

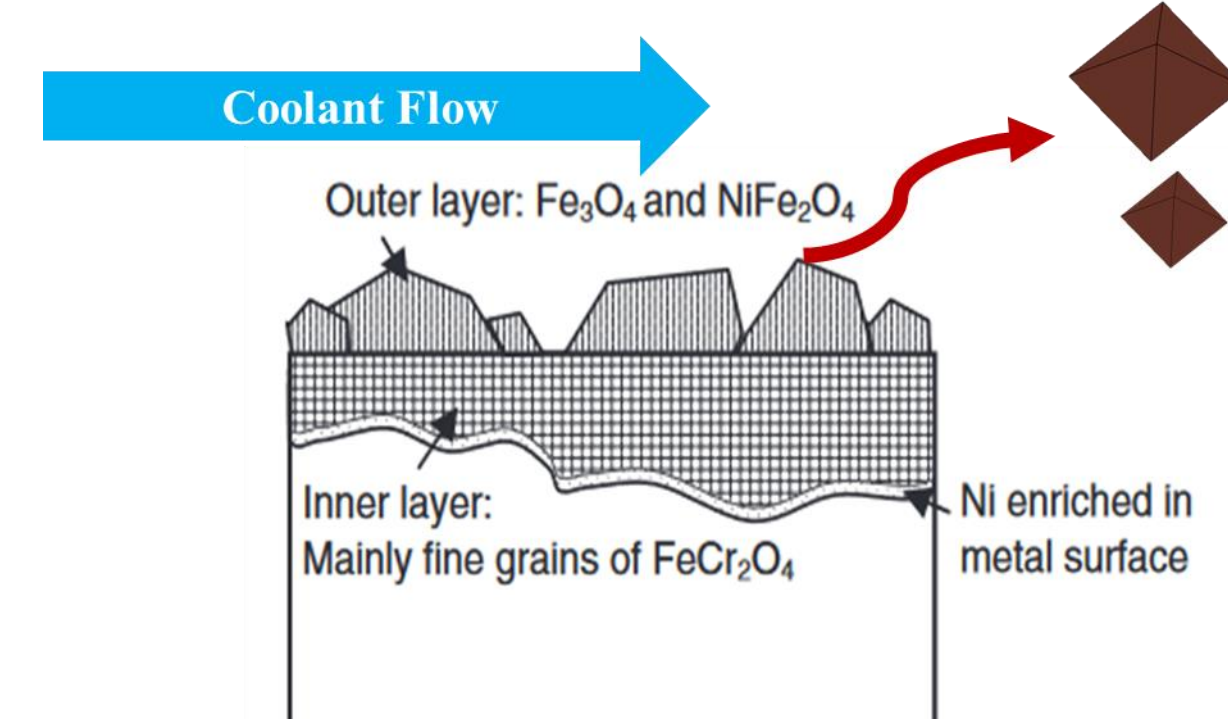
# Multiscale Multiphysics Model of CRUD Transport and Deposition in Pressurized Water Reactors: Formulation and Preliminary Results Examining the Effect of Surface Potentials

Sri Saravana Konganapuram Narasimma Bharathi<sup>1</sup>, Bhavani Sasank Nagothi<sup>1</sup>, Eric Wales<sup>1</sup>, Matthew Armstrong<sup>2</sup>, John Arnason<sup>2</sup>, Kathleen Dunn<sup>1</sup>  
<sup>1</sup>Department of Nanoscale Science & Engineering, University at Albany - SUNY, NY, USA  
<sup>2</sup>Naval Nuclear Laboratory, Niskayuna, New York, USA

## Motivation

Fouling caused by Corrosion products (CRUD) in PWR affect the operational performance of the Reactor (CRUD-induced power shifts, CRUD-induced localized corrosion, etc.,)

**Goal:** To develop a model in COMSOL™ that simulates the deposition of suspended particles inside a pipe representing a reactor environment

