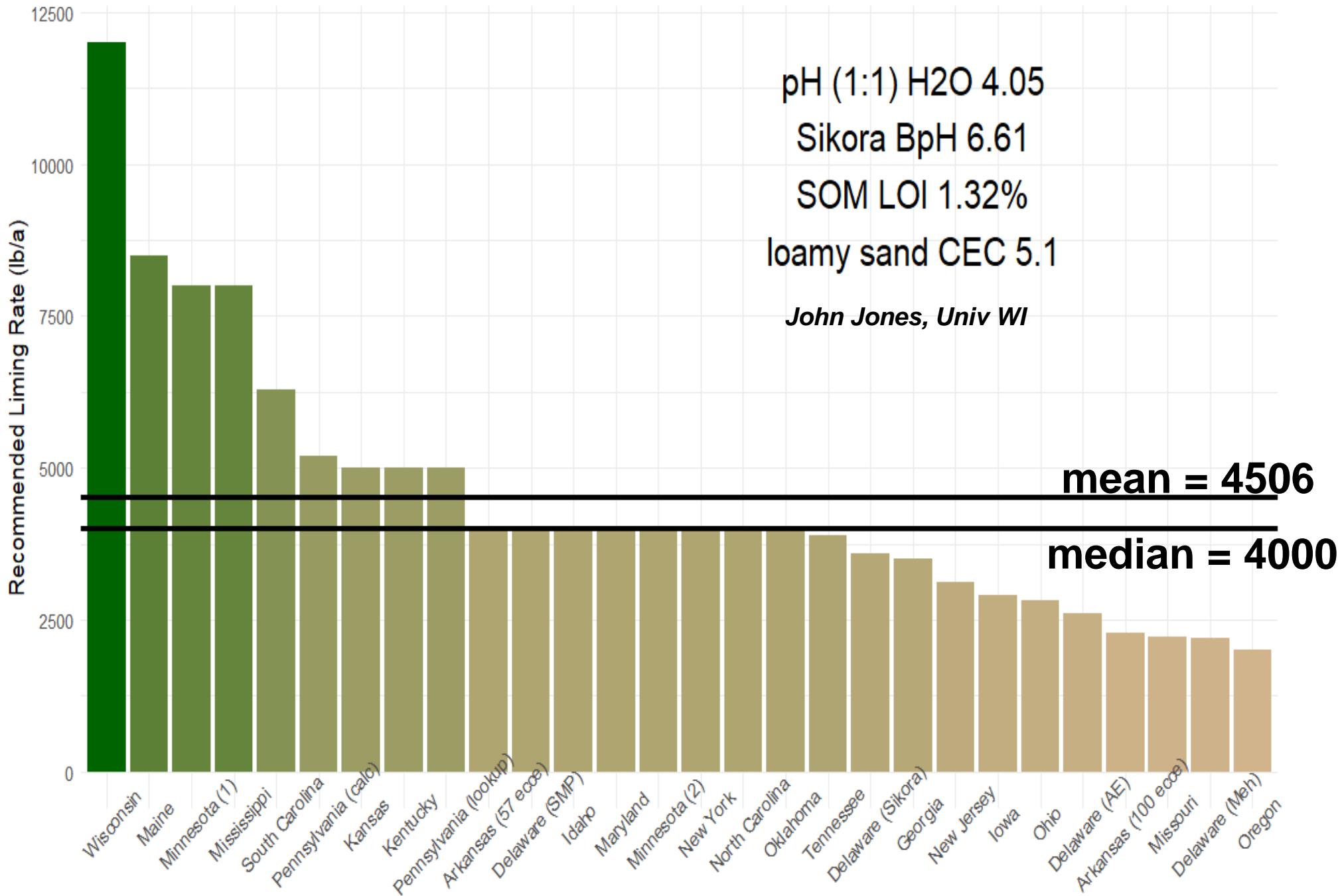
Target pH 6.5 (0-8"), Lime Rate tons per acre 67% ECCE.

# FRST Lime Project: recommendation survey

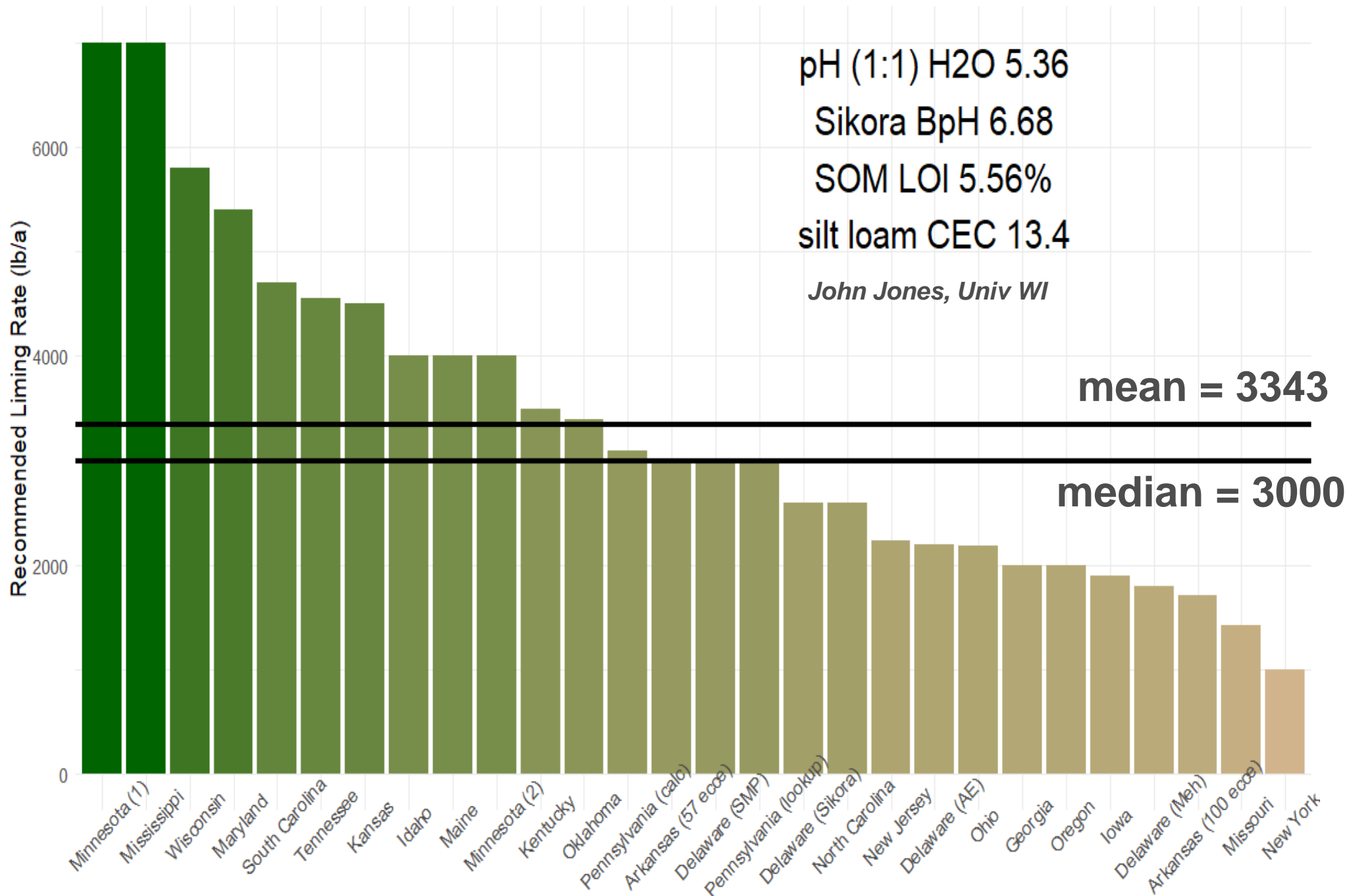
ALP pH and BpH data for six PT soils was submitted to 28 LGU labs for lime recommendations. Soils pH ranged in pH 4.05 – 5.89. Soil Properties:

<b>ALP Soil ID</b>	<b>pH</b>	<b>BpH</b>	<b>CEC</b>	<b>SOM-LOI</b>
			cmol kg <sup>-1</sup>	%
SRS-2113	4.05	6.61	5.1	1.32
SRS-2102	4.23	6.36	6.6	1.19
SRS-1614	4.60	5.90	18.4	5.47
SRS-1903	5.36	6.68	13.4	5.56
SRS-1604	5.52	6.90	12.3	2.27
SRS-2115	5.84	6.81	12.7	1.94

