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HIRANUMA APPLICATION DATA		Automatic Titrator	Data No.	L11	Jun.6. 2017
Lubricant petroleum products	Base number in Lubricant oil (Perchloric acid / Back-titration)				

1. Abstract

The base number of lubricant oil is one of the important index for judging its quality.

Measurement of base number is defined in several standard test methods. It is indicated by “milligrams of potassium hydroxide equivalent weight to acid required to neutralize basic components contained in 1 g of the sample”. There are two methods of base number, hydrochloric acid method and perchloric acid method. In addition, there are two types of perchloric acid titration, forward-titration and back-titration. In this article, perchloric acid method with back-titration will be introduced. The international standard methods for base number with perchloric acid method are shown as bellow.

- JIS K2501 2003: Petroleum products and lubricants – Determination of neutralization number
- ASTM D2896-06 : Standard Test Method for Base Number of Petroleum Products by Potentiometric Perchloric Acid Titration

The potentiometric titration process is as follows:

- 1) Weigh sample exactly corresponding to base number and dissolve it in a titration solvent.
- 2) Immerse glass electrode and reference electrode.
- 3) Add fixed amount of perchloric acid in acetic acid solution to the sample.
- 4) Start titration with sodium acetate in acetic acid solution.

There are two procedures for perchloric acid titration with back-titration, A and B on ASTM D2896. Procedure A and B use different titration solvent volume and sample weight. In this article, measurement with procedure B will be applied.

2. Configuration of instruments and Reagents

(1) Configuration of instruments.

Main unit	: Hiranuma Automatic Titrator COM Series One buret Option
Electrode	: Glass electrode GE-101B : Reference electrode RE-201Z Inner solution should be changed; it is described below.

(2) Reagents

Titrant	: 0.1 mol/L sodium acetate in glacial acetic acid standard solution
Additive	: 0.1 mol/L perchloric acid in glacial acetic acid standard solution
Titration solvent	: Mixture of 500 mL of glacial acetic acid and 1 L of chlorobenzene Refer; Toluene can be used instead of chlorobenzene.
Inner solution	: Saturated sodium perchlorate in glacial acetic acid

3. Measurement procedure

(1) Take 1 g of sample into 100 mL beaker and weigh accurately to 0.1 mg digits.

Note that the weight of sample will be changed depending on the base number.

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pages. the solution.
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 CONTENTS OF BEAKER IS SUITED FOR 2 ml. Standard solution is added with buret, and the

- (4) Titrate by 0.1 mol/L sodium acetate in acetic acid standard solution.
 Also, perform the blank test with the same procedure of sample measurement.

4. Measurement conditions and results

Examples of titration conditions

(1) Measurement of blank

i) Dispense 0.1 mol/L perchloric acid in acetic acid standard solution.

Cndt No.	43
Method	Disp
Buret No.	1
S-Timer	0 sec
Disp.Vol.	4 mL

ii) Titration with 0.1 mol/L sodium acetate in acetic acid standard solution.

Cndt No	45	ConstantNo.	45	Mode No.	8
Method	Auto	Size	0 g	Pre Int	0 sec
Buret No.	2	Blank	0 mL	Del K	5
Amp No.	1	Molarity	0.1 mol/L	Del Sens	0 mV
D. Unit	mV	Factor	1.006	Int Time	5 sec
S-Timer	20 sec	K	0	Int Sens	3 mV
C.P. mL	0 mL	L	0	BrT Speed	2
T Timer	0 sec	Unit	mL	Pulse	40
D.P. mL	0 mL	Formula	D		
End Sens	300	Digits	4		
Over mL	0.3 mL	Auto In Pram.	None		
Max.Vol.	20 mL				

(2) Measurement of sample

i) Dispense 0.1 mol/L perchloric acid in acetic acid standard solution.

Cndt No.	43
Method	Disp
Buret No.	1
S-Timer	0 sec
Disp.Vol.	4 mL

ii) Titration with 0.1 mol/L sodium acetate in acetic acid standard solution.

