

Problem C2: Given equation of state as $P(v - b) = RT$, show that the relation between P and v during an adiabatic process is $P(v - b)^\gamma = \text{constant}$, where γ is the ratio of C_p and C_v .

$$\begin{aligned} dU &= \delta Q - P dV \\ \text{at } dV &= 0 \\ dU &= \delta Q = C_v dT \end{aligned} \tag{1}$$

At $P = \text{const}$

$$C_v dT = \delta Q = \frac{RT}{v - b} dV \tag{2}$$

$$P dV = R dT$$

$$\Rightarrow \frac{RT}{v - b} dV = R dT \tag{3}$$

Inserting Eq. (3) into Eq. (2)

$$C_V dT = dQ - PdV$$

$$dQ = (C_V + R) dT$$

$\approx C_P$

• Adiabatic $dQ = 0$

$$dU = -PdV$$

$$C_V dT = - \frac{RT}{V-b} dV$$

$$\frac{dT}{T} = - \frac{R}{C_V} \frac{dV}{V-b}$$

$$\ln\left(\frac{T}{T_0}\right) = - \frac{R}{C_V} \ln\left(\frac{V-b}{V_0-b}\right)$$

$$\frac{T}{T_0} = \left[\frac{V-b}{V_0-b} \right]^{-\frac{R}{C_V}}$$

$$T (V-b)^{\frac{R}{\alpha}} = T_0 (V_0-b)^{\frac{R}{\alpha}} = \text{const.}$$

$$\left(\frac{P(V-b)}{R} \right) V^{\frac{R}{\alpha}} = \text{const.}$$

$$P (V-b)^{\frac{R}{\alpha} + 1} = \text{const.}$$

$$P (V-b)^{\frac{R+C_v}{C_v}} = \text{const.}$$

$\underbrace{\hspace{1cm}}_{\gamma = C_p/C_v}$

$$P (V-b)^{\gamma} = \text{const.}$$