

# Utilizing Density Functional Theory to Investigate Polaron Formation in Compositionally Complex Oxides



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## Background

- Polarons: self-trapped or localized charge carrier coupled with a lattice distortion. [1]
- Utilized to explain properties of photovoltaics and piezoelectrics. [2, 3]
- Significance of polarons in new materials such as multifunctional compositionally complex oxides (CCO) is not well understood.

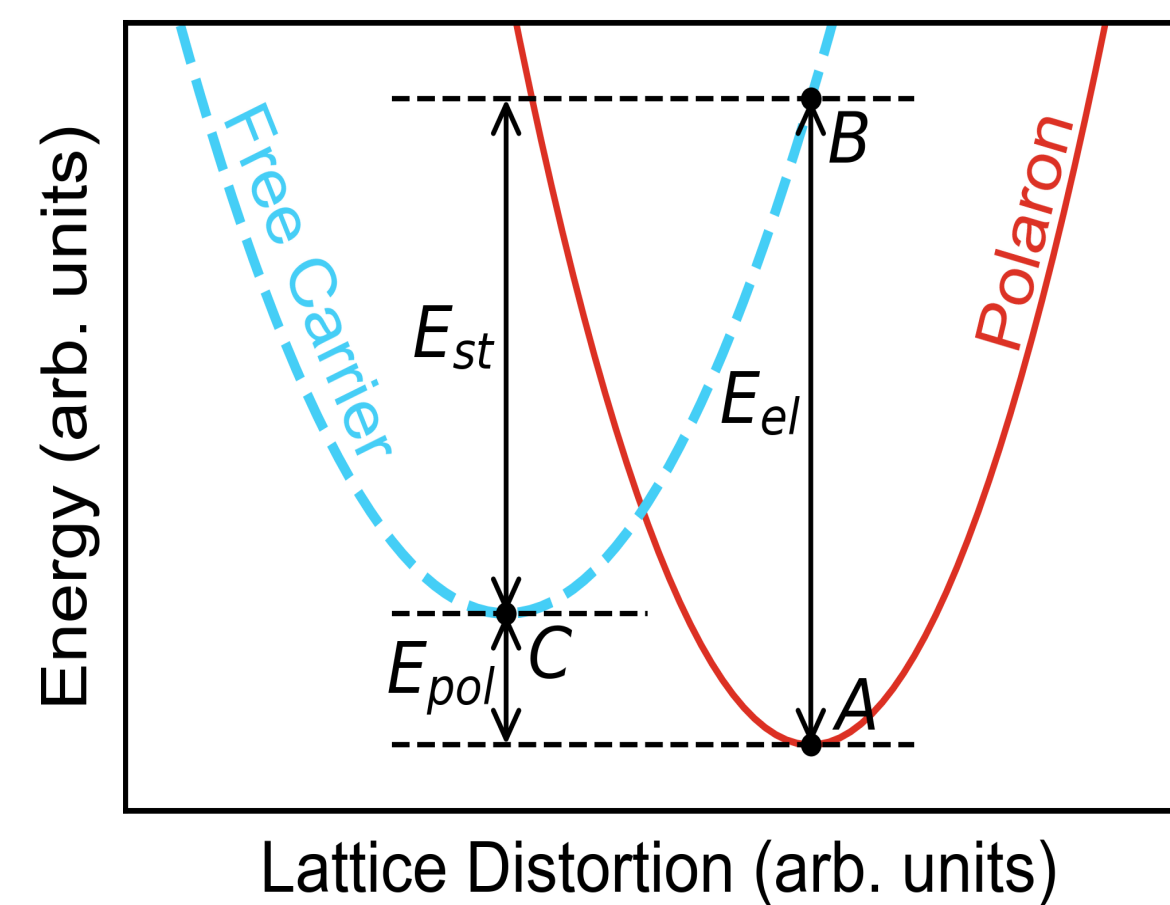
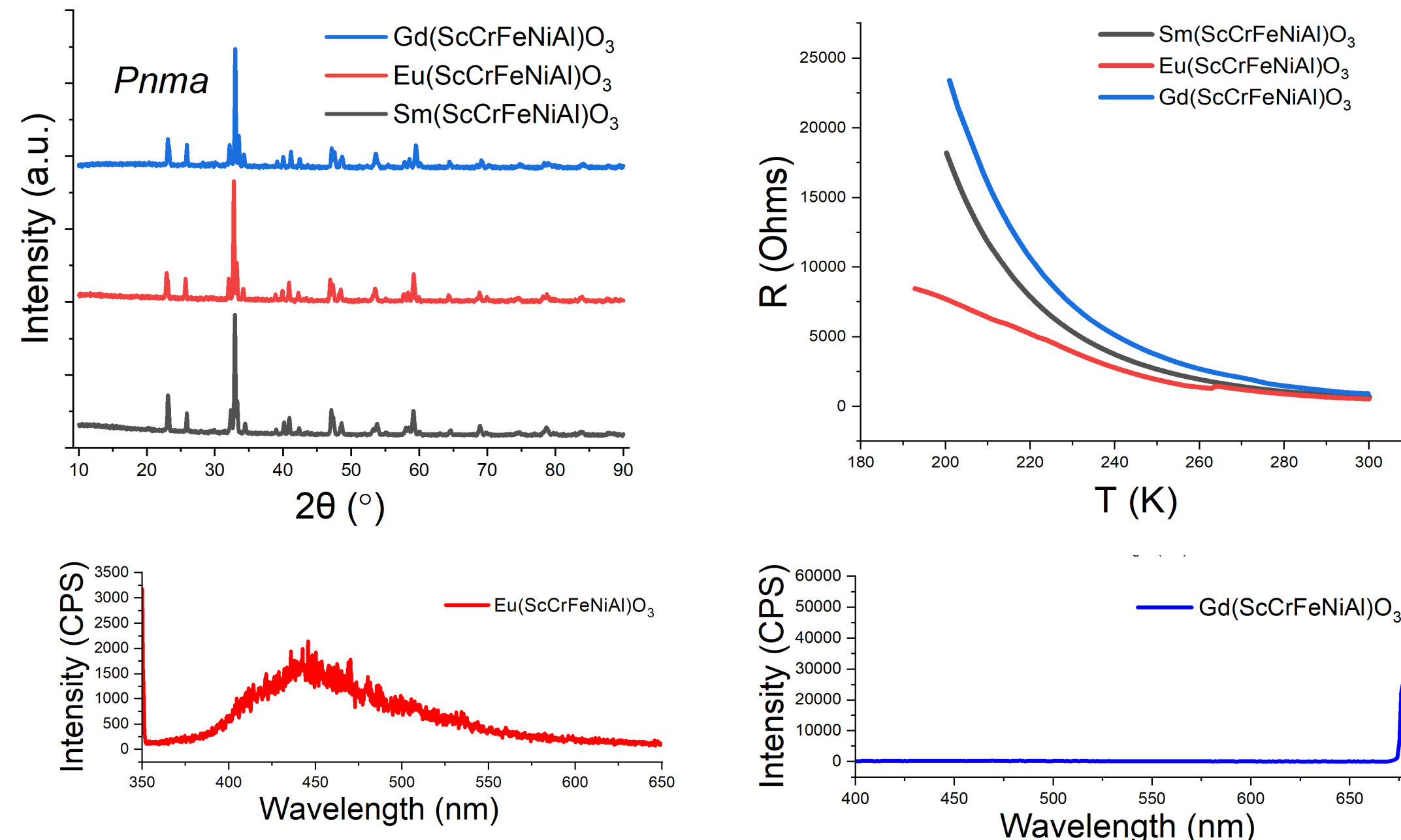
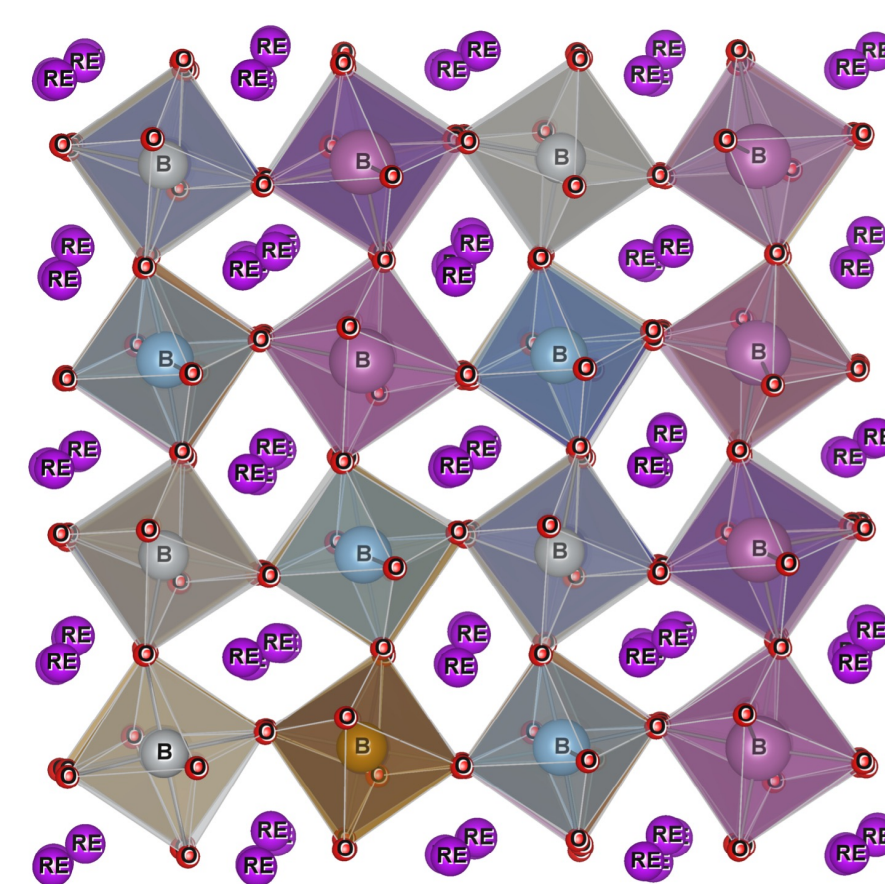