Cndt No.	44	ConstantNo.	44	Mode No.	8
Method	Auto	Size	1.0307 g	Pre Int	0 sec
Buret No.	2	Blank	3.979 mL	Del K	5
Amp No.	1	Molarity	0.1 mol/L	Del Sens	0 mV
D. Unit	mV				

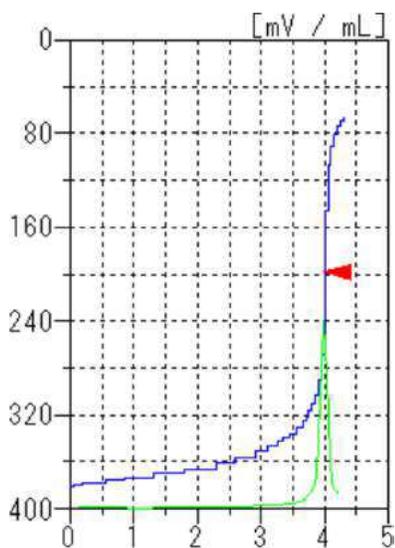
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<p>pages.</p> <p>ABOUT US PRODUCT APPLICATIONS</p>		<p>Int time 5 sec</p> <p>Int Sens 3 mV</p> <p>Brk Speed 2</p> <p>Pulse 40</p>
<p>Unit mg/g</p> <p>Formula ((B-D)*K*F*M/S</p> <p>Digits 4</p> <p>Auto In Pram. Non</p>	<p>CONTACT</p>	

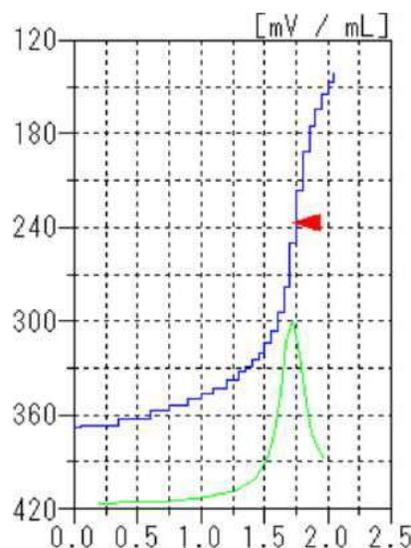
Measurement results

Measurement of blank		
Measurement number	Size (g)	Titer (mL)
1	-	3.979
2	-	3.978
Avg. (Blank)		3.979 mL

Measurement of sample			
Measurement number	Size (g)	Titer (mL)	Base number (mgKOH/g)
1	1.0307	1.719	12.375
2	1.0015	1.779	12.398
3	1.0108	1.763	12.373
Statistic calculation		Avg.	12.4 mgKOH/g
		SD	0.0138 mgKOH/g
		RSD	0.11 %



Measurement of blank



Measurement of sample

Examples of measurement curves

5. Note

(1) Management of the electrode It is recommended to activate the electrodes for about 5 minutes to pure water for each measurement. This is because when glass electrode is used for a long time in a nonaqueous solvent, the response speed and electromotive force decrease. Since the electrolyte of the inner solution and the oil sample may adhere around the liquid junction of reference electrode, it causes the fluctuation of the potential. Therefore it is recommended to clean the liquid junction periodically.

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nt. Temperature change of 1 ° C causes the
surement should be performed at the same

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room temperature as much as possible.

(3) Preparation of inner solution for reference electrode

The inner solution of the reference electrode RE-201Z is filled with saturated KCl aqueous solution when purchased. For this measurement, it is necessary to replace inner solution to saturated sodium perchlorate in acetic acid solution. Replacement procedure is described below.

- i) Prepare the saturated solution of sodium perchlorate in acetic acid with reagent grade of these.
- ii) Discharge inner solution from reference electrode RE-201Z and wash inside it with water and then acetic acid.
- iii) Fill the prepared inner solution into reference electrode from the supply port.
- iv) Leave the electrode for one day before use.

Keywords: JIS K2501, ASTM D2896, Lubricant oil, Neutralization number, Base number, Potentiometric titration, Perchloric titration, Back-titration

*Some measurement would not be possible depending on optional configuration of system.



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