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Ewh! What Stinks?!!!

Odor Control in the 21st Century



Fact:

There will be odors wherever there is the Collection Transport & Treatment of Wastewater



Odor Control Why is it Important?

- Steady encroachment of development on treatment plants
- Address community complaints
- Health and safety of plant operators
- Good neighbor policy

Primary Sources of Odors

Collection System

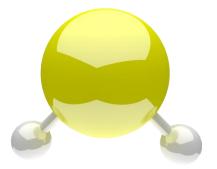
- Headworks and Primary Treatment
- Solids Handling



Typical Odors Found in Wastewater Treatment

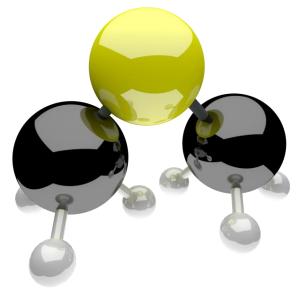
Hydrogen Sulfide H₂S

- Formed by organic reduction of organic sulfur compounds under anaerobic conditions
- Rotten egg smell
- Low solubility in wastewater
- Colorless gas
- Low concentration odor threshold
- Corrosive
- Potentially toxic



Typical Odors Found in Wastewater Treatment Dimethyl-Sulfide

- Volatile organic compound
- Slightly soluble in water
- Disagreeable odor cabbage like
- Component of bad breath



Typical Odors Found in Wastewater Treatment

Ammonia NH₃

- Putrid smell associated with cat urine
- Sources contributing to Ammonia in wastewater are human urea, food processing, pharmaceutical manufacturing
- Produced in anaerobic digestion



Typical Odors Found in Wastewater Treatment

Mercaptans

- Methanethiol
- Pungent smell of rotting cabbage
- Added to natural gas to give it an odor
- Very low odor threshold ppb

Collection System





Dimethyl Sulfide (CH₃)₂S



Hydrogen Sulfide H₂S

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Primary Treatment



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Dimethyl Sulfide (CH₃)₂S



Hydrogen Sulfide H₂S

Solids Handling



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Ammonia NH₃



Hydrogen Sulfide H₂S

Quantifying Odors



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Basic Guide to Odor Control Technologies

There are a limitless number of odor control problems. It is impossible to identify one technology as the best solution for all of them.



