

# *Seven Perfect Days*



Presented by Mr. Derek J. Marshall

At

IDMR; Lansing, Michigan

9/21/2007

# Introduction

- É Four major thrusts of my work in Creation Physics since 1995:
1. Literal 7-Day Creation by Yahweh through Yahshua.
  2. Literal Flooding of Creation as reported in Genesis
  3. The Fall of Man
  4. The Revelation
- É This presentation is an overview of Creation, the *Seven Perfect Days*.



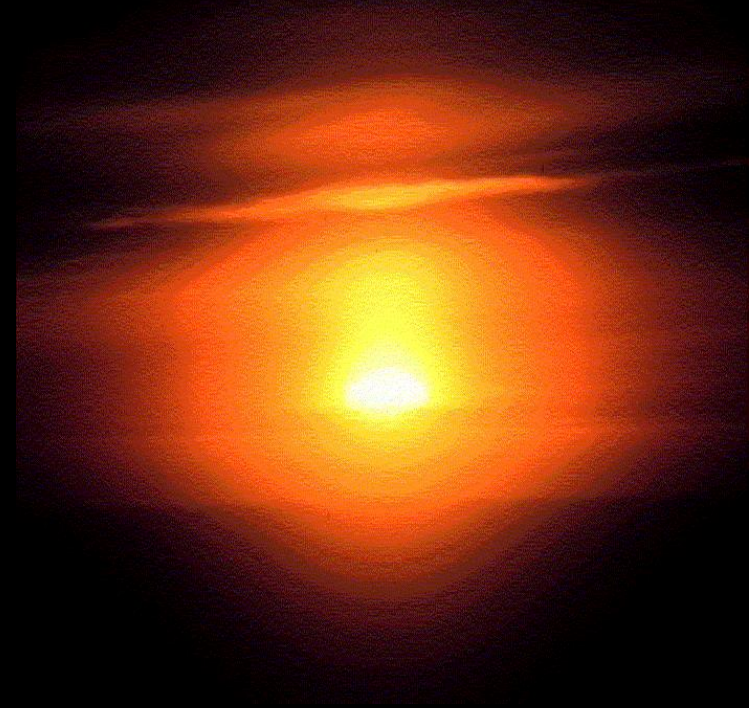
# *My Background*

- É High School: Evergreen Baptist Academy, Class of 1983.
- É USMC; 1983-1989, SGT, E-5, Protestant Lay Leader, Radio Repairman
- É Trigon Electronics; Corona, CA. 1989-1996, Electronics Engineer
- É Bachelor of Arts; Physics, Michigan State University, 2006



# *Common Questions...*

- É Seven Days vs. 4.5 Billion Years????
- É Evolution???
- É Fossils?
- É Isotopic Dating?
- É Noah's Ark?
- É Adam and Eve?
- É Garden of Eden?
- É The 666 in our future?



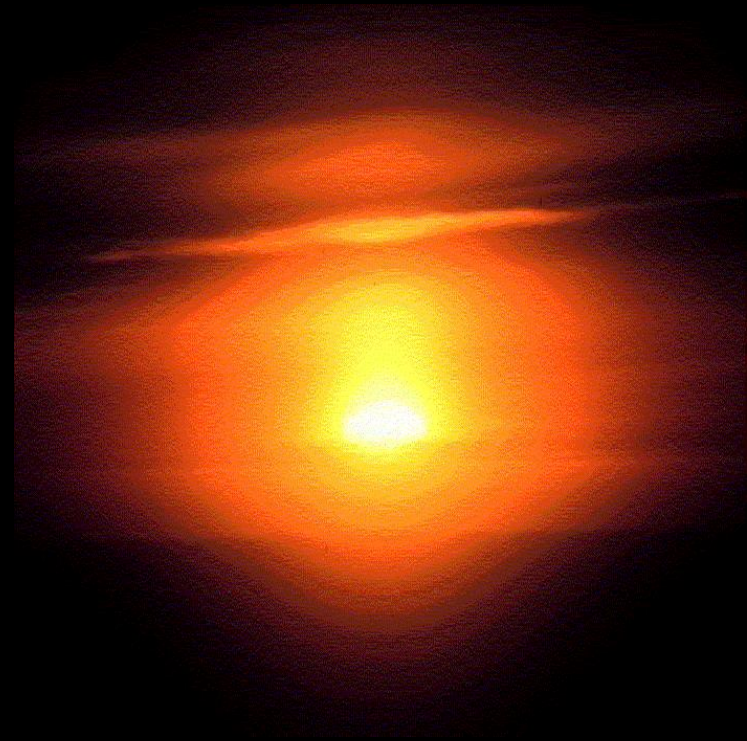
# *...and where we will start*

## É Seven Days?

We will see how much was accomplished, using

- 1) Scripture KJV Bible
- 2) Published scientific data

And I will do my best to use the proper names and titles  
For our Savior and Creator.



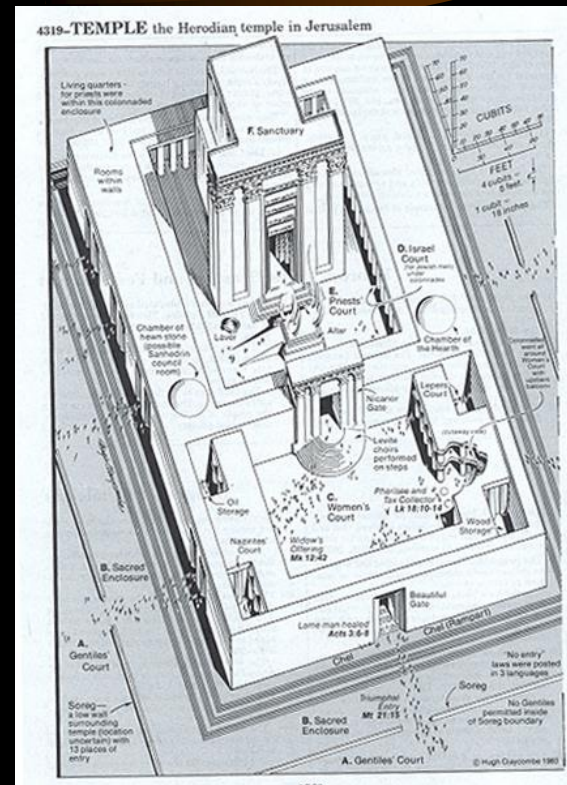
## *What will be Discussed:*

- É Mathematical basis of Creation described by the Bible.
- É Compare this basis to the Period Table of the Elements.
- É Compare this basis to the Tabernacle Pattern.



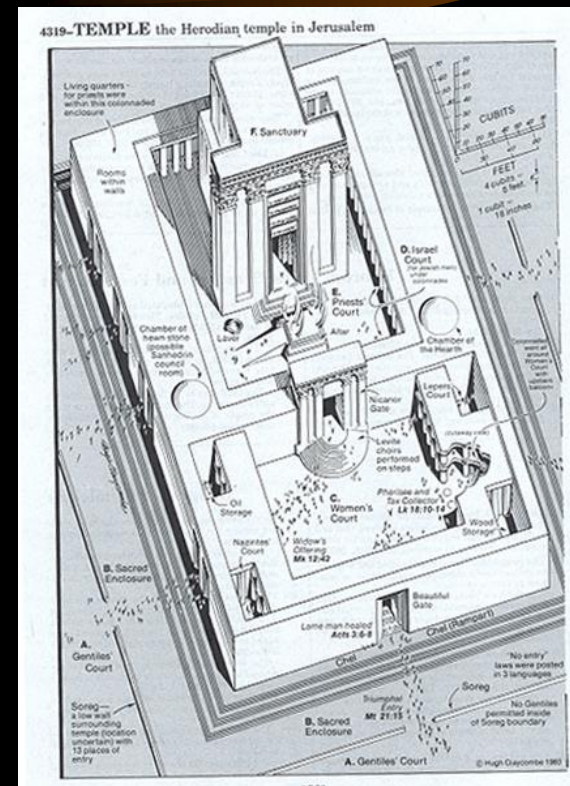
## *Then we can discuss:*

- É The Fallen Position(s) of Man and Woman.
- É Compare these positions to Solar System data.
- É Compare these positions to the Temple at Jerusalem.



# *What this presentation will accomplish:*

- É It will prove that the Bible describes the pattern of the Periodic Table
- É It will prove that the tabernacle and temple also exhibit the pattern of the Periodic Table
- É Thereby validating IDMR claim that the Tabernacle Pattern is, in fact, the Pattern of the Universe.





## ...*PROOOVE???*

É It will prove that the Bible describes the pattern of the Periodic Table!

- ó Since I will be using numbers from the Bible to match-up to numbers from published scientific works, I can produce a probability. (Like winning Power Ball)
- ó I will keep track of the number of matches JUST PRESENTED HERE and derive a probability.
- ó Keep in mind, this is just an OVERVIEW of my work in ONE AREA and there is much more evidence.

# *The Numerical Basis for Creation*



## É Please Read Scriptures:

- ó Is. 40:26      ðBY NUMBERö
- ó Prov. 8:27      ðUSED COMPASSö
- ó Ps. 139:14      ðSUM, IN CONTINUANCEö
- ó Job 38:4      ðMEASURES, LINE, STONEö

# *The Numerical Basis for Creation*

É Let's set this up:

ó ðBY NUMBERö

É {1,2,3,í ,öend of 7-day creationö}

ó ðUSED COMPASSö

É {1° to 360 ° | where 360 ° = 1 Day}

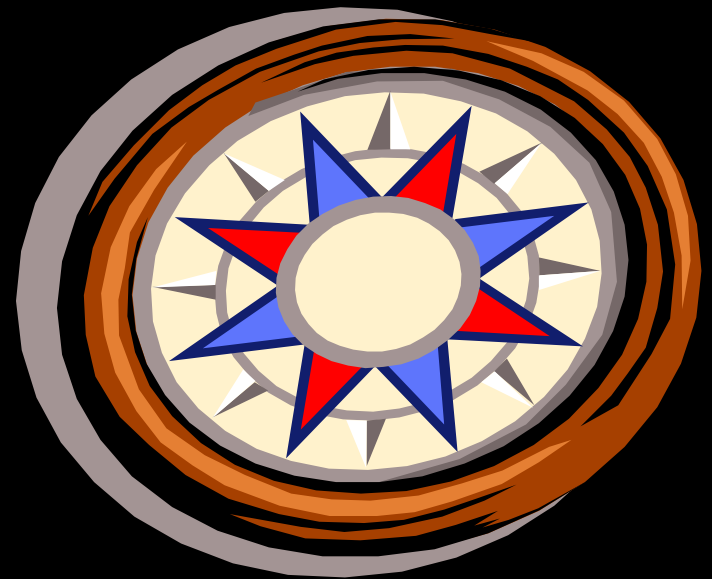
ó ðSUM, IN CONTINUANCEö

É means ðContinually Addö

É Use ö ö function

ó ðMEASURES, LINE, STONEö

É Physical dimensional resultant



# *The Numerical Basis for Creation*



É Here are Creation Function definitions:

- ó Definition: the *ROOT* number is the variable integer used in the function, i.e., {1,2,3,í ,öend 7-day creationö}
- ó Definition: the *SCOPE* angle is the instantaneous resultant angular focus of the Almighty at that COMPASS point in Creation, i.e., {1° to 360 °| where 360 ° = 1 Day}

# *The Numerical Basis for Creation*



É We are going to bring out the ROOT numbers in order, starting with { 1 }, calculate the SCOPE angle generated, and ADD that SCOPE angle to the previous (or initial) SCOPE position.

# *The Numerical Basis for Creation*

É For  $\text{ROOT} = 1$ ;  $\text{SCOPE} = 360^\circ/\text{ROOT}$ , so:

- ó  $\text{SCOPE} = 360^\circ/1 = 360^\circ$  or  $0^\circ$  our starting point.
- ó This corresponds to the Cornerstone.
- ó The spiritual significance of this is beyond the scope of this session.



