# SCIENCE SQP (2024-25) CLASS X (Science 086)

#### Max. Marks: 80

#### **Time Allowed: 3 hours**

#### **General Instructions:**

- 1. All questions would be compulsory. However, an internal choice of approximately 33% would be provided. 50% marks are to be allotted to competency-based questions.
- 2. Section A would have 16 simple/complex MCQs and 04 Assertion-Reasoning type questions carrying 1 mark each.
- 3. Section B would have 6 Short Answer (SA) type questions carrying 02 marks each.
- 4. Section C would have 7 Short Answer (SA) type questions carrying 03 marks each.
- 5. Section D would have 3 Long Answer (LA) type questions carrying 05 marks each.
- 6. Section E would have 3 source based/case based/passage based/integrated units of assessment (04 marks each) with sub-parts of the values of 1/2/3 marks.

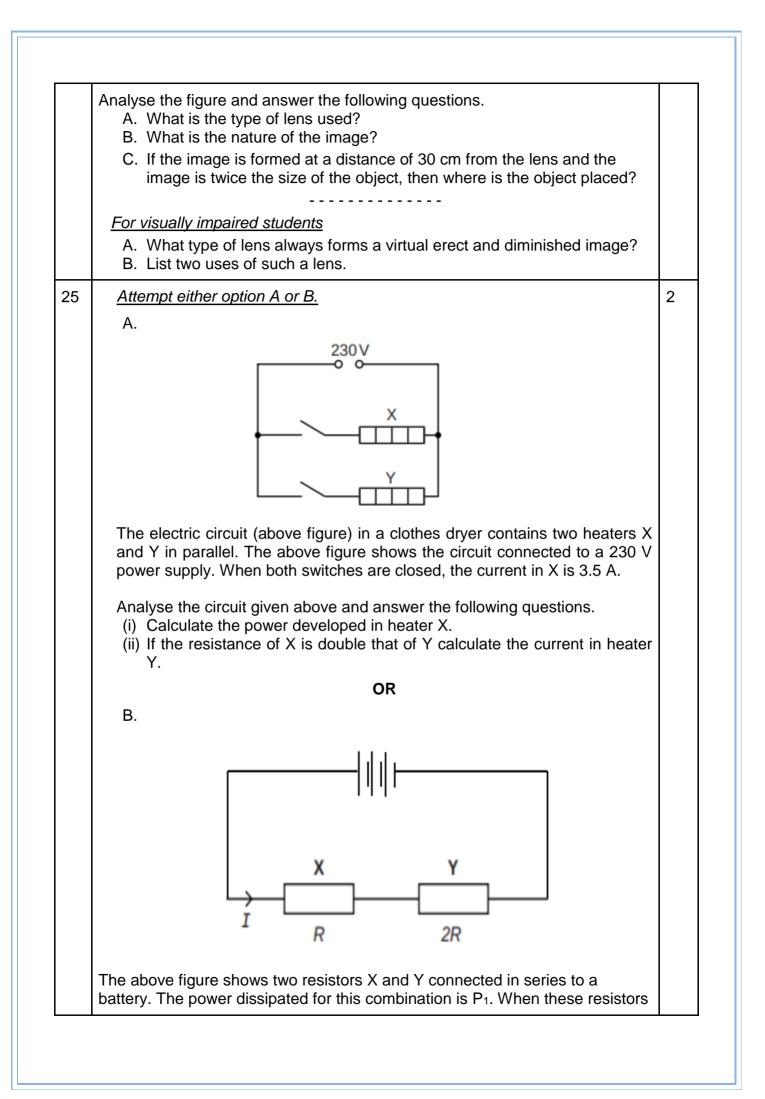
		Section-A 1 to 16 are multiple choice questions. Only one of the choice correct choice as well as the answer to these questions.	es is correct. Sele	ect and	
1	р F / !	entify 'p', 'q' and 'r' in the following balanced reaction Heat Pb (NO <sub>3</sub> ) <sub>2(s)</sub> > q PbO <sub>(s)</sub> + r NO <sub>2(g)</sub> + O <sub>2(g)</sub> A. 2,2,4 B. 2,4,2 C. 2,4,4 D. 4,2,2		1	
2	Match column I with column II and select the correct option using the given codes.				
		Column I	Column II		
		a. A metal that forms amphoteric oxides	(i) Ga		
		b. A metal which melts when kept on our palm	(ii) Au		
		c. A metal that reacts with nitric acid	(iii) Al		
		<ul> <li>A metal which cannot displace hydrogen from acids</li> </ul>	(iv) Mn		
	E	A. a – (ii), b – (i), c – (iii), d – (iv) B. a – (iii), b – (i), c – (iv), d – (ii) C. a – (iv), b – (ii), c – (iii), d – (i) D. a – (iii), b – (ii), c – (i), d – (iv)			

3	Battery HHH Bulb does not glow ?	1
	The solution in the given figure is likely to be A. HNO <sub>3</sub> B. $C_2H_5OH$ C. $H_2SO_4$ D. $CO_2$ in water	
	<i>For Visual Impaired Students</i> Which among the following is considered as the strongest electrolyte? A. Dilute acid B. Dilute sugar solution C. Glucose solution D. Ethanol in water	
1	<ul> <li>An aqueous solution 'A' turns the phenolphthalein solution pink. On addition of an aqueous solution 'B' to 'A', the pink colour disappears. Which of the following statement is true for the solutions 'A' and 'B'.</li> <li>A. A is strongly basic and B is a weak base.</li> <li>B. A is strongly acidic and B is a weak acid.</li> <li>C. A has a pH greater than 7 and B has a pH less than 7.</li> <li>D. A has a pH less than 7 and B has a pH greater than 7.</li> </ul>	1
5	<ul> <li>When 50g of lead powder is added to 300 ml of blue copper sulphate solution, after a few hours, the solution becomes colourless. This is an example of</li> <li>A. Combination reaction</li> <li>B. Decomposition reaction</li> <li>C. Displacement reaction</li> <li>D. Double displacement reaction</li> </ul>	1
6	The electronic configuration of three elements X, Y and Z are X- 2, 8, 7; Y- 2, 8, 2; and Z - 2, 8 A. Y and Z are metals B. Y and X are non-metals C. X is a non -metal and Y is a metal D. Y is a non-metal and Z is a metal	1
7	<ul> <li>Which of the following is an endothermic reaction?</li> <li>A. Burning of candle.</li> <li>B. Cooking of food.</li> <li>C. Decomposition of Vegetable matter.</li> <li>D. Reaction of Sodium with air</li> </ul>	1