Figure 1: Under certain conditions, the formation of a polaron is thermodynamically favorable ( $E_{pol} = E_{el} - E_{st} < 0$ ) [1]

## Experiment

- The multifunctional CCO  $\text{Eu}(\text{Sc}_{0.2}\text{Al}_{0.2}\text{Cr}_{0.2}\text{Fe}_{0.2}\text{Ni}_{0.2})\text{O}_3$  was formulated to combine properties of ferroelectricity and ferromagnetism.
- Figure 2:  $\text{Eu}(\text{Sc}_{0.2}\text{Al}_{0.2}\text{Cr}_{0.2}\text{Fe}_{0.2}\text{Ni}_{0.2})\text{O}_3$  is a variation of  $\text{A}^{2/3}\text{B}^{3/4}\text{O}_3$  structure with rare earth metals (RE) at the A site.



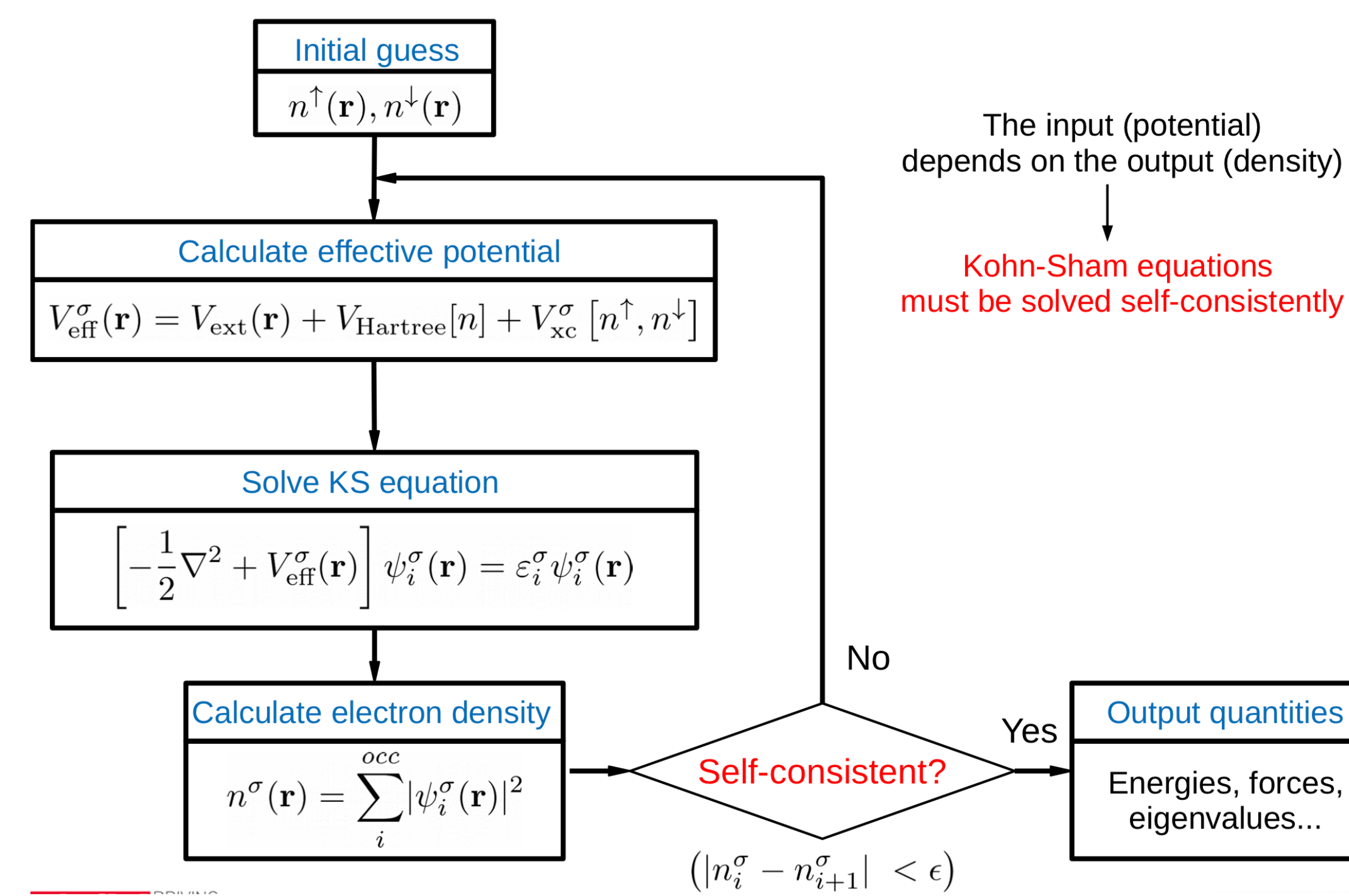
- Figure 3: From top left to bottom right: XRD illustrating the same crystal structure, resistance data highlighting different trends, and photoluminescence spectroscopy data of Eu with extra peaks compared to data collected for Gd.

## Research Questions

- Is there polaron formation on Eu in  $\text{EuXO}_3$ ?
- How can Density Functional Theory (DFT) be used to efficiently model the formation of polarons in compositionally complex oxides?

## Density Functional Theory

Simplifies many body Schrodinger equation to a single variable functional of the total electron density.



- Figure 4: Self consistent field calculations scale based on  $N^3$  where N is the number of electron wavefunctions; many other factors impact computational time such as functional type, density of k-space sampling, and convergence criteria. [4]

Type of Energy Functional	Parameters Used	Computational Cost	Accuracy
Local Density Approximation	$\rho$	Low	Low
Generalized Gradient Approximation	$\rho, \nabla\rho$	Varies	Varies
Hybrid-Functionals	$\rho, \nabla\rho, \nabla^2\rho, \tau$	High	High

- Table 1: The generalized gradient approximation (PBE) was used to reduce computational cost while maximizing accuracy.

## Citations:

- [1] Franchini, C., et al. *Nature Reviews Materials*, 2021. 6(7): p. 560-586.
- [2] Zhang, H., et al. *Journal of Physics: Energy*, 2023. 5(2): p.6-7
- [3] Wang, F, et al. *Proceedings of the National Academy of Science*, 2022. 119(30): p. 2-3
- [4] "The Self-Consistent-Field Cycle". *Siesta Documentation*, docs.siesta-project.org/projects/siesta/en/latest/tutorials/basic/scf-convergence/index.html. Accessed 15 July 2024.
- [5] Pham, T., et al. *American Chemical Society*, 2020. p. 1

## Procedure

- Generate quasirandom unit cell and compare with Monte Carlo simulation.

