

Hazard Communication

What Is Hazard Communication?

The purpose of the OSHA *Hazard Communication* standard is to let you know the hazards of the chemicals in your workplace.

Chemical importers, manufacturers, and reformulators must automatically send chemical hazard information to downstream employers, like yours. They must use labels and Safety Data Sheets (SDSs).

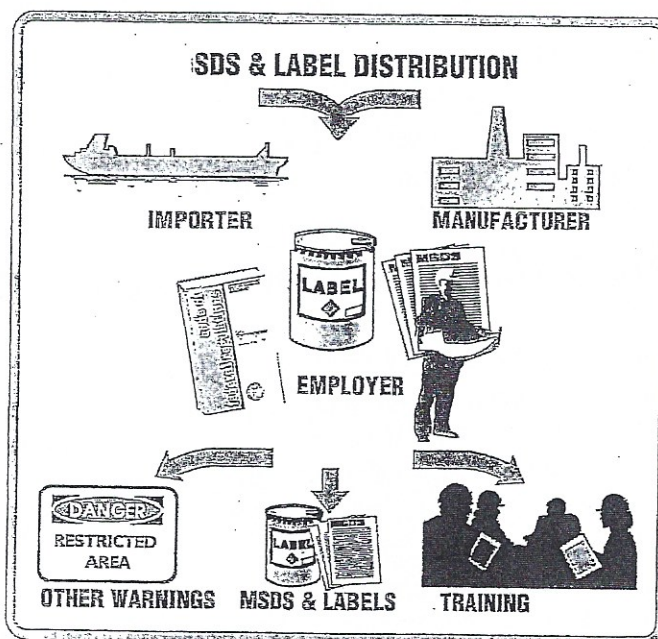
When employers receive the labels and SDSs, they must use these to inform and train workers and to design and run worker protection programs.

The HazCom standard requires your employer to:

- use labeled products,
- get and provide SDSs on all products,
- provide training and information for workers, and
- have a written HazCom program.

A HazCom program requires the employer to provide the information to workers in these ways:

- on container labels,
- on other posters, placards, or warnings the employer may use,
- in training provided to workers, and
- through SDSs available to workers and their representatives.



In this course, we focus on the parts of the Hazard Communication regulation that directly affect your health and safety in the workplace.

Hazard Communication

What Products Require SDSs?

Your employer must have an SDS for every hazardous product used or stored on each job. This includes items such as pipe if it's welded, cut, or ground. It includes treated lumber. It

does not include 'articles' such as a piece of wood, metal, or ceiling tiles – unless the article will be modified. Modification includes cutting, burning, welding, and pulverizing.

How Do We Get SDSs?

OSHA gives you the right to request from your employer an SDS for any chemical with which you work. Your union representative also has the right to request SDSs for you.

OSHA requires your employer to provide the requested SDSs to you or your union representative.

What Information and Training Does HazCom Require?

Your employer must *train* and *inform* you of the hazards of chemicals you use in your job. The employer's training must provide:

- *general training* about HazCom, health and safety hazards, and control methods, and
- *specific training* for the hazards in your workplace.

How Should Products Be Labeled?

All manufacturers, importers, and distributors of hazardous chemicals **MUST label** products so that you can recognize the hazards of the product. If the product is transferred to

another container, it **MUST** be labeled also, unless the amount is for the use of one person during one shift.

Why Do Construction Workers Need HazCom?

A recent study shows that *18% of on-the-job deaths* in construction are due to *exposure to harmful substances*. This does not include the large number of construction workers who die later after getting chronic diseases from exposure to harmful substances.

Before HazCom, manufacturers did not have to tell workers or their doctors about harmful chemicals in products.

Only since the mid-1980s have you had the *right to know*—thanks mainly to the efforts of unions.

This course helps you learn to get and understand chemical information, so you can protect yourself and your family.

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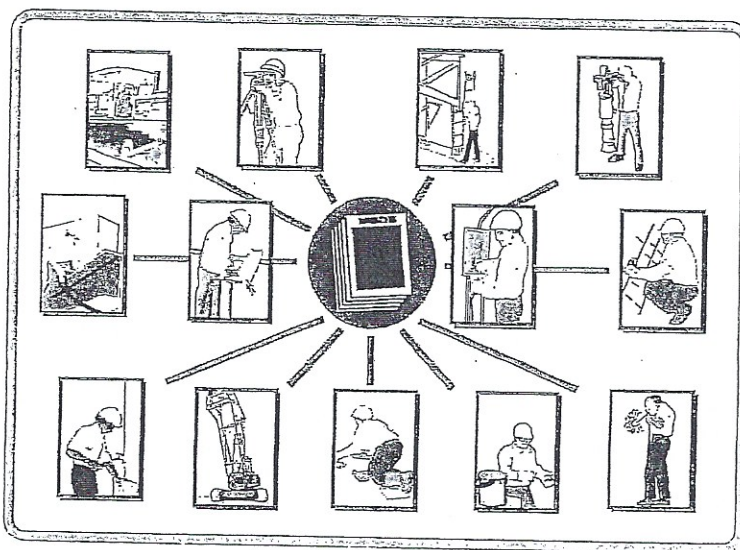
This does not include workers who die off-the-job due to acute or chronic conditions caused by harmful exposures.

What About Multi-Employer Sites?

All employers on a *multi-employer site* must provide information to each other.

Each employer's written HazCom program must include methods to:

- provide other employer(s) with a copy of SDSs or to place them at a central location,
- inform other employer(s) of any precautionary measures to take during normal operating conditions and in foreseeable emergencies, and
- inform the other employer(s) of his or her labeling system.