# *The Numerical Basis for Creation*

É For ROOT = 2; SCOPE =  $360^\circ/\text{ROOT}$ , so:

- ó SCOPE =  $360^\circ/2 = 180^\circ$
- ó This corresponds to dividing Light from Darkness
- ó Since the Creator called out Nights first,  $0^\circ$  to  $180^\circ = \text{NIGHT}$
- ó Therefore, This is this NIGHT 1



# *The Numerical Basis for Creation*

É For  $\text{ROOT} = 3$ ;  $\text{SCOPE} = 360^\circ/\text{ROOT}$

- ó  $\text{SCOPE} = 360^\circ/3 = 120^\circ$
- ó And, previous position was  $180^\circ$
- ó ADDING  $120^\circ$ :  $180 + 120 = 300^\circ$
- ó New position of Creator's focus:  $300^\circ$
- ó Since  $180^\circ$  to  $360^\circ$  is DAY
- ó Therefore this is DAY 1





# *The Numerical Basis for Creation*

- É Here is the Creation Function carried to the end of DAY 3:
  - ó Notice the ROOT numbers at the NIGHT (180°) and DAY (360°) SCOPE positions.
- É This Creation Function forms the ðPattern of the Universeö.
- É The next slide will show the corresponding Creation Function Graph next to a common picture in nature.

## **DAY 1: Night and Day**

ROOT=1:  $360^\circ/1 = 360^\circ$

ROOT=2:  $0^\circ + (360/2)^\circ = 180^\circ$

ROOT=3:  $180^\circ + (360/3)^\circ = 300^\circ$

## **DAY 2: Firmament**

ROOT=4:  $300^\circ + (360/4)^\circ = 390^\circ$  or  $30^\circ$

ROOT=5:  $30^\circ + (360/5)^\circ = 102^\circ$

ROOT=6:  $102^\circ + (360/6)^\circ = 162^\circ$

ROOT=7:  $162^\circ + (360/7)^\circ = 213^\circ$

ROOT=8:  $213^\circ + (360/8)^\circ = 258^\circ$

ROOT=9:  $258^\circ + (360/9)^\circ = 298^\circ$

ROOT=10:  $298^\circ + (360/10)^\circ = 334^\circ$

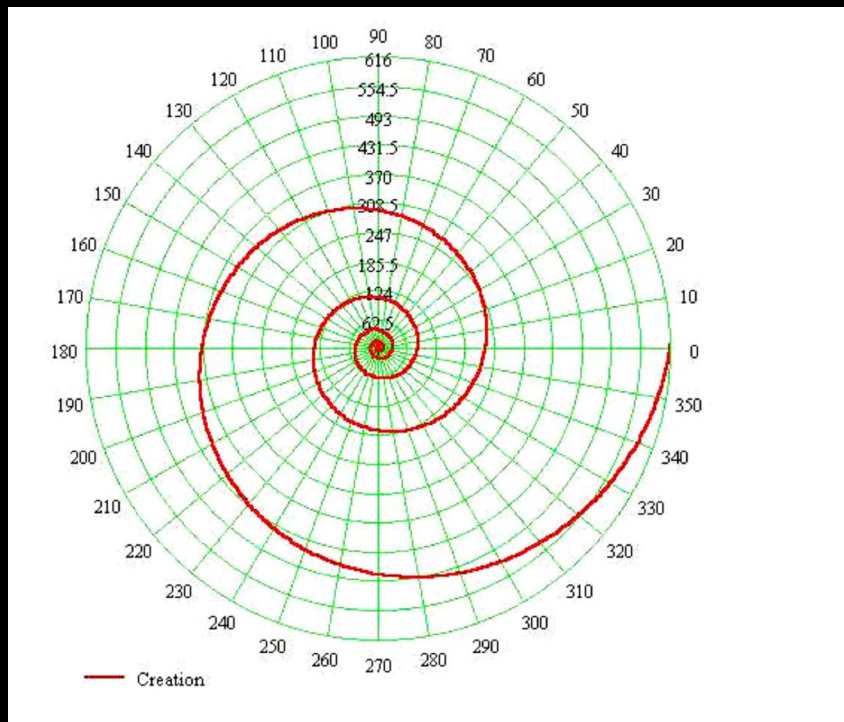
## **DAY 3: Dry Land, Seas, Plants;**

ROOT=11:  $334^\circ + (360/11)^\circ = 7^\circ$

ROOT=18: =  $178^\circ$  end NIGHT 3

ROOT=29: =  $358^\circ$  end DAY 3

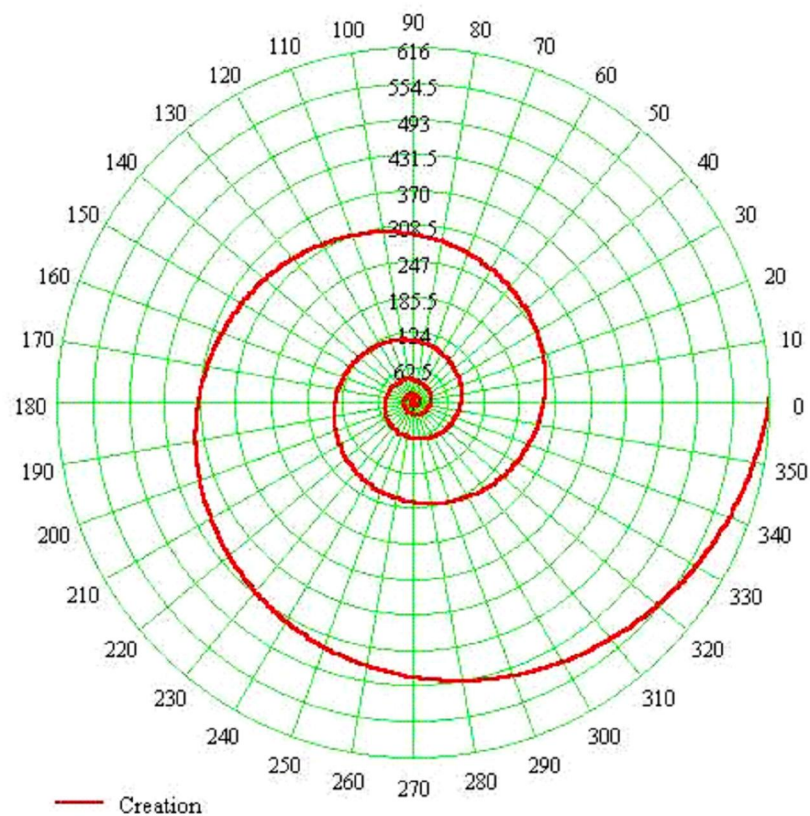
# *The Numerical Basis for Creation*



Messier 51, A Spiral Galaxy

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# *The Numerical Basis for Creation*



A Nautilus shell

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# The Numerical Basis vs Periodic Table

## DAY1: Night and Day

ROOT=1:  $360^\circ/1 = 360^\circ$

ROOT=2:  $0^\circ + (360/2)^\circ = 180^\circ$

ROOT=3:  $180^\circ + (360/3)^\circ = 300^\circ$

## DAY2: Firmament

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## DAY3: Dry Land, Seas, Plants;

ROOT=11:  $334^\circ + (360/11)^\circ = 7^\circ$

ROOT=18: =  $178^\circ$  end NIGHT 3

ROOT=29: =  $358^\circ$  end DAY 3

Main-Group Elements s Subshell fills

Main-Group Elements p Subshell fills

Transition Metals d Subshell fills

Inner-Transition Metals f Subshell fills

\*Lanthanides

\*\*Actinides

Legend: Metal (blue), Metalloid (purple), Nonmetal (orange)

Legend: Atomic number, Symbol, Valence-shell configuration

1	2	Transition Metals										13	14	15	16	17	18					
IA	IIA	III B	IV B	V B	VI B	VII B	VIII B				IX B	X B	III A	IV A	V A	VI A	VII A	VIII A				
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18					
H	He	Li	Be	B	C	N	O	F	Ne	Na	Mg	Al	Si	P	S	Cl	Ar					
1s <sup>1</sup>	1s <sup>2</sup>	2s <sup>1</sup>	2s <sup>2</sup>	2s <sup>2</sup> 2p <sup>1</sup>	2s <sup>2</sup> 2p <sup>2</sup>	2s <sup>2</sup> 2p <sup>3</sup>	2s <sup>2</sup> 2p <sup>4</sup>	2s <sup>2</sup> 2p <sup>5</sup>	2s <sup>2</sup> 2p <sup>6</sup>	3s <sup>1</sup>	3s <sup>2</sup>	3s <sup>2</sup> 3p <sup>1</sup>	3s <sup>2</sup> 3p <sup>2</sup>	3s <sup>2</sup> 3p <sup>3</sup>	3s <sup>2</sup> 3p <sup>4</sup>	3s <sup>2</sup> 3p <sup>5</sup>	3s <sup>2</sup> 3p <sup>6</sup>					
3	4	11	12	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	
Li	Be	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr			
2s <sup>1</sup>	2s <sup>2</sup>	4s <sup>1</sup>	4s <sup>2</sup>	3d <sup>1</sup> 4s <sup>2</sup>	3d <sup>2</sup> 4s <sup>2</sup>	3d <sup>3</sup> 4s <sup>2</sup>	3d <sup>4</sup> 4s <sup>1</sup>	3d <sup>5</sup> 4s <sup>2</sup>	3d <sup>6</sup> 4s <sup>2</sup>	3d <sup>7</sup> 4s <sup>2</sup>	3d <sup>8</sup> 4s <sup>2</sup>	3d <sup>9</sup> 4s <sup>1</sup>	3d <sup>10</sup> 4s <sup>2</sup>	4s <sup>2</sup> 4p <sup>1</sup>	4s <sup>2</sup> 4p <sup>2</sup>	4s <sup>2</sup> 4p <sup>3</sup>	4s <sup>2</sup> 4p <sup>4</sup>	4s <sup>2</sup> 4p <sup>5</sup>	4s <sup>2</sup> 4p <sup>6</sup>			
5	6	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54			
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe					
5s <sup>1</sup>	5s <sup>2</sup>	4d <sup>1</sup> 5s <sup>2</sup>	4d <sup>2</sup> 5s <sup>2</sup>	4d <sup>3</sup> 5s <sup>1</sup>	4d <sup>4</sup> 5s <sup>1</sup>	4d <sup>5</sup> 5s <sup>1</sup>	4d <sup>6</sup> 5s <sup>1</sup>	4d <sup>7</sup> 5s <sup>1</sup>	4d <sup>8</sup> 5s <sup>1</sup>	4d <sup>9</sup> 5s <sup>1</sup>	4d <sup>10</sup> 5s <sup>1</sup>	5s <sup>2</sup> 5p <sup>1</sup>	5s <sup>2</sup> 5p <sup>2</sup>	5s <sup>2</sup> 5p <sup>3</sup>	5s <sup>2</sup> 5p <sup>4</sup>	5s <sup>2</sup> 5p <sup>5</sup>	5s <sup>2</sup> 5p <sup>6</sup>					
6	7	55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86			
Cs	Ba	La*	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn					
6s <sup>1</sup>	6s <sup>2</sup>	5d <sup>1</sup> 6s <sup>2</sup>	5d <sup>2</sup> 6s <sup>2</sup>	5d <sup>3</sup> 6s <sup>2</sup>	5d <sup>4</sup> 6s <sup>2</sup>	5d <sup>5</sup> 6s <sup>2</sup>	5d <sup>6</sup> 6s <sup>2</sup>	5d <sup>7</sup> 6s <sup>2</sup>	5d <sup>8</sup> 6s <sup>2</sup>	5d <sup>9</sup> 6s <sup>1</sup>	5d <sup>10</sup> 6s <sup>2</sup>	6s <sup>2</sup> 6p <sup>1</sup>	6s <sup>2</sup> 6p <sup>2</sup>	6s <sup>2</sup> 6p <sup>3</sup>	6s <sup>2</sup> 6p <sup>4</sup>	6s <sup>2</sup> 6p <sup>5</sup>	6s <sup>2</sup> 6p <sup>6</sup>					
7	8	87	88	89	104	105	106	107	108	109												
Fr	Ra	Ac**	Db	Jl	Rf	Bh	Hn	Mt														
7s <sup>1</sup>	7s <sup>2</sup>	6d <sup>1</sup> 7s <sup>2</sup>	6d <sup>2</sup> 7s <sup>2</sup>	6d <sup>3</sup> 7s <sup>2</sup>	6d <sup>4</sup> 7s <sup>2</sup>	6d <sup>5</sup> 7s <sup>2</sup>	6d <sup>6</sup> 7s <sup>2</sup>	6d <sup>7</sup> 7s <sup>2</sup>	6d <sup>8</sup> 7s <sup>2</sup>													
		Inner-Transition Metals																				
		58	59	60	61	62	63	64	65	66	67	68	69	70	71							
		Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu							
		4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup>	4f <sup>1</sup> 6s <sup>2</sup>	4f <sup>1</sup> 6s <sup>2</sup>	4f <sup>1</sup> 6s <sup>2</sup>	4f <sup>6</sup> 6s <sup>2</sup>	4f <sup>7</sup> 6s <sup>2</sup>	4f <sup>7</sup> 6s <sup>2</sup>	4f <sup>7</sup> 6s <sup>2</sup>	4f <sup>9</sup> 6s <sup>2</sup>	4f <sup>10</sup> 6s <sup>2</sup>	4f <sup>11</sup> 6s <sup>2</sup>	4f <sup>11</sup> 6s <sup>2</sup>	4f <sup>13</sup> 6s <sup>2</sup>	4f <sup>14</sup> 6s <sup>2</sup>	4f <sup>14</sup> 6s <sup>2</sup>						
		90	91	92	93	94	95	96	97	98	99	100	101	102	103							
		Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr							
		6d <sup>1</sup> 7s <sup>2</sup>	5f <sup>1</sup> 6d <sup>1</sup> 7s <sup>2</sup>	5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup>	5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup>	5f <sup>7</sup> 7s <sup>2</sup>	5f <sup>7</sup> 7s <sup>2</sup>	5f <sup>7</sup> 7s <sup>2</sup>	5f <sup>10</sup> 7s <sup>2</sup>	5f <sup>10</sup> 7s <sup>2</sup>	5f <sup>14</sup> 7s <sup>2</sup>	5f <sup>14</sup> 7s <sup>2</sup>	5f <sup>14</sup> 7s <sup>2</sup>	5f <sup>14</sup> 7s <sup>2</sup>	5f <sup>14</sup> 7s <sup>2</sup>	5f <sup>14</sup> 7s <sup>2</sup>						