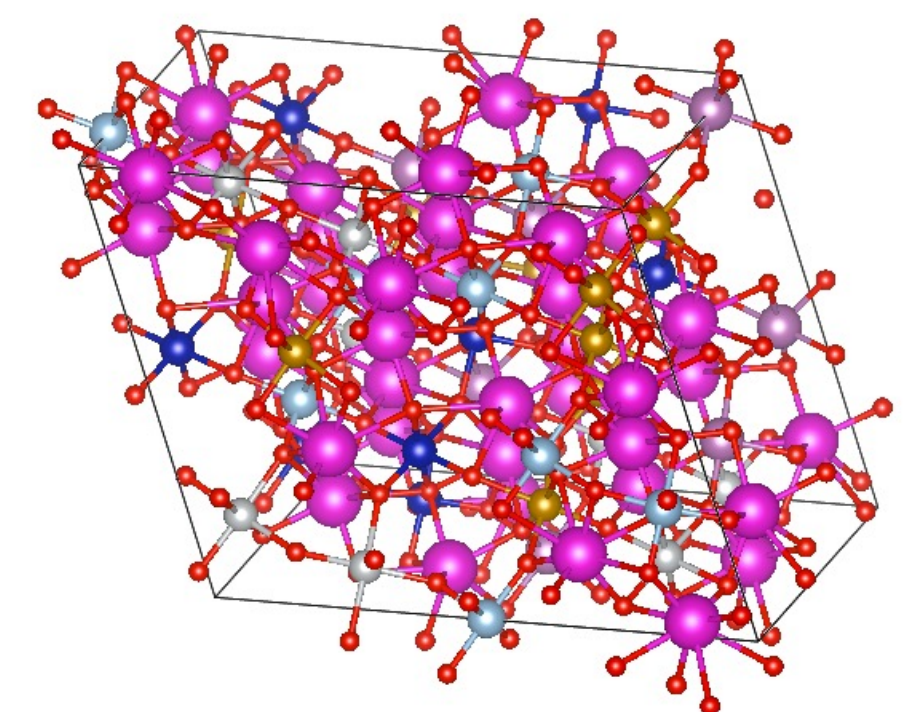


Figure 5: 200 atom unit cell for  $\text{Eu}(\text{Sc}_{0.2}\text{Al}_{0.2}\text{Cr}_{0.2}\text{Fe}_{0.2}\text{Ni}_{0.2})\text{O}_3$