What Is an 'SDS?

The *'SDS* is a detailed technical bulletin. It is the primary source of information about hazardous products used on your job. Almost every product in your workplace has an *'SDS*—solvents, concrete, paint, adhesives, grout, degreaser, diesel fuel, and more.

What Information Is in an 'SDS?

The *'SDS* tells:

- what company makes the product,
- what the physical and health hazards are,
- how to recognize when you are overexposed, and
- how to protect yourself from the product.

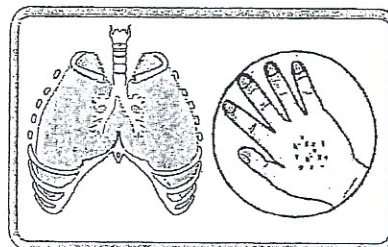
| | |
|-----------------------|--|
| Symptoms of Exposure | |
| _____ | |
| _____ | |
| _____ | |
| Safe Handling and Use | |
| _____ | |
| _____ | |
| _____ | |
| Protective Measures | |
| _____ | |
| _____ | |
| _____ | |
| Safety Data Sheet | |
| Manufacturer's Name | |
| _____ | |
| Product Name | |
| _____ | |
| Physical Hazards | |
| _____ | |
| _____ | |
| Health Hazards | |
| _____ | |
| _____ | |

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What Is Health Hazard Information?

The health hazard information tells you about the harmful effects of a product. Health hazard information includes:

- routes of entry
- types of health hazards
- target organs
- carcinogenicity
- signs and symptoms of exposure
- aggravations to medical conditions
- emergency or first aid procedures



What Are the Main Routes of Entry?

For a chemical to harm you, it must get onto or into your body.

The SDS lists the ways the chemical gets into the body:

- breathe it.
- swallow it.
- absorb it through your skin.

You can't smell or taste all chemicals. Not all chemicals are irritating. Even when chemicals have *warning properties*, such as an odor, you stop smelling them after a while. This doesn't mean they won't harm you. You just don't notice them anymore.

What happens when you breathe, absorb, or swallow a chemical? The chemical may get

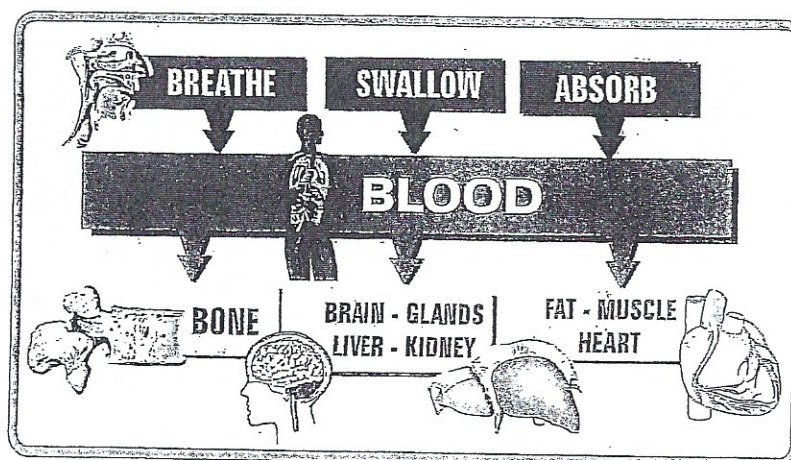
into your bloodstream. From there, it can circulate throughout your body. It can stay in your body for a long time or a short time. It may cause serious health effects.

It may irritate or injure your nose, mouth, throat, or lungs. It may damage:

- your bone
- your brain
- your glands
- your liver
- your kidneys
- your fat
- your muscle
- your heart

These are *target organs*. If a chemical is known to harm certain organs, they are named in the health hazard section. Often, the

SDS uses the organ's name: *heart*. Other times it describes the chemical: *cardiotoxic*.



Hazard Communication

What Are the Two Types of Health Hazards?

SDSs list health hazards as either:

- acute
- chronic

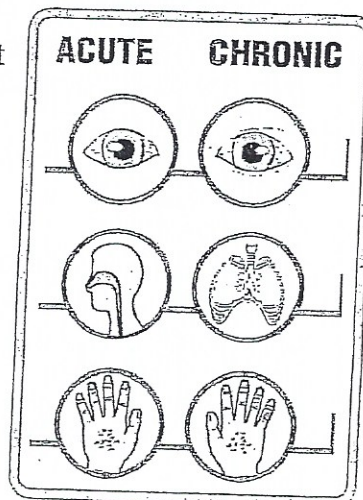
An *acute effect* is immediate. It often happens from a single exposure. Acute does not mean mild. Acute effects of chemical overexposure include irritation of skin, nose, throat, or eyes, nausea or dizziness, heart failure, a coma, or death.

A *chronic effect* is long-term and persistent. It does not always mean a critical condition, though many chronic health problems are.

An acute effect on the eyes might be irritation. A chronic effect could be scarred cornea.

An acute respiratory effect might be irritation. A chronic effect could be bronchitis.

An acute effect on skin might be irritation. A chronic effect could be dermatitis.



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What Is Carcinogenicity?

Carcinogenicity means cancer-causing. If the product contains a carcinogen, the SDS must say so. If any chemical in the product is

a suspected or confirmed animal or human carcinogen, ask your supervisor for a safer product. Treat the product as a carcinogen.

What Are the Signs and Symptoms of Exposure?