# The Numerical Basis vs. Periodic Table

## É Day 1, the Creation of Day and Night

- ó ROOT = 1, 2, (3)
- ó Z = 1, Hydrogen, is Element 1
- ó Z = 2, Helium, is Element 2
- ó The Sun is mostly Hydrogen and Helium.
- ó Fusion to Element 12, Carbon

Main-Group Elements s Subshell fills

Main-Group Elements p Subshell fills

1 — Atomic number  
H — Symbol  
1s<sup>1</sup> — Valence-shell configuration

Period	1 IA	2 IIA	Transition Metals d Subshell fills										13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA								
1	1 H 1s <sup>1</sup>																		2 He 1s <sup>2</sup>							
2	3 Li 2s <sup>1</sup>	4 Be 2s <sup>2</sup>											5 B 2s <sup>2</sup> 2p <sup>1</sup>	6 C 2s <sup>2</sup> 2p <sup>2</sup>	7 N 2s <sup>2</sup> 2p <sup>3</sup>	8 O 2s <sup>2</sup> 2p <sup>4</sup>	9 F 2s <sup>2</sup> 2p <sup>5</sup>	10 Ne 2s <sup>2</sup> 2p <sup>6</sup>								
3	11 Na 3s <sup>1</sup>	12 Mg 3s <sup>2</sup>	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIII B	9 VIII B	10 VIII B	11 IB	12 IIB	13 Al 3s <sup>2</sup> 3p <sup>1</sup>	14 Si 3s <sup>2</sup> 3p <sup>2</sup>	15 P 3s <sup>2</sup> 3p <sup>3</sup>	16 S 3s <sup>2</sup> 3p <sup>4</sup>	17 Cl 3s <sup>2</sup> 3p <sup>5</sup>	18 Ar 3s <sup>2</sup> 3p <sup>6</sup>								
4	19 K 4s <sup>1</sup>	20 Ca 4s <sup>2</sup>	21 Sc 3d <sup>1</sup> 4s <sup>2</sup>	22 Ti 3d <sup>2</sup> 4s <sup>2</sup>	23 V 3d <sup>3</sup> 4s <sup>2</sup>	24 Cr 3d <sup>5</sup> 4s <sup>1</sup>	25 Mn 3d <sup>5</sup> 4s <sup>2</sup>	26 Fe 3d <sup>6</sup> 4s <sup>2</sup>	27 Co 3d <sup>7</sup> 4s <sup>2</sup>	28 Ni 3d <sup>8</sup> 4s <sup>2</sup>	29 Cu 3d <sup>10</sup> 4s <sup>1</sup>	30 Zn 3d <sup>10</sup> 4s <sup>2</sup>	31 Ga 4s <sup>2</sup> 4p <sup>1</sup>	32 Ge 4s <sup>2</sup> 4p <sup>2</sup>	33 As 4s <sup>2</sup> 4p <sup>3</sup>	34 Se 4s <sup>2</sup> 4p <sup>4</sup>	35 Br 4s <sup>2</sup> 4p <sup>5</sup>	36 Kr 4s <sup>2</sup> 4p <sup>6</sup>								
5	37 Rb 5s <sup>1</sup>	38 Sr 5s <sup>2</sup>	39 Y 4d <sup>1</sup> 5s <sup>2</sup>	40 Zr 4d <sup>2</sup> 5s <sup>2</sup>	41 Nb 4d <sup>4</sup> 5s <sup>1</sup>	42 Mo 4d <sup>5</sup> 5s <sup>1</sup>	43 Tc 4d <sup>5</sup> 5s <sup>2</sup>	44 Ru 4d <sup>7</sup> 5s <sup>1</sup>	45 Rh 4d <sup>8</sup> 5s <sup>1</sup>	46 Pd 4d <sup>10</sup>	47 Ag 4d <sup>10</sup> 5s <sup>1</sup>	48 Cd 4d <sup>10</sup> 5s <sup>2</sup>	49 In 5s <sup>2</sup> 5p <sup>1</sup>	50 Sn 5s <sup>2</sup> 5p <sup>2</sup>	51 Sb 5s <sup>2</sup> 5p <sup>3</sup>	52 Te 5s <sup>2</sup> 5p <sup>4</sup>	53 I 5s <sup>2</sup> 5p <sup>5</sup>	54 Xe 5s <sup>2</sup> 5p <sup>6</sup>								
6	55 Cs 6s <sup>1</sup>	56 Ba 6s <sup>2</sup>	57 La*	72 Hf 5d <sup>2</sup> 6s <sup>2</sup>	73 Ta 5d <sup>3</sup> 6s <sup>2</sup>	74 W 5d <sup>4</sup> 6s <sup>2</sup>	75 Re 5d <sup>5</sup> 6s <sup>2</sup>	76 Os 5d <sup>6</sup> 6s <sup>2</sup>	77 Ir 5d <sup>7</sup> 6s <sup>2</sup>	78 Pt 5d <sup>9</sup> 6s <sup>1</sup>	79 Au 5d <sup>10</sup> 6s <sup>1</sup>	80 Hg 5d <sup>10</sup> 6s <sup>2</sup>	81 Tl 6s <sup>2</sup> 6p <sup>1</sup>	82 Pb 6s <sup>2</sup> 6p <sup>2</sup>	83 Bi 6s <sup>2</sup> 6p <sup>3</sup>	84 Po 6s <sup>2</sup> 6p <sup>4</sup>	85 At 6s <sup>2</sup> 6p <sup>5</sup>	86 Rn 6s <sup>2</sup> 6p <sup>6</sup>								
7	87 Fr 7s <sup>1</sup>	88 Ra 7s <sup>2</sup>	89 Ac**	104 Db 6d <sup>3</sup> 7s <sup>2</sup>	105 JI 6d <sup>4</sup> 7s <sup>2</sup>	106 Rf 6d <sup>4</sup> 7s <sup>2</sup>	107 Bh 6d <sup>5</sup> 7s <sup>2</sup>	108 Hn 6d <sup>6</sup> 7s <sup>2</sup>	109 Mt 6d <sup>7</sup> 7s <sup>2</sup>	Inner-Transition Metals f Subshell fills																
			*Lanthanides										58 Ce 4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup>	59 Pr 4f <sup>2</sup> 6s <sup>2</sup>	60 Nd 4f <sup>3</sup> 6s <sup>2</sup>	61 Pm 4f <sup>4</sup> 6s <sup>2</sup>	62 Sm 4f <sup>6</sup> 6s <sup>2</sup>	63 Eu 4f <sup>7</sup> 6s <sup>2</sup>	64 Gd 4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup>	65 Tb 4f <sup>9</sup> 6s <sup>2</sup>	66 Dy 4f <sup>10</sup> 6s <sup>2</sup>	67 Ho 4f <sup>11</sup> 6s <sup>2</sup>	68 Er 4f <sup>12</sup> 6s <sup>2</sup>	69 Tm 4f <sup>13</sup> 6s <sup>2</sup>	70 Yb 4f <sup>14</sup> 6s <sup>2</sup>	71 Lu 4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup>
			**Actinides										90 Th 6d <sup>2</sup> 7s <sup>2</sup>	91 Pa 5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup>	92 U 5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup>	93 Np 5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup>	94 Pu 5f <sup>6</sup> 7s <sup>2</sup>	95 Am 5f <sup>7</sup> 7s <sup>2</sup>	96 Cm 5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup>	97 Bk 5f <sup>9</sup> 7s <sup>2</sup>	98 Cf 5f <sup>10</sup> 7s <sup>2</sup>	99 Es 5f <sup>11</sup> 7s <sup>2</sup>	100 Fm 5f <sup>12</sup> 7s <sup>2</sup>	101 Md 5f <sup>13</sup> 7s <sup>2</sup>	102 No 5f <sup>14</sup> 7s <sup>2</sup>	103 Lr 5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup>

Legend: Metal (Blue), Metalloid (Purple), Nonmetal (Orange)

# The Numerical Basis vs. Periodic Table

## É Day 2, the Creation of the Firmament

- ó ROOT = (3) to 10
- ó Z = 6, Carbon, is essential for life
- ó Z = 7, Nitrogen, is 78% of our atmosphere
- ó Z = 8, Oxygen, is 21% of our atmosphere
- ó The first 10 elements: commonly referred to as the ðrganic chemistð periodic table.

