How Biomaterials Work Their Magic in Dentistry

Souad Menif



Photo by Matthew Poetker on Unsplash

Have you ever wondered how dentists can restore a tooth to look so natural, as if it always belonged? I know I have. And what about how our bodies accept these new teeth, even though they're not technically part of us? No health risks, no immune reactions—just a perfect fit.

Today, we'll uncover the fascinating secrets behind this dental magic. The answer lies in biomaterials, a cornerstone of modern dentistry.

In this article, we'll break down what dental biomaterials are and why they're so important. We'll explore their types, properties, and the science behind them. By the end, you'll have a clear understanding of how biomaterials shape the future of dentistry.

What Are Biomaterials?

Biomaterials are specialized materials designed to interact safely with the body. In dentistry, they play a crucial role in fixing, replacing, or protecting teeth. But why is it so important to understand them?

Because biomaterials determine two things:

- 1. How long dental treatments last
- 2. How natural they look and feel.

For a material to be classified as a biomaterial, it must meet three essential criteria:

- **Biocompatibility:** It should not harm the body. No inflammation, no allergies, no infections. Biocompatible materials should also be sterile to avoid complications.
- **Mechanical Strength:** It needs to withstand daily forces like chewing and grinding without breaking down.
- Aesthetics: The material should blend seamlessly with natural teeth.

Types of Dental Biomaterials



Now that we know the criteria, let's look at the main types of biomaterials used in dentistry:

1. Metals and Alloys:

Metals like titanium and chrome alloys are strong, durable, and resistant to rust. Titanium, in particular, is biocompatible and commonly used in dental implants.

2. Ceramics:

Ceramics are prized for their tooth-like appearance and safety. Some, like hydroxyapatite, even encourage bone growth.

3. Polymers:

Flexible and versatile, polymers like composite resins mimic natural dentin and are ideal for fillings and repairs.

4. Composites:

These materials combine the strengths of different substances. They're both strong and aesthetically pleasing, making them a go-to for cosmetic work.

5. Natural Materials:

Derived from biological sources, these materials sometimes support bone healing or regeneration.

With this foundation, let's dive deeper into each category to understand their unique roles. (check this <u>link</u> for a broader overview of all the materials existing in dentistry).

Polymers: The LEGO Blocks of Dentistry



Photo by <u>Nik</u> on <u>Unsplash</u>

Imagine building with LEGO blocks. Each block represents a small molecule (a monomer), and when you snap them together, you create a chain. That's essentially how polymers are made. These long chains give polymers their unique flexibility and durability, making them the go-to material for many dental applications.

How Polymers Are Made

Polymers come to life through a process called **polymerization**, which works in two main ways:

- 1. **Step-Growth Polymerization (Polycondensation):** Think of this as assembling a necklace, bead by bead, where each addition releases something small, like water or alcohol. This method is common in adhesives and resins.
- 2. **Chain-Growth Polymerization (Polyaddition):** Picture stacking a tower with blocks, one at a time, without anything extra falling off. This is how dental composites are created.



source

Key Properties and Applications

Polymers can adapt to many roles while being lightweight and safe for the human body. Here are some ways they're used:

- Acrylic Resins: acrylic resins create denture bases and temporary crowns that fit perfectly.
- **Composite Resins:** Like a touch-up paint for your teeth, these materials blend seamlessly with your natural tooth color.
- **Sealants:** Think of them as raincoats for your teeth, shielding them from decay.

Recent Advancements in Polymer Technology

Polymers are also getting smarter, with recent innovations making them more functional and long-lasting:

- **Smart Composites:** These materials release helpful ions like calcium and fluoride when the pH in your mouth drops, helping to rebuild and strengthen teeth. <u>1.2</u>
- **Self-Healing Composites:** When damaged, these materials can repair themselves, making dental restorations last longer.<u>2.3</u>
- **Bioactive Glass Ionomer Cement:** This material acts like human dentin and releases fluoride, helping to prevent cavities while blending naturally with teeth.<u>4.5</u>

Types of Polymers

Polymers also come in two main categories:

- **Thermoplastics:** Think of wax; you can reshape it when heated, making it handy for adaptable applications.
- **Thermosetting Polymers:** Once set, they're as solid as cured concrete, providing the strength needed for durable restorations.

Advantages and Drawbacks

While polymers are versatile and aesthetically pleasing, they're not without limitations. They lack the strength of metals or ceramics and can degrade over time, especially in moist environments like the mouth. Researchers are continuing to experiment with polymers, but for now, we rely on other materials that offer greater mechanical resistance. The most obvious choice? Metals and alloys!