Many *acute, reversible* effects of exposure are listed on the SDS as signs and symptoms of exposure:

- dizziness
- skin irritation
- headaches
- cold sweats
- difficulty breathing
- nausea
- fatigue
- sneezing
- irritability
- eye, nose, or throat irritation

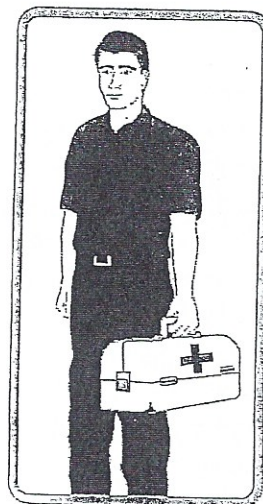
These signs and symptoms are your body's way of reacting to foreign substances. They are meant to alert you that you may be harmed. They are listed on the SDS to serve as

early warnings of overexposure.



What Medical Conditions Are Aggravated by Exposure?

Sometimes chemicals can make a certain *medical condition* worse. Many solvents aggravate asthma or bronchitis. One degreaser, methylene chloride, breaks down to carbon monoxide in the blood stream and may aggravate heart conditions. If the product aggravates any medical condition, the manufacturer must list it.



What Emergency or First Aid Measures Are Given?

The SDS will offer some recommendations for emergency or first aid procedures.

These recommendations are usually not sufficient for a treating physician.

What Are Hazardous Ingredients?

This part usually contains three types of information:

- the hazardous components,
- the exposure limits, and
- the percent of the chemical in the product.

Let's review each one:

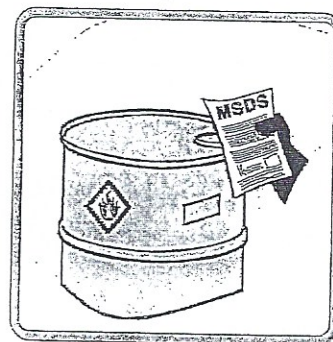
Hazardous Components. Lists the product's chemical names and synonyms and may list common names.

PEL or TLV. Gives the legal or recommended exposure limit for each hazardous chemical in the product.

The *PEL (Permissible Exposure Limit)* is the airborne concentration of a chemical to which workers may be *legally* exposed day after day for a lifetime. PELs are set by OSHA. PELs are

not necessarily 'safe' exposure levels. The TLV (Threshold Limit Value) is a *recommended* airborne concentration of a chemical set by a private organization of safety and health professionals.

The percent of the hazardous ingredient in the product is usually the percent by weight or volume. The higher the percent of an ingredient in a product the greater your potential exposure to it.



Hazard Communication

How Do We Measure PELs and TLVs?

Both PELs and TLVs are usually measured in *parts per million*.

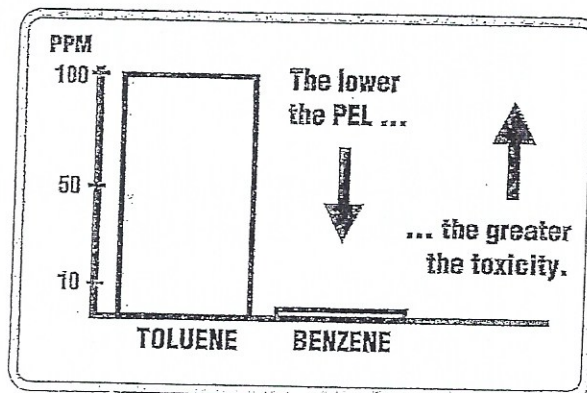
Parts per million (ppm) is a measure of small amounts of *gases or vapors* in air. As a fraction, a ppm is one-millionth. This is an extremely small amount of material. It doesn't take a lot of a chemical to make you sick.

A PEL of 1 ppm may be 1,000 times more toxic than a PEL of 1,000 ppm.

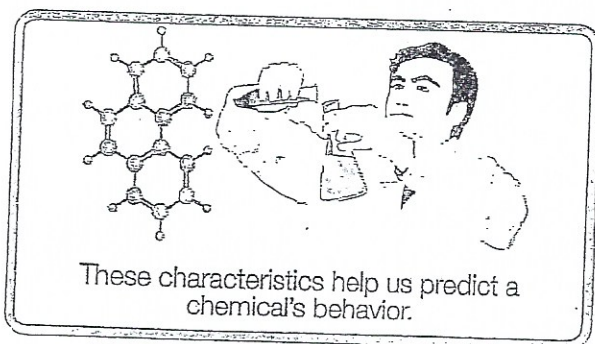
Use the PEL or TLV as a *relative comparison* of the hazard. Knowing the PEL can help you understand how important it is to keep the chemical out of your body.

Often, the *lower* the PEL, the *more dangerous* the chemical. PELs are like fish

hooks or wire—the smaller the number, the bigger the hazard. Chemicals with smaller PELs are more harmful because smaller amounts of the chemicals will cause harm.



What Are Physical and Chemical Characteristics?



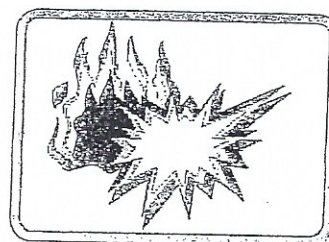
Like people, chemicals have characteristics that predict how they will behave. A chemical's characteristics can help us understand when, how, and why it becomes most hazardous.

These characteristics include:

- boiling point
- specific gravity
- vapor pressure
- melting point
- vapor density
- evaporation rate
- solubility in water
- appearance/odor

What Are Fire and Explosion Hazard Data?

Use the SDS to find out about the *fire and explosion hazards* of a product. Precautions for fire fighting and fire-related physical and health hazards are also listed here.



What Does 'Extinguishing Media' Mean?