Main-Group Elements s Subshell fills																		Main-Group Elements p Subshell fills									
1 IA		2 IIA		Transition Metals d Subshell fills										13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA								
Period	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18									
1	1 H 1s <sup>1</sup>																	2 He 1s <sup>2</sup>									
2	3 Li 2s <sup>1</sup>	4 Be 2s <sup>2</sup>											5 B 2s <sup>2</sup> 2p <sup>1</sup>	6 C 2s <sup>2</sup> 2p <sup>2</sup>	7 N 2s <sup>2</sup> 2p <sup>3</sup>	8 O 2s <sup>2</sup> 2p <sup>4</sup>	9 F 2s <sup>2</sup> 2p <sup>5</sup>	10 Ne 2s <sup>2</sup> 2p <sup>6</sup>									
3	11 Na 3s <sup>1</sup>	12 Mg 3s <sup>2</sup>											13 Al 3s <sup>2</sup> 3p <sup>1</sup>	14 Si 3s <sup>2</sup> 3p <sup>2</sup>	15 P 3s <sup>2</sup> 3p <sup>3</sup>	16 S 3s <sup>2</sup> 3p <sup>4</sup>	17 Cl 3s <sup>2</sup> 3p <sup>5</sup>	18 Ar 3s <sup>2</sup> 3p <sup>6</sup>									
4	19 K 4s <sup>1</sup>	20 Ca 4s <sup>2</sup>	21 Sc 3d <sup>1</sup> 4s <sup>2</sup>	22 Ti 3d <sup>2</sup> 4s <sup>2</sup>	23 V 3d <sup>3</sup> 4s <sup>2</sup>	24 Cr 3d <sup>5</sup> 4s <sup>1</sup>	25 Mn 3d <sup>5</sup> 4s <sup>2</sup>	26 Fe 3d <sup>6</sup> 4s <sup>2</sup>	27 Co 3d <sup>7</sup> 4s <sup>2</sup>	28 Ni 3d <sup>8</sup> 4s <sup>2</sup>	29 Cu 3d <sup>10</sup> 4s <sup>1</sup>	30 Zn 3d <sup>10</sup> 4s <sup>2</sup>	31 Ga 4s <sup>2</sup> 4p <sup>1</sup>	32 Ge 4s <sup>2</sup> 4p <sup>2</sup>	33 As 4s <sup>2</sup> 4p <sup>3</sup>	34 Se 4s <sup>2</sup> 4p <sup>4</sup>	35 Br 4s <sup>2</sup> 4p <sup>5</sup>	36 Kr 4s <sup>2</sup> 4p <sup>6</sup>									
5	37 Rb 5s <sup>1</sup>	38 Sr 5s <sup>2</sup>	39 Y 4d <sup>1</sup> 5s <sup>2</sup>	40 Zr 4d <sup>2</sup> 5s <sup>2</sup>	41 Nb 4d <sup>4</sup> 5s <sup>1</sup>	42 Mo 4d <sup>5</sup> 5s <sup>1</sup>	43 Tc 4d <sup>5</sup> 5s <sup>2</sup>	44 Ru 4d <sup>7</sup> 5s <sup>1</sup>	45 Rh 4d <sup>8</sup> 5s <sup>1</sup>	46 Pd 4d <sup>10</sup>	47 Ag 4d <sup>10</sup> 5s <sup>1</sup>	48 Cd 4d <sup>10</sup> 5s <sup>2</sup>	49 In 5s <sup>2</sup> 5p <sup>1</sup>	50 Sn 5s <sup>2</sup> 5p <sup>2</sup>	51 Sb 5s <sup>2</sup> 5p <sup>3</sup>	52 Te 5s <sup>2</sup> 5p <sup>4</sup>	53 I 5s <sup>2</sup> 5p <sup>5</sup>	54 Xe 5s <sup>2</sup> 5p <sup>6</sup>									
6	55 Cs 6s <sup>1</sup>	56 Ba 6s <sup>2</sup>	57 La* 5d <sup>1</sup> 6s <sup>2</sup>	72 Hf 5d <sup>2</sup> 6s <sup>2</sup>	73 Ta 5d <sup>3</sup> 6s <sup>2</sup>	74 W 5d <sup>4</sup> 6s <sup>2</sup>	75 Re 5d <sup>5</sup> 6s <sup>2</sup>	76 Os 5d <sup>6</sup> 6s <sup>2</sup>	77 Ir 5d <sup>7</sup> 6s <sup>2</sup>	78 Pt 5d <sup>9</sup> 6s <sup>1</sup>	79 Au 5d <sup>10</sup> 6s <sup>1</sup>	80 Hg 5d <sup>10</sup> 6s <sup>2</sup>	81 Tl 6s <sup>2</sup> 6p <sup>1</sup>	82 Pb 6s <sup>2</sup> 6p <sup>2</sup>	83 Bi 6s <sup>2</sup> 6p <sup>3</sup>	84 Po 6s <sup>2</sup> 6p <sup>4</sup>	85 At 6s <sup>2</sup> 6p <sup>5</sup>	86 Rn 6s <sup>2</sup> 6p <sup>6</sup>									
7	87 Fr 7s <sup>1</sup>	88 Ra 7s <sup>2</sup>	89 Ac** 6d <sup>1</sup> 7s <sup>2</sup>	104 Rf 6d <sup>2</sup> 7s <sup>2</sup>	105 Db 6d <sup>3</sup> 7s <sup>2</sup>	106 Sg 6d <sup>4</sup> 7s <sup>2</sup>	107 Bh 6d <sup>5</sup> 7s <sup>2</sup>	108 Hs 6d <sup>6</sup> 7s <sup>2</sup>	109 Mt 6d <sup>7</sup> 7s <sup>2</sup>	Inner-Transition Metals f Subshell fills																	
			*Lanthanides							58 Ce 4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup>	59 Pr 4f <sup>3</sup> 6s <sup>2</sup>	60 Nd 4f <sup>4</sup> 6s <sup>2</sup>	61 Pm 4f <sup>5</sup> 6s <sup>2</sup>	62 Sm 4f <sup>6</sup> 6s <sup>2</sup>	63 Eu 4f <sup>7</sup> 6s <sup>2</sup>	64 Gd 4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup>	65 Tb 4f <sup>9</sup> 6s <sup>2</sup>	66 Dy 4f <sup>10</sup> 6s <sup>2</sup>	67 Ho 4f <sup>11</sup> 6s <sup>2</sup>	68 Er 4f <sup>12</sup> 6s <sup>2</sup>	69 Tm 4f <sup>13</sup> 6s <sup>2</sup>	70 Yb 4f <sup>14</sup> 6s <sup>2</sup>	71 Lu 4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup>				
			**Actinides							90 Th 6d <sup>2</sup> 7s <sup>2</sup>	91 Pa 5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup>	92 U 5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup>	93 Np 5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup>	94 Pu 5f <sup>6</sup> 7s <sup>2</sup>	95 Am 5f <sup>7</sup> 7s <sup>2</sup>	96 Cm 5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup>	97 Bk 5f <sup>9</sup> 7s <sup>2</sup>	98 Cf 5f <sup>10</sup> 7s <sup>2</sup>	99 Es 5f <sup>11</sup> 7s <sup>2</sup>	100 Fm 5f <sup>12</sup> 7s <sup>2</sup>	101 Md 5f <sup>13</sup> 7s <sup>2</sup>	102 No 5f <sup>14</sup> 7s <sup>2</sup>	103 Lr 5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup>				

Legend:

- Metal
- Metalloid
- Nonmetal

# The Numerical Basis vs. Periodic Table

## É NIGHT 3, the Creation of Earth, ROOT 11 to 18

- 6 Z=11: **Sodium:** The sixth most abundant metal of earth.
- 6 Z=12: **Magnesium:** The seventh most abundant metal of earth
- 6 Z=13: **Aluminum:** The third most abundant metal of earth.
- 6 Z=14: **Silicon:** The second most abundant (semi)metal of earth.
- 6 Z=15: **Phosphorus:** An abundant non-metal of earth.
- 6 Z=16: **Sulfur:** An abundant non-metal of earth
- 6 Z=17: **Chlorine:** An abundant non-metal of earth.
- 6 Z=18: **Argon:** The end of the 3<sup>rd</sup> period of the periodic table

Main-Group Elements s Subshell fills

Main-Group Elements p Subshell fills

1 — Atomic number  
H — Symbol  
1s<sup>1</sup> — Valence-shell configuration

Period	1 IA	2 IIA	Transition Metals d Subshell fills										13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA								
1	1 H 1s <sup>1</sup>																									
2	3 Li 2s <sup>2</sup>	4 Be 2s <sup>2</sup>											5 B 2s <sup>2</sup> 2p <sup>1</sup>	6 C 2s <sup>2</sup> 2p <sup>2</sup>	7 N 2s <sup>2</sup> 2p <sup>3</sup>	8 O 2s <sup>2</sup> 2p <sup>4</sup>	9 F 2s <sup>2</sup> 2p <sup>5</sup>	10 Ne 2s <sup>2</sup> 2p <sup>6</sup>								
3	11 Na 3s <sup>2</sup>	12 Mg 3s <sup>2</sup>	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIII B	9 VIII B	10 VIII B	11 IB	12 IIB	13 Al 3s <sup>2</sup> 3p <sup>1</sup>	14 Si 3s <sup>2</sup> 3p <sup>2</sup>	15 P 3s <sup>2</sup> 3p <sup>3</sup>	16 S 3s <sup>2</sup> 3p <sup>4</sup>	17 Cl 3s <sup>2</sup> 3p <sup>5</sup>	18 Ar 3s <sup>2</sup> 3p <sup>6</sup>								
4	19 K 4s <sup>1</sup>	20 Ca 4s <sup>2</sup>	21 Sc 3d <sup>1</sup> 4s <sup>2</sup>	22 Ti 3d <sup>2</sup> 4s <sup>2</sup>	23 V 3d <sup>3</sup> 4s <sup>2</sup>	24 Cr 3d <sup>5</sup> 4s <sup>1</sup>	25 Mn 3d <sup>5</sup> 4s <sup>2</sup>	26 Fe 3d <sup>6</sup> 4s <sup>2</sup>	27 Co 3d <sup>7</sup> 4s <sup>2</sup>	28 Ni 3d <sup>8</sup> 4s <sup>2</sup>	29 Cu 3d <sup>10</sup> 4s <sup>1</sup>	30 Zn 3d <sup>10</sup> 4s <sup>2</sup>	31 Ga 4s <sup>2</sup> 4p <sup>1</sup>	32 Ge 4s <sup>2</sup> 4p <sup>2</sup>	33 As 4s <sup>2</sup> 4p <sup>3</sup>	34 Se 4s <sup>2</sup> 4p <sup>4</sup>	35 Br 4s <sup>2</sup> 4p <sup>5</sup>	36 Kr 4s <sup>2</sup> 4p <sup>6</sup>								
5	37 Rb 5s <sup>1</sup>	38 Sr 5s <sup>2</sup>	39 Y 4d <sup>1</sup> 5s <sup>2</sup>	40 Zr 4d <sup>2</sup> 5s <sup>2</sup>	41 Nb 4d <sup>4</sup> 5s <sup>1</sup>	42 Mo 4d <sup>5</sup> 5s <sup>1</sup>	43 Tc 4d <sup>5</sup> 5s <sup>2</sup>	44 Ru 4d <sup>7</sup> 5s <sup>1</sup>	45 Rh 4d <sup>8</sup> 5s <sup>1</sup>	46 Pd 4d <sup>10</sup>	47 Ag 4d <sup>10</sup> 5s <sup>1</sup>	48 Cd 4d <sup>10</sup> 5s <sup>2</sup>	49 In 5s <sup>2</sup> 5p <sup>1</sup>	50 Sn 5s <sup>2</sup> 5p <sup>2</sup>	51 Sb 5s <sup>2</sup> 5p <sup>3</sup>	52 Te 5s <sup>2</sup> 5p <sup>4</sup>	53 I 5s <sup>2</sup> 5p <sup>5</sup>	54 Xe 5s <sup>2</sup> 5p <sup>6</sup>								
6	55 Cs 6s <sup>1</sup>	56 Ba 6s <sup>2</sup>	57 La*	72 Hf 5d <sup>2</sup> 6s <sup>2</sup>	73 Ta 5d <sup>3</sup> 6s <sup>2</sup>	74 W 5d <sup>4</sup> 6s <sup>2</sup>	75 Re 5d <sup>5</sup> 6s <sup>2</sup>	76 Os 5d <sup>6</sup> 6s <sup>2</sup>	77 Ir 5d <sup>7</sup> 6s <sup>2</sup>	78 Pt 5d <sup>9</sup> 6s <sup>1</sup>	79 Au 5d <sup>10</sup> 6s <sup>1</sup>	80 Hg 5d <sup>10</sup> 6s <sup>2</sup>	81 Tl 6s <sup>2</sup> 6p <sup>1</sup>	82 Pb 6s <sup>2</sup> 6p <sup>2</sup>	83 Bi 6s <sup>2</sup> 6p <sup>3</sup>	84 Po 6s <sup>2</sup> 6p <sup>4</sup>	85 At 6s <sup>2</sup> 6p <sup>5</sup>	86 Rn 6s <sup>2</sup> 6p <sup>6</sup>								
7	87 Fr 7s <sup>1</sup>	88 Ra 7s <sup>2</sup>	89 Ac**	104 Db 6d <sup>3</sup> 7s <sup>2</sup>	105 JI 6d <sup>4</sup> 7s <sup>2</sup>	106 Rf 6d <sup>4</sup> 7s <sup>2</sup>	107 Bh 6d <sup>5</sup> 7s <sup>2</sup>	108 Hn 6d <sup>6</sup> 7s <sup>2</sup>	109 Mt 6d <sup>7</sup> 7s <sup>2</sup>	Inner-Transition Metals f Subshell fills																
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			**Actinides										90 Th 6d <sup>2</sup> 7s <sup>2</sup>	91 Pa 5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup>	92 U 5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup>	93 Np 5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup>	94 Pu 5f <sup>6</sup> 7s <sup>2</sup>	95 Am 5f <sup>7</sup> 7s <sup>2</sup>	96 Cm 5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup>	97 Bk 5f <sup>9</sup> 7s <sup>2</sup>	98 Cf 5f <sup>10</sup> 7s <sup>2</sup>	99 Es 5f <sup>11</sup> 7s <sup>2</sup>	100 Fm 5f <sup>12</sup> 7s <sup>2</sup>	101 Md 5f <sup>13</sup> 7s <sup>2</sup>	102 No 5f <sup>14</sup> 7s <sup>2</sup>	103 Lr 5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup>