8	During cellular oxidation of Glucose, ATP is produced along with formation of other products in this reaction. Which of the following events is associated with production of maximum ATP molecules per molecule of Glucose during this process? Synthesis of A. ethanol in yeast B. lactic acid in muscle cells C. carbon dioxide in yeast cells D. carbon dioxide in human cells	1
9	During which of the following stages of the circulation of blood in a normal human being, the oxygenated blood is pumped to all parts of the body? A. contraction of the left atrium B. contraction of left ventricle C. relaxation of the right atrium D. relaxation of the right ventricle	1
10	<ul> <li>Which of the following adaptations in herbivores helps in digestions of cellulose?</li> <li>A. Longer large intestine</li> <li>B. Smaller large intestine</li> <li>C. Smaller small intestine</li> <li>D. Longer small intestine</li> </ul>	1
11	<ul> <li>There was a cerebellar dysfunction in a patient. Which of the following activities will get disturbed in this patient as a result of this?</li> <li>A. Salivation</li> <li>B. Hunger control</li> <li>C. Posture and balance</li> <li>D. Regulation of blood pressure</li> </ul>	1
12	<ul> <li>In snails individuals can begin life as male and depending on environmental conditions they can become female as they grow. This is because</li> <li>A. male snails have dominant genetic makeup.</li> <li>B. female snails have dominant genetic makeup.</li> <li>C. expression of sex chromosomes can change in a snail's life time.</li> <li>D. sex is not genetically determined in snails.</li> </ul>	1
13	<ul> <li>In the following cases, a ray is incident on a concave mirror. In which case is the angle of incidence equal to zero?</li> <li>A. A ray parallel to the principal axis.</li> <li>B. A ray passing through the centre of curvature and incident obliquely.</li> <li>C. A ray passing through the principal focus and incident obliquely.</li> <li>D. A ray incident obliquely to the principal axis, at the pole of the mirror.</li> </ul>	1

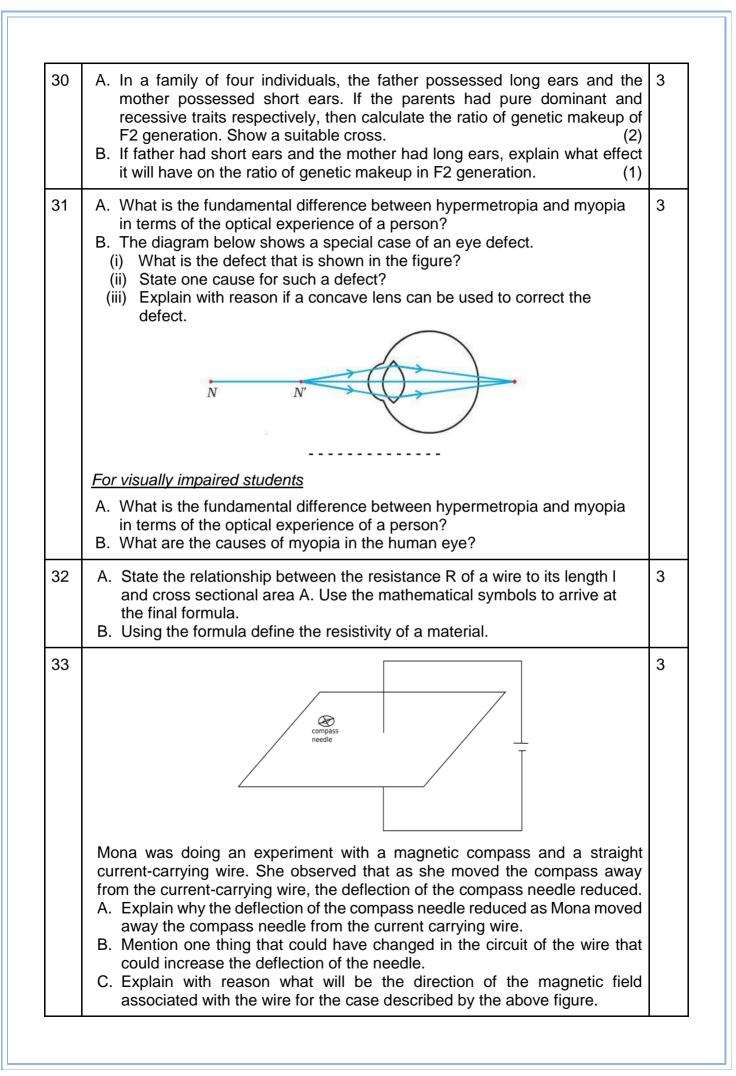
14		White light	Glass prism	A B hys for A and B.	1	1
			Colour of Ray A	Colour of Ray B		
		A.	Blue	Red		
		B.	Green	Yellow		
		C.	Red	Violet		
		D.	Violet	Indigo		
16	the sou B. It utilize reprod C. It utiliz level. D. It trans	urce. es the most o uction, mover es 10% of lig sfers only 10%	f the chemical energy ment etc. ht energy and transf 6 of light energy to th	1% of light energy dir / for its own respiratio fers the rest to the ne e next trophic level. sers in the ecosystem	n, growth, ext trophic	1
10	A. Natura B. Enrich C. Waste	I replenishme	ent of soil. en in atmosphere. on.	sers in the ecosystem		I
thes / E (	e questions by A. Both A and	v selecting the R are true, ar R are true, ar t R is false.	two statements – <b>As</b> e appropriate option g nd R is the correct ex nd R is not the correc	planation of A.	o <b>n</b> (R). Answ	ve
17	colourless ga stick is broug	is is produce ht near it.		e containing a substa sound when a burni ed by Hydrogen.		1
18	Assertion (A	): The numb	er of chromosomes ir	n a cell and in a germ	cell is not 1	1

19	Assertion (A): A convex mirror always forms an image behind it and the image formed is virtual. Reason (R): According to the sign convention, the focal length of a convex mirror is positive.				
20	Assertion (A): If the lions are removed from a food chain it will not affect the food chain, however if the plants are removed from a food chain it will disturb the ecosystem. Reason (R): Plants are producers who can make food using sunlight, while lions are consumers.				
Que	Section-B stion No. 21 to 26 are very short answer questions				
21	Identify the type of each of the following reactions stating the reason for your answers. A. $Fe_2O_3 + 2AI \rightarrow AI_2O_3 + 2Fe + heat$ B. $Pb (NO_3)_2 + 2KI \rightarrow PbI_2(\downarrow) + 2KNO_3$	2			
22	Differentiate between alveoli and nephron on the basis of the following points:	2			
	S. No. Feature Alveoli Nephron				
	1 Structure and location				
	2 Function				
23	<ul> <li><u>Attempt either option A or B.</u></li> <li>A. List the steps for the synthesis of glucose by the plants. What special feature is found in desert plants related to this process?</li> <li><b>OR</b></li> <li>B. Explain the role of the following enzymes in the process of digestion of food in humans: <ul> <li>(i) Salivary amylase</li> <li>(ii) Pepsin</li> <li>(iii) Trypsin</li> <li>(iv) Lipase</li> </ul> </li> </ul>				
24	The above figure shows the formation of an image by a lens shown by a thick line.	2			