Metals and Alloys: Strength in Diversity

Why use metal mixtures instead of pure metals? The answer lies in **enhancing mechanical properties** and addressing material weaknesses. Let's dive into how metals and their alloys contribute to dentistry.

What Are Metals?

Crystalline Solids: Metals belong to a group of materials called crystalline solids, where atoms are arranged in an orderly pattern. This structured arrangement directly impacts properties like **strength** (resistance to deformation) and **ductility** (ability to stretch without breaking).



source

Crystallography: This is the study of how atoms are arranged in crystalline materials. In dentistry, understanding crystallography is key to predicting how metals and alloys perform under stress.

Common Crystal Structures

Metals and alloys arrange their atoms into specific patterns, each with unique traits:



• Body-Centered Cubic (BCC):

Picture a cube with atoms at each corner and one in the center. This structure is strong but not very flexible, making it great for rigid applications

• Face-Centered Cubic (FCC): Imagine stacking oranges in a box—atoms sit at each corner and the center of each face. This arrangement is strong and flexible, perfect for durable restorations like dental crowns.

• Hexagonal Close-Packed (HCP): Think of a honeycomb made of layers of atoms stacked tightly in a hexagon shape. It's incredibly strong but less bendable, ideal for materials that need to hold their shape under pressure.

What Are Alloys, and Why Use Them?

An **alloy** is a combination of two or more metals designed to create a material that's stronger, more flexible, or more resistant to wear and tear than the original metals alone. Alloys form two types of solid solutions:



source

- 1. **Substitutional Solid Solution:** Atoms of the added metal replace atoms in the base metal.
- 2. **Interstitial Solid Solution:** Smaller atoms fit into the gaps between atoms of the base metal.

Types of Alloys in Dentistry

Dental alloys are classified based on their composition:

- **Precious Alloys:** Contain over 60% noble metals (like gold) with at least 40% gold content.
- Semi-Precious Alloys: Contain at least 25% noble metals.
- Non-Precious Alloys: Contain less than 25% noble metals.

• **Titanium and Titanium Alloys:** Known for **biocompatibility** (minimal adverse reactions in the body) and strength, making them ideal for implants.

The **American Dental Association (ADA)** further categorizes alloys based on their strength:

- **Type 1:** Low strength, used for inlays.
- **Type 2:** Medium strength, suitable for crowns.
- **Type 3:** High strength, ideal for bridges.
- **Type 4:** Extra-high strength, used in removable partial dentures.

Heat Treatments

Heat treatments make dental alloys stronger, harder, and more flexible, helping them last longer in crowns, bridges, and implants.



source

Homogenization Treatment smooths out the inside of the alloy. First, it's heated to about 1100 °C, held at that temperature for a while, and then cooled down slowly. This process evens out the alloy's structure, reduces stress, and makes it stronger and more flexible. <u>9.11</u>

Hardening Treatment focuses on making the alloy harder. The alloy is heated to 800–1200 °C and then quickly cooled by dipping it in water or oil (called quenching). This locks the atoms in place, creating a harder structure that's still flexible enough to avoid breaking. <u>10.12</u>

These treatments ensure dental alloys can handle everyday chewing and biting, staying strong and reliable over time.

Titanium: A Game-Changer in Dentistry



Photo by <u>Yuheng Ouyang</u> on <u>Unsplash</u>

Titanium, commonly found in phones, is also one of the most essential materials in dentistry. Known for being light, strong, and highly resistant to corrosion, it offers a perfect combination of durability and biocompatibility, making it ideal for dental implants, orthodontic tools, and prosthetics.

What Makes Titanium Special?

One of titanium's most fascinating features is how its atomic structure changes with temperature. At high temperatures, it has a loose cubic structure (called the β -phase). When it cools below 882°C, this shifts to a tighter hexagonal structure (α -phase). This adaptability gives it the strength and flexibility needed for complex dental applications.



Titanium also comes in grades based on its oxygen content. These variations affect its strength and ductility, making it versatile enough to suit different dental uses, from lightweight tools to robust implants.

Tailored for Dentistry

Titanium alloys are designed for specific needs:

- Alpha alloys are lightweight and strong, suited for simpler applications.
- **Beta alloys** offer high mechanical resistance and are easy to machine for intricate designs.
- Alpha+Beta alloys combine strength and flexibility, making them ideal for implants and prosthetics.

Why Titanium Works in Dentistry

Titanium is a favorite in dentistry for its compatibility with the body. It resists corrosion, avoids irritating tissues, and even discourages plaque buildup. For implants, it bonds seamlessly with bone, creating a strong, lasting connection. Its flexibility also makes it ideal for orthodontic tools, aligning teeth with comfort and precision.