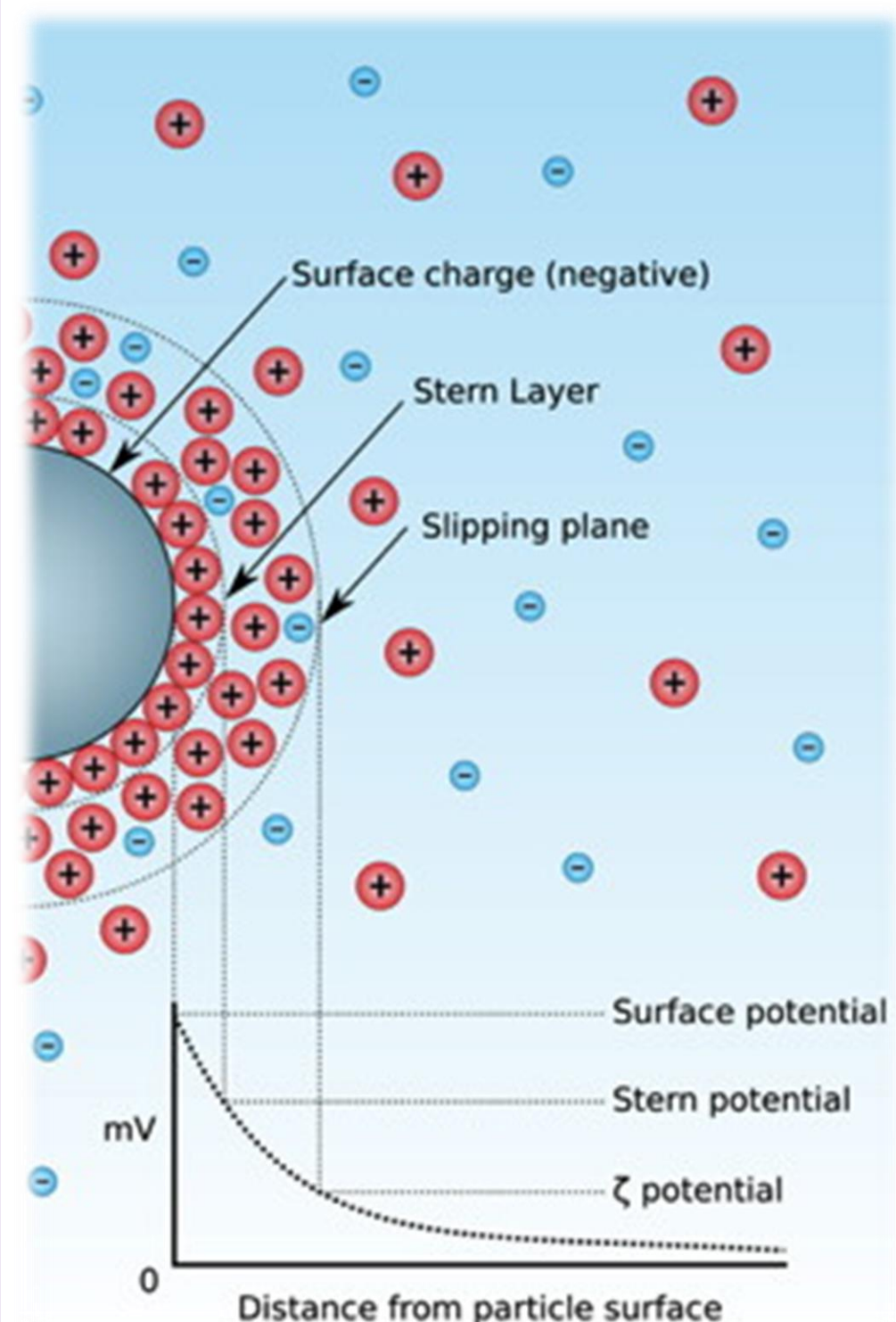


## Transport and Deposition of CRUD

### Thermodynamics

Will particles deposit on the wall?