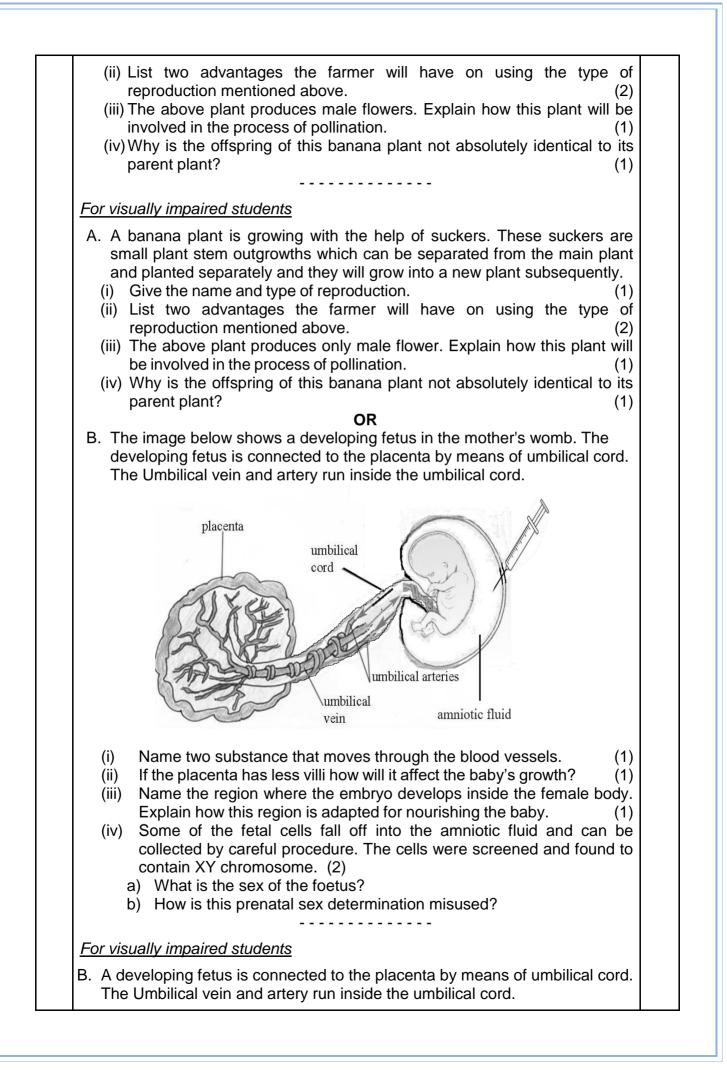


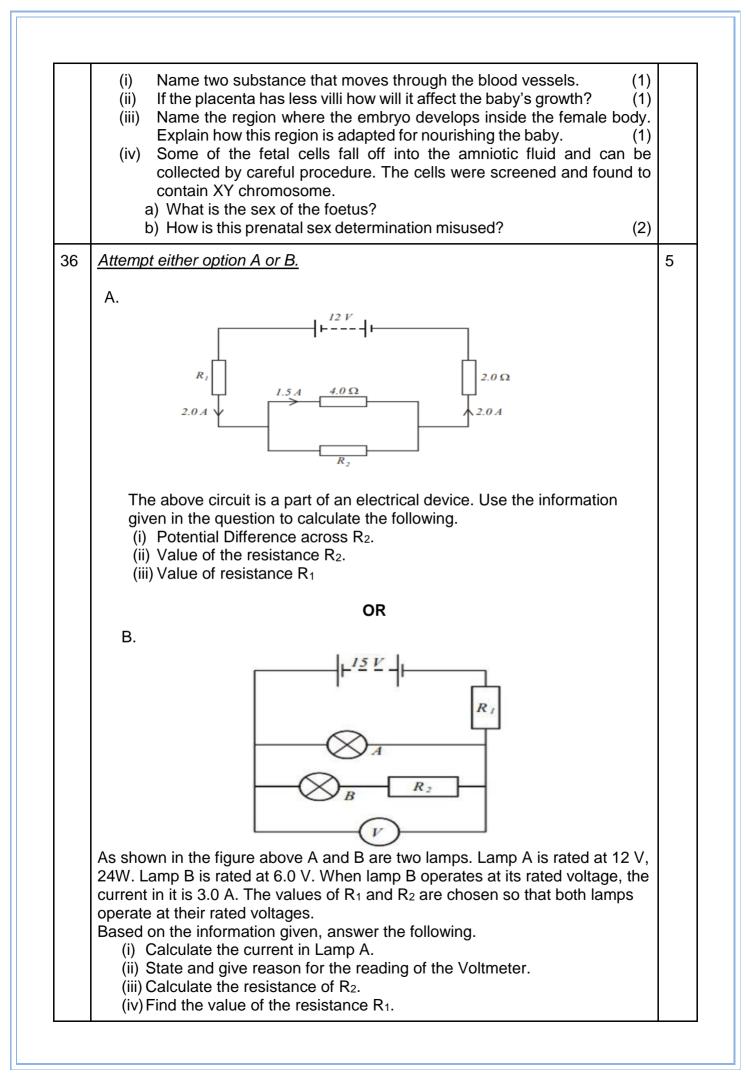
	given by P <sub>2</sub> . Find out the ratio $\frac{P_1}{P_2}$ .	
	For visually impaired students	
	A. We have four resistors A, B, C and D of resistance 3 $\Omega$ , 6 $\Omega$ , 9 $\Omega$ and 12 $\Omega$ respectively. Find out the lowest resistance which can be obtained by combining these four resistors.	
	OR B. You are given 2 fuse wires A and B with current ratings 2A and 5A respectively. Justify with reason which of the two would you use with a 1000W, 220V room heater?	
26	The cartoon below addresses a growing concern:	2
	pesticite main.jpg (1148×574) (frontiersin.org)	
	What impact will the process shown in the image have on Humans if they occupy the last trophic level? Explain.	
	For viewally imposed atudenta	
	<i>For visually impaired students</i> Create a food chain with more than 2 trophic levels that exists in the cabbage farm. If Humans occupy the last trophic level, then how would spraying pesticide affect the humans? Explain.	
Que	Section-C stion No. 27 to 33 are short answer questions	<u>.</u>
27	A. Anirudh took two metal oxides; aluminium oxide and magnesium oxide as shown in the pictures given below. But he forgot to label them. How will you guide/ help Anirudh to identify the oxides and label them?	3