# Soil SRS-2113

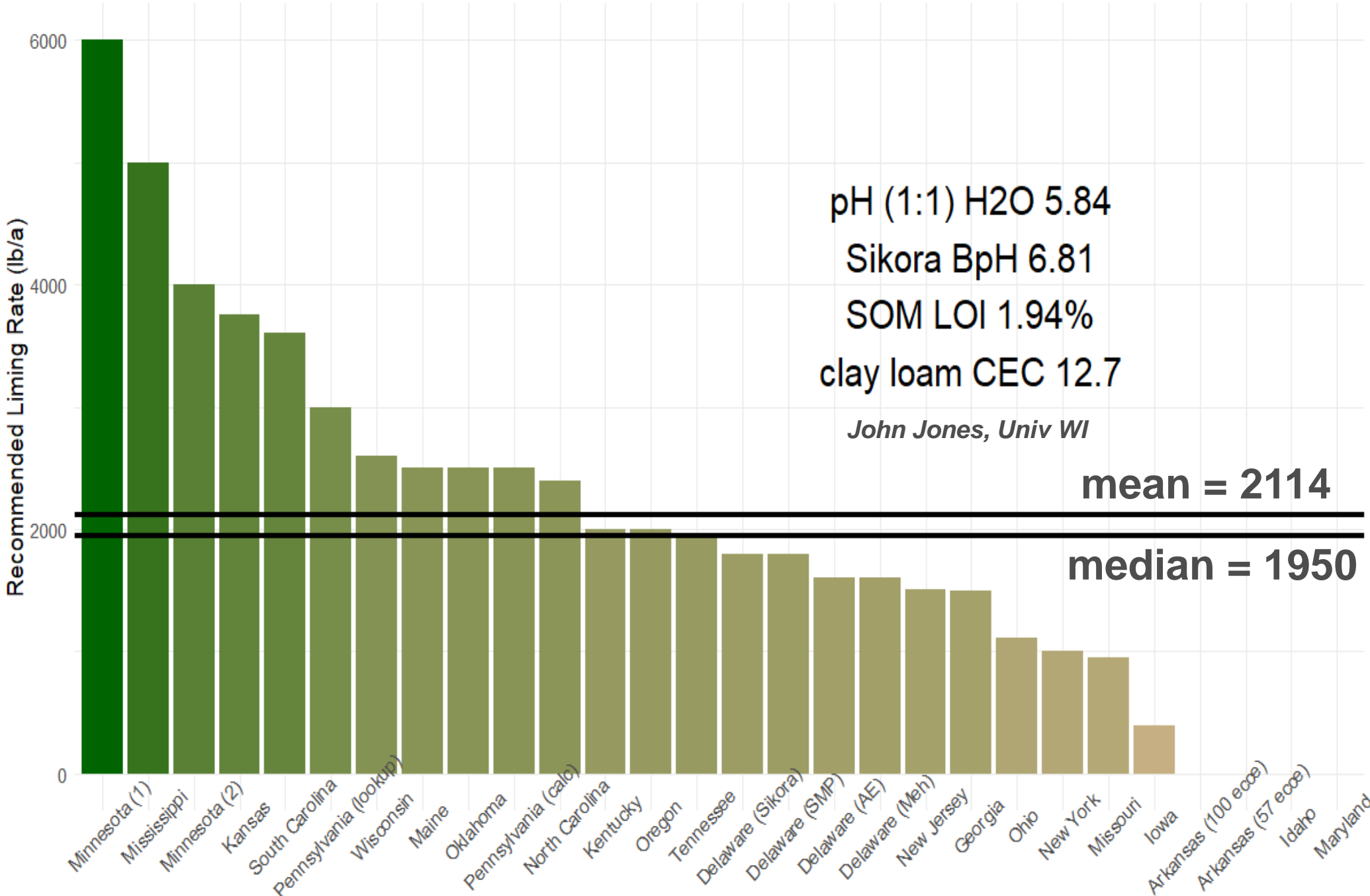


# Soil SRS-1903



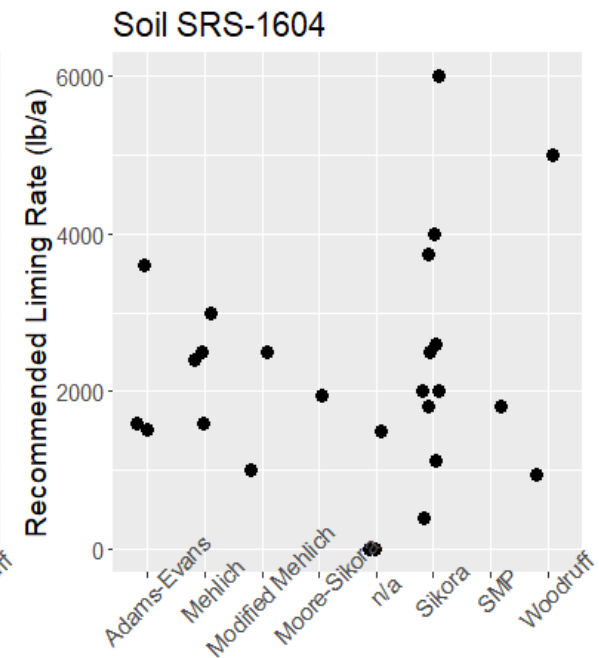
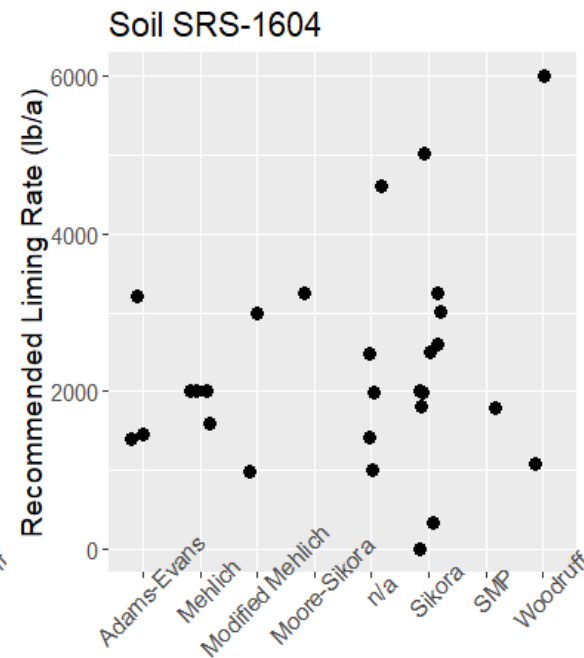
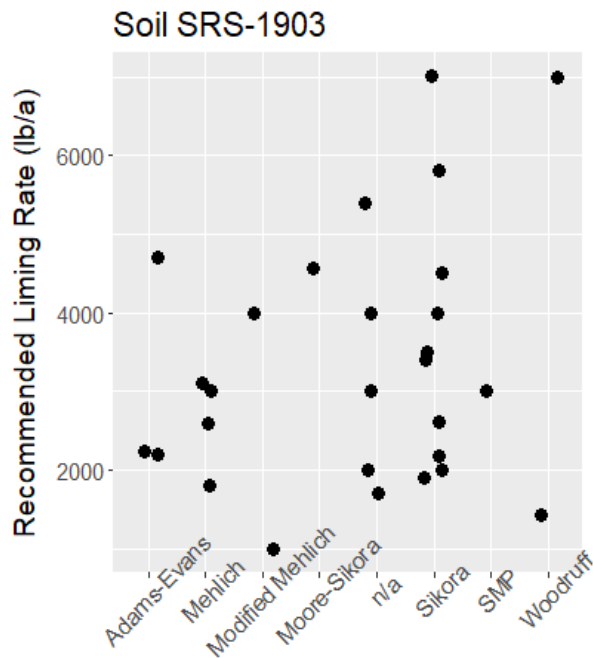
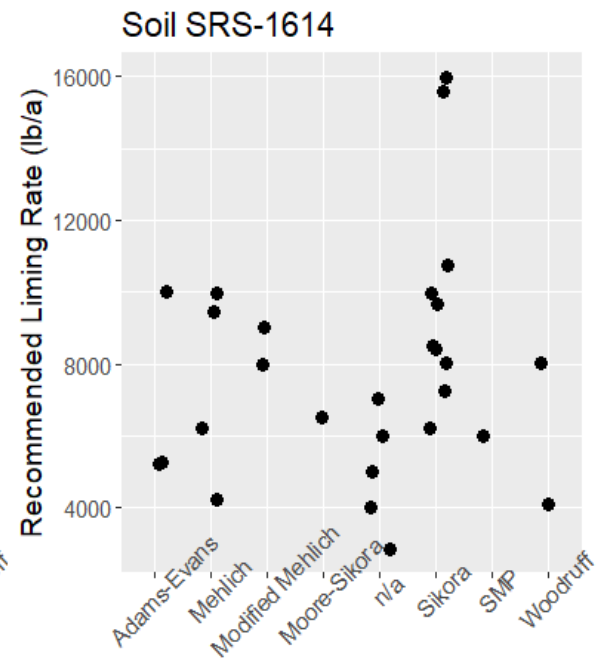
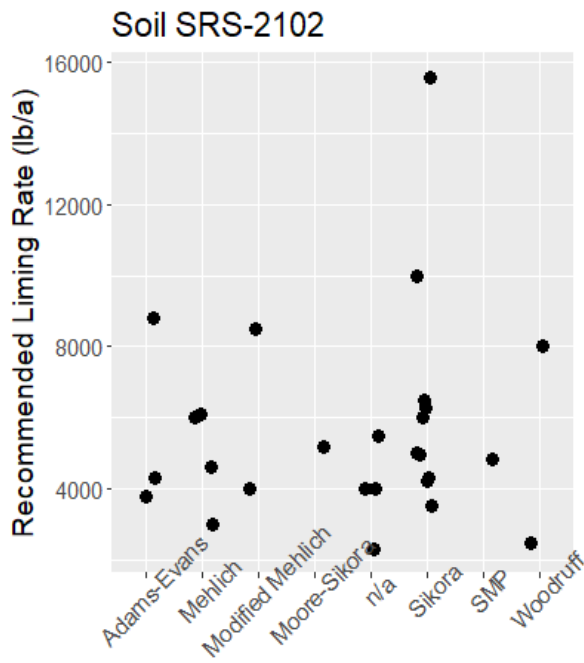
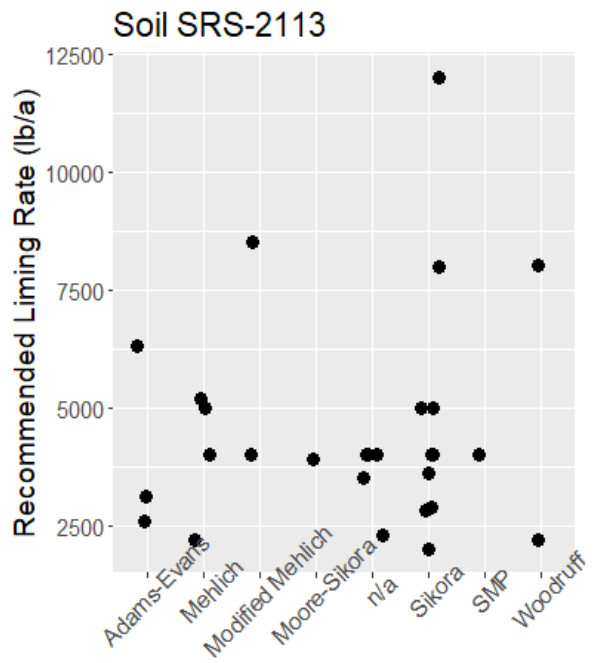


# Soil SRS-2115





*John Jones,  
Univ WI*



# Lime Survey Summary



- Liming recommendations represent considerable research effort and resources.
- As an agronomist advising farmers, how would you interpret 4.5, 5.0, 5.5 soil pH values?
- Variability by state and buffer solution type is great.
- Investigation methods focused on regional soils, but also biased towards state cropping systems? (e.g., WI considering alfalfa requirements and not simply lime to increase pH?)

# FRST Lime Project: Incubation study

Collect 120 soils across the US representing major soil physiographic units across the four major soil regions (WERA-102, SERA-6, NEC-67, NECRA-13). Soils ranging in pH 4.0-6.4, loamy sand to clay loam, CEC, SOM and mineralogy. *In progress*

Soil acidity is independent of soil type or location.



# FRST Lime Project: Incubation study



Soil Analysis: pH (2 methods), BpH (4 methods), M3 analytes, SOM, SOC, CEC, titrateable acidity, and exch Al.

Lime incubation study: 7 rates of lime application, source  $\text{Ca}(\text{OH})_2$ , 10 day equilibration, and assess pH and  $\text{NO}_3\text{-N}$ .

Develop lime recommendation algorithms based on the four 4 buffer pH methods (Sikora, Adams Evans, Mehlich Modified and Sikora-2) and multi- regression analysis.

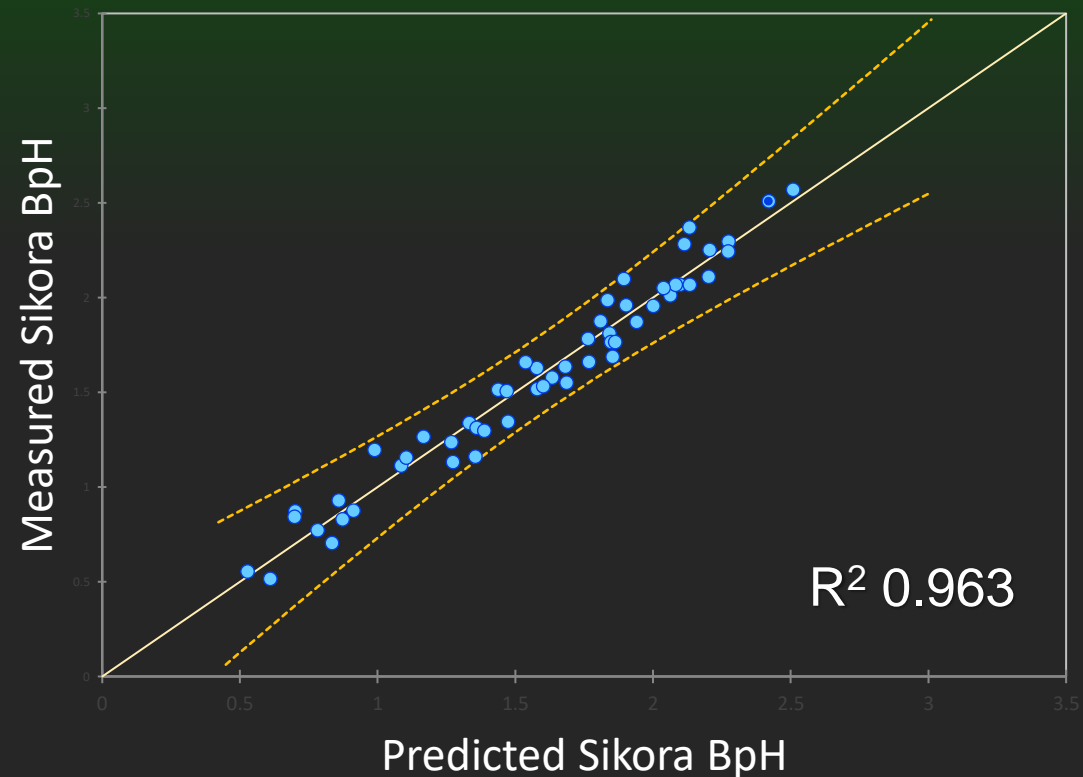
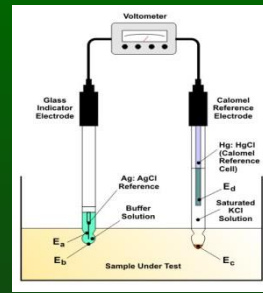
Verify lime recommendation algorithms on 12 new soils.

# FRST Lime: BpH model

BpH model based on ALP data, multi linear regression utilizing soil  $\text{pH}_{\text{salt}}$ ; CEC and M3-Ca,  $n=54$  soils,  $\text{pH} < 6.3$ , BpH Sikora  $< 7.0$ .