Legend: Metal (blue), Metalloid (purple), Nonmetal (orange)

# The Numerical Basis vs. Periodic Table

## É DAY 3, Creation of Earth, ROOT 19 to 30

- 6 Z=19: **Potassium:** The eighth most abundant metal of earth.
- 6 Z=20: **Calcium:** The fifth most abundant metal of earth
- 6 Z=22: **Titanium:** An abundant metal of earth.
- 6 Z=23: **Vanadium:** An abundant metal of earth.
- 6 Z=24: **Chromium:** An abundant metal of earth
- 6 Z=25: **Manganese:** An abundant metal of earth.
- 6 Z=26: **Iron:** The fourth most abundant metal of earth.
- 6 Z=27: **Cobalt:** An abundant metal of earth.
- 6 Z=28: **Nickel:** An abundant metal of earth
- 6 Z=29: **Copper:** An abundant metal of earth
- 6 Z=30: **Zinc:** An abundant metal of earth

The periodic table is color-coded by element classification: Metals (blue), Metalloids (purple), and Nonmetals (orange). A legend indicates:
 

- Blue box: Metal
- Purple box: Metalloid
- Orange box: Nonmetal

 A callout box for Hydrogen (H) shows:
 

- 1: Atomic number
- H: Symbol
- 1s<sup>1</sup>: Valence-shell configuration

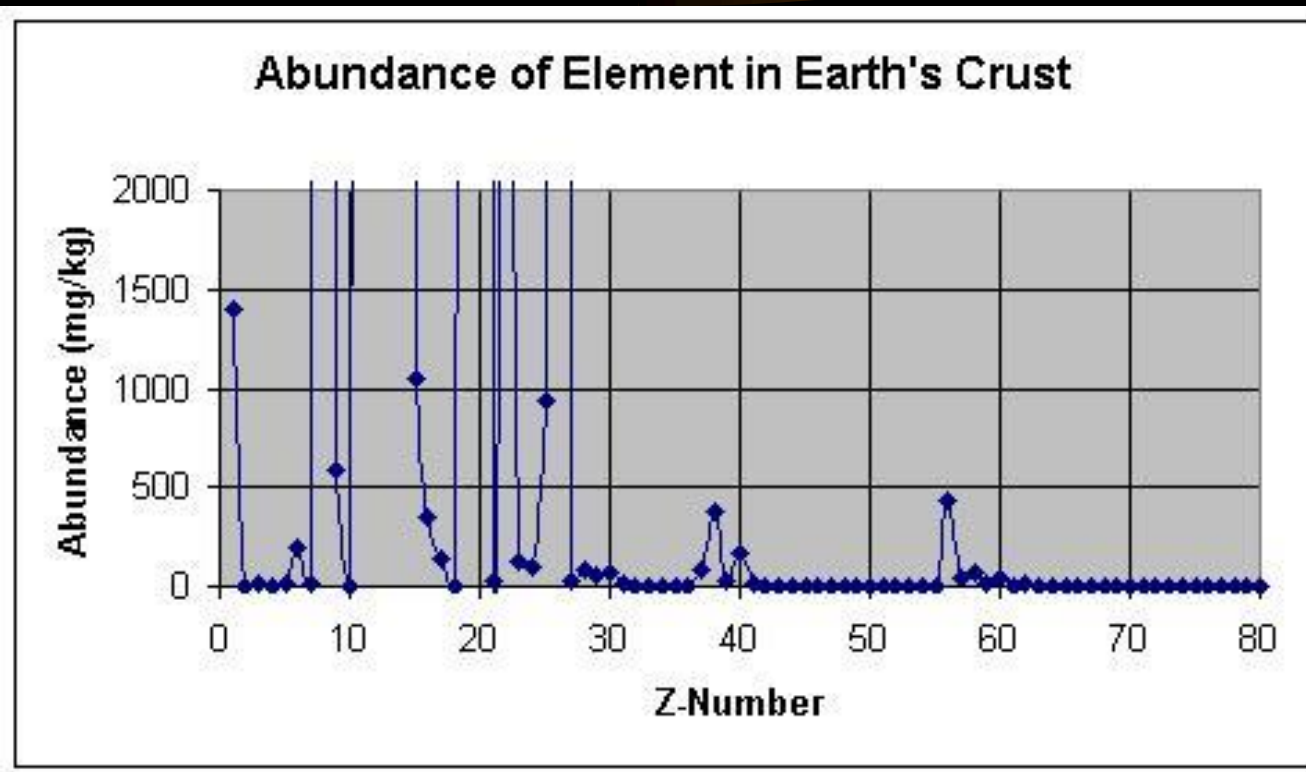
Period	Main-Group Elements s Subshell fills										Main-Group Elements p Subshell fills											
	1 IA	2 IIA	Transition Metals d Subshell fills								13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA						
1	1 H 1s <sup>1</sup>																2 He 1s <sup>2</sup>					
2	3 Li 2s <sup>1</sup>	4 Be 2s <sup>2</sup>															5 B 2s <sup>2</sup> 2p <sup>1</sup>	6 C 2s <sup>2</sup> 2p <sup>2</sup>	7 N 2s <sup>2</sup> 2p <sup>3</sup>	8 O 2s <sup>2</sup> 2p <sup>4</sup>	9 F 2s <sup>2</sup> 2p <sup>5</sup>	10 Ne 2s <sup>2</sup> 2p <sup>6</sup>
3	11 Na 3s <sup>1</sup>	12 Mg 3s <sup>2</sup>															13 Al 3s <sup>2</sup> 3p <sup>1</sup>	14 Si 3s <sup>2</sup> 3p <sup>2</sup>	15 P 3s <sup>2</sup> 3p <sup>3</sup>	16 S 3s <sup>2</sup> 3p <sup>4</sup>	17 Cl 3s <sup>2</sup> 3p <sup>5</sup>	18 Ar 3s <sup>2</sup> 3p <sup>6</sup>
4	19 K 4s <sup>1</sup>	20 Ca 4s <sup>2</sup>	21 Sc 3d <sup>1</sup> 4s <sup>2</sup>	22 Ti 3d <sup>2</sup> 4s <sup>2</sup>	23 V 3d <sup>3</sup> 4s <sup>2</sup>	24 Cr 3d <sup>5</sup> 4s <sup>1</sup>	25 Mn 3d <sup>5</sup> 4s <sup>2</sup>	26 Fe 3d <sup>6</sup> 4s <sup>2</sup>	27 Co 3d <sup>7</sup> 4s <sup>2</sup>	28 Ni 3d <sup>8</sup> 4s <sup>2</sup>	29 Cu 3d <sup>10</sup> 4s <sup>1</sup>	30 Zn 3d <sup>10</sup> 4s <sup>2</sup>	31 Ga 4s <sup>2</sup> 4p <sup>1</sup>	32 Ge 4s <sup>2</sup> 4p <sup>2</sup>	33 As 4s <sup>2</sup> 4p <sup>3</sup>	34 Se 4s <sup>2</sup> 4p <sup>4</sup>	35 Br 4s <sup>2</sup> 4p <sup>5</sup>	36 Kr 4s <sup>2</sup> 4p <sup>6</sup>				
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7	87 Fr 7s <sup>1</sup>	88 Ra 7s <sup>2</sup>	89 Ac**	104 Db 6d <sup>3</sup> 7s <sup>2</sup>	105 Jt 6d <sup>4</sup> 7s <sup>2</sup>	106 Rf 6d <sup>5</sup> 7s <sup>2</sup>	107 Bh 6d <sup>6</sup> 7s <sup>2</sup>	108 Hn 6d <sup>7</sup> 7s <sup>2</sup>	109 Mt 6d <sup>8</sup> 7s <sup>2</sup>													
			*Lanthanides																			
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			58 Ce 4f <sup>1</sup> 5d <sup>1</sup> 6s <sup>2</sup>	59 Pr 4f <sup>3</sup> 6s <sup>2</sup>	60 Nd 4f <sup>4</sup> 6s <sup>2</sup>	61 Pm 4f <sup>5</sup> 6s <sup>2</sup>	62 Sm 4f <sup>6</sup> 6s <sup>2</sup>	63 Eu 4f <sup>7</sup> 6s <sup>2</sup>	64 Gd 4f <sup>7</sup> 5d <sup>1</sup> 6s <sup>2</sup>	65 Tb 4f <sup>9</sup> 6s <sup>2</sup>	66 Dy 4f <sup>10</sup> 6s <sup>2</sup>	67 Ho 4f <sup>11</sup> 6s <sup>2</sup>	68 Er 4f <sup>12</sup> 6s <sup>2</sup>	69 Tm 4f <sup>13</sup> 6s <sup>2</sup>	70 Yb 4f <sup>14</sup> 6s <sup>2</sup>	71 Lu 4f <sup>14</sup> 5d <sup>1</sup> 6s <sup>2</sup>						
			90 Th 6d <sup>2</sup> 7s <sup>2</sup>	91 Pa 5f <sup>2</sup> 6d <sup>1</sup> 7s <sup>2</sup>	92 U 5f <sup>3</sup> 6d <sup>1</sup> 7s <sup>2</sup>	93 Np 5f <sup>4</sup> 6d <sup>1</sup> 7s <sup>2</sup>	94 Pu 5f <sup>6</sup> 7s <sup>2</sup>	95 Am 5f <sup>7</sup> 7s <sup>2</sup>	96 Cm 5f <sup>7</sup> 6d <sup>1</sup> 7s <sup>2</sup>	97 Bk 5f <sup>9</sup> 7s <sup>2</sup>	98 Cf 5f <sup>10</sup> 7s <sup>2</sup>	99 Es 5f <sup>11</sup> 7s <sup>2</sup>	100 Fm 5f <sup>12</sup> 7s <sup>2</sup>	101 Md 5f <sup>13</sup> 7s <sup>2</sup>	102 No 5f <sup>14</sup> 7s <sup>2</sup>	103 Lr 5f <sup>14</sup> 6d <sup>1</sup> 7s <sup>2</sup>						




# *The Numerical Basis vs. Periodic Table*

Notice how the abundances\* of Elements in the Earth Crust drop off after  $Z=30$ :

Data From CRC Handbook of Chemistry and Physics, 2001 ed.