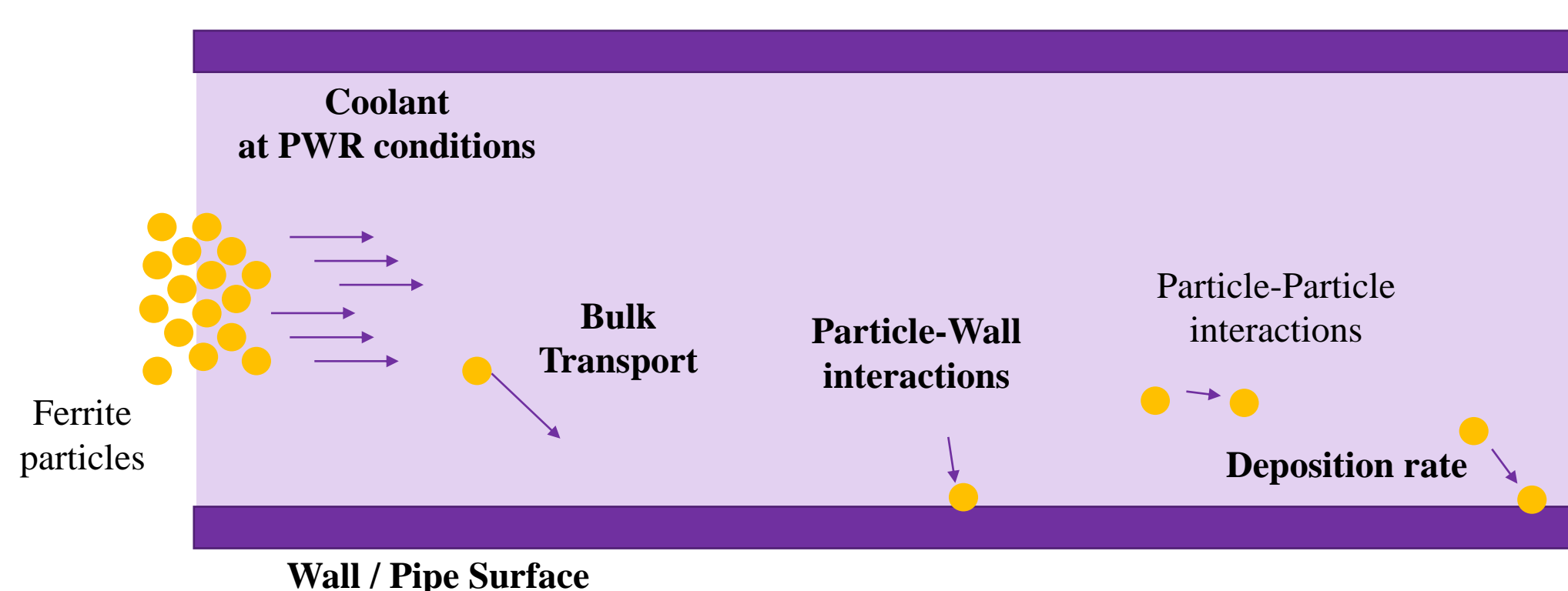
- Colloidal forces
  - Dipole interactions: van der Waal's (VDW) force
  - Electric double-layer (EDL) interactions
- Highly sensitive to surface properties such as Stern potential



### Kinetics

At what rate do particles approach the wall?

- Bulk transport of particles due to turbulence and convective transport in the coolant
- Fluid flow solved with k- $\omega$  turbulence model taken for the Stokes' drag force calculations