Sikora BpH, as a measurement of reserve acidity, is highly related to pH, CEC and M3-Ca for ALP soils.

Results suggest a similar model relating lime neutralization of soil acidity could be developed using pH, M3 cations and CEC.



# FRST Lime Project: support

The FRST project is supported through a USDA grant, and is funding for lime project soil collections for the initial incubation research.

Additional support will be need to complete soils analysis incubation research and data analysis.

Support will be sought from the fertilizer industry, lime manufacturers and commodity workgroups.



# Summary

Acid soils are extensive in the US, with agriculture utilizing 79 M metric tons of aglime in 2018, valued at \$3.3 billion. Across the US soil s with pH < 6.2 constitute 20-80% of soils tested.



FRST lime project survey of lime recommendations used in the US show high variability in recommendations.

Given pH and BpH, LGU lime Recs are highly variable across the US ranging from 1000 – 6000 lbs/ac on an acid soil. Even labs using the same BpH mehtod had highly variable lime recs.



The goal of the FRST Lime project is to develop new lime recommendations based on current BpH methods, basd on a national US database.



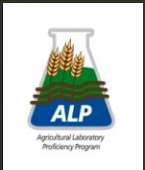


**Thank you for your time  
and attention**

**E-mail:**

**Robert.Miller@cts-interlab.com**

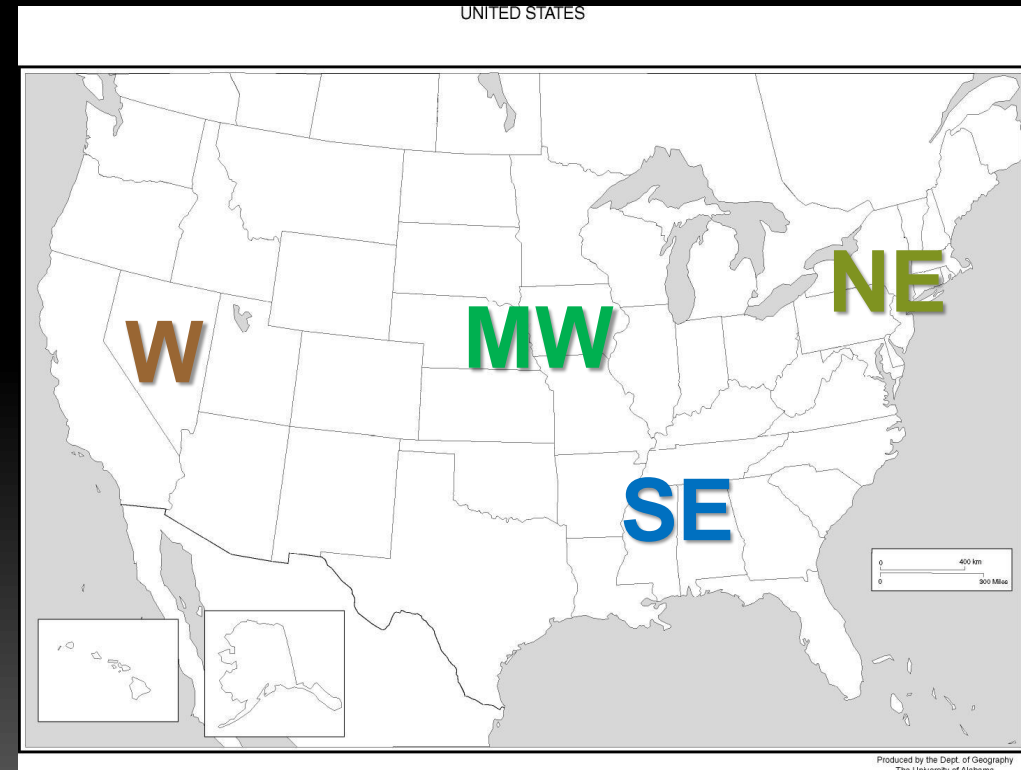
**Txt: 970-217-2572**



# Soil buffer pH methods, North America

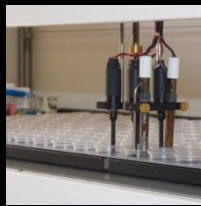
## Measurement of reserve acidity

- 1) SMP BpH (MW, W)
- 2) Sikora BpH (SE, MW)
- 3) Adams Evans BpH (SE, MW, W)
- 4) Moore-Sikora BpH (SE)
- 5) Mehlich BpH<sub>orig</sub> (SE)
- 6) Mehlich BpH<sub>mod</sub> (NE, SE)
- 7) Sikora 2 BpH (SE, MW)
- 8) Woodruff BpH (MW)
- 9) CaOH<sub>2</sub> Titration (SE)



A Sikora BpH method was developed on 2002 to replace hazardous reagents components of the SMP BpH method, and has been widely adopted. Moore-Sikora represents a modification of the Adams Evans BpH implemented at Clemson University, 2012. Modified Mehlich BpH introduced in 2002, substituting Ca for Ba in the BpH reagent.

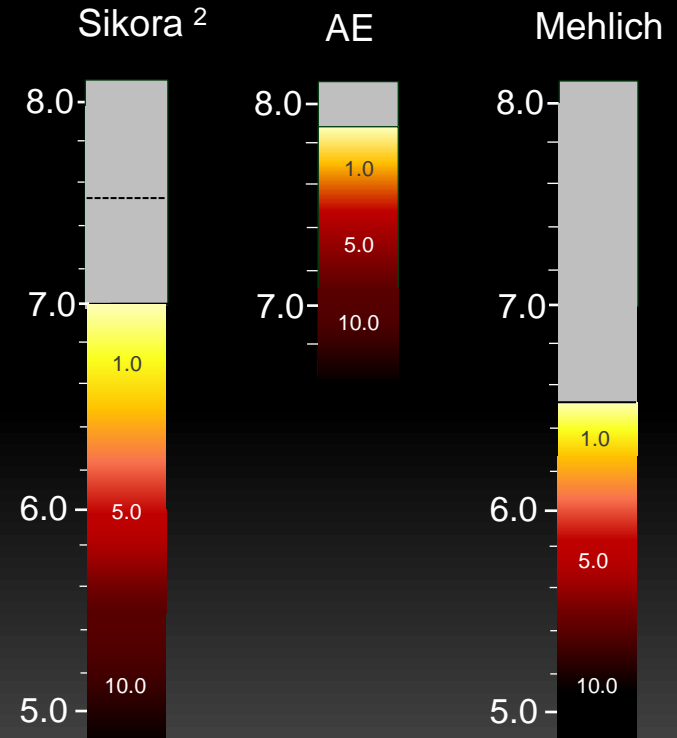
# Soil BpH methods



Lab BpH methods<sup>1</sup> are based on soil/buffer equilibration: SMP, 10 minutes stirring and 30 minutes settle time; Mehlich BpH is 5 seconds stir and 30 min settle.

The Adam Evans (AE) BpH was developed for soils with low CEC, low in 2:1 clay minerals. Whereas, SMP was developed for soils CEC 10 -25 cmol/kg dominant in 2:1 soil clay minerals.

Response range for AE BpH is narrower 7.90 – 7.20 than SMP or Sikora BpH. Mehlich BpH range 6.6 – 5.1.



<sup>2</sup> Max range of lime rates 0-10 ton/ac

Measurement uncertainty

Sikora BpH:  $\pm 0.05$  pH

AE BpH:  $\pm 0.03$

<sup>1</sup> Sikora, F. 2014. Soil test Methods for the Southeastern United States. Southern Cooperative Series Bulletin No. 419 ISBN# 1581614195

# FRST Lime Project: Identify need

Need of updating Recs. Lime rate recommendations, with few exceptions, are more than three decades old. Over this period tillage systems have shifted and both nutrient applications rates and yields increased. Need to update lime recommendations.

Lime-Rec inconsistencies. Recommendations are mostly state specific, based on researcher/grad student studies, and as a result rates are variable.

Lab pH and BpH quality. Today more than 80% of soil testing today is done by commercial labs. Method issues of both inter and intra lab consistency.



# Comparison of lime recommendations



Soil ID <sup>1</sup>	SRS-1910	SRS-2001	SRS-2006	SRS-2003
pH <sub>H2O</sub>	<b>5.78</b>	<b>5.19</b>	<b>5.48</b>	<b>6.01</b>
Sikora BpH	<b>6.87</b>	<b>6.65</b>	<b>6.50</b>	<b>6.85</b>
AE BpH	<b>7.54</b>	<b>7.44</b>	<b>7.30</b>	<b>7.45</b>

	Calculated lime rate lbs/ac <sup>2</sup>			
Kansas St Univ	<b>1000</b>	<b>1600</b>	<b>2100</b>	<b>200</b>
Tri-State (OH-MI-IN)	<b>500</b>	<b>2100</b>	<b>3200</b>	<b>700</b>
Mid West LGU	<b>200</b>	<b>16700</b>	<b>22300</b>	<b>0</b>
Clemson Univ	<b>1100</b>	<b>1900</b>	<b>2100</b>	<b>1100</b>
Univ Delaware	<b>1400</b>	<b>2500</b>	<b>2800</b>	<b>1400</b>

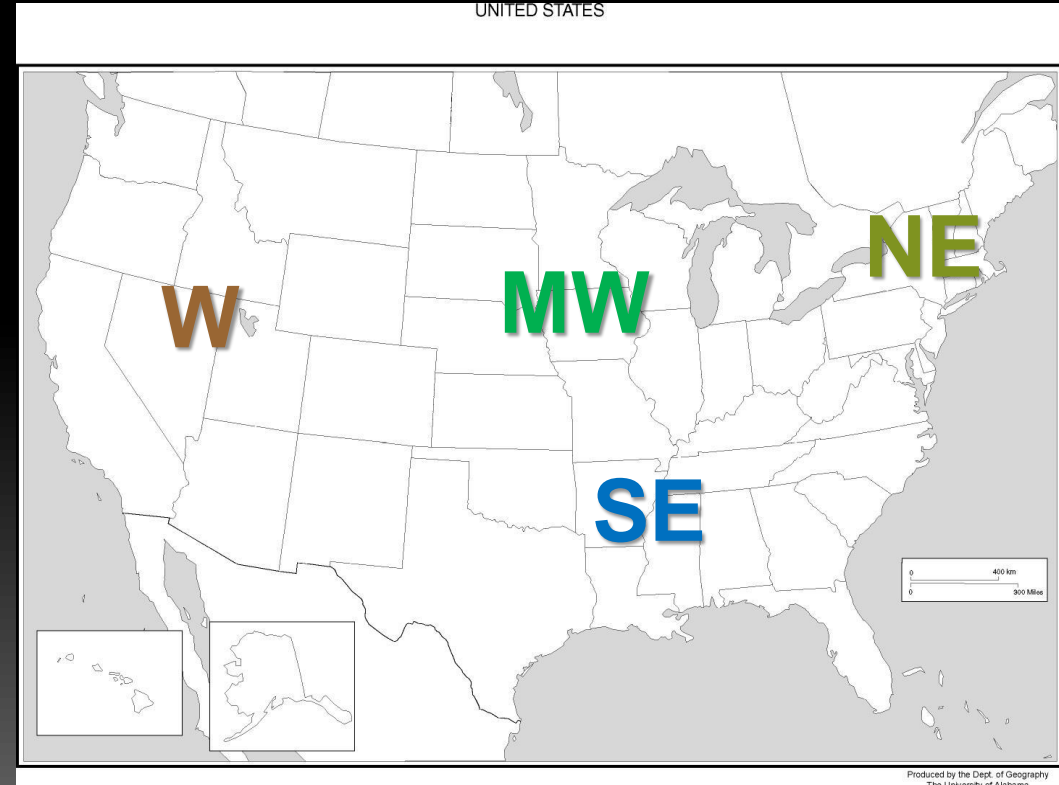
<sup>1</sup> Soil data source ALP program 2019-2020, population median values.

<sup>2</sup> Recommendations based on a 3" depth, target pH of 6.5, and 100% ECCE.