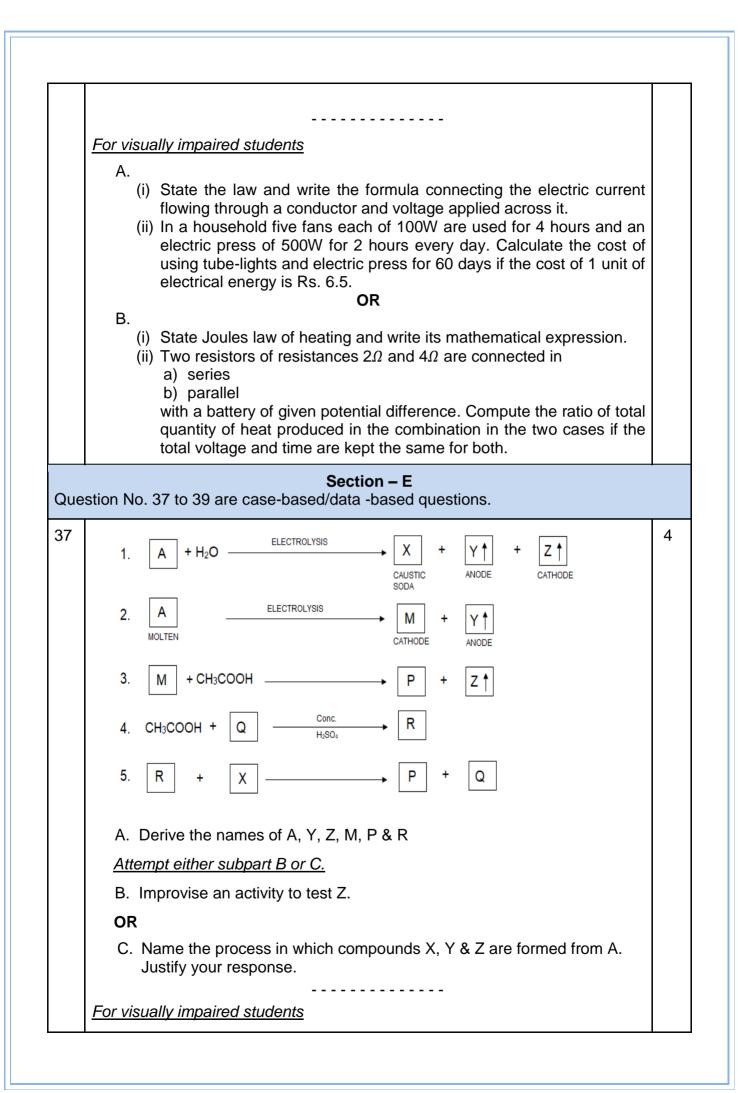
	<ul> <li>B. In an activity Aishu was given two substances; Copper Sulphide (Cu<sub>2</sub>S) and Copper Oxide (Cu<sub>2</sub>O) to obtain copper from these compounds. She was able to extract Copper successfully. Illustrate with the help of chemical equations how Aishu might have completed the activity.</li> </ul>	
	<ul> <li><u>For visually impaired students</u></li> <li>Give reasons for the following</li> <li>i. Certain metals are used for making cooking utensils.</li> <li>ii. Hydrogen gas does not evolve when certain metals except Mg &amp; Mn react with nitric acid.</li> </ul>	
28	Attempt either option A or B. A. (i) In the given series of reactions, name the compounds X and Z. (ii) Which type of reaction is X to Z? NaCl + H <sub>2</sub> O + CO <sub>2</sub> ↑ + NH <sub>3</sub> ↑ $X$ + Y $\downarrow \Delta$ Z + H <sub>2</sub> O + CO <sub>2</sub> ↑ Q $\leftarrow$ 10H <sub>2</sub> O	3
	<ul> <li>(iii) You are given 3 unknown solutions A, B, and C with pH values of 6, 8 and 9.5 respectively. In which solution will the maximum number of hydronium ions be present? Arrange the given samples in the increasing order of H<sup>+</sup> ion concentration. OR</li> <li>B. Comment on the following statements: <ul> <li>(i) Bee sting is treated with baking soda paste whereas wasp sting is treated with dilute vinegar.</li> <li>(ii) Farmers treat soil with quicklime when tilling.</li> <li>(iii) Ancient sculptures and marble structures are conserved by treating them with certain chemicals."</li> </ul> </li> </ul>	
29	Water is used by the leaves of the plants for photosynthesis but rather than watering the leaves, we water the plant through the soil. How does this water reach the leaves of the plant?	3