COMSOL  
Modules

- Computational Fluid Dynamics
- Particle Tracing Module

## Multiscale Multiphysics Model Formulation

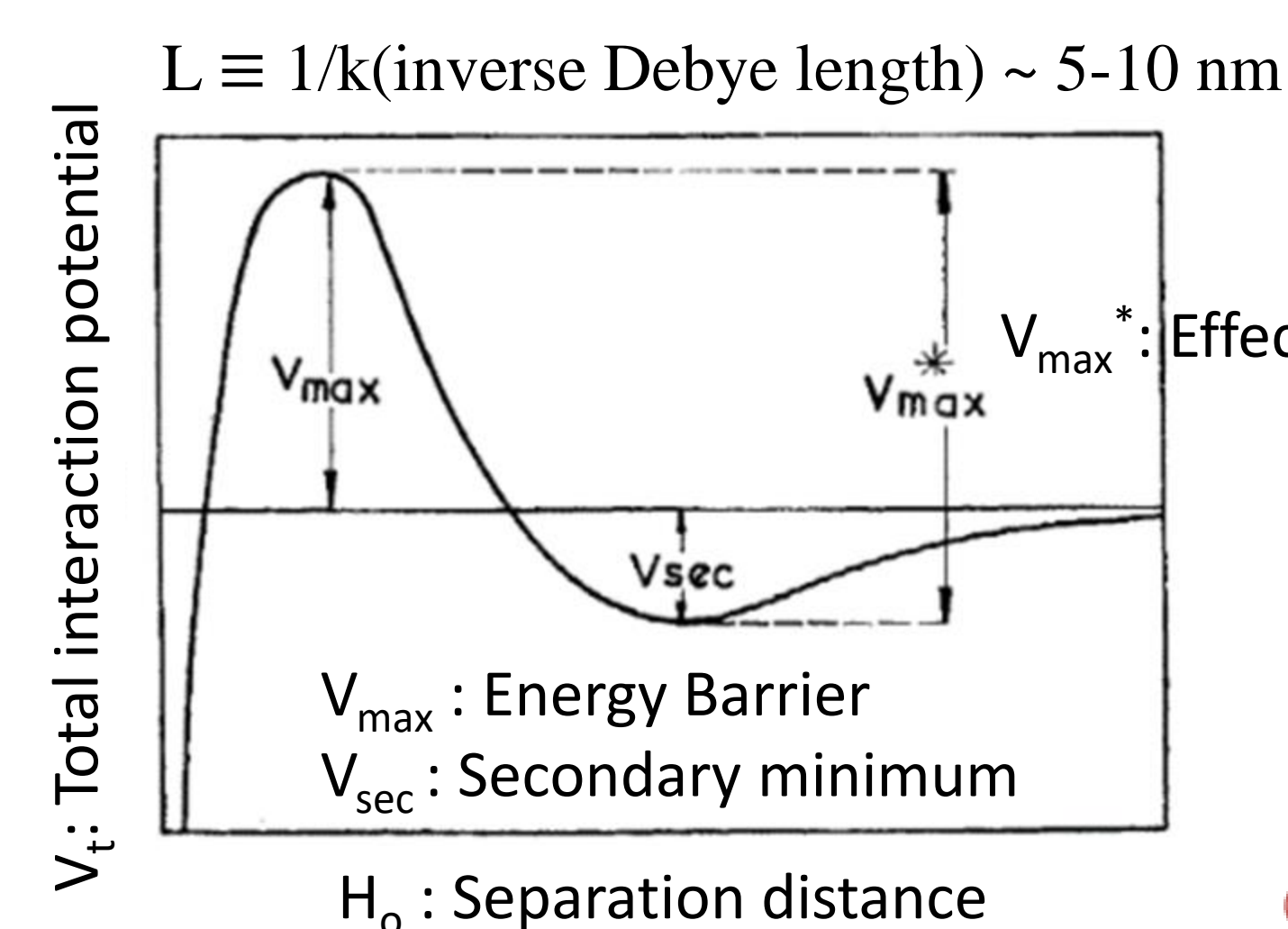
Time scales of events