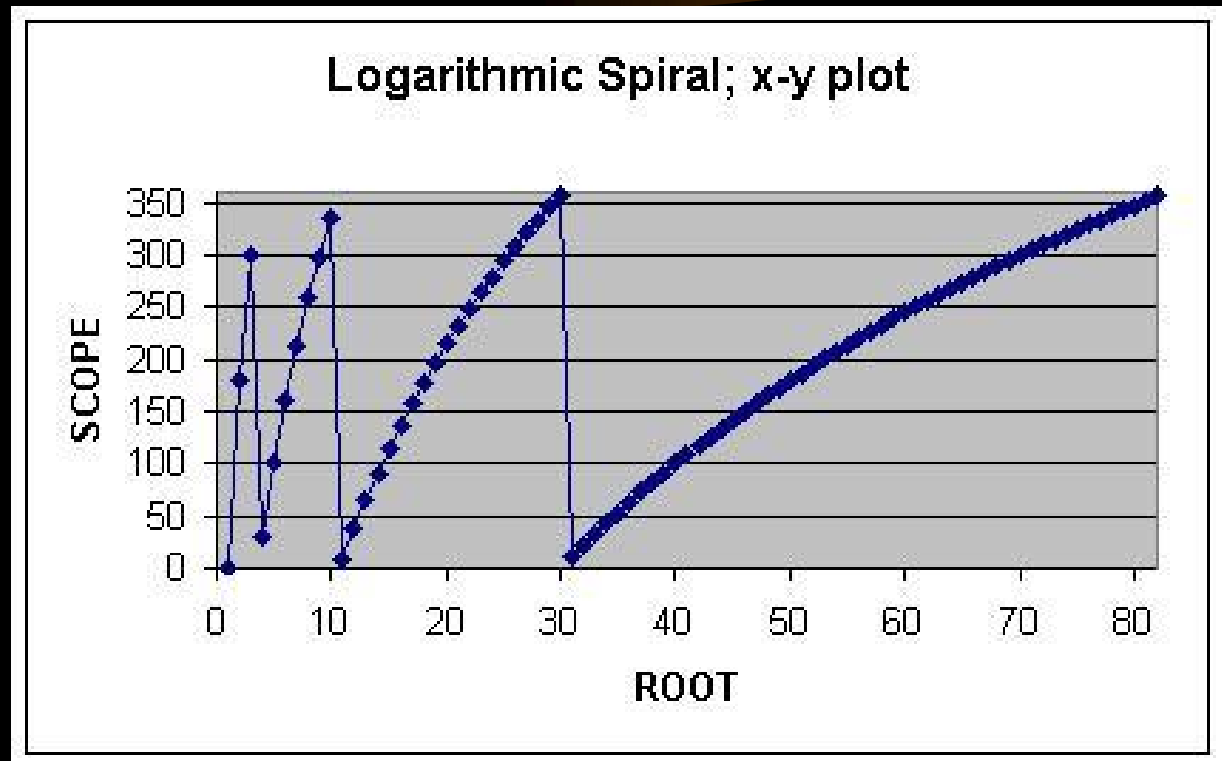


## *The Numerical Basis vs. Periodic Table*

- 
- É On the next slide, I will re-graph my Creation function. The spiral graph is a òradial plotö, i.e. one with an axis on the radius.
  - É I will now plot the ROOT numbers on the horizontal axis, and the SCOPE angles on the vertical axis.

# *The Numerical Basis vs. Periodic Table*

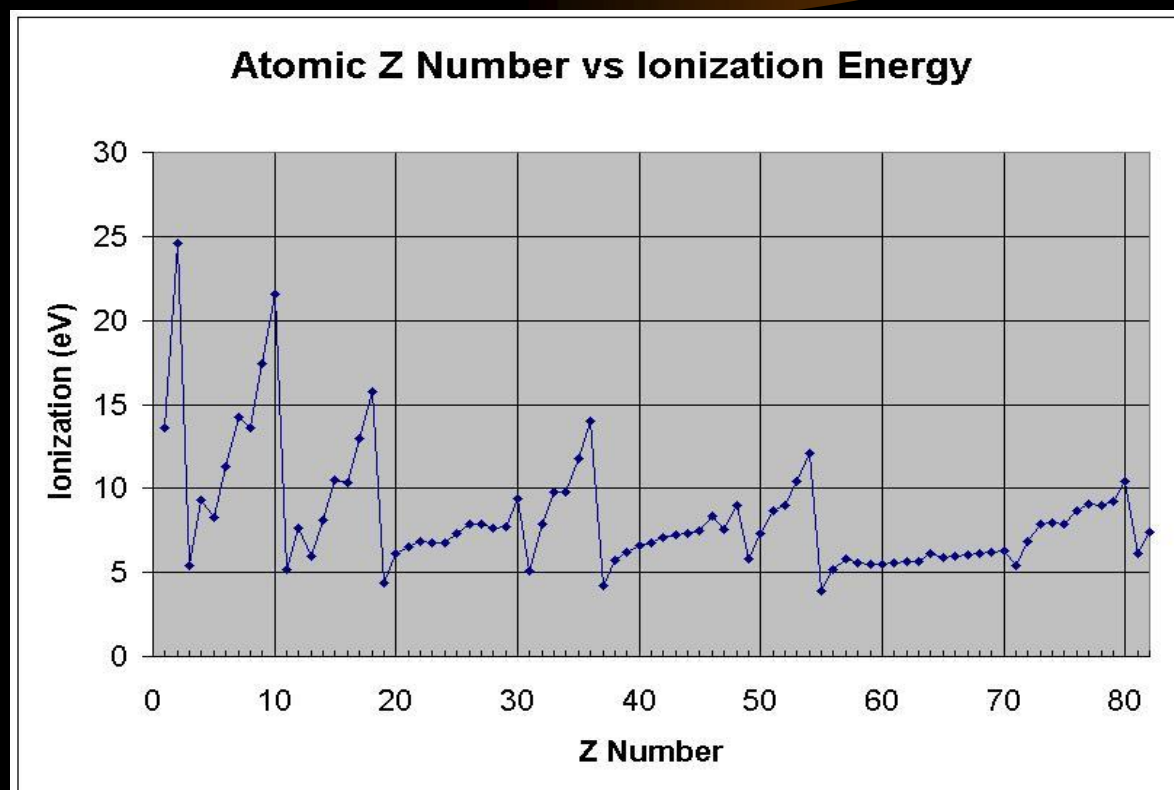
- É This is the Creation Function graphed on an x-y plot.
- É This shows Day 1 to Day 4.
- É NIGHTS begin at  $\text{SCOPE} = 0^\circ$
- É DAYS end when  $\text{SCOPE} = 360^\circ$  is passed through.



# *The Numerical Basis vs. Periodic Table*

- É This is the actual Pattern of the Periodic Table\*
- É This shows  $Z=1$  to  $Z=82$  (Lead).
- É Vertical axis shows stability-less likely to be changed.
- É More exposure to nucleus (i.e. not eclipsed) means more stable atom.

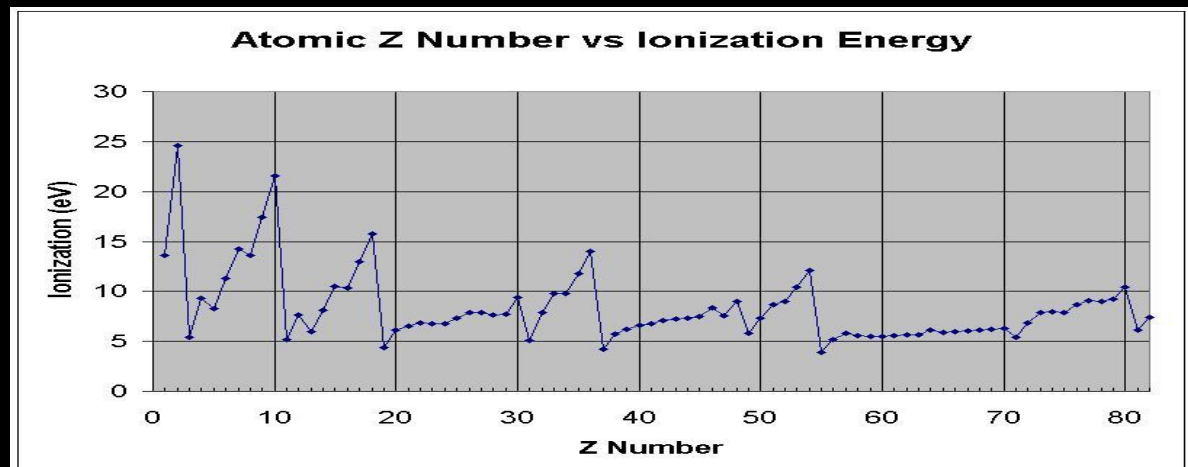
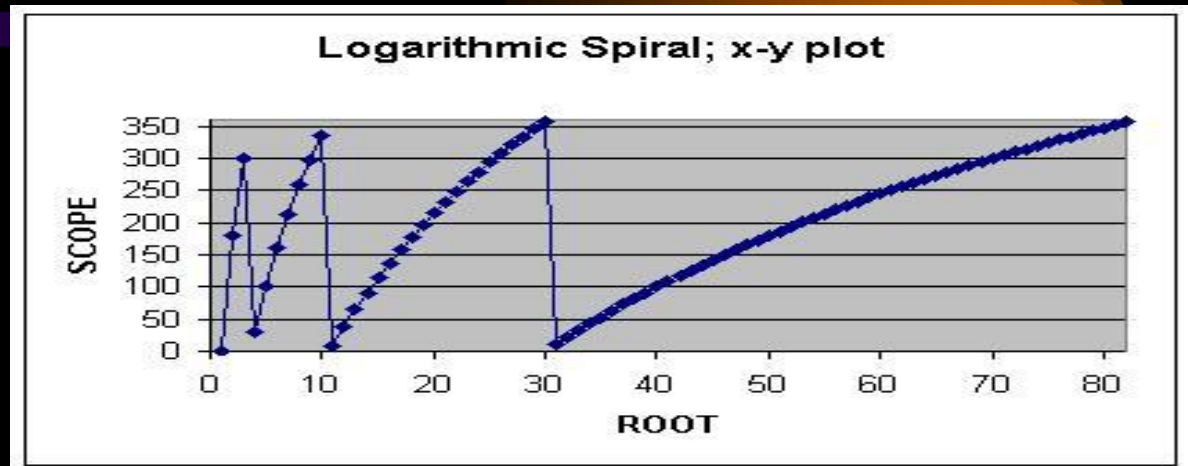
É Data from CRC Handbook of Chemistry and Physics, 2001



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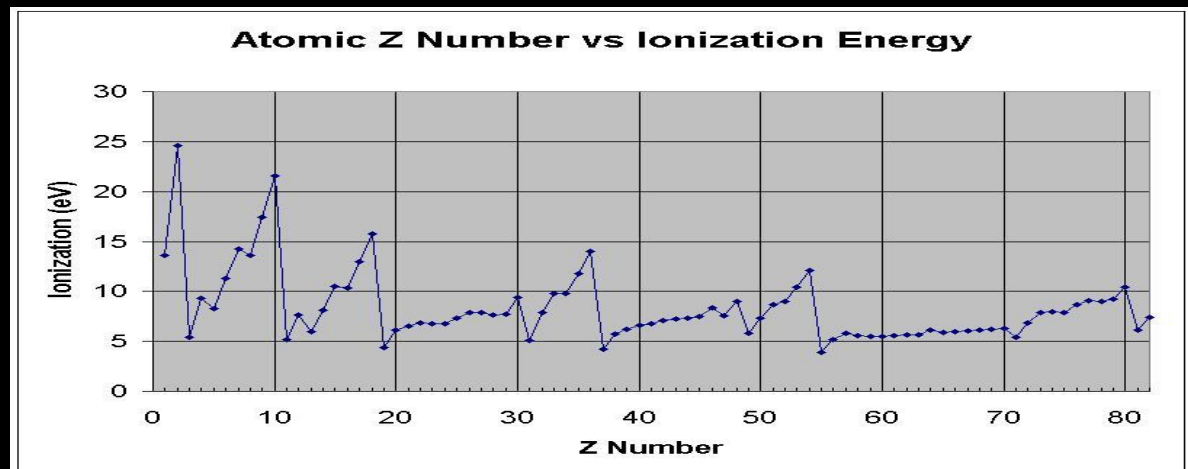
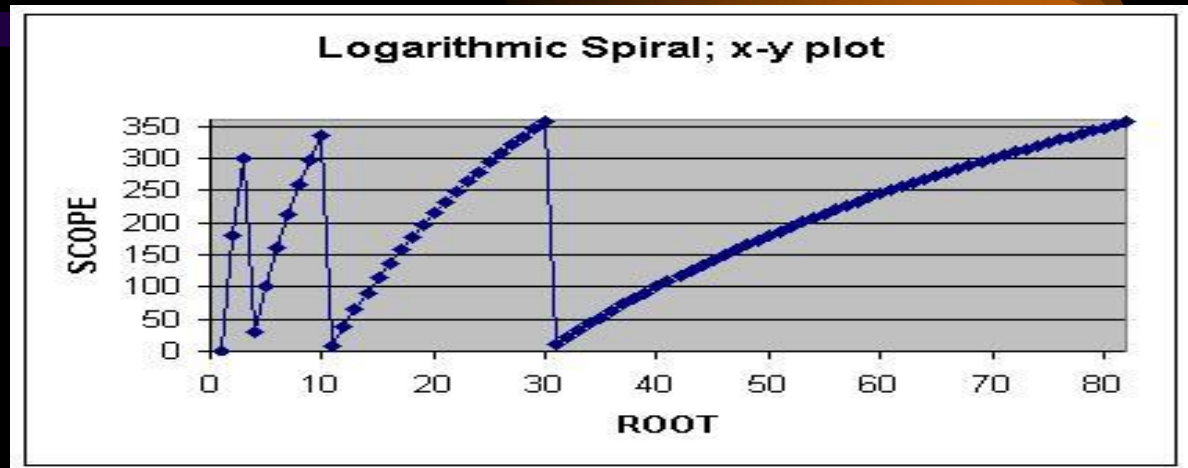
# *The Numerical Basis vs. Periodic Table*

- É Compare these two graphs:
- É Notice they are similar until  $\text{ROOT} = 18$ .
- É Notice bottom chart has  $Z = 1$  and  $Z = 2$  only as Day 1
- É Notice that the 18 points on bottom chart are "scrunched" compared to those on upper chart.
- É What does it all mean?