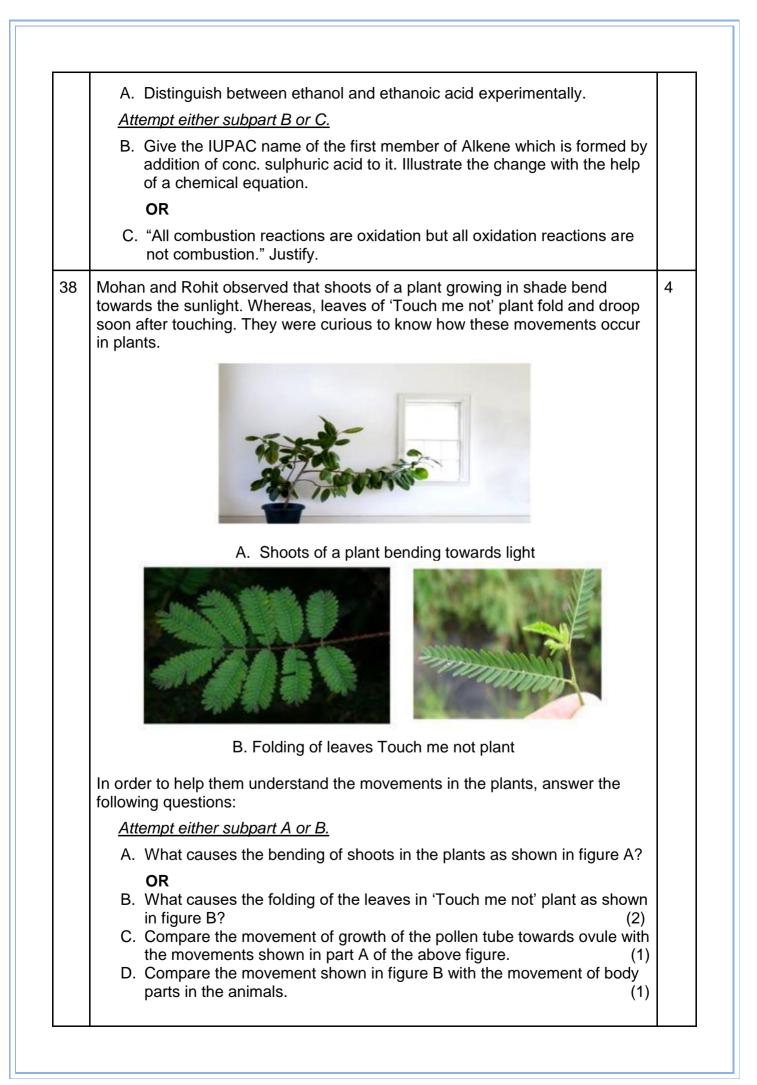


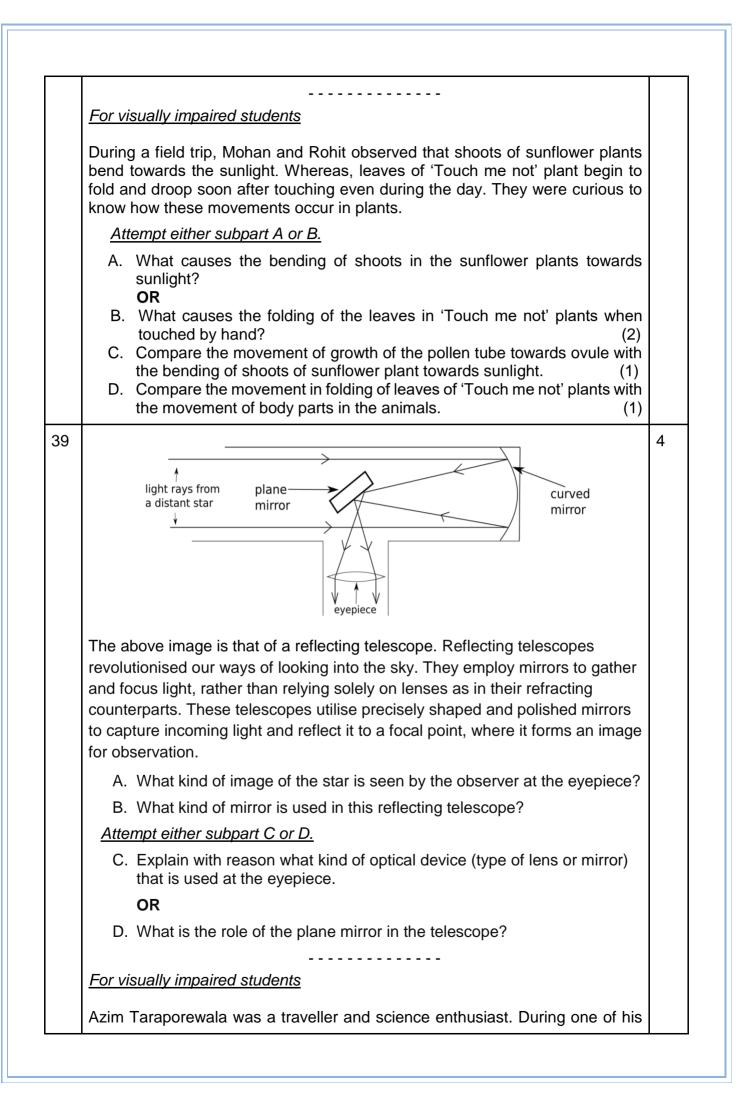
Que	Section-D stion No. 34 to 36 are long answer questions.	
34	Attempt either option A or B.	5
	Α.	ľ
	<ul> <li>(i) "Keerthi thinks that Substitution reaction occurs in saturated Hydrocarbons, on the contrary Krishi thinks, it occurs in unsaturated Hydrocarbons." Justify with valid reasoning whose thinking is correct.</li> <li>(ii) "Methane and Propane and their Isomers are used as fuels" Comment. Draw the electron dot structure of the immediate lower homologue of Propane. Give any two characteristics of homologues of a given</li> </ul>	
	<ul><li>homologous series.</li><li>(iii) A mixture of oxygen and ethyne is burnt for welding. Can you predict why a mixture of ethyne and air is not used?</li></ul>	
	OR	
	<ul> <li>B.</li> <li>(i) 'A' &amp; 'B' are sodium salts of long-chain carboxylic acid and long chain Sulphonic acid respectively. Which one of A or B will you prefer as a cleansing agent while using underground water (hand pump water)?</li> </ul>	
	<ul><li>Give the reason for your answer.</li><li>(ii) Elaborate on the process of cleansing action. Illustrate micelle with the</li></ul>	
	<ul> <li>help of labelled diagram.</li> <li>(iii) Write the chemical equation of the preparation of soap from an ester CH<sub>3</sub>COOCH<sub>3</sub>. What is the name of this process?</li> </ul>	
35	Attempt either option A or B.	5
	A. The image below shows a banana plant which is growing with the help of suckers. These suckers are small plant stem outgrowths which can be separated from the main plant and planted separately and they will grow into a new plant subsequently.	
	Suckers	
	<ul> <li>Fig1-Parts-of-Banana-plant-FAO-2021.png (623×609) (wp.com)</li> <li>(i) Give the name and type of reproduction that is shown in the image above.</li> </ul>	











travels he found himself on the edge of an island without any mode of communication. As he had read in many stories, he thought he would light a fire on the beach and travelling boats or ships could see that fire and come to give him a ride. He had run out of lighters and match-sticks but had a reading glass. Being a science enthusiast he knew some tricks and used that lens and a scrap of paper to light a fire, with the help of scorching rays from the sun.

- A. Which lens can be used by Azim to create the fire?
- B. What property of the lens helps Azim to create the fire?

Attempt either subpart C or D.

C. List two more uses of this kind of lens.

### OR

D. Explain with reason the condition under which the lens can form both real as well as virtual images.

#### \*\*\*\*\*\*\*\*\*

## MARKING SCHEME 2024 -25 Class X Science (086)

	Section-A	
1	A. 2,2,4	1
2	B. a – (iii), b – (i), c – (iv), d – (ii)	1
3	<ul> <li>A. C<sub>2</sub>H<sub>5</sub>OH</li> <li>Alternate question for VI</li> <li>A. Dilute acid</li> </ul>	1
4	C. A has a pH greater than 7 and B has a pH less than 7	1
5	C. Displacement reaction	1
6	C. X is a non-metal and Y is a metal.	1
7	B. Cooking of food	1
8	D. carbon dioxide in human cells	1
9	B. contraction of left ventricle.	1
10	D. Longer small intestine.	1
11	C. Posture and balance.	1
12	D. sex is not genetically determined in snails.	1
13	B. A ray passing through the centre of curvature and incident obliquely.	1
14	C. Red Violet	1
15	C. It utilizes 10% of light energy and transfers the rest to the next trophic level.	1
16	B. Enrichment of oxygen in the atmosphere.	1
17	A. A is true but R is false	1
18	B. Both A and R are true, and R is not the correct explanation of A	1
19	B. Both A and R are true, and R is not the correct explanation of A	1
20	D. A is false but R is true.	1