# Soil pH methods, North America

## Measurement of active acidity

- 1) pH (1:1) H<sub>2</sub>O (**SE, NE, MW, W**)
- 2) pH (1:1) 0.01 M CaCl<sub>2</sub> (**SE, MW**)<sup>1</sup>
- 3) pH Saturated Paste (**W**)
- 4) pH (1:2) 0.01 M CaCl<sub>2</sub> (**MW**)
- 5) pH (1:1) 1.0 N KCl (**SE**)
- 6) pH (1:2) H<sub>2</sub>O (**W**)
- 7) pH (1:5) H<sub>2</sub>O

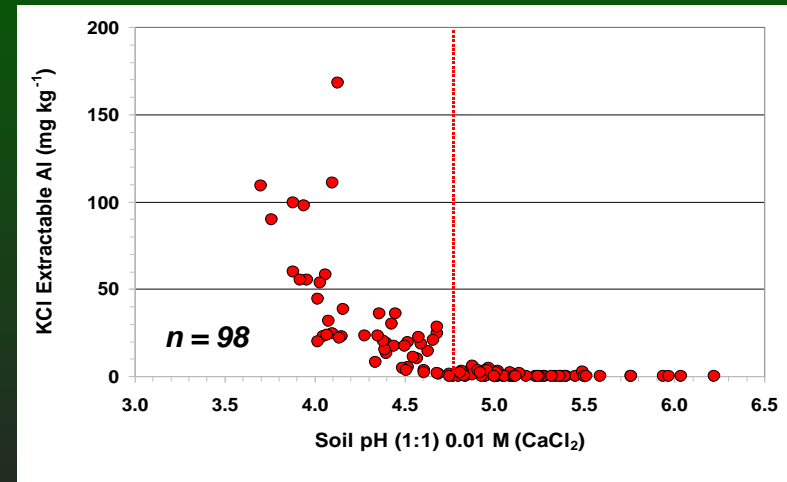


<sup>1</sup> Usage of the pH salt method has increased in the last decade to address low ionic strength soil slurry solutions and improve accuracy.

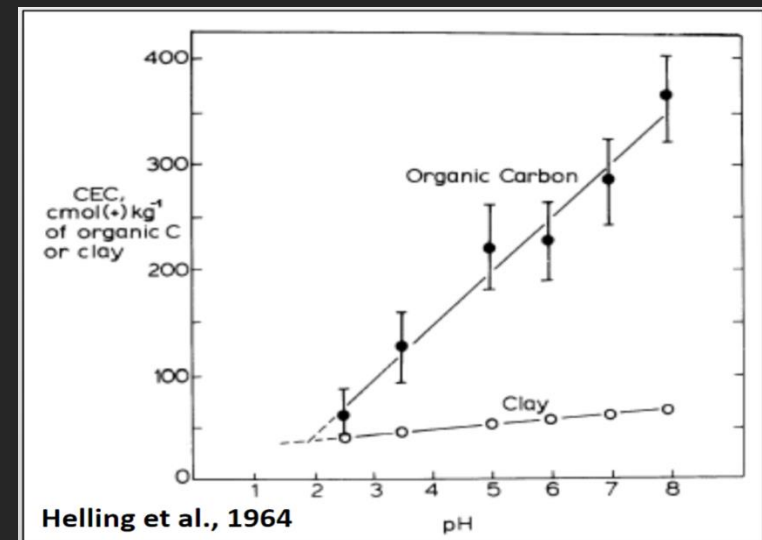
# Soil pH: Al<sup>3+</sup> and CEC

Extractable Al<sup>3+</sup> increases as pH (salt) declines below 5.0. Low pH suppresses plant vigor and may results in Al<sup>3+</sup> toxicity for specific crops.

Soil CEC is pH dependent. Soil minerals show a slight increase in CEC transitioning from pH 5.0-7.0, whereas organic carbon, shows a dramatic change, increasing 32%.



Miller R.O. 2005. California Department of food and Agriculture, Project Report.



# Lab quality of BpH analysis

## Upper Midwest Region

Double-blind ALP reference soils were submitted to 7 labs and replicated 3 times, 2019. Lime rates calculated based on ISU recommendations. Lab ID CIA BpH resulted in lime rates range 3800 lbs/ac, \$114/ac.

## Southeast Region

Double-blind reference soils submitted to 5 labs, and replicated 3 times, 2000. Lime rates calculated based on SERA-6 recommendations. Lab ID AYT BpH values range in lime rate 5,700 lbs/ac, \$171/ac.

\* Lime cost estimate \$60/ton

Soil ID ALP SRS-1814 Lab ID CIA			
<i>Date of Analysis</i>	$\text{pH}_{\text{H}_2\text{O}}$	Sikora BpH	Lime rate (lbs/ac) <sup>1</sup>
10 / 2019	6.10	6.48	2,900
11 / 2019	5.60	6.10	5,700
12 / 2019	6.20	6.63	1,900
<b>Ref Value</b>	<b>6.08</b>	<b>6.67</b>	<b>1,600</b>

<sup>1</sup> Lime recs bases on ISU lime recommendations, based on 1550 ECCE.

Soil ID ALP SRS-1913 Lab ID AYT			
<i>Date of Analysis</i>	$\text{pH}_{\text{H}_2\text{O}}$	BpH	Lime rate (lbs/ac) <sup>2</sup>
5 / 2020	4.5	6.1	11,500
6 / 2020	4.5	6.2	10,000
7 / 2020	4.8	6.5	5,800
<b>Ref Value</b>	<b>4.6</b>	<b>6.5</b>	<b>5,800</b>

<sup>2</sup> Lime recs based on SERA-6, SMP method (KY) lime recommendations, based on 1550 ECCE.



# pH, liming and VRT

Liming has increased with the advent of precision Ag and VRT specific placement across the field.

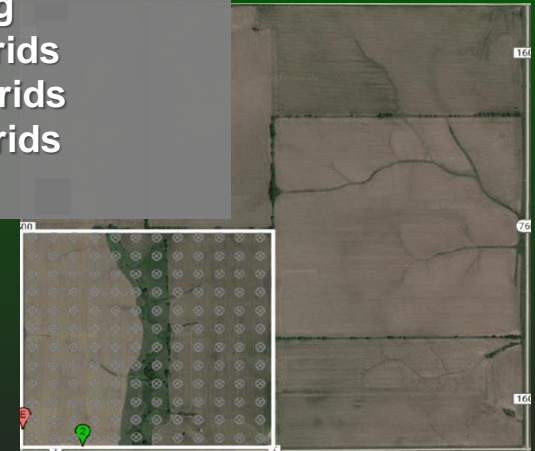
Lime evaluation on a 155 acre field shows a 40% reduction in cost, for 2.5 ac grid relative to composite. Lime cost savings pays for sampling.

Soil Density Sampling	Ag Lime Tons	Field Lime Cost
Grid - 1.1 ac	280	\$11,780
Grid - 2.5 ac	240	\$10,008
Grid - 4.4 ac	272	\$11,424
Field composite	399	\$16,721
Zone : n=6	401	\$16,842

Ag Lime: Rec based on SMP BpH and LGU Rec, lime used in formulas is 1,500 ECCE, \$42.00/ton

Nebraska Field  
Sampled April 2017  
155 acres

Field sampling split into:  
One composite sample  
Grid sampling  
1.1 Acre grids  
2.5 Acre Grids  
4.4 acre Grids  
Zones: 6



Cost of Lime \$10 - \$20/tn - OH

Trucking delivery: \$10/tn

VRT application cost: \$8 - \$12/tn

VRT cost \$40 - \$50/tn – NE, IA, NC

Lime application frequency:

Midwest: 3 yrs new ground;

7 yrs when BpH reaches 6.9/7.0;

High rainfall regions, annually

# Quality of BpH analysis



## Southern Region

Six double-blind reference soils were submitted to five labs, and replicate 3 times in the spring of 2020. Lime rates were calculated based on SERA-6 lime recommendations.

Lab ID AYT (at right) showed a wide range in SMP BpH values resulting in a range in lime rate of 5,700 lbs/ac. BpH repeatability was an issue on 2 of 6 soils evaluated.

Soil ID ALP SRS-1913			
<i>Date of Analysis</i>	<i>Soil pH<sub>H2O</sub></i>	<i>BpH</i>	<i>Lime rate (lbs/ac) <sup>1</sup></i>
5 / 2020	4.5	6.10	11,500
6 / 2020	4.5	6.21	10,000
7 / 2020	4.8	6.52	5,800
<b>Ref Value</b>	<b>4.6</b>	<b>6.54</b>	<b>5,500</b>

<sup>1</sup> Lime recs based on SERA-6, SMP method (KY) lime recommendations, based on 1550 ECCE.

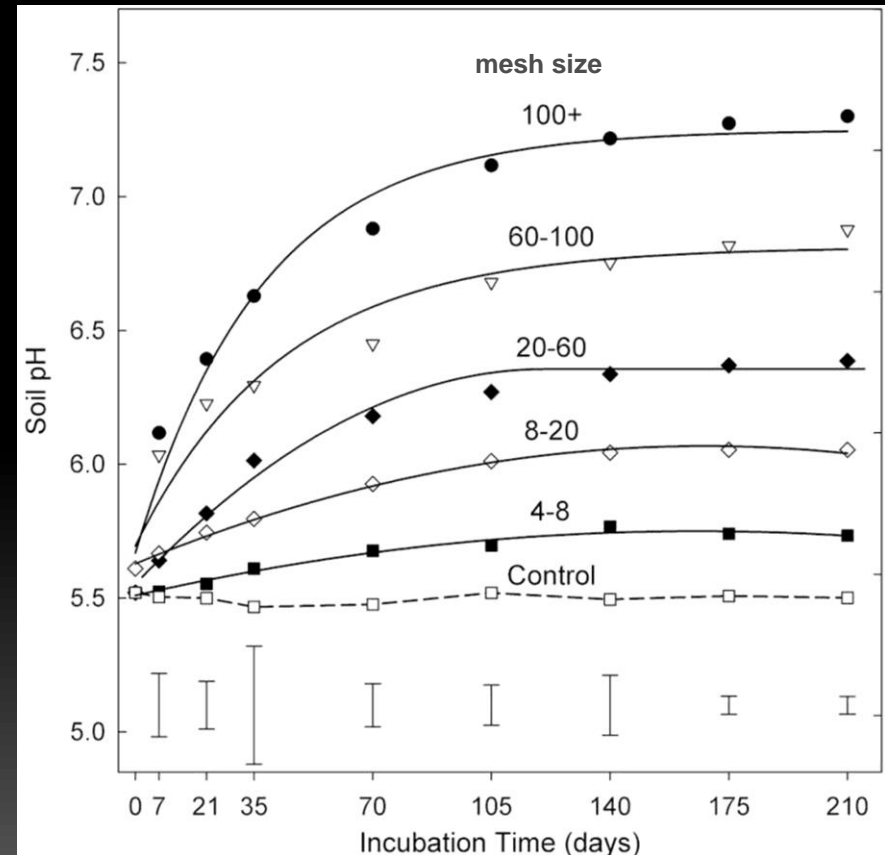
**5,700 lbs/ac results in a difference in \$128/ac, @ a lime cost: \$45/ton**

# Lime quality - finess

Lime finess impact the reaction time.  
Efficiency of lime at raising pH relative to  $\text{CaCO}_3$  increases with incubation time and particle finess.

Finer particle sizes reduce incubation time for neutralizing soil acidity.

Pelleted lime, calcitic aglime, and dolomitic aglime efficiencies across all incubation periods were 60 to 90, 47 to 65, and 12 to 47% respectively (Jones and Mallarino, 2018).



Jones and Mallarino. 2018. Influence of Source and Particle Size on Agricultural Limestone Efficiency at Increasing Soil pH. SSSAJ 82: 271-282.

[https://www.google.com/url?sa=i&url=https://doi.org/10.2136/sssaj2017.06.0207&psig=AOvVaw1O8dGcBQhvwme0rES2uL&usq=\\_\\_1601230474287000&source=imag&es&cd=16&ved=0CAMQJBJqFwoTCND6yYV3h-wCFQAAAAAAGAAAAABAT](https://www.google.com/url?sa=i&url=https://doi.org/10.2136/sssaj2017.06.0207&psig=AOvVaw1O8dGcBQhvwme0rES2uL&usq=__1601230474287000&source=imag&es&cd=16&ved=0CAMQJBJqFwoTCND6yYV3h-wCFQAAAAAAGAAAAABAT)

# Soil depth and liming

Stratification of pH and BpH provide challenges for making lime recommendations.

N-8 field on the left has low lime rate on surface but higher rec in sub soil. Challenge in no-Till field.

K-9 is more typical with acid soil at surface and trend upward with depth.

