### **Chemistry of Cyanotoxins in Surface** Waters

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Agenda

- Cyanotoxins in water
  - Microcystins
  - Cylindrospermopsins
  - Anatoxin-a
  - Saxitoxins
  - Nodularins



- Observations from Case Studies
- AWWA CyanoTox 2.0 vs. Oxidation Studies

Chemistry of Cyanotoxins in Surface Waters

## Microcystins

### Microcystins in Water

- Most common cyanotoxin
  - Presence of nitrate increases toxin concentration within cells
  - Maximum toxin concentration at 20°C to 25°C
  - Intracellular toxins released as cell lysing or mortality occurs
    - Natural life cycle or chemical means
    - UV irradiation slowly breaks down microcystins



Microcystis Aeruginosa

### Microcystins in Water

- Made up of 7 amino acids in unique structure
  - $C_{49}H_{74}N_{10}O_{12}$  (MC-LR)
    - Similar for other variants
  - 995 g/mole to 1,040 g/mole
  - Solubility in water ~1,500 mg/L
  - Half-life in water environment is up to 10 weeks
  - Henry's Law constant
    - Reduction by aeration not likely
  - Toxicity from enzyme reaction stripping phosphate from proteins that leads to liver damage
  - LD50
    - 5 mg/kg



### Microcystins in Water



Microcystin-LR (7) amino acids

### Microcystin-LR



Oxidation at double bonds degrades structure to amino acids groups

### Microcystin-RR



### Microcystin-YR



### Microcystin-LA



#### Other congeners include MC-H4YR, MC-WR, MC-FR

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# Cylindrospermopsins

- Common cyanotoxin
  - Presence of nitrate increases toxin concentration within cells
  - Maximum toxin concentration at 20°C to 25°C
  - Intracellular toxins released as cell lysing or mortality occurs
    - Natural life cycle or chemical means
    - More toxic than microcystins
    - UV irradiation breaks down cylindrospermopsins



Cylindropspermopsis Raciborskii

- Polycyclic uracil derivative in unique structure
  - $\bullet C_{15}H_{21}N_5O_7S$
  - 415 g/mole
  - Solubility in water very high
    - Stable over wide pH range
  - Half-life in water environment is about 8 weeks
  - Toxicity from inhibited protein synthesis leading to cell mortality
  - LD50
    - 4.4 mg/kg to 6.9 mg/kg



Cylindrospermopsin



Guanidine and Uracil rings



Guanidine and Uracil rings



Guanidine and Uracil rings

Non-toxic metabolites



5-chloro-cylindrospermopsin uracil ring altered with Cl





Deoxycylindrospermopsin no toxicity - OH stripped

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## Anatoxin-a



Anabeana Aequalis

- Less common cyanotoxin
  - Maximum toxin concentration at 20°C to 25°C
  - Intracellular toxins released as cell lysing or mortality occurs
    - Natural life cycle or chemical means
  - Strong neurotoxin with acute toxicity
    - Named very fast death factor (VFDF) aquatic toxin
    - LD50 0.25 mg/kg
    - Toxicity from attack of neuromuscular receptors resulting in paralysis
  - Half-life in water environment
    - <24 hours, degrades within hours at pH greater than 8</p>



Anatoxin-a

- Bicyclic amine alkaloid in unique structure
  - $C_{10}H_{15}NO$
  - 165 g/mole
  - Solubility in water very low
    - <50 mM/L
  - Unstable in water, UV irradiation leads to degradation into non-toxic forms
    - Dihydroanatoxin-a
    - Epoxyanatoxin-a



Carbonyl (C=O) near nitrogen responsible for toxicity



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## Saxatoxins



Aphanizomenon flos-aquae

- Less common cyanotoxin
  - Maximum toxin concentration at 20°C to 25°C
  - Intracellular toxins released as cell lysing or mortality occurs
    - Natural life cycle or chemical means
  - Strong neurotoxin with acute toxicity
    - Named paralytic shellfish toxin (PST)
    - LD50 ≈ 0.26 mg/kg
    - Toxicity from attack of neuromuscular receptors resulting in paralysis
  - Half-life in water environment
    - Up to 10 weeks



Aphanizomenon flos-aquae

- Reduced purine and guanidine rings in unique structure
  - $C_{10}H_{17}N_7O_4$
  - 299 g/mole
  - Solubility in water lower than other toxins
    - <0.14 moles/L</p>
    - 193  $\mu$ g/L found in one surface source (WHO)
  - More research needed related to saxitoxins



Guanidine ring structures



 $H_2N \rightarrow O$   $H_1N \rightarrow H$   $H_1N \rightarrow G \rightarrow H_2$   $H_1N \rightarrow H_2$   $H_2$   $H_2$  $H_$ 

7,8,9-Guanadine ring responsible for toxic bonding at neurons -Possible substitution may alter toxicity, Oxidation possible for substitution?

#### Other metabolites

- Neosaxitoxin
- Gonyautoxins
- Decarbamoylsaxitoxin
- More than 57 metabolites known







Gonyautoxin

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## Nodularins

### Nodularins in Water

- Uncommon cyanotoxin
  - Microcystin-like structure
  - Maximum toxin concentration at 20°C to 25°C
  - Intracellular toxins released as cell lysing or mortality occurs
    - Natural life cycle or chemical means
    - UV irradiation breaks down nodularins
    - Toxicity like microcystins



Nodularia spumigena

### Nodularins in Water

- Made up of 5 amino acids in unique structure
  - $C_{41}H_{60}N_8O_7$ 
    - Similar for other variants
  - 825 g/mole
  - Solubility in water unknown
  - Half-life in water environment is up to 18 days
  - Toxicity from enzyme attack of proteins resulting in liver damage
  - LD50
    - 5 mg/kg



Nodularin-R

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## **Oxidative Treatments**

### Permanganate Treatment



### Permanganate Treatment















### **Ozone Treatment**



### **Ozone Treatment**



### **AOP Treatment**



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## Adsorptive Treatments

### **Activated Carbon Treatment**



### PAC Treatment



### PAC Treatment



### GAC Treatment



### **GAC** Treatment

$$CUR, lbs/1,000 gallons = \frac{EBCT * \rho GAC * 10^{3}}{T * 7.48 * 1,440}$$

Where CUR = Carbon Usage Rate, pounds per 1,000 gallons EBCT= empty bed contact time, minutes QGAC = carbon density, pounds/cubic foot 7.48 = 7.48 gallons per cubic foot 1,440 = 1,440 minutes per day

1998 WRF report - 'Removal of DBP Precursors by GAC Adsorption"

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### **GAC Treatment**



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# AWWA CyanoTox 2.0

- Oxidation model works for some water systems, not others
  - KMnO<sub>4</sub> trials
  - Chlorine trials
- Input variables and run model
- Output gives remaining cyanotoxin based on research data in lab water
  - Does not account for natural water conditions



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# Questions

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