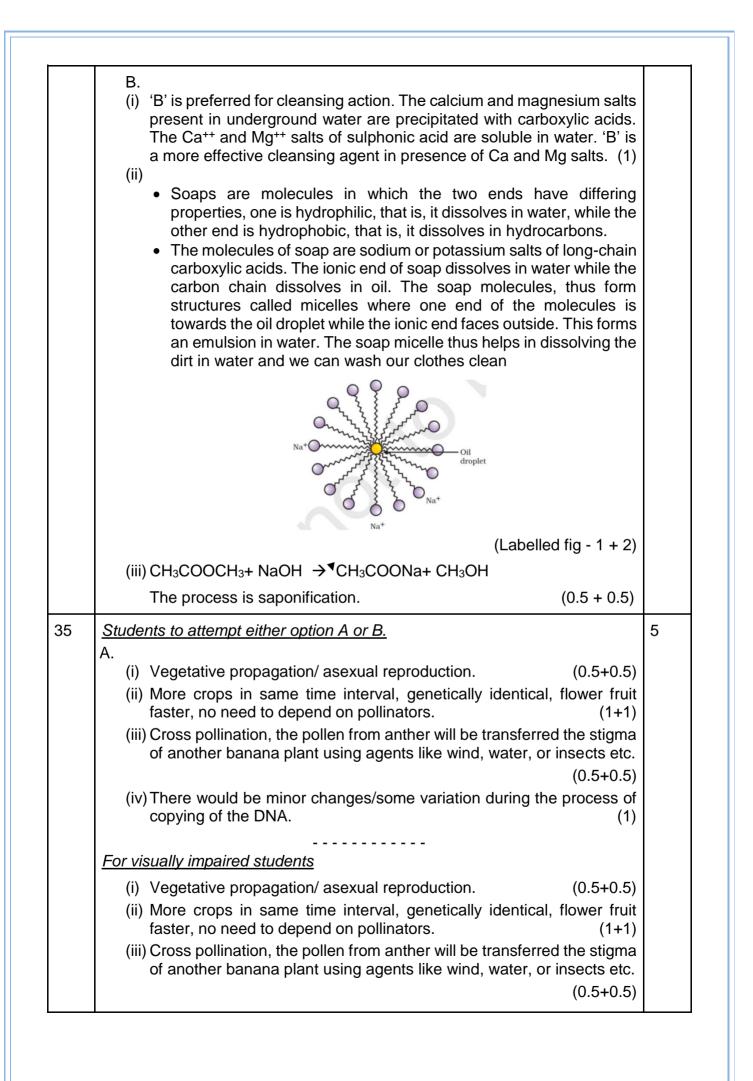
			Section-B			
21	Heat i eleme B. Doubl As the	s evolved or a nt or aluminiur e displacemen ere is an excha	ement reaction/Redox reaction More reactive element displ m reduces iron (II) oxide to in t / Precipitation reaction ange of ions between reactar (of Lead iodide) is formed	aces a less reactive on ( ( nts and products /	/	2
22	S. No.	Feature	Alveoli	Nephron	2	2
	1	Structure and location	Balloon like structures present at the terminal ends of bronchioles in lungs	Tubular structure present in kidneys		
	2	Function	Exchange of gases	Filtration of blood to form urine		
	A. Ste 	ps of synthesis Absorption of I Conversion of I molecules into Reduction of c Desert plants ntermediate w chlorophyll dur Salivary amyla molecule to su Pepsin – Helps Trypsin – It hel Lipase – Brea glycerol.	OR ase – breaks down starch gar. s to digest proteins in stomad lps in digesting proteins to an aking down of emulsified f	rates. night and prepare energy absorbed by (0.5 n which is a comp ch. mino acids.	ater an the x 4) olex and ( 4)	2
24	B. The in C. Magni <u>-30c</u> u Her <u>For visual</u> A. conca	<sup>m</sup> = 2 nce u = -15 cm l <u>y impaired stu</u> ive lens	$S = \frac{v}{u} = \frac{h_i}{h_o} = 2.$		(1)	2

25	Student to attempt either A or B.		2
	A. P = VI	(1)	
	$= 230 \times 3.5 = 805 \text{ W}$		
	I $\propto \frac{1}{R}$ , so half the resistance means double the current.		
	Therefore, current in $Y = 7.0 A$ .	(1)	
	OR		
	B. For series total resistance is $R+2R = 3R$	(0.5)	
	$P_1 = \frac{V^2}{3R}.$	(0.5)	
	For parallel total Resistance is $\frac{2R}{3}$ .	(0.5)	
	$P_2 = \frac{V^2}{2R/3} = \frac{3V^2}{2R} .$		
	$\frac{P_1}{P_2} = \frac{2}{9}.$	(0.5)	
	For visually impaired students		
	A. The resistance will be lowest/minimum if all the resistors are connected in parallel.		
	The equivalent resistance in parallel combination is given by $1 \qquad 1 \qquad 1 \qquad 1 \qquad 1$	(0.5)	
	$\frac{1}{R_{eqv}} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3} + \frac{1}{R_4}$	(0.5)	
	Substituting the values, we get $\frac{1}{R_{eqv}} = \frac{1}{3} + \frac{1}{6} + \frac{1}{9} + \frac{1}{12}$	~ ,	
		(0.5)	
	$= \frac{12+6+4+2}{36} = \frac{24}{36} = \frac{2}{3}$ $R_{eqv} = \frac{3}{2} = 1.5 \Omega$		
	$R_{eqv} = 2$ = 1.5 52	(0.5)	
	OR		
	B. $P = V \times I$	(0.5)	
	$I = \frac{P}{V} = \frac{1000}{220} = 4.54 A$	(0.5)	
	We will be using fuse B with is rated as 5A. This is because it able to sustain the current (4.54 A) passing through it. Wherea A will melt and break the circuit as the current exceeds its ratir	is fuse	
:6	Pesticides are non-biodegradable/ keep getting accumulated at each level, / persist for longer time/ and thus last trophic level has concentration/ humans will have the highest concentration of persist for longer time/ and the highest concentration	trophic highest	2
	For visually impaired students		

	Cabbage/plant-> rabbit-> snake-> owl (Any other relevant food chain) (1) Pesticides are non-biodegradable and persist for long. So when humans consume plants or any animal that consume this plant, the pesticide enters the food chain and keeps getting accumulated at each trophic level, thus the organism in the last trophic level-human being has the highest concentration of pesticide and this is called <b>biological magnification</b> . (1)	
	Section-C	
27	A. Aqueous solution of magnesium oxide turns red litmus to blue. Aluminium oxide is amphoteric and insoluble in water. Thus, it does not change the colour of either blue or red litmus. <b>OR</b> Magnesium oxide reacts with acid only whereas Aluminium oxide reacts with acids and bases, which are amphoteric. (1) B. $\frac{2Cu_2S + 3O_2(g) - \frac{Heat}{2} + 2Cu_2O(g) + 2SO_2(g)}{2Cu_2O + Cu_2S - \frac{Heat}{2} + 3Cu(g) + SO_2(g)}$ (1+1)	3
	<ul> <li>For visually impaired students</li> <li>A. Certain metals (like aluminium/ copper) are used for making cooking utensils as they are good conductors of heat and have high melting points. (0.5 + 0.5)</li> <li>B. Hydrogen gas is not evolved when a metal reacts with nitric acid. This is because HNO3 is a strong oxidising agent. It oxidises the H2 produced to water and itself gets reduced to any of the nitrogen oxides (N<sub>2</sub>O, NO, NO<sub>2</sub>). But magnesium (Mg) and manganese (Mn) react with very dilute HNO<sub>3</sub> to evolve H<sub>2</sub> gas. (1+1)</li> </ul>	
28	Students to attempt either A or B.A.(i) $X = NaHCO_3; Z = Na_2CO_3$ (0.5+0.5)(ii) Decomposition reaction(0.5)(iii) Solution A(0.5)(iv) Increasing order or H+ ions C <b<a< td="">(1)ORB.(i) As bee sting is acidic and wasp sting is basic.(1)(ii) To change the nature of soil to (neutral or basic).(1)(iii) To protect sculptures from the effects of certain gases present in</b<a<>	3
29	<ul> <li>In plants, the water is absorbed by the plants from the soil through the</li> </ul>	3
•	<ul> <li>roots. Xylem tissue of the roots, stems and leaves are interconnected to form a continuous system of water conducting channels. (1)</li> <li>During the day, when stomata are open, the transpiration pull becomes</li> </ul>	-