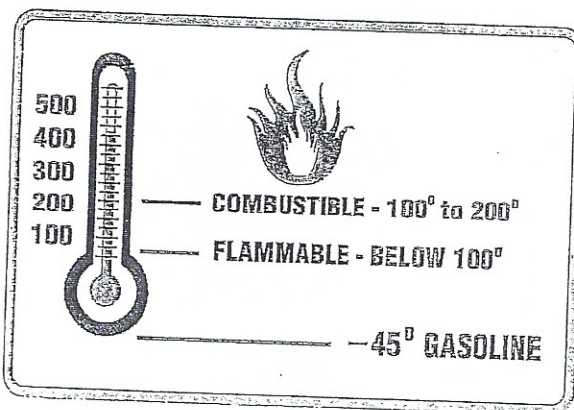
Extinguishing media is what will put the fire out. Typical extinguishing media are water fog, foam, alcohol foam, CO₂, and dry chemical.

Never make the mistake of assuming that water works for everything.

What Is Flash Point?

Flash point is the *temperature* at which a chemical releases enough vapors to ignite. Liquids evaporate and give off vapors. At the flash point temperature, a spark or other source of ignition will ignite the vapors. The *lower* the flash point of a chemical you use, the more likely that a heat source will cause it to burn or explode.

A product with a flash point under 100° is classified as *flammable*. A product with a flash point above 100° but below 200° is classified as *combustible*.



What Are Special Fire Fighting Procedures?

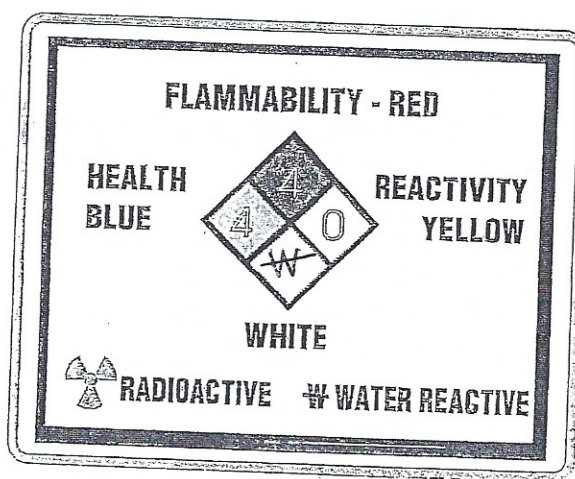
This tells the equipment and procedures for fighting a fire involving the chemical. Typical recommendations include methods for cooling containers of chemicals and personal protective equipment needed for fighting the fire. Generally, it is better to leave the fire fighting to the fire fighters.

What Is the NFPA Label?

The *National Fire Protection Association (NFPA) label* was developed to warn fire fighters about the hazards of chemicals in a fire. Today, the NFPA label appears on many product containers and on some SDSs.

The NFPA label is a diamond containing four squares of different colors. The squares contain a number from 0 to 4, with 0 meaning no hazard and 4 meaning a severe hazard. The internal squares are red, blue, yellow, and white.

- **RED** is fire,
- **BLUE** is health,
- **YELLOW** is reactivity.
- **WHITE** is reserved for special hazards such as 'use no water,' represented by **W**.

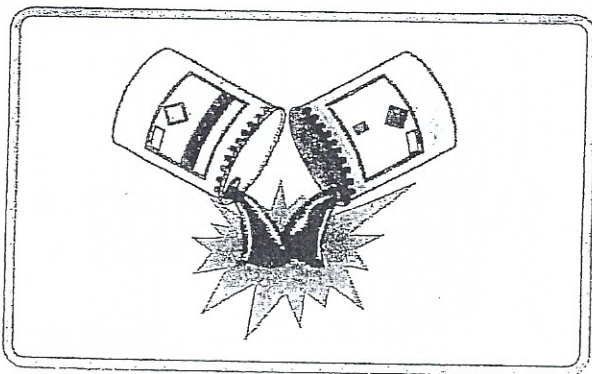


The NFPA does not cover chronic health effects or give the name of the chemical, the product, or the manufacturer, which is required by the OSHA regulations.

What are the hazards for the product of the NFPA label pictured above?

Hazard Communication

What Is Reactivity?



A chemical labeled *reactive* tends to undergo chemical change and release energy. These chemical changes may cause pressure build-up, temperature increase, or formation of toxic

or corrosive by-products. These chemical changes can be started by:

- heat,
- improper storage or handling, or
- direct contact with other incompatible chemicals.

Polymerization is one example of chemical reactivity. Have you noticed that the bond of a two-part glue gets hot as the glue sets? This is a controlled *polymerization*. But when a reactive undergoes polymerization, it can get out of control. The result is usually a fire or explosion. If the chemical is capable of *hazardous polymerization*, then it's noted here in this part of the SDS.

What Is Safe Handling and Use?

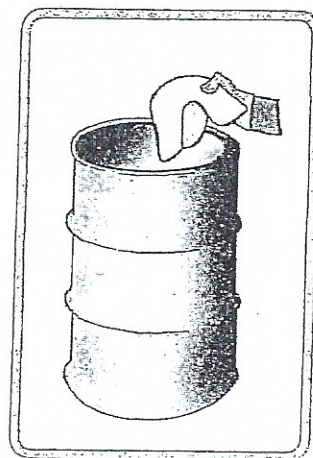
The SDS must give advice on:

- accidental spill or release
- disposal of materials
- recommended handling and storage precautions

What Are the Precautions for Spill or Release?