# The Numerical Basis vs. Periodic Table

- É Remember this upper chart is the first 4 Days of Original "Good" Creation:
- É Smooth days and nights
- É This lower chart is published scientific measurements made recently of our "Cursed" and "Flood-Wracked" world.
- É Scrunched, much of it FALLEN below Day level.



# *The Creation is Fallen, Fallen*



## É Please Read Scriptures:

- ó Dan. 2:31-35      õGold, Silver, Brass, Iron, Clayö
- ó Dan. 3:1      õH = 60, W = 6ö
- ó Job 38:4      õMEASURES, LINE, STONEö
- ó Is. 24:1      õEarth is upside down, scatteredö
- ó Is. 24:19      õEarth is fallen, brokenö

# *The Creation is Fallen, Fallen*

## É Daniel Image accounting

ó Gold,	Z = 79	Head
ó Silver,	Z = 47	Breast, Arms
ó Brass,	Z = 29 (Cu) ,30 (Zn)	Belly, Thighs
ó Iron,	Z = 26	Legs (Knee and down)
ó Clay,	Z = 13 (Al) ,14 (Si) to 19	Feet

É Height is head to foot, 79 to 19 = 60 cubits.

É Width is foot size, 19 to 13 = 6 cubits.

É Please read Webster's Definition of "Clay".

É Probability of 8 matches (in order) out of 100 possibles is 1:7,503,063,898,176,000. *Lotto is only 5 matches (any order) out of 46 possibles.*



# *The Creation is Fallen, Fallen*

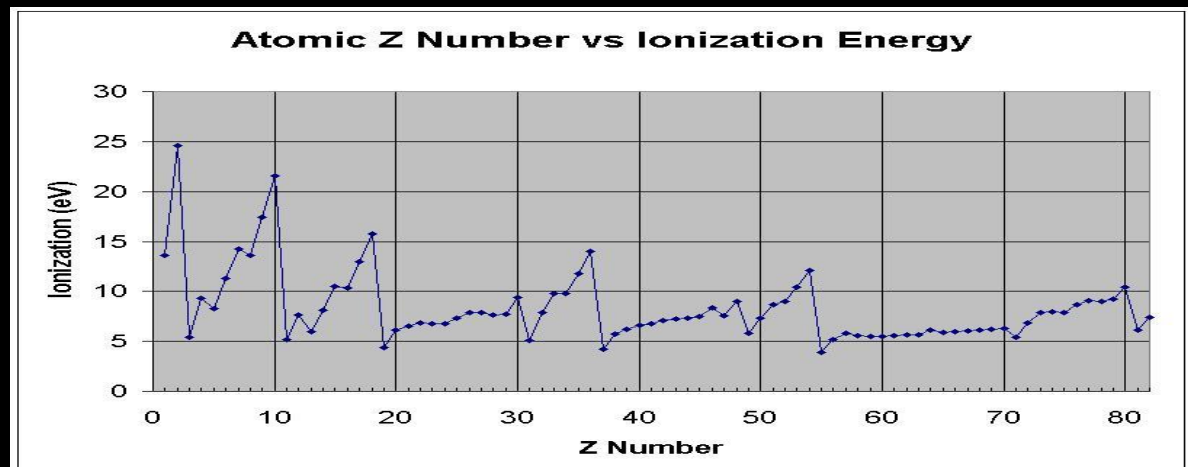
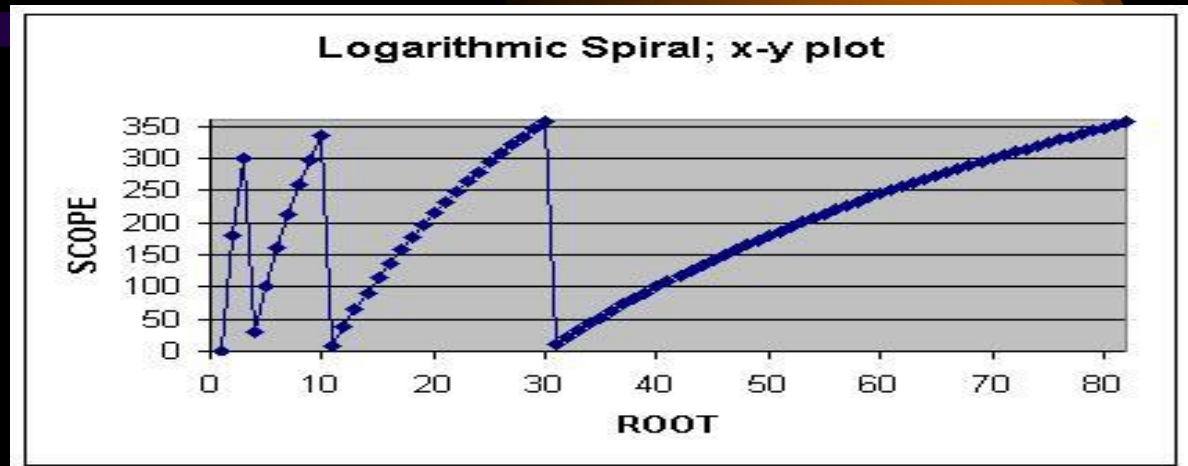


## É Daniel Image Story

- ó STONE hits FOOT of Iron/Clay
- ó Since shin is iron, and foot is attached to shin at Z=19, this is where STONE hits.
- ó Image FALLS.
- ó Image shatters.
- ó STONE fills EARTH.
- ó Earth is turned upside down (SCOPE rotate 180°)

# *Creation is Fallen, Earth is upside down*

- É The ROOT = 1 STONE is thrown.
- É This creates a chain-reaction, such that ultimately the structure of Creation breaks between ROOT = 18 and ROOT = 19.
- É The Creation after ROOT = 19 FALLS, SCOPE rotated 180°
- É Notice the breaks at foot (19), thighs (30), breast (47), head (79).



## *The fallen position of Man (males)*

É Daniel Image is a MALE.

ó Gold,	Z = 79	Head
ó Silver,	Z = 47	Breast, Arms
ó Brass,	Z = 29 (Cu) ,30 (Zn)	Belly, Thighs
ó Iron,	Z = 26	Legs (Knee and down)
ó Clay,	Z = 13 (Al) ,14 (Si) to 19	Feet

É Height is head to foot,  $79$  to  $19 = 60$  cubits.

É Width is foot size,  $19$  to  $13 = 6$  cubits.

É We have established that  $ROOT=19$  to  $ROOT=30$  is the Earth plane.

É Therefore, males stand (feet at  $19$ ) on the Earth plane.

# *The Fallen Position of Man*



É Please Read Scriptures:

ó Rom. 12:1

õLIVING SACRIFICEö

ó 1 Cor. 6:19,20

õBODY IS TEMPLEö

É We will now place the man in the Tabernacle.

# *The Fallen Position of Man*



- É Sacrifice killed at Gate, put on brazen altar
  - ó Where stone hit foot, ðkilled Earthö, ROOT=19
  - ó Man is standing on altar (living sacrifice) at ROOT=19.  
Burning bush not consumed.
- É The altar and laver are made of brass: ROOT=29,30
  - ó Earth plane is ROOT=19 to ROOT=30
  - ó Earth plane includes the Moon.

# Solar System vs. Periodic Table

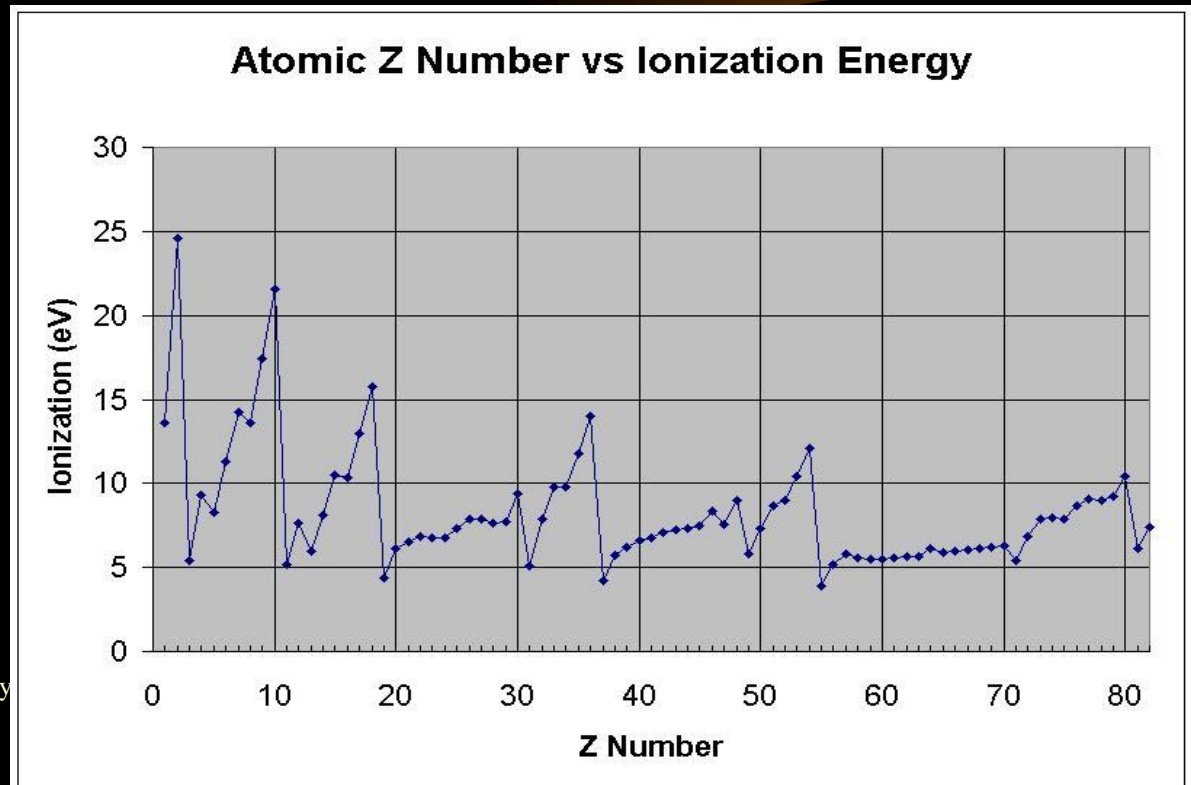
É How long does it take light to get to planets from the Sun?\*

- ó Mercury 193 (200) Sec.
- ó Venus 361 (360) Sec.
- ó Earth 499 (500) Sec.
- ó Mars 760 (760) Sec.

É Divide each (rounded time) by 20 and compare to the graph!