Vapor Phase Technologies

Treats the Air Stream

- Chemical Scrubbers
- Biofilters
- Carbon Filters
- Photoionization





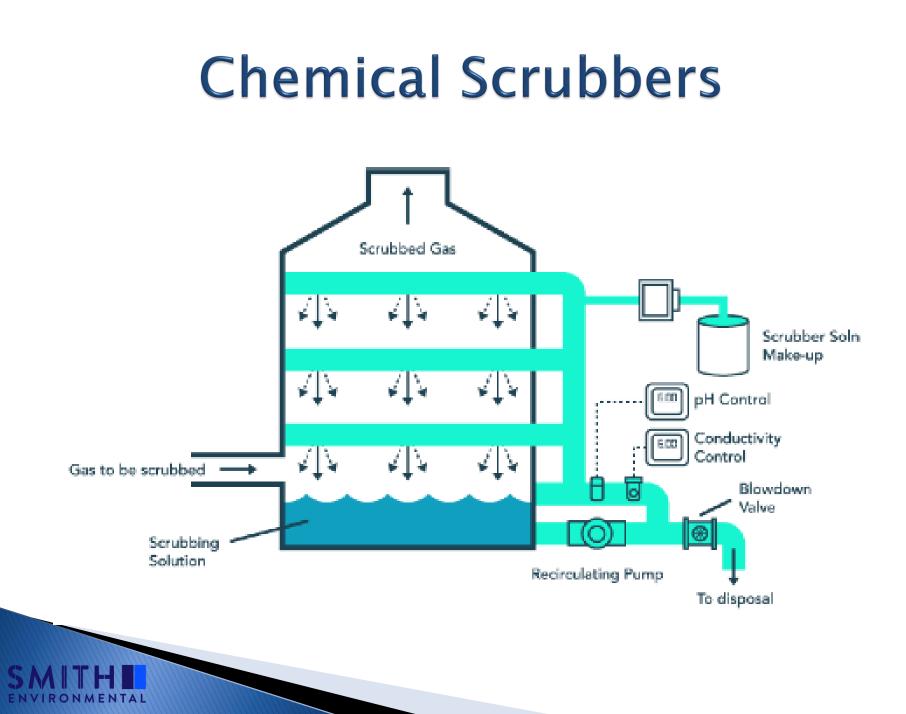


- Works on water soluble compounds
- Reliable technology
- Purely chemical process
- 90% removal efficiency



Mechanism of Action

- Sodium Hydroxide Solubilizes H₂S
- Sodium Hypochlorite oxidizes remaining sulfur compounds
- Scrubber is packed with media
- Chemicals flow down through packed bed while odorous air flows up through the bed
- Chemicals are recirculated through tower



Pros:

- Purely chemical process provides consistent results
- 90% removal efficiency
- Cons:
- Chemicals are expensive
- Mechanical components, recirc pump and chem feed pumps must be maintained
- Mechanical components must be protected from freezing



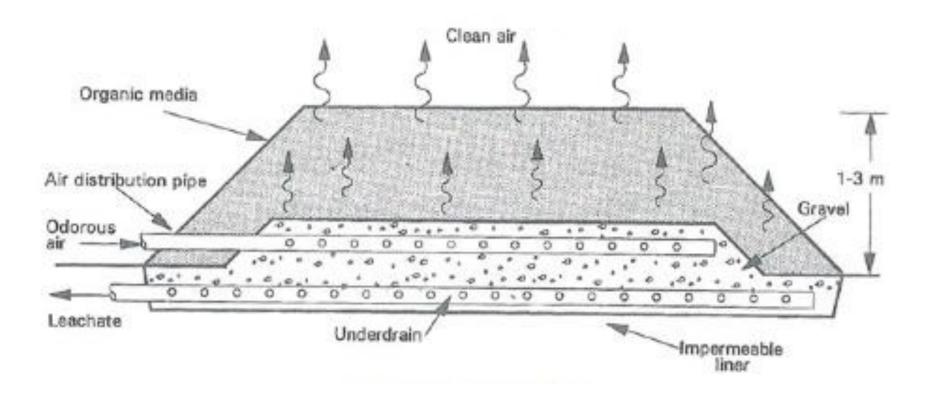
Removes biodegradable water soluble odors





Mechanism of Action

- Odorous compounds are solubilized on filter media
- Compounds are then degraded by bacteria on the media
- Sized based Empty Bed Residence Time EBDT
- Typical EBDT is 45 seconds



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Media is typically shredded wood or mulch. Synthetic media is also used to allow for shorter bed residence time.





Pros:

- Simple process good for consistent odor compounds and concentrations
- Typically low capital cost

Cons:

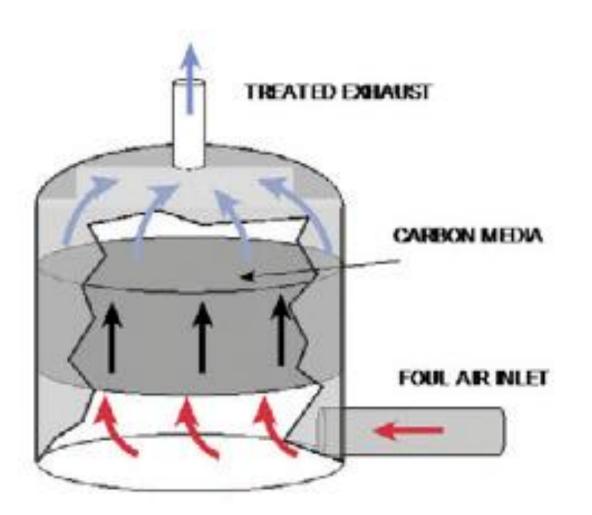
- Biological process may be prone to upsets
- Performance impacted by temperature
- Needs time to acclimate to odors to be treated
- Requires large footprint
- 85% removal efficiency