These are the procedures to use if the material is *released or spilled* in an uncontrolled or unplanned manner. These usually include such things as:

- avoid breathing gases and vapor
- avoid skin contact with liquid or solid
- remove sources of ignition



What Does Disposal Mean?

Disposal means how to get rid of the material or its waste when you are finished. Methods must always follow federal, state, and local regulations.

What Are Handling and Storage Requirements?

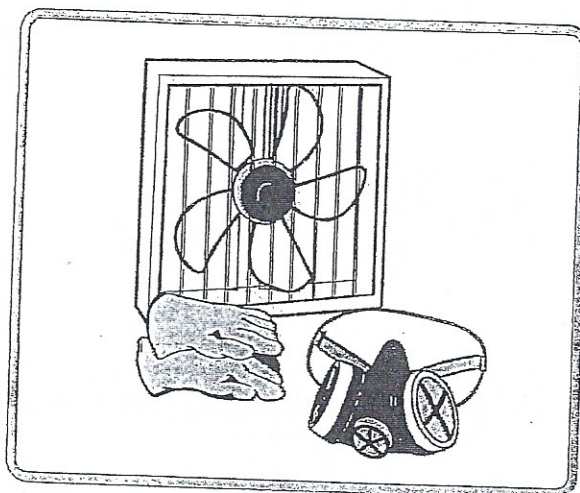
Conditions for *storage* are given here, such as:

- safe storage life
- temperature
- ventilation
- no smoking
- sources of ignition

What Are Control Measures?

If you look at nothing else in the SDS, check this out. These are the *control measures* you can use to reduce your exposure to hazardous chemicals. Control measures are methods used by employers or workers to reduce exposures to hazardous substances.

One important control measure that probably will not be listed on an SDS is to *substitute* a safer product.



What Are the Limitations of the SDS?

SDSs cover chemical hazards of products you use in your workplace. But construction workers face many other hazards that aren't described on SDSs. Among them are:

- heat stress
- noise
- vibration
- electrocution
- CO from combustion

Some of these exposures may aggravate chemical exposures from the products you work with. Your employer is responsible for informing you about all the hazards in your workplace.

Hazard Communication

ABC SOLVENT - MAKE-BELIEVE SOLVENT Material Safety Data Sheet for the U.S.A. and Canada

SECTION I - PRODUCT INFORMATION

| | |
|------------------------|-------------------------------------|
| IDENTITY (TRADE NAME): | ABC SOLVENT |
| SYNONYMS: | STODDARD SOLVENT |
| ABC PART NUMBER(S): | 6617 |
| FAMILY/CHEMICAL NAME: | ALIPHATIC HYDROCARBON |
| PRODUCT USE: | Cleaning and degreasing metal parts |

SECTION II - MANUFACTURER INFORMATION

| | |
|---------------|--------------------------|
| MANUFACTURER: | ABC COMPANY |
| | 777 Big Timber Road |
| | Parker, Oregon 90210 |
| TELEPHONE: | 1-800-421-5144 |
| ISSUE DATE: | March 19, 1998 |
| PREPARER: | Product MSDS Coordinator |

SECTION III - HAZARDOUS COMPONENTS

| Name | Synonym | WT% | CAS | PEL |
|----------------------|------------------|------|-------|-----|
| Parts Washer Solvent | Stoddard Solvent | 85.0 | 64741 | 100 |
| Xylene | Dimethylbenzene | 15.0 | 7890 | 100 |

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SAFETY DATA SHEET

Issuing Date January 5, 2015

Revision Date June 12, 2015

Revision Number 1

1. IDENTIFICATION OF THE SUBSTANCE/PREPARATION AND OF THE COMPANY/UNDERTAKING

Product identifier

Product Name Clorox® Regular-Bleach₁

Other means of identification

EPA Registration Number 5813-100

Recommended use of the chemical and restrictions on use

Recommended use Household disinfecting, sanitizing, and laundry bleach

Uses advised against No information available

Details of the supplier of the safety data sheet

Supplier Address
The Clorox Company
1221 Broadway
Oakland, CA 94612

Phone: 1-510-271-7000

Emergency telephone number

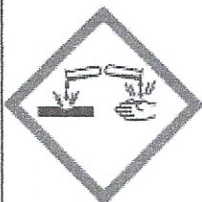
Emergency Phone Numbers For Medical Emergencies, call: 1-800-446-1014
For Transportation Emergencies, call Chemtrec: 1-800-424-9300

2. HAZARDS IDENTIFICATION**Classification**

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

| | |
|-----------------------------------|------------|
| Skin corrosion/irritation | Category 1 |
| Serious eye damage/eye irritation | Category 1 |

GHS Label elements, including precautionary statements**Emergency Overview**

| | |
|--|--------------------|
| Signal word | Danger |
| Hazard Statements Causes severe skin burns and eye damage Causes serious eye damage | |
|  | |
| Appearance | Clear, pale yellow |
| Physical State | Thin liquid |
| Odor | Bleach |

Precautionary Statements - Prevention

Wash face, hands and any exposed skin thoroughly after handling.

Wear protective gloves, protective clothing, face protection, and eye protection such as safety glasses.

Precautionary Statements - Response

Immediately call a poison center or doctor.

If swallowed: Rinse mouth. Do NOT induce vomiting.

If on skin (or hair): Take off immediately all contaminated clothing. Rinse skin with water.

Wash contaminated clothing before reuse.

If inhaled: Remove person to fresh air and keep comfortable for breathing.

Specific treatment (see supplemental first aid instructions on this label).

If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Precautionary Statements - Storage

Store locked up.

Precautionary Statements - Disposal

Dispose of contents in accordance with all applicable federal, state, and local regulations.