Nitz 80 - 2017

Depth <sup>1</sup>	pH	BpH
0 – 2"	6.3	6.7
2 – 4"	5.9	6.5
4 – 6"	5.3	6.1
6 – 8"	5.4	6.3

K9 - 2017

Depth <sup>1</sup>	pH	BpH
0 – 2"	5.0	6.3
2 – 4"	5.1	6.4
4 – 6"	5.5	6.6
6 – 8"	6.0	6.8

Ag Lime: Application based on 100% ECCE, lime used in formulas is 1,500 ECCE corrected to 100%, \$42.00/ton

# Soil BpH methods



BpH methods based on soil + buffer reagent, stirring, equilibration time with subsequent measurement of pH with ISE. Equilibration times: 10 – 15 minutes.

Response range for Adams Evans (AE) BpH is narrower 7.20 – 7.90 than other BpH methods.

Across methods BpH lab std errors ( $\sigma_{x^-}$ ) range from 0.01 – 0.03 BpH units across the working range 5.0 - 7.5 .

Method <sup>1</sup>	Mean	Std error
pH 1:1 <sub>H2O</sub>	5.91	0.05
pH 1:1 <sub>0.01 M CaCL2</sub>	7.70	0.01
SMP BpH	6.73	0.02
Sikora BpH	6.72	0.01
Adam Evans BpH	7.44	0.02
Mehlich BpH	6.13	0.02

<sup>1</sup> Data Source: ALP program, soil ID SRS-1814.

# Amelioration of Acidity - Liming

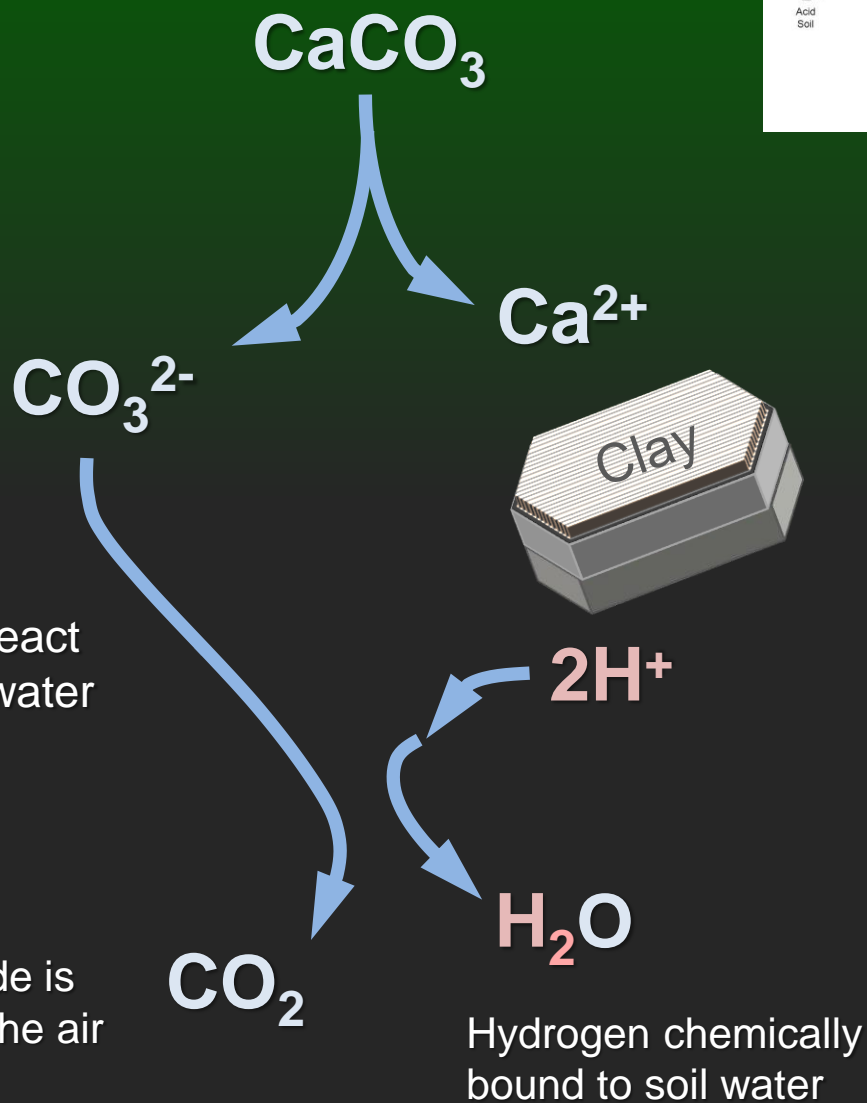
In moist soil calcium carbonate dissolves into calcium and carbonate ions

Calcium replaces hydrogen on soil constituents

Excess hydrogen ions react with carbonate to form water and carbon dioxide

Carbon dioxide is Released to the air

Hydrogen chemically bound to soil water



Soil Acidity

- Soil Liming Reaction

$$\text{CaCO}_3 \rightarrow \text{Ca}^{2+} + \text{CO}_3^{2-}$$

Acid Soil

Exchange

$$\text{H}_2\text{O} + \text{Al}(\text{OH})_3$$

Ionise to soil

Excess (causing with carbon water an

Carbon d into the s

# Soil BpH methods



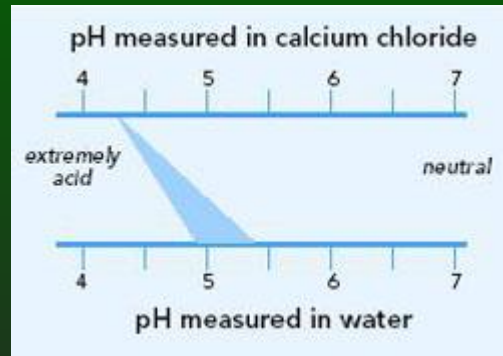
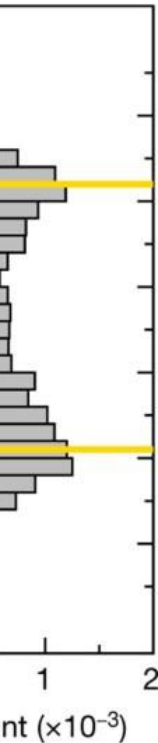
BpH methods based on soil + buffer reagent, stirring, equilibration time with subsequent measurement of pH with ISE. Equilibration times: 10 – 15 minutes.

Response range for Adams Evans (AE) BpH is narrower 7.20 – 7.90 than other BpH methods.

Across methods BpH lab std errors ( $\sigma_{x^-}$ ) range from 0.01 – 0.03 BpH units across the working range 5.0 - 7.5 .

Method <sup>1</sup>	Mean	Std error
pH 1:1 <sub>H2O</sub>	5.91	0.05
pH 1:1 <sub>0.01 M CaCL2</sub>	7.70	0.01
SMP BpH	6.73	0.02
Sikora BpH	6.72	0.01
Adam Evans BpH	7.44	0.02
Mehlich BpH	6.13	0.02

<sup>1</sup> Data Source: ALP program, soil ID SRS-1814.



[https://agriculture.vic.gov.au/\\_\\_data/assets/image/0008/560834/figure-1\\_ph-measured-calcium-water.jpg](https://agriculture.vic.gov.au/__data/assets/image/0008/560834/figure-1_ph-measured-calcium-water.jpg)

The pH<sub>w</sub> may be higher by 0.6 to 1.2 in low salinity soils and higher by 0.1 to 0.5 in high salinity soils. Research has shown a difference of 0.7 for a wide range of soils.

Higher pH<sub>w</sub> values to around 10 may be associated with alkali mineral soils containing sodium carbonates and bicarbonates.

Research has shown that seasonal variation of pH<sub>w</sub> can vary up to 0.6 of a pH unit in any one year. In comparison, soil pH<sub>Ca</sub> measurements are less affected by seasons.



# Soil BpH methods



BpH methods based on soil + buffer reagent, stirring, equilibration time with subsequent measurement of pH with ISE.

Across methods BpH uncertainty averages  $\pm 0.06$  BpH units across the working range 5.0 -7.5 .

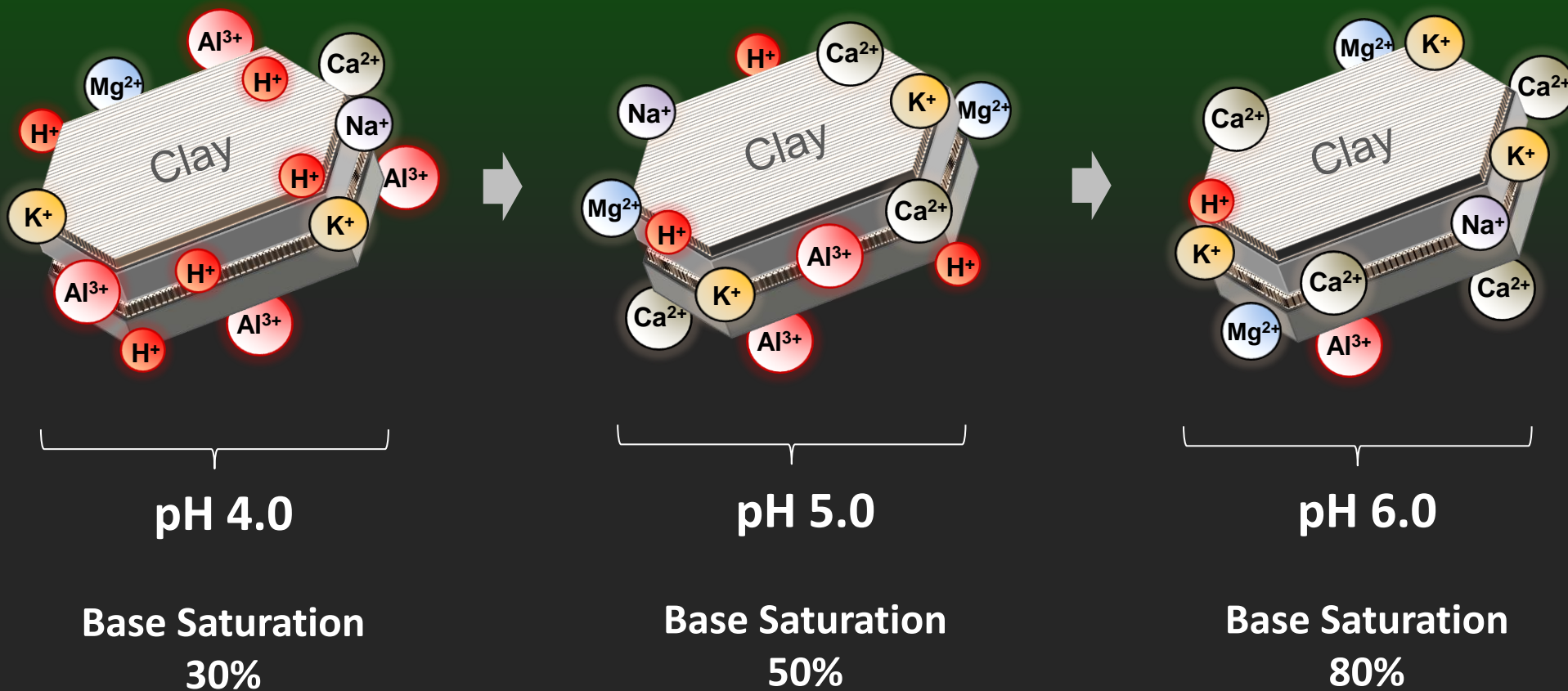
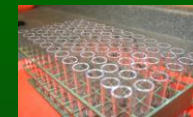
The difference between BpH - pH reflects the amount of reserve acidity neutralized, and is proportional to CEC.

Lime recommendations, based on buffer pH, are calibrated based on soil  $\text{CaCO}_3$  equilibration research and assessment of neutralization of acidity.

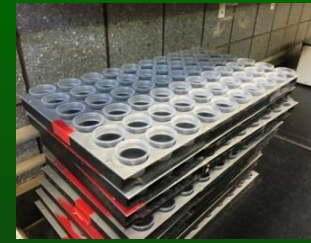
Soil ID <sup>1</sup>	CEC Cmol/kg	pH (1:1) H <sub>2</sub> O	BpH Sikora
SRS-1706	2.7	5.34	7.07
SRS-1511	6.7	5.60	6.82
SRS-1705	11.5	5.50	6.60
SRS-2006	20.2	5.48	6.48

<sup>1</sup> Data Source: ALP program, 2015-2020, median values.