Effective on H₂S and sulfurbased compounds



Mechanism of Action

- Odorous air stream is passed through bed of adsorbent carbon
- Sulfur based odorous compounds adhere to the surface of the carbon removing them from the air stream
- 98% removal efficiency





Pros:

- Very simple and stable process
- 98% removal efficiency

Cons:

- H₂S concentrations above 10 ppm result in high operating costs
- Does not remove nitrogen based odors
- Humid air impacts performance so high humidity air must be dried prior to entering carbon filter



Carbon Adsorption

Carbon Filter with de-mister to remove particulate water

ΙΜΕΝΤΑΙ



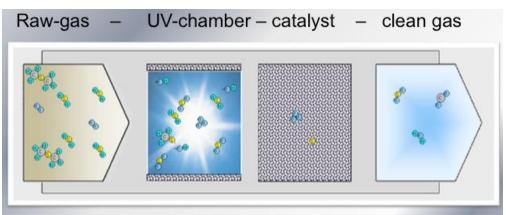
- Odor control technology that's been around for more than 20 years, commonly used in Europe
- Newer odor control technology in North America
- Process uses UV Light in combination with a Catalyst





Mechanism of Action

- UV Lamps produce reactive Oxygen ions O⁻²
 O₂⁻ O₂⁻², Hydroxyl radicals OH⁻, and Ozone
 O₃ to react and oxidize odorous compounds
- Catalyst provides a place for the reactions to go to completion.
- 98% removal efficiency



Pros:

- Reacts with all odorous compounds
- Low MX requirements
- Small footprint
- On/Off operation works immediately
- Handles high concentrations
- Low operating costs compared to other technologies
- 98% efficient

Cons:

Higher up front capital cost





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... and the air is clean.

Liquid Phase Technologies

Treats the liquid stream Typically used in the collection system

- Iron Salts
- Bioxide
- Hybrid technology



Iron Salts

Method of Action

- Ferric Salts are metered into the liquid stream
- Chemical reaction oxidizes some of the Sulfides and precipitates the rest



Iron Salts

Pros:

- Chemical reaction provides consistent results
- Ferric released at the treatment plant will precipitate phosphorous

Cons:

- Requires chemical delivery and storage
- MX of chemical metering pumps
- Must be protected from freezing weather

Bioxide

Method of Action

 Proprietary Nitrate solution is injected in liquid stream in sufficient quantities to biochemically oxidize sulfides



Bioxide

Pros:

 Multiple point odor control in collection system

Cons:

- Proprietary Process
- Expensive chemical
- Chemical dosing pump need attention
- Must protect equipment from weather



New hybrid odor control technology targeted for Lift Stations, Wet Wells, and Manholes

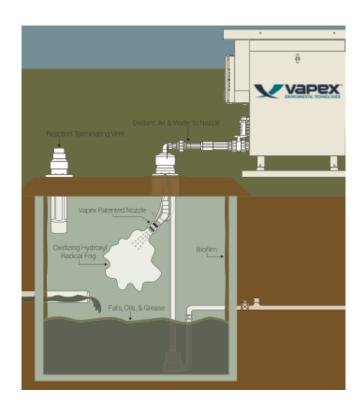






Mechanism of Action

- Combination of Ozone and H₂O produces reactive Hydroxyl radical OH⁻
- Sprayed in the headspace of wet well to oxidize odorous compounds
- Hybrid technology blending Liquid and Vapor Phase odor removal



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Pros:

 In addition to oxidizing odorous compounds, it will break down FOG in the wet well

Cons:

- Power requirements
- New technology with little track record

Questions?

Comments?

General Excitement?

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