Hazards not otherwise classified (HNOC)

Although not expected, heart conditions or chronic respiratory problems such as asthma, chronic bronchitis, or obstructive lung disease may be aggravated by exposure to high concentrations of vapor or mist.

Product contains a strong oxidizer. Always flush drains before and after use.

Unknown Toxicity

Not applicable.

Other information

Very toxic to aquatic life with long lasting effects.

Interactions with Other Chemicals

Reacts with other household chemicals such as toilet bowl cleaners, rust removers, acids, or products containing ammonia to produce hazardous irritating gases, such as chlorine and other chlorinated compounds.

3. COMPOSITION/INFORMATION ON INGREDIENTS

| Chemical Name | CAS-No | Weight % | Trade Secret |
|---------------------|-----------|----------|--------------|
| Sodium hypochlorite | 7681-52-9 | 5 - 10 | * |

* The exact percentage (concentration) of composition has been withheld as a trade secret.

4. FIRST AID MEASURES**First aid measures****General Advice**

Call a poison control center or doctor immediately for treatment advice. Show this safety data sheet to the doctor in attendance.

Eye Contact

Hold eye open and rinse slowly and gently with water for 15 - 20 minutes. Remove contact lenses, if present, after the first 5 minutes, then continue rinsing eye. Call a poison control center or doctor for treatment advice.

Skin Contact

Take off contaminated clothing. Rinse skin immediately with plenty of water for 15-20 minutes. Call a poison control center or doctor for treatment advice.

Inhalation

Move to fresh air. If breathing is affected, call a doctor.

Ingestion

Have person sip a glassful of water if able to swallow. Do not induce vomiting unless told to do so by a poison control center or doctor. Do not give anything by mouth to an unconscious person. Call a poison control center or doctor immediately for treatment advice.

Protection of First-aiders

Avoid contact with skin, eyes, and clothing. Use personal protective equipment as required. Wear personal protective clothing (see section 8).

Most important symptoms and effects, both acute and delayed**Most Important Symptoms and Effects**

Burning of eyes and skin.

Indication of any immediate medical attention and special treatment needed**Notes to Physician**

Treat symptomatically. Probable mucosal damage may contraindicate the use of gastric lavage.

5. FIRE-FIGHTING MEASURES

Suitable Extinguishing Media

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Unsuitable Extinguishing Media

CAUTION: Use of water spray when fighting fire may be inefficient.

Specific Hazards Arising from the Chemical

This product causes burns to eyes, skin, and mucous membranes. Thermal decomposition can release sodium chlorate and irritating gases and vapors.

Explosion Data

Sensitivity to Mechanical Impact None.

Sensitivity to Static Discharge None.

Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Personal Precautions

Avoid contact with eyes, skin, and clothing. Ensure adequate ventilation. Use personal protective equipment as required. For spills of multiple products, responders should evaluate the MSDSs of the products for incompatibility with sodium hypochlorite. Breathing protection should be worn in enclosed and/or poorly-ventilated areas until hazard assessment is complete.

Other Information

Refer to protective measures listed in Sections 7 and 8.

Environmental precautions

Environmental Precautions

This product is toxic to fish, aquatic invertebrates, oysters, and shrimp. Do not allow product to enter storm drains, lakes, or streams. See Section 12 for ecological Information.

Methods and material for containment and cleaning up

Methods for Containment

Prevent further leakage or spillage if safe to do so.

Methods for Cleaning Up

Absorb and containerize. Wash residual down to sanitary sewer. Contact the sanitary treatment facility in advance to assure ability to process washed-down material.

7. HANDLING AND STORAGE**Precautions for safe handling****Handling**

Handle in accordance with good industrial hygiene and safety practice. Avoid contact with skin, eyes, and clothing. Do not eat, drink, or smoke when using this product.

Conditions for safe storage, including any incompatibilities**Storage**

Store away from children. Reclose cap tightly after each use. Store this product upright in a cool, dry area, away from direct sunlight and heat to avoid deterioration. Do not contaminate food or feed by storage of this product.

Incompatible Products

Toilet bowl cleaners, rust removers, acids, and products containing ammonia.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION**Control parameters****Exposure Guidelines**

| Chemical Name | ACGIH TLV | OSHA PEL | NIOSH IDLH |
|----------------------------------|-----------|----------|------------|
| Sodium hypochlorite 7681-52-9 | None | None | None |

ACGIH TLV: American Conference of Governmental Industrial Hygienists - Threshold Limit Value. OSHA PEL: Occupational Safety and Health Administration - Permissible Exposure Limits. NIOSH IDLH: Immediately Dangerous to Life or Health.

Appropriate engineering controls**Engineering Measures**

Showers
Eyewash stations
Ventilation systems

Individual protection measures, such as personal protective equipment**Eye/Face Protection**

If splashes are likely to occur: Wear safety glasses with side shields (or goggles) or face shield.

Skin and Body Protection

Wear rubber or neoprene gloves and protective clothing such as long-sleeved shirt.

Respiratory Protection

If irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be provided in accordance with current local regulations.

Hygiene Measures

Handle in accordance with good industrial hygiene and safety practice. Wash hands after direct contact. Do not wear product-contaminated clothing for prolonged periods. Remove and wash contaminated clothing before re-use. Do not eat, drink, or smoke when using this product.