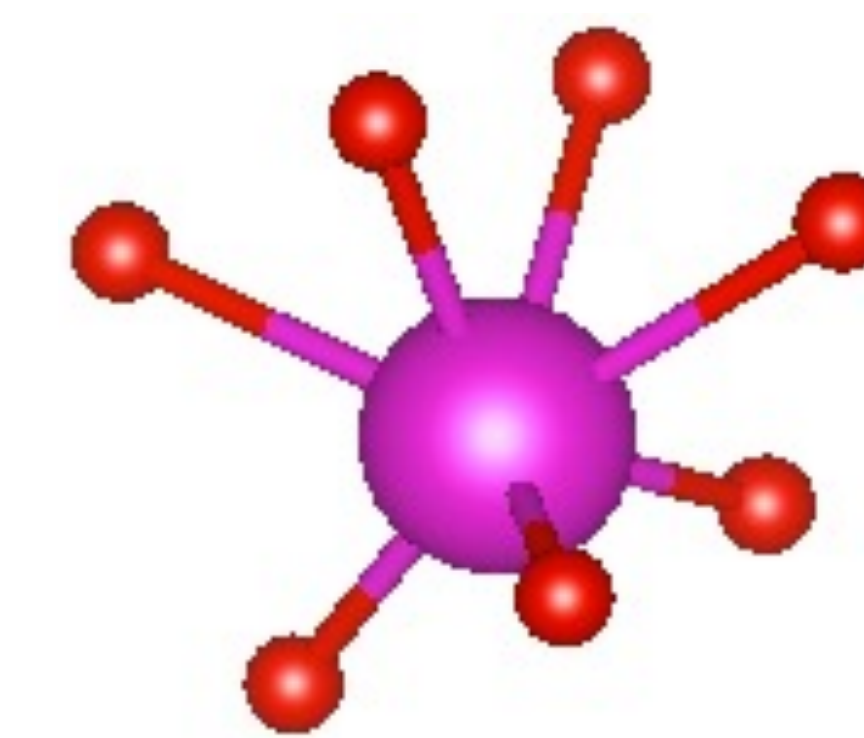
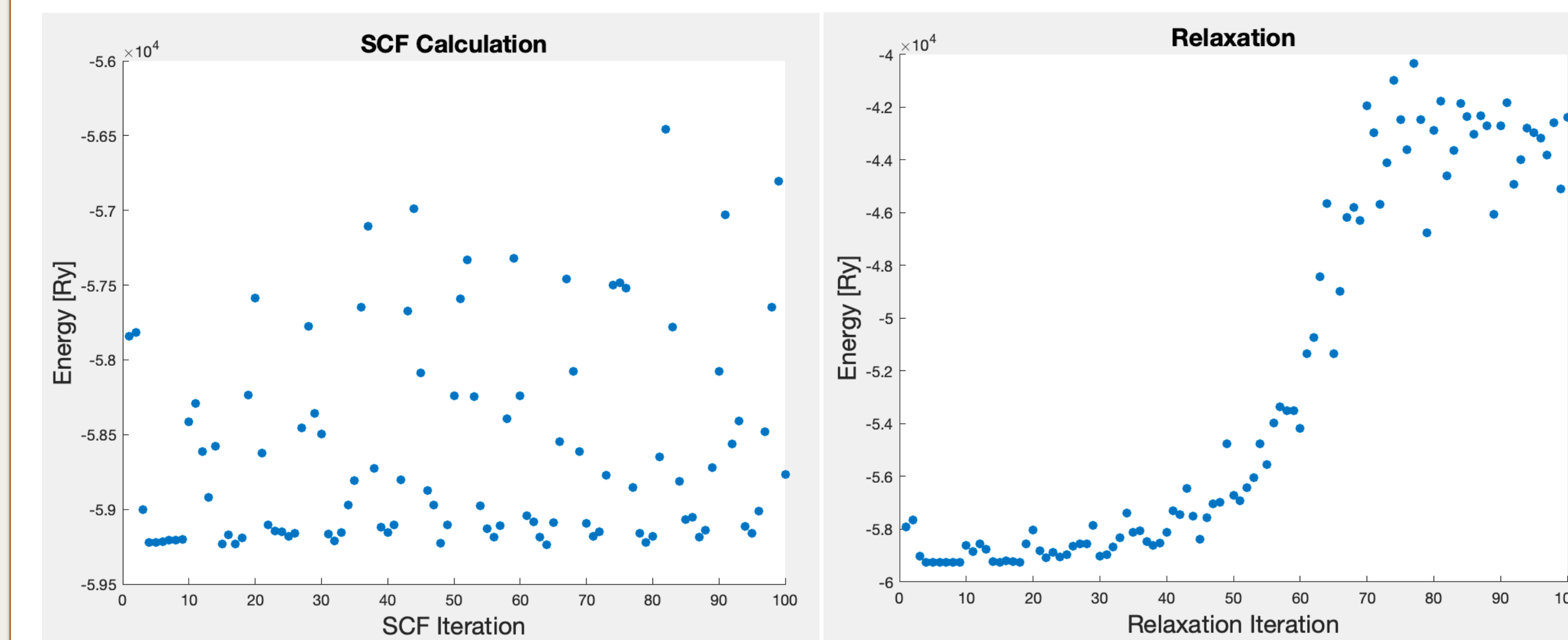


Figure 6: Eu atom bonding with adjacent oxygen atoms

- Bond distortion method<sup>[5]</sup>: increase the distance between oxygen atoms and an Europium atom.
- Increase the Hubbard correction, DFT+U, around the selected Europium atom.
- Add an extra electron.
- Perform self-consistent field calculations to determine electron wavefunctions.

## Results



- Energy vs algorithm iterations illustrate the difficulty of converging CCOs in quantum ESPRESSO.

## Conclusions

- Due to the experimental indication of both  $\text{Eu}^{2+}$  and  $\text{Eu}^{3+}$ , it is very likely that polarons form in this material. However, DFT calculations are currently inconclusive.

## Future Work

- Randomize wavefunctions, compare polaron formation energies for different types of polarons, various magnitudes of bond distortion, and models of  $\text{Eu}(\text{Sc}_{0.2}\text{Al}_{0.2}\text{Cr}_{0.2}\text{Fe}_{0.2}\text{Ni}_{0.2})\text{O}_3$
- Demonstrate that the procedure does not form polarons in  $\text{Gd}(\text{Sc}_{0.2}\text{Al}_{0.2}\text{Cr}_{0.2}\text{Fe}_{0.2}\text{Ni}_{0.2})\text{O}_3$  and  $\text{Sm}(\text{Sc}_{0.2}\text{Al}_{0.2}\text{Cr}_{0.2}\text{Fe}_{0.2}\text{Ni}_{0.2})\text{O}_3$