

pH in gaseous substances

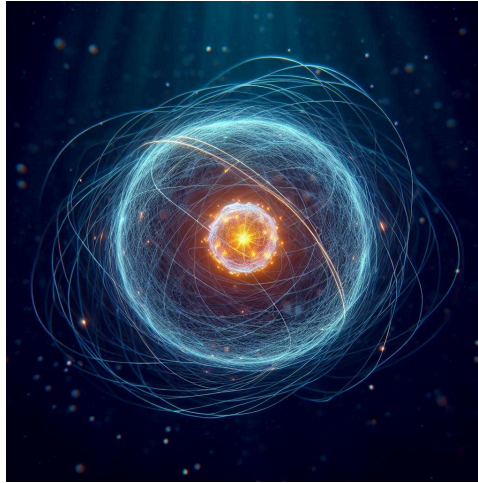


Illustration: *Hydrogen - The electron seems missing, since it is for perception.*

Pressure defines pH. The weak force is controlled. Under pressure, which might operate differently as we have learned, Hydrogen-electrons will operate differently. The strong force comes into the matter. It is interactions.

Question: *How much pressure is within gaseous substances vs. solid matter? Is this happening within a vacuum?*

Low (acid) pH has a tendency to destroy biological matter. This is a result of the charge. Electrons are a building block for life, and matter, which means that perception comes into the picture. It is classified as a Lepton. Leptons equals perception based subatomic particles. By this meaning what is needed to live and survive. Fix the system.

Lepton - Perception

Photon (electron+positron') - Perception

Touchable matter - Quark

The Boson is a solid proof of vibrations and how wave functions operate.

pH has more energy in gaseous substances than solid matter,