9. PHYSICAL AND CHEMICAL PROPERTIES

Physical and Chemical Properties

| | | | |
|----------------|-------------|----------------|--------------------------|
| Physical State | Thin liquid | Odor | Bleach |
| Appearance | Clear | Odor Threshold | No information available |
| Color | Pale yellow | | |

| <u>Property</u> | <u>Values</u> | <u>Remarks/ Method</u> |
|--|-------------------|------------------------|
| pH | ~12 | None known |
| Melting/freezing point | No data available | None known |
| Boiling point / boiling range | No data available | None known |
| Flash Point | Not flammable | None known |
| Evaporation rate | No data available | None known |
| Flammability (solid, gas) | No data available | None known |
| Flammability Limits in Air | | |
| Upper flammability limit | No data available | None known |
| Lower flammability limit | No data available | None known |
| Vapor pressure | No data available | None known |
| Vapor density | No data available | None known |
| Specific Gravity | ~1.1 | None known |
| Water Solubility | Soluble | None known |
| Solubility in other solvents | No data available | None known |
| Partition coefficient: n-octanol/water | No data available | None known |
| Autoignition temperature | No data available | None known |
| Decomposition temperature | No data available | None known |
| Kinematic viscosity | No data available | None known |
| Dynamic viscosity | No data available | None known |
| Explosive Properties | Not explosive | |
| Oxidizing Properties | No data available | |

Other Information

| | |
|----------------------------|-------------------|
| Softening Point | No data available |
| VOC Content (%) | No data available |
| Particle Size | No data available |
| Particle Size Distribution | No data available |

10. STABILITY AND REACTIVITY

Reactivity

Reacts with other household chemicals such as toilet bowl cleaners, rust removers, acids, or products containing ammonia to produce hazardous irritating gases, such as chlorine and other chlorinated compounds.

Chemical stability

Stable under recommended storage conditions.

Possibility of Hazardous Reactions

None under normal processing.

Conditions to avoid

None known based on information supplied.

Incompatible materials

Toilet bowl cleaners, rust removers, acids, and products containing ammonia.

Hazardous Decomposition Products

None known based on information supplied.

11. TOXICOLOGICAL INFORMATION**Information on likely routes of exposure****Product Information**

| | |
|---------------------|---|
| Inhalation | Exposure to vapor or mist may irritate respiratory tract and cause coughing. Inhalation of high concentrations may cause pulmonary edema. |
| Eye Contact | Corrosive. May cause severe damage to eyes. |
| Skin Contact | May cause severe irritation to skin. Prolonged contact may cause burns to skin. |
| Ingestion | Ingestion may cause burns to gastrointestinal tract and respiratory tract, nausea, vomiting, and diarrhea. |

Component Information

| Chemical Name | LD50 Oral | LD50 Dermal | LC50 Inhalation |
|----------------------------------|------------------|-----------------------|-----------------|
| Sodium hypochlorite 7681-52-9 | 8200 mg/kg (Rat) | >10000 mg/kg (Rabbit) | - |

Information on toxicological effects

| | |
|-----------------|--|
| Symptoms | May cause redness and tearing of the eyes. May cause burns to eyes. May cause redness or burns to skin. Inhalation may cause coughing. |
|-----------------|--|

Delayed and immediate effects as well as chronic effects from short and long-term exposure

| | |
|--------------------------|--|
| Sensitization | No information available. |
| Mutagenic Effects | No information available. |
| Carcinogenicity | The table below indicates whether each agency has listed any ingredient as a carcinogen. |

| Chemical Name | ACGIH | IARC | NTP | OSHA |
|----------------------------------|-------|---------|-----|------|
| Sodium hypochlorite 7681-52-9 | - | Group 3 | - | - |

*IARC (International Agency for Research on Cancer)
Group 3 - Not Classifiable as to Carcinogenicity in Humans*

| | |
|---------------------------------|--|
| Reproductive Toxicity | No information available. |
| STOT - single exposure | No information available. |
| STOT - repeated exposure | No information available. |
| Chronic Toxicity | Carcinogenic potential is unknown. |
| Target Organ Effects | Respiratory system, eyes, skin, gastrointestinal tract (GI). |
| Aspiration Hazard | No information available. |

Numerical measures of toxicity - Product Information

The following values are calculated based on chapter 3.1 of the GHS document

ATEmix (oral)

54 g/kg

ATEmix (inhalation-dust/mist)

58 mg/L

12. ECOLOGICAL INFORMATION**Ecotoxicity**

Very toxic to aquatic life with long lasting effects.

This product is toxic to fish, aquatic invertebrates, oysters, and shrimp. Do not allow product to enter storm drains, lakes, or streams.

Persistence and Degradability

No information available.

Bioaccumulation

No information available.

Other adverse effects

No information available.

13. DISPOSAL CONSIDERATIONS**Disposal methods**

Dispose of in accordance with all applicable federal, state, and local regulations. Do not contaminate food or feed by disposal of this product.

Contaminated Packaging

Do not reuse empty containers. Dispose of in accordance with all applicable federal, state, and local regulations.

14. TRANSPORT INFORMATION**DOT**

Not restricted.

TDG

Not restricted for road or rail.

ICAO

Not restricted, as per Special Provision A197, Environmentally Hazardous Substance exception.

IATA

Not restricted, as per Special Provision A197, Environmentally Hazardous Substance exception.

IMDG/IMO

Not restricted, as per IMDG Code 2.10.2.7, Marine Pollutant exception.

15. REGULATORY INFORMATION**Chemical Inventories****TSCA**

All components of this product are either on the TSCA 8(b) Inventory or otherwise exempt from listing.

DSL/NDSL

All components are on the DSL or NDSL.

