

- (i) Compound (A) is
- (a) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Br}$ (b) $\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{Br}$
- (c) $\text{CH}_3\text{C}(\text{CH}_3)_2\text{Br}$ (d) $\text{CH}_3\text{CH}_2\text{CH}_2\text{Br}$
- (ii) Which type of isomerism is present in compound (A) and (C)?
- (a) Positional (b) Functional (c) Chain (d) Both (a) and (c)

OR

Identify compound (B).

- (a) $\text{CH}_3\text{C}(\text{CH}_3)=\text{CH}_2$ (b) $\text{CH}_3\text{CH}=\text{CHCH}_3$
- (c) $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2$ (d) None of these
- (iii) IUPAC name of compound (D) is
- (a) *n*-octane (b) 2,5-dimethylhexane
- (c) 2-methylheptane (d) 3,4-dimethyl hexane.
- (iv) When compound (C) is treated with alc. KOH and then treated with HBr in presence of peroxide, the compound obtained is
- (a) $\text{CH}_3\text{C}(\text{CH}_3)_2\text{Br}$ (b) $\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{Br}$
- (c) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Br}$ (d) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{CH}_3)\text{Br}$

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