É Periodic Table Data from CRC Handbook of Chemistry and Physics, 2001

É Solar System Data from *An Introduction to the Solar System*, Cambridge Press, 2003



# *The Prophetic Power of the Pattern*



- É This Creation work is unique because it makes predictions based upon an established pattern.
- É Remember how I stated that Daniel image was òmaleö?
- É I will now use the Tabernacle Pattern, Scripture, and Science to place the post-Fall òfemaleö relative to the Fallen male.

# *The Post-Fall Position of Women*



## É Please Read Scriptures:

- ó Rev. 12:1                   ōSTANDING ON MOONö
- ó Micah 4:13               ōBRASS HOOVESö
- ó Gen. 2:22                 ōRIB FROM MANö
- ó 1 Cor. 11:12           ōMAN -> WOMAN, WOMAN -> MAN

É We will now place the woman in relation to the Tabernacle and Temple in Jerusalem.

É We will also confirm with a *very brief* female anatomy review.

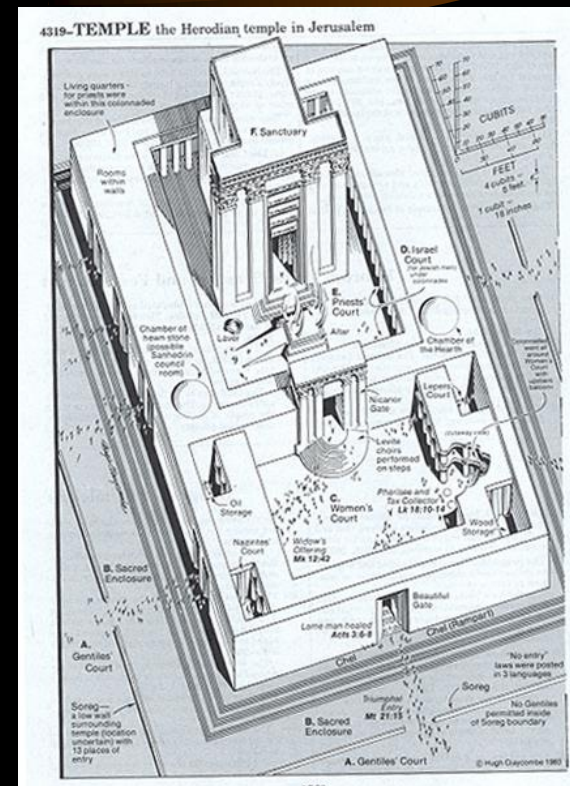


# The position of Woman

É The Temple at Jerusalem has a Tabernacle Pattern inside

É í And it has an additional area in front of the Tabernacle Gate

É This area is the Women's Court.



# The position of Woman

Also in the Woman  
Court:

ÉOil Storage

ÉOlive Oil

ÉOleic Acid:  $C_{18}H_{34}O$

ÉROOT=1 Hydrogen

ÉROOT=6 Carbon

ÉROOT=8 Oxygen

ÉSpirit of Yahshua

ÉWood Storage

ÉOrganics

ÉCellulose polymer

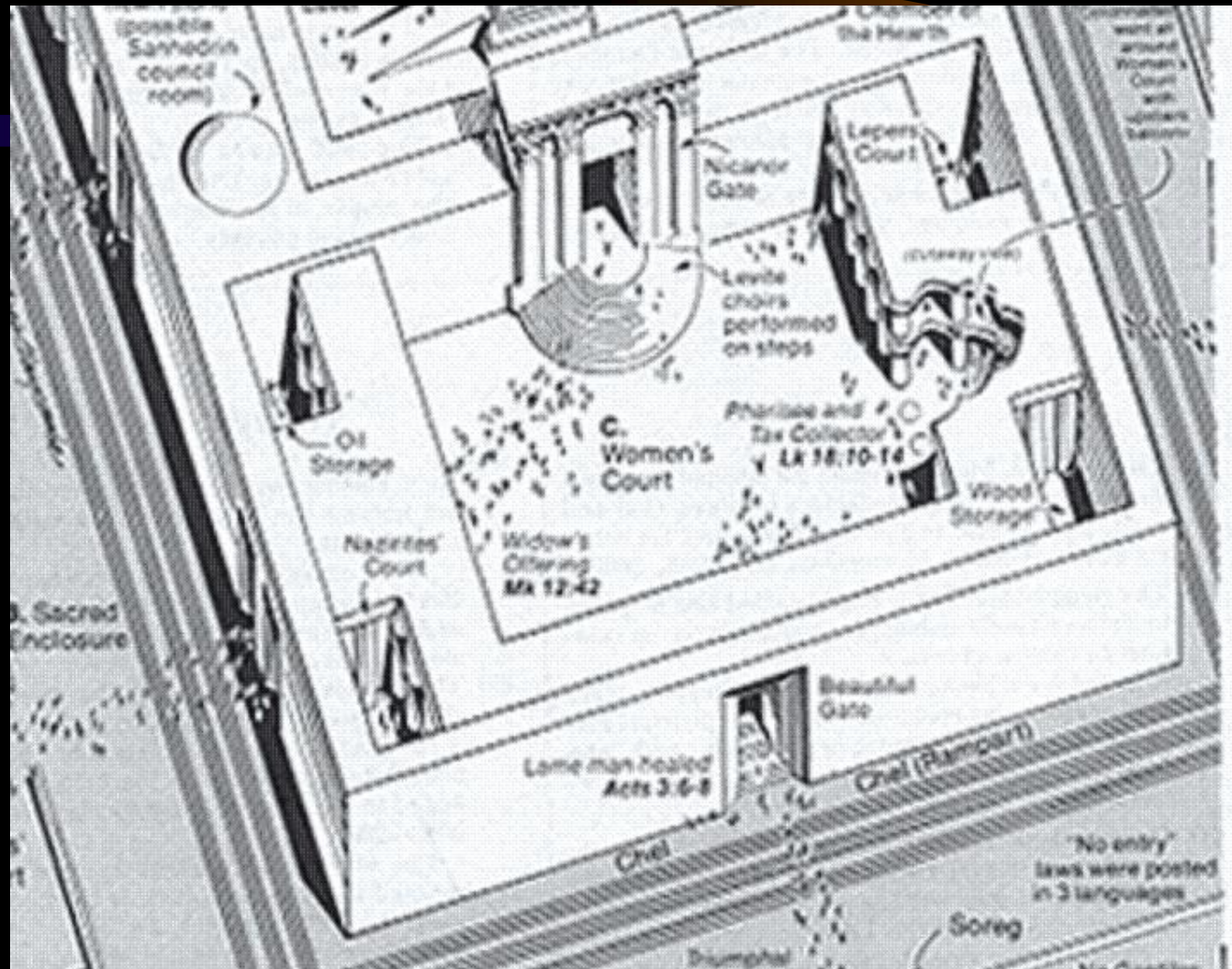
É $C_6H_{12}O_6$ ]n

ÉAll <ROOT=18

ÉCross material

ÉNoah's Ark material

ÉEtc.



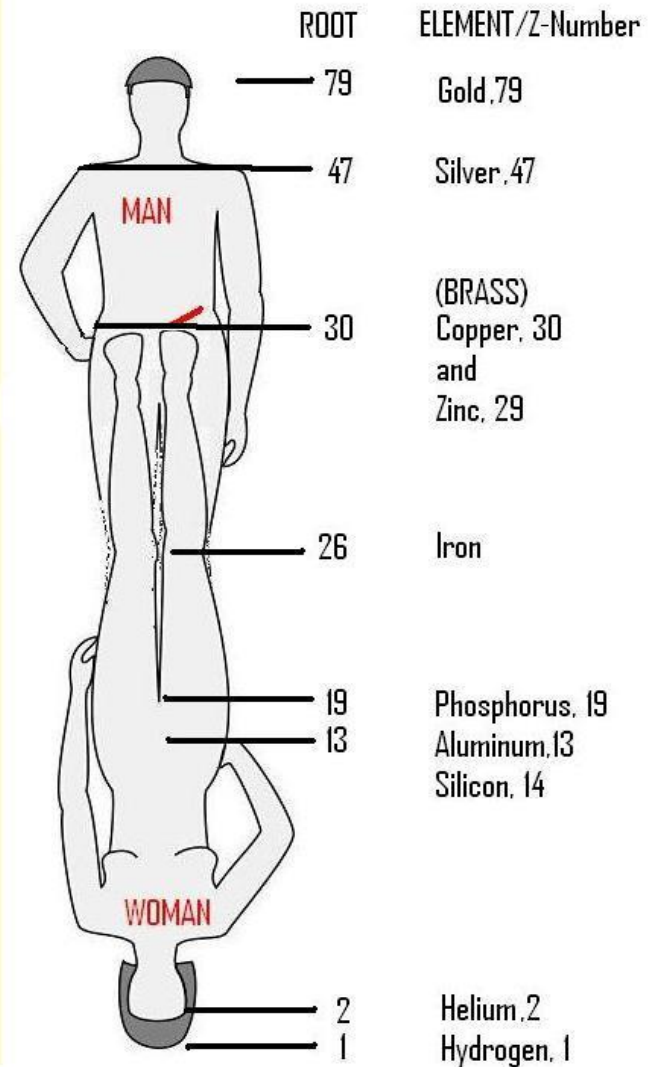
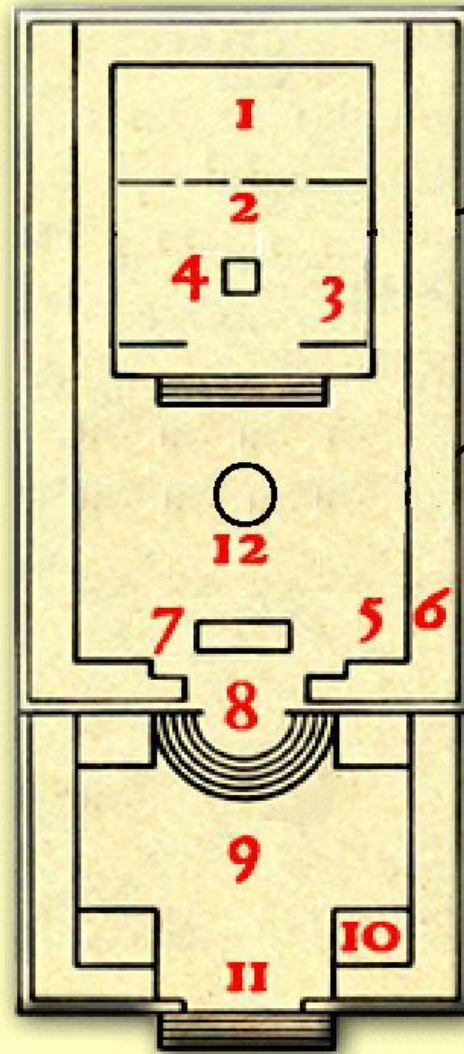
# *The Post-Fall Position of Women*

## É Brief anatomy review: Woman's Cycle

- ó Resembles Moon Cycle, 28 days.
- ó Woman's position: ROOT=2 (Sun) to ROOT=30 (Moon)
- ó Women have approximately 30 years of reproductive maturity.
- ó Ovulation occurs for different women Day 15 to Day 19 of cycle, consistent with Tabernacle Gate (death) at ROOT=19.
- ó Position of oil storage (Holy Spirit) correct to fertilize Virgin Mary before the Gate, before ROOT=19.
- ó Blood flow into Gate, into bottom of Brazen Altar.
- ó Cleansing action by ROOT=30, position of brazen laver (water)
- ó Samaritan woman leaves water vessel at well with Yahshua, ROOT=30
- ó It goes on and oní ..

Relative Positions of Post-Fall man and Woman in the Creation

1. Holy of Holies
2. Veil (Curtain)
3. Holy Place
4. Altar of Incense
5. Court of Priests
6. Court of Israel
7. Altar of Sacrifice
8. Nicanor Gate
9. Court of Women
10. Wood
11. Gate Beautiful
12. Laver



## *What's Next?*

É This is ñentry-levelö information to understand:

- ó The Flood
- ó The Fall
- ó The Revelation
- ó Other Mysteries/Miracles
- ó Paul understood all this.
- ó Peter understood all this.
- ó It is our turn to understand and ACT!