While titanium is incredibly durable, its high melting point can make it tricky to work with. That's where ceramics step in. With their natural look and strength, ceramics complement titanium perfectly, offering solutions that are both functional and aesthetically pleasing.

Ceramics: Natural Beauty for Dental Restorations

Ceramics are the best for when we need to mimic natural teeth (aesthetic restorations). They are biocompatible, meaning safe for use in the body, and resist both wear and corrosion, making them ideal for long-lasting and aesthetic dental solutions.

What Are Ceramics Made Of?

Ceramics have a unique structure that combines:



- Glassy Matrix: Provides translucency and a natural tooth-like appearance.
- **Crystalline Phases:** Reinforce the material, improving strength and resistance to temperature changes.

In dentistry, ceramics are made primarily of oxides (around 99%) and are shaped through **sintering**.

Sintering is a process where ceramic powders are heated just enough to bond together without fully melting—similar to baking clay for pottery.

Types of Ceramics

Different ceramics are used for different dental applications:

- **Feldspathic Ceramics:** Traditional porcelains known for their high translucency, perfect for veneers and front-tooth crowns.
- **Glass-Ceramics:** A mix of glass and crystals, offering both strength and aesthetic qualities for back-tooth restorations.
- **Zirconia Ceramics:** Extremely tough and durable, often used for crowns, bridges, and implant abutments to handle heavy chewing forces.

Why Are Ceramics So Popular in Dentistry?

Ceramics have unique properties that make them ideal for dental use:

- **Natural Appearance:** They closely mimic the color, translucency, and shine of real teeth.
- Strength: They can withstand significant biting and chewing forces.
- **Resistance:** Ceramics are resistant to wear, corrosion, and staining.
- **Biocompatibility:** They are safe and non-toxic for use in the human body.

However, ceramics can be brittle, which means they may crack under extreme stress, like biting something very hard.

Where Are Ceramics Used in Dentistry?

Ceramics are used in a variety of dental treatments:

- **Veneers:** Thin shells that enhance the appearance of front teeth.
- **Crowns:** Caps that protect and restore damaged teeth.
- **Bridges:** Fixed prosthetics that replace missing teeth by anchoring to adjacent teeth.

• **Implants:** Ceramic components that connect dental implants to crowns or dentures.

Ceramics are a key material in modern dentistry, offering unmatched aesthetics and durability. While they closely mimic natural teeth and resist wear, their brittleness remains a challenge. Despite this, they continue to be a reliable choice for many dental restorations. To better understand their performance under different conditions, we turn to equilibrium diagrams, which help predict and optimize their behavior in various dental applications.

Equilibrium Diagram: A Map for Material Behavior

An equilibrium diagram, or phase diagram, is like a roadmap for materials, showing how their different parts (called phases) shift with changes in temperature and composition.

Think of it as a weather map—but instead of forecasting storms or sunshine, it predicts how metals and alloys react under various conditions. This essential tool in materials science, including dentistry, helps us understand and fine-tune materials for specific applications.

Why Are Equilibrium Diagrams Important in Dentistry?

In dental applications, equilibrium diagrams help in three key ways:

1. Predicting Phase Behavior:

Think of phases as the "states" a material can be in—like ice, water, and steam for water. An equilibrium diagram shows what phases an alloy will have at a given temperature and composition. This is crucial for picking materials that will perform well in a patient's mouth, like ensuring a crown can handle daily wear without breaking.

2. Guiding Alloy Formation:

When creating dental alloys, the diagram acts like a recipe. It shows how to mix metals to get the right "ingredients" for strength, corrosion resistance, or aesthetic appeal. For example, tweaking an alloy's composition might produce a phase that makes it stronger for bridges or crowns.

3. Optimizing Heat Treatment:

Heat treatments like hardening or sintering rely on controlling phases in a material. The equilibrium diagram helps identify the right temperature and

time to create the perfect microstructure. This ensures that metals or ceramics used in restorations are strong, durable, and long-lasting.



Example: The Gold-Copper Phase Diagram

- Adding more copper can make the alloy stronger, which is great for bridges.
- Balancing the mix keeps the alloy shiny and corrosion-resistant, ideal for crowns.

Equilibrium diagrams are like cheat sheets for dental materials. They help predict how materials will behave, guide alloy creation, and optimize treatments. With this knowledge, dental professionals can design restorations that look great, feel strong, and last a long time.