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory

DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

U.S. Federal Regulations**SARA 313**

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372

SARA 311/312 Hazard Categories

| | |
|--|-----|
| Acute Health Hazard | Yes |
| Chronic Health Hazard | No |
| Fire Hazard | No |
| Sudden Release of Pressure Hazard | No |
| Reactive Hazard | No |

Clean Water Act

This product contains the following substances which are regulated pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

| Chemical Name | CWA - Reportable Quantities | CWA - Toxic Pollutants | CWA - Priority Pollutants | CWA - Hazardous Substances |
|----------------------------------|------------------------------------|-------------------------------|----------------------------------|-----------------------------------|
| Sodium hypochlorite 7681-52-9 | 100 lb | | | X |

CERCLA

This material, as supplied, contains one or more substances regulated as a hazardous substance under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302)

| Chemical Name | Hazardous Substances RQs | Extremely Hazardous Substances RQs | RQ |
|----------------------------------|---------------------------------|---|---|
| Sodium hypochlorite 7681-52-9 | 100 lb | - | RQ 100 lb final RQ RQ 45.4 kg final RQ |

EPA Statement

This chemical is a pesticide product registered by the Environmental Protection Agency and is subject to certain labeling requirements under federal pesticide law. These requirements differ from the classification criteria and hazard information required for safety data sheets and for workplace labels of non-pesticide chemicals. Following is the hazard information as required on the pesticide label:

DANGER: CORROSIVE. Causes irreversible eye damage and skin burns. Harmful if swallowed. Do not get in eyes, on skin, or on clothing. Wear protective eyewear and rubber gloves when handling this product. Wash thoroughly with soap and water after handling and before eating, drinking, chewing gum, using tobacco, or using the restroom. Avoid breathing vapors and use only in a well-ventilated area.

US State Regulations**California Proposition 65**

This product does not contain any Proposition 65 chemicals.

U.S. State Right-to-Know Regulations

| Chemical Name | New Jersey | Massachusetts | Pennsylvania | Rhode Island | Illinois |
|----------------------------------|------------|---------------|--------------|--------------|----------|
| Sodium hypochlorite 7681-52-9 | X | X | X | X | |
| Sodium chlorate 7775-09-9 | X | X | X | | |

International Regulations**Canada****WHMIS Hazard Class**

E - Corrosive material

**16. OTHER INFORMATION**

NFPA Health Hazard 3 Flammability 0 Instability 0 Physical and Chemical Hazards -

HMIS Health Hazard 3 Flammability 0 Physical Hazard 0 Personal Protection B

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General Disclaimer

The information provided in this Safety Data Sheet is correct to the best of our knowledge, information and belief at the date of its publication. The information given is designed only as a guidance for safe handling, use, processing, storage, transportation, disposal, and release and is not to be considered a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other materials or in any process, unless specified in the text.

End of Safety Data Sheet

Test yourself on this brief review:

1. Which of the following statements about an SDS is true?

- a. My employer can refuse to give it to me.
- b. I can only get one from the manufacturer.
- c. My union representative can get it for me.
- d. To get it fast, I should write to OSHA.

2. What products require SDSs at the jobsite?

- a. products with a hazard rating of '10'
- b. imports subject to the EPA tax
- c. all hazardous products on the job
- d. only products stored on the job

3. Which of the following is *not* part of a required HazCom program?

- a. warning labels
- b. training
- c. 'SDSs
- d. hazardous duty pay

4. The OSHA Hazard Communication Standard has been in effect since _____?

- a. the late 1800s
- b. the mid-1980s
- c. 1992
- d. 1950

Test yourself on this brief review:

1. The three main routes of entry are:

- a. breathing, skin absorption, swallowing.
- b. breathing, eyes, ears.
- c. breathing, skin, immersion.
- d. nose, blood, skin.

2. Choose the effect that is *not* an acute effect:

- a. cancer
- b. suffocation
- c. dizziness
- d. heart attack

3. Choose the effect that is probably *not* a chronic effect:

- a. cancer
- b. jaundice
- c. nausea
- d. allergic dermatitis

4. Four chemicals cause respiratory problems. Each has a different PEL. Circle the one that is probably most harmful.

- a. PEL = 200 ppm
- b. PEL = 1,000 ppm
- c. PEL = 10 ppm
- d. PEL = 1 ppm

5. The OSHA Hazard Communication Standard was created mainly through the efforts of:

- a. labor unions
- b. employer associations
- c. manufacturing associations
- d. insurance companies

Hazard Communication

Test yourself on this brief review:

1. Which of these products is the most dangerous fire hazard?
 - a. flash point = 80°
 - b. flash point = -55° F
 - c. flash point = 100° F
 - d. flash point = 150° F
2. Which of these products is classified as flammable?
 - a. flash point = 300° F
 - b. flash point = 200° F
 - c. flash point = 100° F
 - d. flash point = 50° F
3. In the National Fire Protection Association (NFPA) warning label system, the number in the BLUE diamond indicates which type of hazard?
 - a. red
 - b. health
 - c. fire
 - d. yellow
4. A reactive product has a tendency to:
 - a. harden quickly.
 - b. release energy.
 - c. disperse.
 - d. dry rapidly.

Test yourself on this brief review:

1. Which of the following is a control measure that probably will not be listed on an SDS?
 - a. respirator
 - b. ventilation
 - c. substitute safer product
 - d. gloves
2. Which one of the following would require an SDS?
 - a. a faulty guardrail on a 15-foot scaffold
 - b. frayed electrical cord
 - c. a confined space on a new building site
 - d. a two-part epoxy adhesive
3. The sample MSDS used in the exercises is:
 - a. official SDS format required by OSHA
 - b. a very unique and unusual SDS
 - c. just one of many formats for SDSs
 - d. official SDS format required by DOE