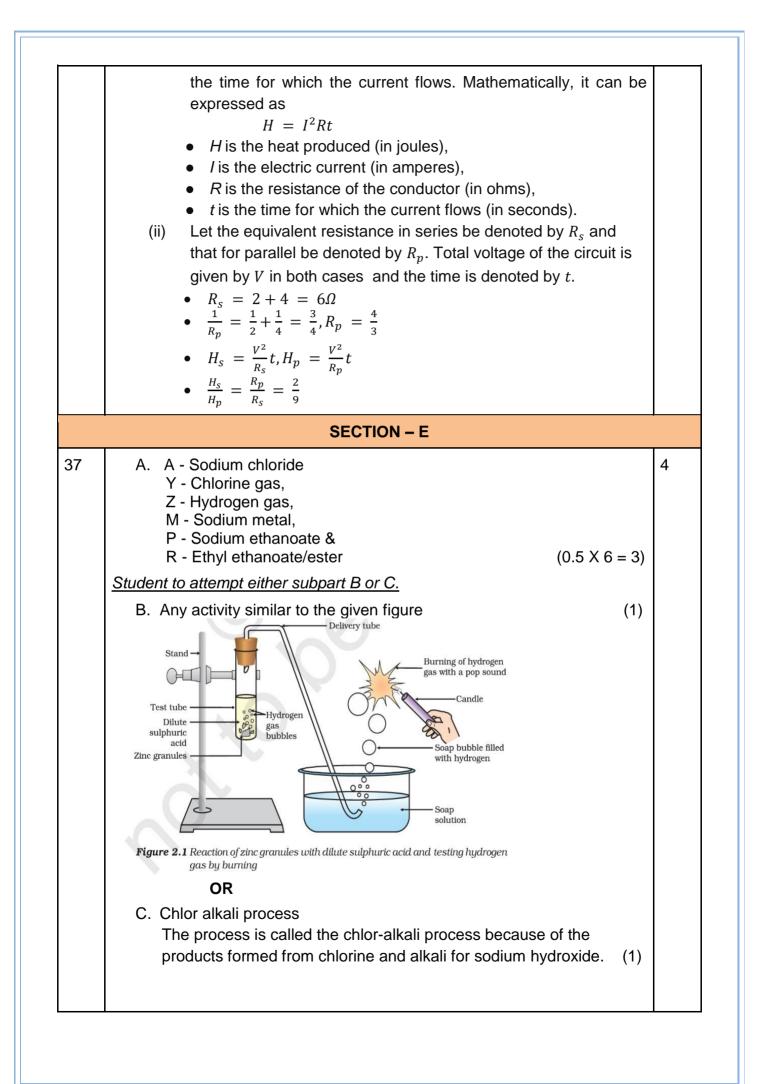
istant objects appear blurry while near objects are seen clearly. (1) The image shows a case of hypermetropia. (0.5) shortening of the eyeball or focal length of the eye lens becomes too long. (0.5) No the concave lens would diverge the rays coming to the eyeball and will push the image even further, but a convex lens should be used which will help to converge the rays and create the image at the exact place on the retina. (1) <i>risually impaired students</i> Hypermetropia is a defect that causes difficulty in focusing on near objects, with clearer vision observed for distant objects. In Myopia distant objects appear blurry while near objects are seen clearly. (1)
Hypermetropia is a defect that causes difficulty in focusing on near objects, with clearer vision observed for distant objects. In Myopia distant objects appear blurry while near objects are seen clearly.
bjects, with clearer vision observed for distant objects. In Myopia distant objects appear blurry while near objects are seen clearly. (1)
High converging power of eye lens, elongation of eye-ball. (2)
$\begin{array}{c} \propto l \\ \propto \frac{l}{A} \end{array} $
$= \rho \frac{l}{A}.$ $(1.5)$ $R = \rho \frac{l}{A}$
$\propto \frac{l}{A} = \rho \frac{l}{A}.$ (1.5)

	<ul> <li>current carrying wire. Hence when Mona moved the compass away from the current carrying wire, the magnetic effect was less on it and hence the deflection was less. (1)</li> <li>B. Magnetic field strength is directly proportional to the current in the wire. So, Mona could increase the current in the circuit to observe a greater deflection in the compass needle. (1)</li> <li>C. The battery suggests that the current is going from top of the plane to the bottom of the plane. Using the right hand thumb rule we can say that the magnetic field will be clockwise. (1)</li> </ul>	
	Section-D	T
34	<ul> <li>Student to attempt either option A or B.</li> <li>A.</li> <li>(i) Keerthi's thinking is correct as substitution reactions occur in saturated hydrocarbons, hydrogen atoms are replaced with heteroatoms in saturated hydrocarbons. Whereas in unsaturated hydrocarbons an addition reaction occurs, simple molecules are added across double and/or triple bonds. (1)</li> <li>(ii) Methane and propane undergo combustion reaction in presence of oxygen and produce large amount of energy. (1) The lower homologue of propane is ethane has the following electron dot structure:</li> </ul>	5
	(1)	
	<ul> <li>ANY TWO CHARACTERISTICS</li> <li>Difference in -CH2- / 14u molecular mass of any two adjacent homologues.</li> <li>Same general formula/ functional group</li> <li>Similar chemical properties</li> <li>Gradual change in physical properties (1)</li> <li>(iii) The mixture of ethyne and oxygen in sufficient amounts undergoes complete combustion to fire a clean blue flame. In pressure of insufficient supply of oxygen or in presence of air, ethyne does not undergo complete combustion and produces sooty flame. (1)</li> </ul>	
	OR	



	(iv) There would be minor changes/some variation during the process of copying of the DNA. (1)	
	OR	
	В.	
	<ul> <li>(i) Nutrients /glucose/oxygen/ waste. (any two)</li> <li>(1)</li> <li>(ii) less surface area for nutrients (glucose/oxygen) to pass from mother to embryo slow growth.</li> </ul>	
	(iii) uterus; has thick lining with rich supply of blood to nourish the embryo. (1)	
	(iv)	
	<ul> <li>a) male child</li> <li>b) misused as if the foetus is female, some people engage in aborting the child leading to female foeticide. (2)</li> </ul>	
	For visually impaired students A.	
	(i) Nutrients /glucose/oxygen/ waste. (any two) (1)	
	(ii) less surface area for nutrients (glucose/oxygen) to pass from mother to embryo slow growth. (1)	
	<ul><li>(iii) uterus; has thick lining with rich supply of blood to nourish the embryo.</li><li>(1)</li></ul>	
	(iv)	
	<ul> <li>a) male child</li> <li>b) misused as if the foetus is female, some people engage in aborting the child leading to female foeticide. (2)</li> </ul>	
36	Students to attempt either option A or B.	5
	A. (i) p.d. across 4 $\Omega$ resistor = p.d. Across R2 as both are in parallel. (0.5)	
	(0.5) $1.5(A) \times 4(\Omega) = 6 V$ (0.5) (ii) Total Current through 4 $\Omega$ and R <sub>2</sub> = 2.0 A (given).	
	Current through 4 $\Omega$ = 1.5 A (given) (0.5) Hence current through R <sub>2</sub> = 2-1.5= 0.5 A	
	Using Ohm's law for $R_2$ we get (0.5) $6 V = 0.5 A \times R_2$ Hence $R_2 = 6/0.5 = 12 \Omega$ (0.5)	
	(iii) p.d. across $R_1$ = Total p.d (p.d. across $R_2$ ) - (p.d. across 2.0 $\Omega$ ) (0.5)	
	p.d. across 2.0 $\Omega$ = 2x2 = 4 V p.d. across R <sub>2</sub> = 6 V (calculated before) (0.5)	
	Hence p.d. across $R_1 = 12 - 6 - 4 = 2 V$ (0.5) Current through $R_1 = 2A$ (0.5)	
	Using Ohm's Law, we get $R_1 = 2V/2A = 1\Omega$ (0.5)	