- Kolmogorov length and time scales for turbulence
- Range of EDL forces

$$\eta \equiv \left(\frac{\nu^3}{\varepsilon}\right)^{1/4}$$

$$\tau_\eta \equiv \left(\frac{\nu}{\varepsilon}\right)^{1/2}$$

$\nu$  Viscosity  
 $\varepsilon$  Dissipation rate



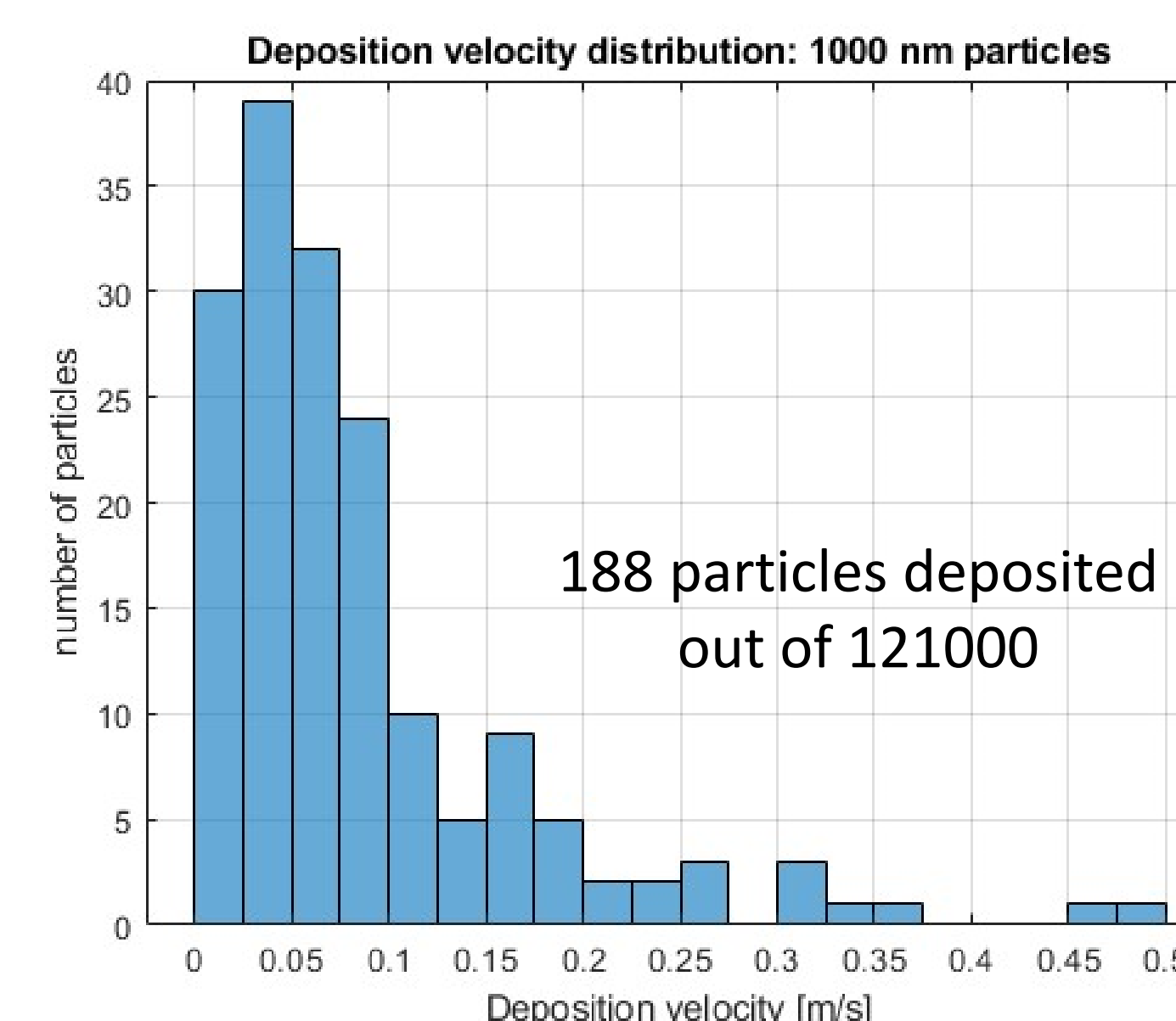
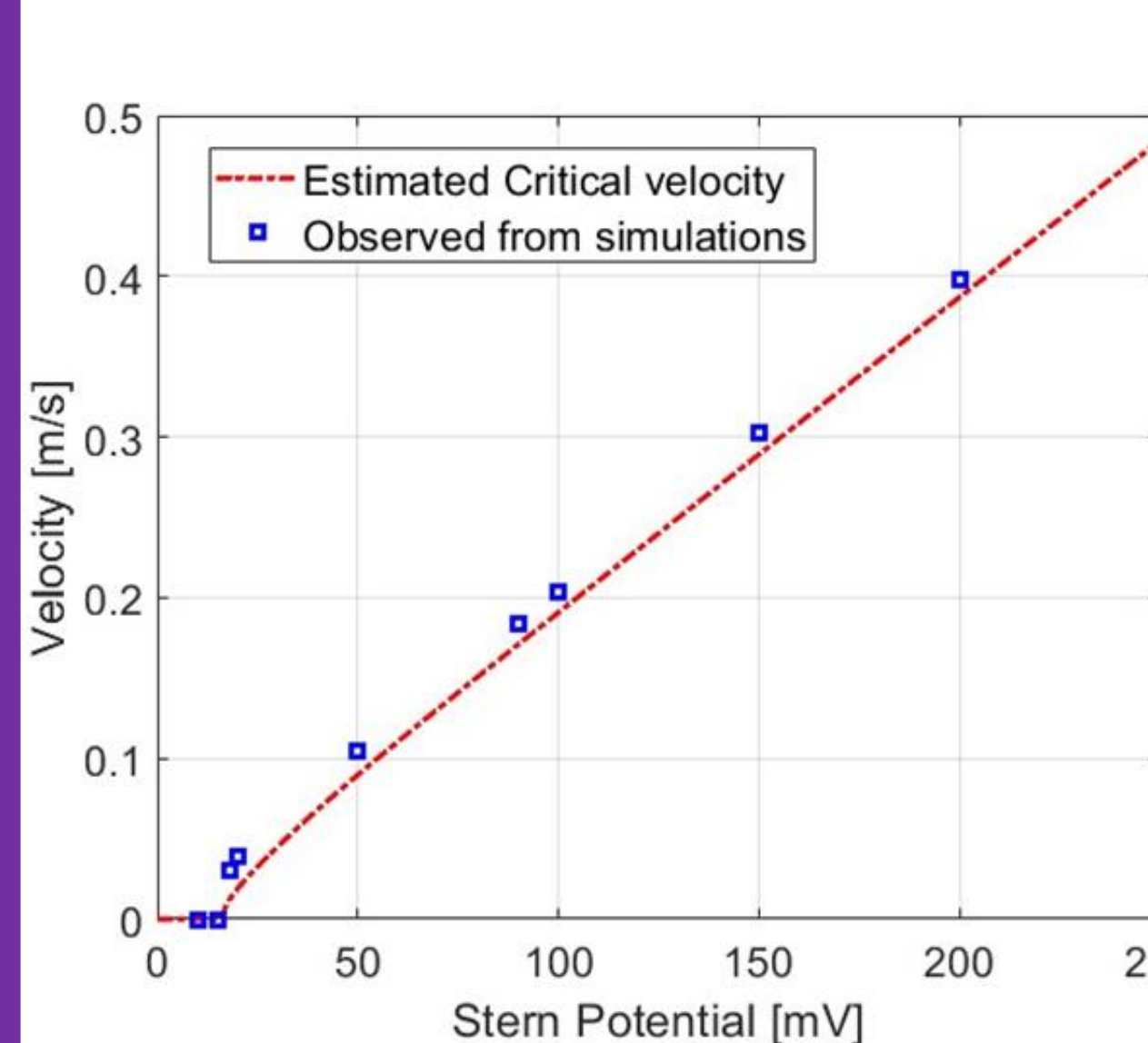
Critical velocity: minimum (radial) velocity for a particle to overcome barrier to deposit on the wall