# Soil pH and Base Saturation (BS)



# Impact of soil pH on crop tolerance



		Soil pH							
Tolerance	Crop	4.0	4.5	5.0	5.5	6.0	6.5	7.0	7.5
Least	Alfalfa						██████████		
	Barley						██████████		
	Sweetclover						██████████		
Medium	Red Clover					██████████			
	Corn					██████████			
	Soybean					██████████			
	Oats, Rye				██████████				
Most	Blueberries	██████████							
	Buckwheat			██████████					
	Potatoes				██████████				

[https://www.agry.purdue.edu/ext/forages/publications/AY-267\\_fig2.jpg](https://www.agry.purdue.edu/ext/forages/publications/AY-267_fig2.jpg)

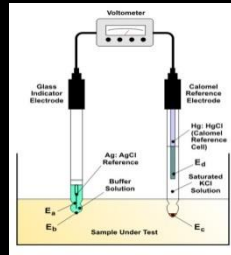
Table 1. Crop yields relative to pH.

Crop	Soil pH				
	4.7	5	5.7	6.8	7.5
Relative Yield (100 is the best, 0 is the worst)					
Corn	34	73	83	100	85
Wheat	68	78	89	100	99
Soybeans	65	79	80	100	93
Oats	77	93	99	98	100
Barley	0	23	80	95	100
Alfalfa	2	9	43	100	100
Timothy (grass)	31	47	59	100	95

Methods for Assessing Soil Quality, page 173 (SSSA, 1996)

<https://www.fssystem.com/Portals/0/resize566x333.jpg>

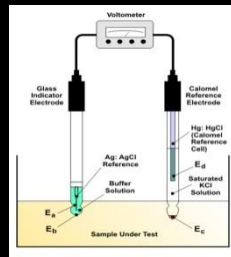
# Soil pH (1:1) method performance



Soil ID <sup>1</sup>	pH (1:1) H <sub>2</sub> O		pH (1:1) 0.01 M CaCl <sub>2</sub>	
	Mean	Stdev	Mean	Stdev
<b>SRS-1911</b>	<b>5.46</b>	<b>0.06</b>	<b>5.05</b>	<b>0.03</b>
<b>SRS-1912</b>	<b>8.13</b>	<b>0.05</b>	<b>8.10</b>	<b>0.06</b>
<b>SRS-1913</b>	<b>4.60</b>	<b>0.06</b>	<b>4.30</b>	<b>0.03</b>
<b>SRS-1914</b>	<b>5.43</b>	<b>0.08</b>	<b>5.10</b>	<b>0.03</b>
<b>SRS-1915</b>	<b>7.46</b>	<b>0.05</b>	<b>7.30</b>	<b>0.04</b>

<sup>1</sup> Source Agricultural Laboratory Proficiency program cycle 41 data base, 2019. Data intra-lab method performance, three replications.

# Soil buffer pH method comparison



Soil ID <sup>1</sup>	pH (1:1) H <sub>2</sub> O	SMP Buffer		Sikora Buffer		Adams Evans Buffer	
		Mean	Stdev	Mean	Stdev	Mean	Stdev
SRS-1911	5.36	6.62	0.04	6.68	0.05	7.43	0.03
SRS-1912	6.16	6.90	0.04	6.92	0.06	7.53	0.03
SRS-1913	6.88	6.88	0.06	7.01	0.05	7.67	0.03
SRS-1914	4.60	6.54	0.03	6.61	0.09	7.46	0.03
SRS-1915	5.43	6.56	0.04	6.64	0.04	7.39	0.04

<sup>1</sup> Source Agricultural Laboratory Proficiency program cycle 39-41 data base, 2019. Data intra-lab method performance, three replications.

# ***Nitrogen fertilizer and equivalent acidity***

---

<b>Material</b>	<b>% Nitrogen</b>	<b>kg of Lime per kg of N</b>
<b>Ammonium Sulfate</b>	<b>20.5</b>	<b>5.35</b>
<b>Calcium Nitrate</b>	<b>16.0</b>	<b>0.0</b>
<b>Urea</b>	<b>46.6</b>	<b>1.8</b>
<b>UAN 32</b>	<b>45.5</b>	<b>1.8</b>
<b>Organic Biomass</b>	<b>2.2</b>	<b>0.9</b>

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<sup>1</sup> Source: Andrews, 1954. The Response of Crops and Soils to Fertilizer and Manures.



# BpH lab data quality

Reference soils (double –blind) were submitted to 7 Midwest labs fall 2019, and replicate three times over 3 months.

Lab ID CIA (at right) showed a wide range in Sikora BpH values resulting in variance in lime rate of 2200 lbs/ac for soil ID SRS-1612 and 3848 lbs/ac for SRS-1814, BpH repeatability was an issue on 5 of 6 soils evaluated.

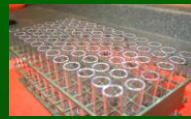
Buffer pH repeatability was an issue for 2 of 7 labs tested in 2019.

Soil ID ALP SRS-1612			
<i>Date of Analysis</i>	Soil pH <sub>H2O</sub>	Sikora BpH	Lime Rate (lbs/ac) <sup>1</sup>
10 / 2019	5.40	6.67	1565
11 / 2019	5.10	6.53	2581
12 / 2019	5.80	6.79	694
Ref Value	5.11	6.62	1780

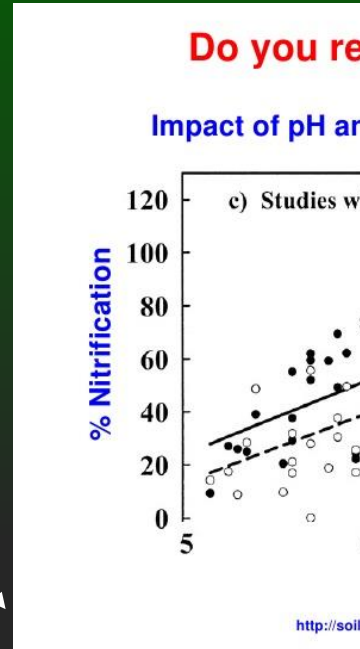
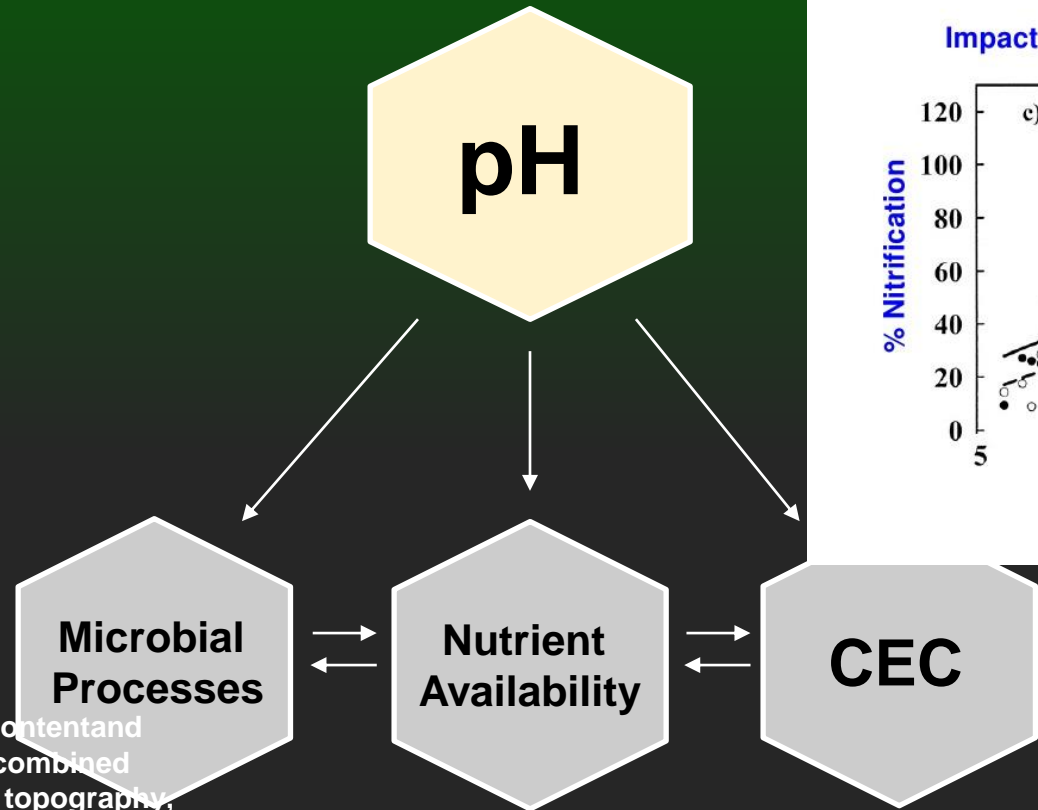
<sup>1</sup> Lime recs bases on ISU lime recommendations, based on 1550 ECCE.

Soil ID ALP SRS-1814			
<i>Date of Analysis</i>	Soil pH <sub>H2O</sub>	Sikora BpH	Lime Rate (lbs/ac) <sup>1</sup>
10 / 2019	6.10	6.48	2944
11 / 2019	5.60	6.10	5703
12 / 2019	6.20	6.63	1855
Ref Value	6.08	6.67	1565

# Impact of Soil pH



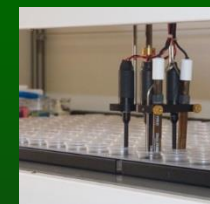
pH impacts multiple soil properties and is influenced by all three soil components. It is associated with clay and SOM and modified by pH.



Inherent factors affecting soil pH such as climate, mineral content and soil texture cannot be changed. Natural soil pH reflects the combined effects of soil-forming factors (parent material, time, relief or topography, climate, and organisms). The pH of newly formed soils is determined by minerals in the soil's parent material. Temperature and rainfall control leaching intensity and soil mineral weathering. In warm, humid environments, soil pH decreases over time in a process called soil acidification, due to leaching from high amounts of rainfall. In dry climates, however, soil weathering and leaching are less intense and pH can be neutral or alkaline. Soils with high clay and organic matter content are more able to resist a drop or rise in pH (have a greater



# Soil BpH methods



Lab BpH methods<sup>1</sup> are based on soil/buffer equilibration: 10 minutes stirring and 30 minutes settle time, (note Mehlich BpH is 5 seconds stir and 30 min settle).

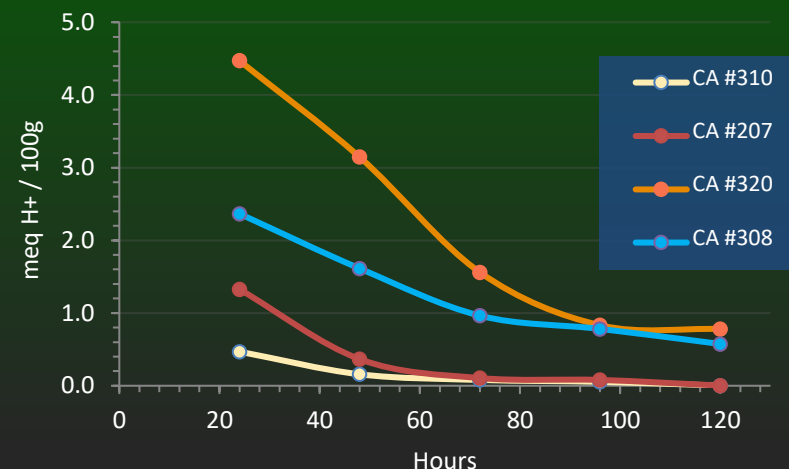
BpH is an approximate measure of soil reserve acidity ( $H^+$ ), and the value represents a short term equilibrium.

Response range for Adams Evans (AE) BpH is narrower 7.20 – 7.90 than SMP and Sikora BpH methods.

Intra-lab BpH method std errors ( $\sigma x^-$ ) range from 0.02 – 0.03 BpH units.