Alternative method	
Total Resistance = R1 + $\frac{(4 \times 12)}{(4+12)}$ + 2 = 12V/2 A = 6 $\Omega$ (0.	5)
$R1 = 6 - (3 + 2) = 1 \Omega $ (0.	.5)
OR B.	
(i) Use of P = IV $I = P \div V = 24 W \div 12V$ (0.5)	5)
Current in lamp A – 2 A (0.8 (ii) Voltmeter reading = 12 V (0. Lamp A and Lamp B are in parallel.	'
Hence p.d. across the arm containing A = p.d. across arm containing B	
= 12  V (from a) (0. (iii) p.d. across R <sub>2</sub> + p.d. across B = 12 V. (0.	'
p.d. across $B = 6 V$ (given) Hence p.d. across $R_2 = 12 V - 6 V = 6V$ (0.	5)
Current through $R_2$ = Current through B = 3A (given) (0.3 Use of R = V/I	5)
$R_{2} = 6V/3A = 2\Omega \qquad (0.$ (iv) Current through $R_{1} = \text{Total Current} = 3A+2A = 5A$ (0.3)	-
p.d. across $R_1 = 15V - 12V = 3V$ $R_1 = 3V/5A = 0.6 \Omega$ (0.	5)
A. (i) Ohm's Law is the law, which states that the electric current flowing through a conductor is directly proportional to the voltag (V) applied across it and inversely proportional to the resistant (R) of the conductor. Mathematically, it can be represented as: ( $V \propto I$	ge ce
$V = IR \tag{(}$	1)
(ii) Let the energy consumed by the fans be $E_f$ and the energy consumed by the electric press be $E_p$ . • $E_f = P \times t = \frac{100 \times 4}{1000} = 0.4  kWh$	
• $E_f = P \times t = \frac{100 \times 4}{1000} = 0.4  kWh$ • $E_p = P \times t = \frac{500 \times 2}{1000} = 1  kWh$ • Total energy consumed in 1 day = $E_p + E_f = 1.4  kWh$	(1)
Total energy consumed in 60 days = $1.4 \times 60 = 84  kWh$ .	(2)
OR	
 <ul> <li>B.</li> <li>(i) Joule's Law of Heating states that the amount of heat produce in a conductor is directly proportional to the square of the electric current passing through it, the resistance of the conductor, ar</li> </ul>	ric



ETH			
	IANOIC ACID	ETHANOL	
Etha carb	ction with carbonates and hydrogen carbonates: noic acid reacts with carbonates and hydrogen onates to give rise to salt, carbon dioxide and er. The salt produced is commonly called	No reaction	
sodi	um acetate.		
2C⊢	$_{3}$ COOH + Na <sub>2</sub> CO <sub>3</sub> $\rightarrow$ 2CH <sub>3</sub> COONa + H <sub>2</sub> O + CO <sub>2</sub>		
		(1-	+1)
B. Et	nene		
are c relea	combustion reactions oxygen is added hence a xidation, whereas in oxidation reactions, ener sed (along with the products), hence all oxida ustion reactions.	gy may or may not ation reactions are	be
A. •	Bending of shoots of plants is a response directional, growth-related movement. When growing plants detect sunlight, a h synthesized at shoot tip helps the cells to grow When light is coming from one side of the pla shady side of the shoot. This concentration of auxin stimulates the ce longer on the side of the shoot which is awa plant appears to bend towards light. <b>OR</b> Leaves of 'Touch me not' plant respond to th growth independent movement.	normone called au v longer. nt, auxin diffuses to lls of the shoot to gr ny from the light. Th (0.5 x 4 s	xin, the row ius, =2)

	<ul> <li>D.</li> <li>i) Although both plants and animals show electrical-chemical means to convey the information from cell to cell but unlike nerve cells in animals there is no specialized tissue in plants for conduction of information. (0.5)</li> <li>ii) In animal cells, change in shape occurs because of the specialized proteins found in muscle cells; plant cells change shape by changing the amount of water in them. (0.5)</li> </ul>	
	For visually impaired students	
	Students to attempt either subpart A or B.	
	Α.	
	<ul> <li>Bending of shoots of plants is a response to the stimulus and a directional, growth-related movement.</li> </ul>	
	• When growing plants detect sunlight, a hormone called auxin,	
	synthesized at shoot tip helps the cells to grow longer.	
	<ul> <li>When light is coming from one side of the plant, auxin diffuses to the shady side of the shoot.</li> </ul>	
	• This concentration of auxin stimulates the cells of the shoot to grow	
	longer on the side of the shoot which is away from the light. Thus,	
	plant appears to bend towards light. $(0.5 \times 4 = 2)$	
	OR	
	B.	
	<ul> <li>Leaves of 'Touch me not' plant respond to the stimulus by showing</li> </ul>	
	growth independent movement.	
	<ul> <li>These plants use electrical-chemical means to convey the information from cell to cell.</li> </ul>	
	<ul> <li>Movement happens at a point different from the point of touch.</li> </ul>	
	<ul> <li>Plant cells change shape by changing the amount of water in them,</li> </ul>	
	resulting in swelling or shrinking, and therefore in changing shape.	
	$(0.5 \times 4 = 2)$	
	C. Growth of pollen tubes towards the ovule is an example of chemotropism whereas bending of shoots towards sunlight is an	
	example of phototropism. (1) D.	
	<ul> <li>i) Although both plants and animals show electrical-chemical means</li> </ul>	
	to convey the information from cell to cell but unlike nerve cells in	
	animals there is no specialized tissue in plants for conduction of	
	information. (0.5)	
	ii) In animal cells, change in shape occurs because of the specialized	
	proteins found in muscle cells; plant cells change shape by changing	
	the amount of water in them. (0.5)	
39	A. Real Image (the final image is formed due to the lens at the eye-piece)	4
	(1)	

В.	Concave Mirror (1)
<u>St</u>	udent to attempt either subpart C or D.
C.	A converging lens is used at the eyepiece to collect the rays from the plane mirror and help the viewer to see a real erect image of the star. (2)
	OR
D.	The plane mirror laterally inverts the image formed by the curved mirror and its position helps to direct the rays towards the eye-piece. (2)
For vi	sually impaired students
Α.	Convex lens. (1)
В.	Converging property. The lens can converge parallel rays to one point. (1)
<u>St</u>	udent to attempt either subpart C or D
C.	To correct hypermetropia, lenses of telescopes, microscopes and slide projectors. (2)
	OR
D.	If the object is kept between the optical centre and the focus the image obtained is virtual, rest in all cases the image is real.

\*\*\*\*\*\*