Shaping Dental Materials: Subtractive Manufacturing, Additive Manufacturing, and CAD/CAM



Modern dentistry relies on precise shaping techniques to create functional and aesthetic dental restorations. Two main methods are used to shape materials: **subtractive manufacturing** and **additive manufacturing**, both of which are enhanced by **CAD/CAM technology**.

Subtractive Manufacturing: Shaping by Removing

Subtractive manufacturing is like sculpting from a block of stone—it starts with a solid piece of material, and excess is removed until the desired shape is achieved.

- How It Works: A milling machine uses a rotating tool to carve away material with incredible precision.
- Best For: Metals and ceramics, where durability and accuracy are essential.
- **Example:** This method is often used to create crowns and bridges that must fit perfectly and withstand heavy chewing forces.

Additive Manufacturing: Building Layer by Layer

Additive manufacturing is more like assembling a puzzle, where layers are added one by one to form the final object.

- How It Works: 3D printing techniques, such as **stereolithography (SLA)** and **fused deposition modeling (FDM)**, use digital models to create complex designs.
- **Best For:** Custom dental appliances, implants, and prototypes with intricate shapes.
- **Example:** A dental implant guide made through 3D printing ensures precision during surgery while being fully customized for the patient.

CAD/CAM Technology:

At the heart of these processes is **CAD/CAM technology** (Computer-Aided Design and Manufacturing). This technology connects the design and fabrication steps seamlessly:

- 1. **Digital Design:** A dentist scans the patient's tooth to create a detailed 3D model using CAD software.
- 2. **Fabrication:** CAM software sends the design to a milling machine or 3D printer, which shapes materials like zirconia or porcelain into the restoration.

Why It Matters:

- **Precision:** Every restoration fits perfectly, minimizing adjustments.
- **Speed:** Crowns and bridges can be fabricated in a single appointment.
- **Aesthetics:** Materials like porcelain achieve a natural look that matches the patient's teeth.

Conclusion:

Understanding biomaterials isn't just for dentists. Patients benefit from knowing how these materials work, why they're chosen, and how they impact the longevity and appearance of dental treatments. Biomaterials ensure that dentistry isn't just functional but also beautiful and enduring. From polymers to ceramics, every material has a role in creating the perfect smile. And as technology advances, the possibilities for biomaterials in dentistry continue to expand. But how might environmental concerns shape the development and disposal of dental biomaterials in the future?

—

Have questions or thoughts? Let's discuss them below! Connect to me via <u>LinkedIn</u> or email me at : <u>souad.menif@gmail.com</u>.

Sources

- 1. <u>Smart Biomaterials: An Evolving Paradigm in Dentistry PMC</u>
- 2. Smart Material for Smarter Dentistry-PMC
- 3. <u>Smart composites The new era in smart dentistry</u>
- 4. <u>Smart materials in dentistry: A review IJARnD</u>
- 5. <u>Smart materials in dentistry</u>
- 6. <u>Structure and Properties of Cast Dental Alloys Pocket Dentistry</u>
- 7. <u>Crystal Structures of Metals Chemistry LibreTexts</u>
- 8. <u>Structure of Metals and Alloys Pocket Dentistry</u>
- 9. "Homogenization Heat Treatment Influence on Microstructure Evolution and Mechanical Properties for an Alloy Used in Lightweight Aerospace Applications." Thermal Processing. <u>Link</u>
- 10. "Heat Treatment of Dental Alloys: A Review." ResearchGate. Link
- 11. "Study on the Surface of Cobalt-Chromium Dental Alloys." PMC. Link
- 12. "Influence of Homogenization on Phase Transformations during Heat Treatment." MDPI. Link
- 13. "Dental Ceramics: A Comprehensive Guide To Modern Restorative Dentistry." Land of Smile. <u>Link</u>
- 14. "Ceramics overview | British Dental Journal." Nature. Link
- 15. "Dental ceramics: An update." PMC. Link
- 16. "Dental Ceramics." Pocket Dentistry. Link
- 17. "Ceramics in dentistry." ResearchGate. Link
- "An overview of dental glass-ceramics: From material design to clinical application." Wiley Online Library. <u>Link</u>
- 19. "Phase Diagrams in Materials Science." ASM International. Link
- 20. "Understanding Phase Diagrams." Materials Science & Engineering. Link
- 21. "Phase Diagrams for Ceramics." Wiley Online Library. Link

- 22. "Procédés modernes de mise en forme des matériaux en odontologie." HAL. Link
- 23. "Les différents matériaux utilisées dans la CFAO dentaire." Kreos Dental. Link
- 24. "La CFAO dentaire L'incontournable de la dentisterie moderne." Owandy. Link
- 25. "Matériaux et CFAO dentaire." Le Fil Dentaire. Link