<sup>1</sup> Sikora, F. 2014. Soil test Methods for the Southeastern United States. Southern Cooperative Series Bulletin No. 419 ISBN# 1581614195

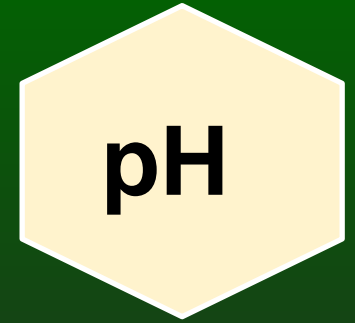
Soil acid neutralization,  $Ca(OH)_2$



Method <sup>1</sup>	Mean	Std error
SMP BpH	6.73	0.03
Sikora BpH	6.72	0.02
Adam Evans BpH	7.44	0.03
Mehlich BpH	6.13	0.03

<sup>1</sup> Data Source: ALP program, soil ID SRS-1814.

# Impact of Soil pH



Soil pH reflects the combined effects of soil-forming factors (parent material, time, relief or topography, climate, and biology).

pH of newly formed soils is determined by soil parent material. Temperature and precipitation impact leaching intensity and soil mineral weathering.

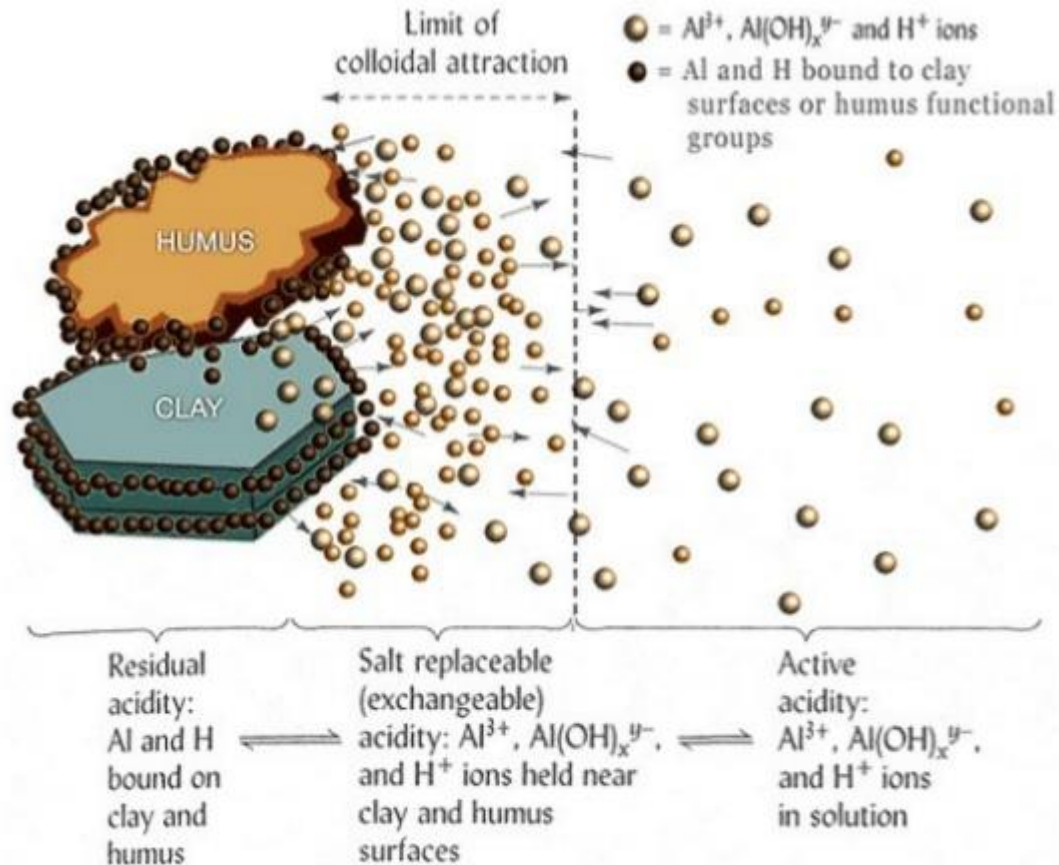
In warm, humid environments, soil pH decreases over time due to leaching and mineral weathering. Dry climates, however, soil weathering and leaching are less and pH may be neutral or alkaline.

Soils high in clay and SOM content have greater buffering capacity and resist pH changes. Sandy soils commonly have low SOM content, resulting in a low buffering capacity, and have high rates of H<sub>2</sub>O infiltration making them more vulnerable to acidification.

# Impact of EC on soil pH method differences

ALP ID	EC (1:1) dS/m	pH (1:1) <sub>H<sub>2</sub>O</sub>	pH (1:1) <sub>0.01 M CaCl<sub>2</sub></sub>	$\Delta$ pH
SRS-1204	0.07	5.23	4.46	0.77
SRS-0907	0.15	5.52	4.93	0.59
SRS-1414	0.31	5.40	4.95	0.45
SRS-1702	0.45	5.25	4.90	0.35
SRS-1803	0.61	5.13	4.87	0.26
SRS-1814	1.07	5.91	5.70	0.21

<sup>1</sup> Source Agricultural Laboratory Proficiency program data base 2009 and 2018, median pH (1:1)<sub>H<sub>2</sub>O</sub> results based on 62 laboratories reporting.



[http://www.landfood.ubc.ca/soil200/images/15\\_3acidity.jpg](http://www.landfood.ubc.ca/soil200/images/15_3acidity.jpg)

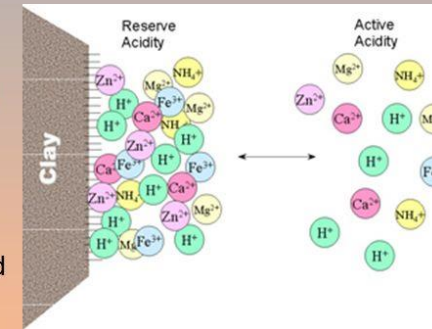
### Active vs. Reserve Acidity

active acidity

- $\text{H}^+$  ions in soil solution
- what is actually measured by a pH test

reserve acidity

- $\text{H}^+$  and  $\text{Al}^{+++}$  ions adsorbed to soil colloids





# PAC Program



In February 2019 The Illinois Soil Testing Association (ISTA) and the Soil and Plant Analysis Council (SPAC), agreed to develop a memorandum of understanding (MOU) to develop a plant analysis certification (PAC) program for labs providing plant/botanical analyses in North America.

In July 2019 the MOU approved by both organizations and a PAC sub-committee was formed from members of the organizations. A draft of PAC protocols was developed in the fall of 2019 and a program logo was approved in February 2020.

**PAC Committee:** Vernon Pabst, Dustin Sawyer, Bryan Thayer, John Spargo



# Lab plant performance, 2018



Analysis <sup>1</sup>	Confidence Limits as percent of the median	Total number of results <sup>2</sup>	Percent of labs with > 2 failures
N Comb	6	300	36.0 %
P	13	408	23.5 %
K	16	420	17.1 %
S	15	420	27.3 %
Zn	19	396	21.2%
B	21	408	26.5 %
Cu	18	348	17.2 %

<sup>1</sup> Plant analysis method failures based on 95% CL of median, all reporting labs, 12 samples, 2018.

<sup>2</sup> Total based on number of labs x number of plant samples evaluated.

# Dissociation of H<sub>2</sub>O



- $\text{H}_2\text{O} \rightarrow \text{H}^+ + \text{OH}^-$
- $K_{\text{eq}} = [\text{H}^+] [\text{OH}^-]$
- $\log K_{\text{eq}} = -14 = \log K_w$

$$K_{\text{eq}} = \frac{[\text{H}^+] [\text{OH}^-]}{[\text{H}_2\text{O}]}$$

- $\text{pH} = -\log [\text{H}^+]$
- $\text{pOH} = -\log [\text{OH}^-]$
- $\text{pK} = \text{pOH} + [\text{H}^+] = 14$

$$\text{pH} = 2$$
$$[\text{H}^+] = 10^{-2}$$



# Soil vs plant method performance



ISTA Results	Method	Total number of results <sup>1</sup>	Lab method failures <sup>2</sup>	Percent of labs with > 2 failures
	pH (1:1) <sub>H2O</sub>	330	30	9.5 %
	M3-P ICP	255	45	17.6 %
	M3-K	300	44	15.0 %

ALP Results	Method	Total number of results <sup>1</sup>	Lab method failures <sup>2</sup>	Percent of labs with > 2 failures
	N-Comb	300	108	36.0 %
	P	408	96	23.4 %
	K	420	72	17.6 %
	S	420	115	27.3 %

<sup>1</sup> Total based on number of labs x number of results reported.

<sup>2</sup> Number of PT sample results exceeding 95% confidence limits of the median.





# PAC Program Outline



## PAC assessment of lab performance

- Laboratory ALP plant analysis data assessed annually across three cycles, total twelve (12) plant materials.
- Certification requirements. A lab must successfully pass on ten of twelve analyte results. Lab performance failure will require automatic retest. Re-test failure, removal from PAC web site.
- PAC nutrient classes:
  - Macro: N, P, K, Ca, Mg, S
  - Micro: B, Cu, Fe, Mn, Zn
  - Anion option:  $\text{NO}_3\text{-N}$ ,  $\text{PO}_4\text{-P}$ , Cl



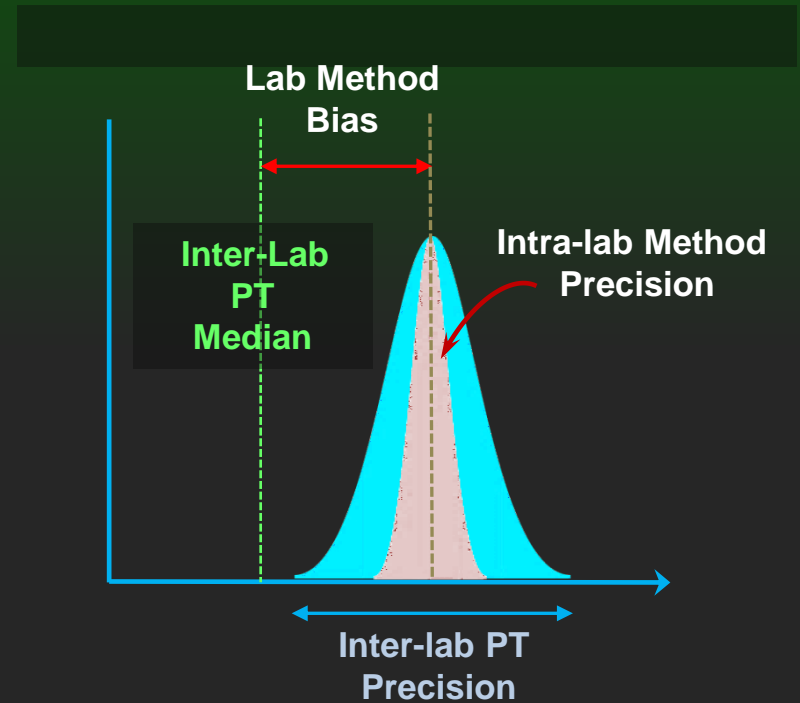
# PAC Program



Assessment of lab results will be based on PAC lab participant data.

Outliers deleted based on IQR statistical analysis, and lab performance evaluated based on mean and 95% confidence range limits for each analyte.

Laboratory precision will be documented, but will not be utilized to evaluate performance.





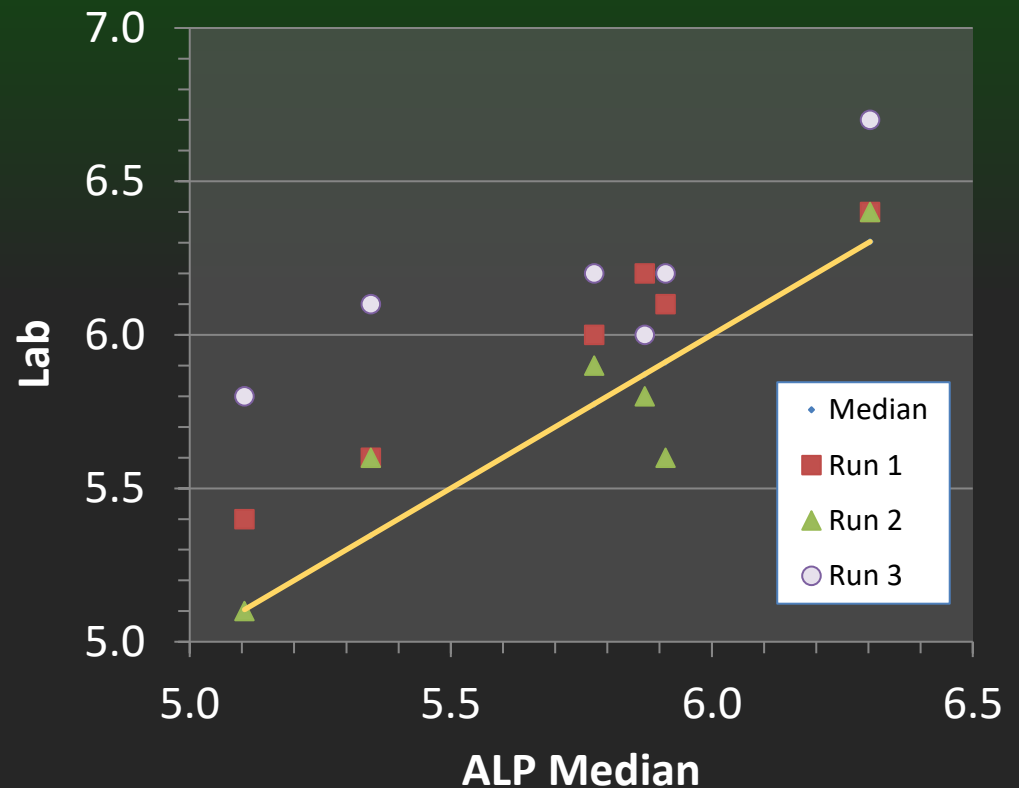
# DB - Summary



Soil pH for the six soils is at left for lab CIA. Yellow line designates the median pH.

The lab has consistent high bias on strongly acid soils, and inconsistent in reported values across the three submissions.

### pH - Lab CIA



# Measurement of pH

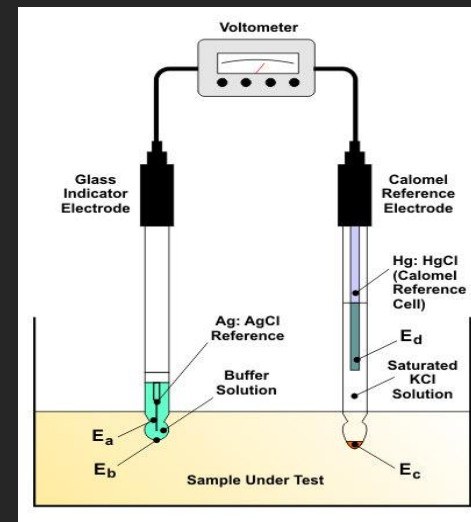
The hydrogen ion activity of a soil solution slurry can be measured visual using indicators photometric dyes, and potentiometric methods.

Potentiometric methods determine pH by using the electrical voltage potential of a pH-sensitive electrode (ISE) as a measurement signal in millivolts.

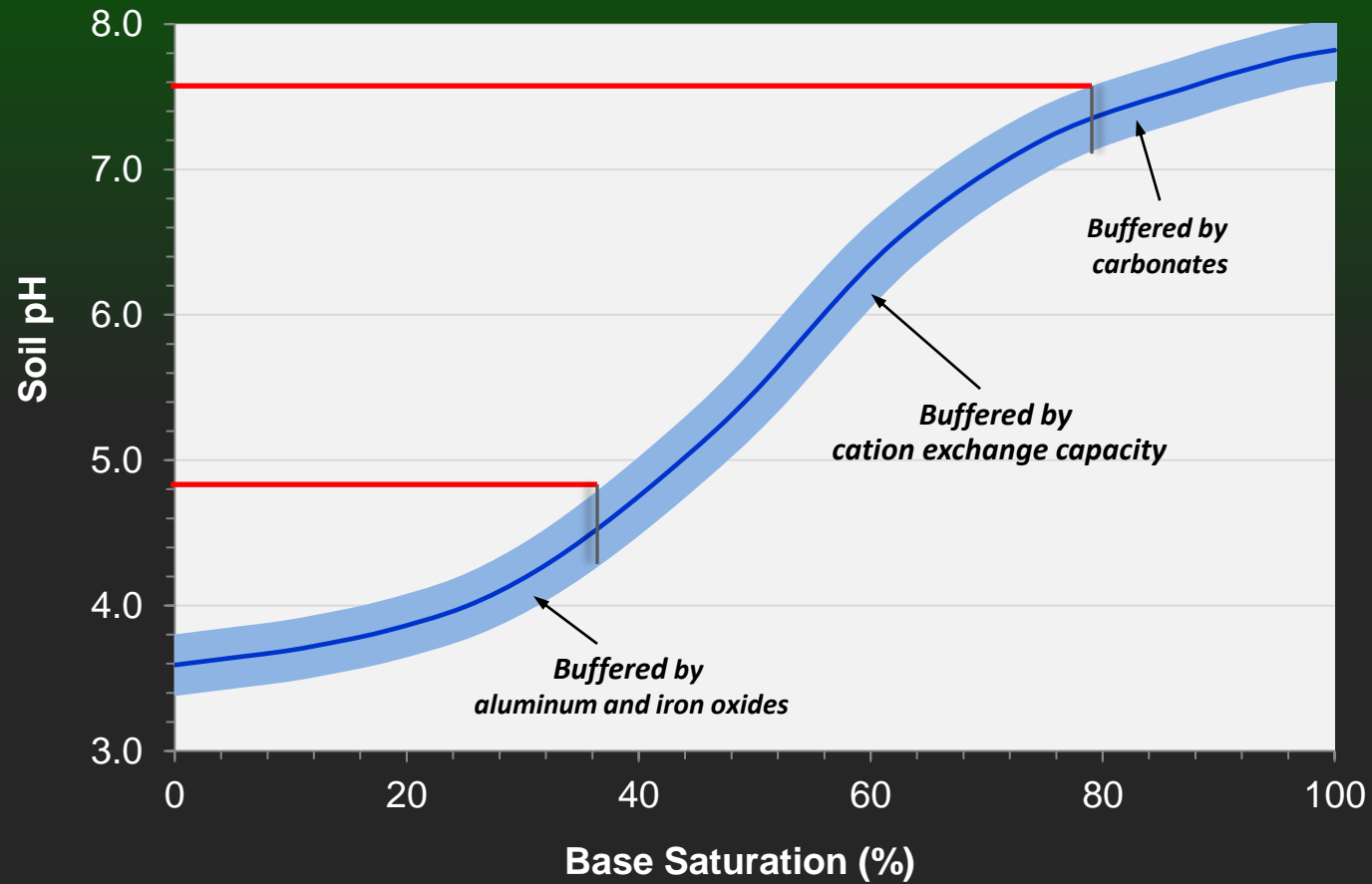
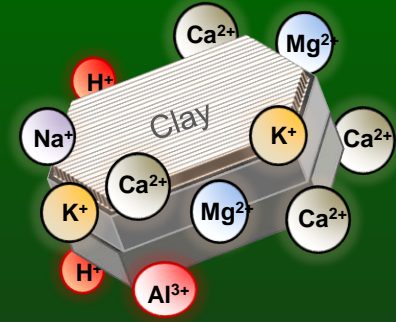
Soil pH is commonly measured by ISE, but extractants and ratios vary by region.



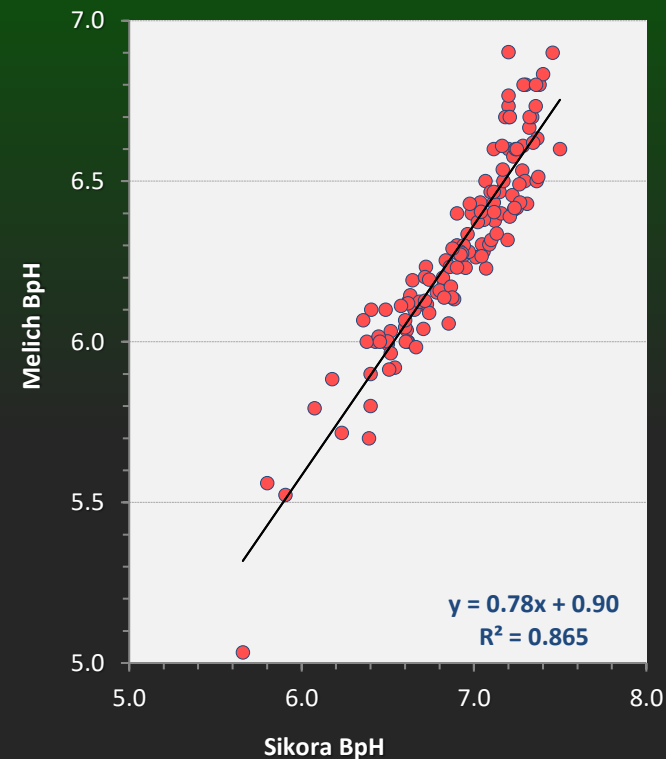
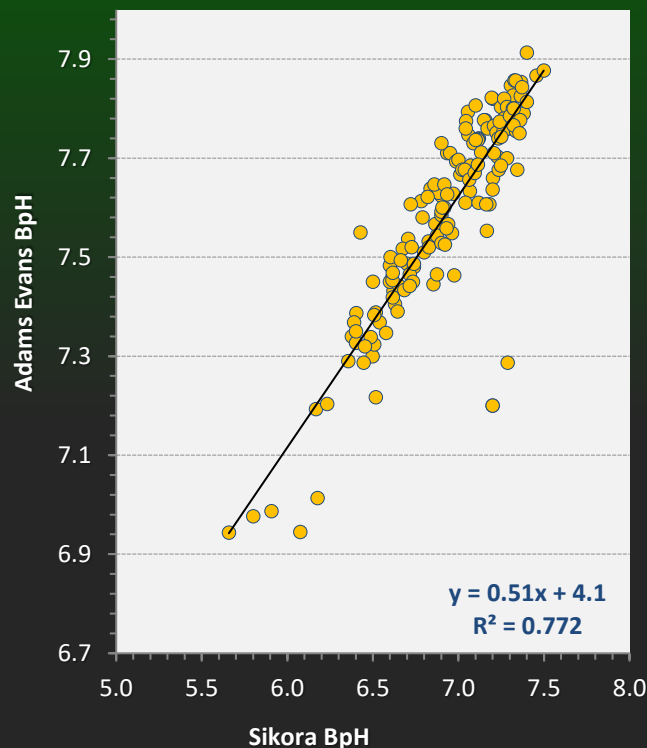
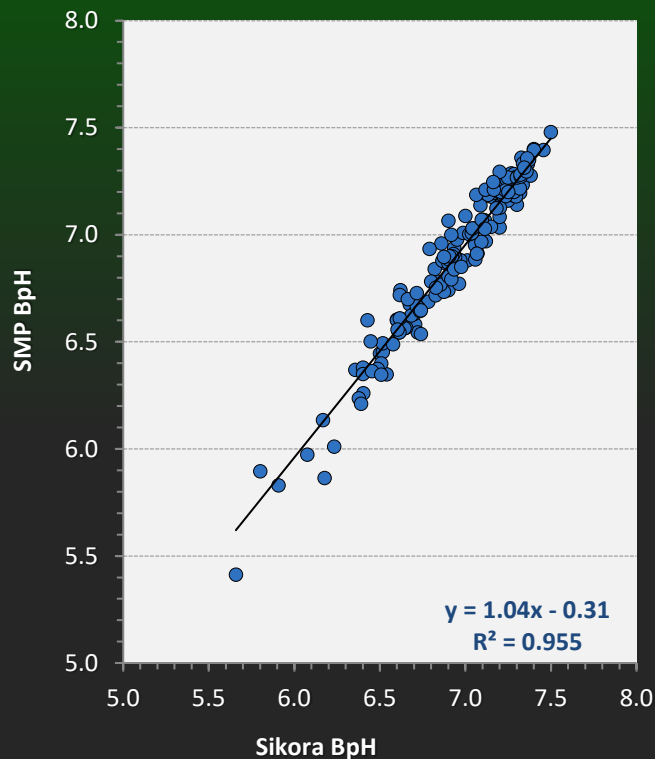
<https://images.ctfassets.net/zmar7hmmcinb/4psmPQM66koakK4aik04uE775776999614becd9346caab5b8aa3e2/n-do-a-soil-3.jpg>



# Soil Acidity and Base Saturation



# BpH method comparison



Data Source: Agricultural Laboratory Proficiency program cycle data base, 2006-2020, 157 soils pH 1:1 0.01M CaCl<sub>2</sub> < 7.2 collected across North America.