Coolant (Water) Parameters	
Density	660 kg/m <sup>3</sup>
Viscosity	1.01 x 10 <sup>-4</sup> Pa.s
Temperature	598.15 K
Inlet velocity (average)	13.5 m/s
Outlet pressure	15.5 MPa
Length and radius of pipe geometry	10 m, 0.3685 m

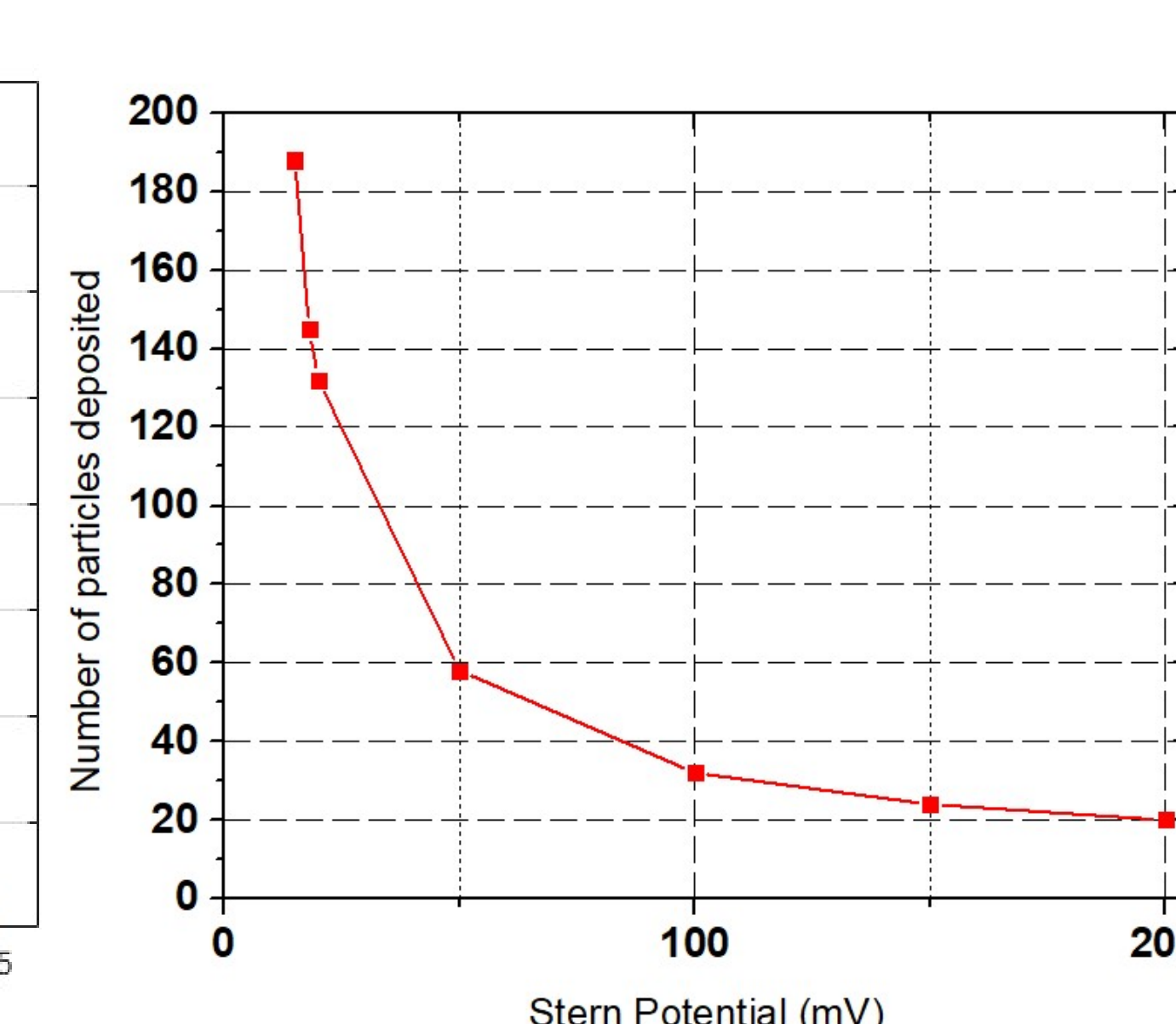
Nickel ferrite particles	
Particle diameter	1 $\mu$ m

## Preliminary Results: Sensitivity Analysis

- Simulations were performed to understand the effect of the Stern potential on deposition
- Fine-scale simulations show a linear trend between critical velocity and the Stern potential, in agreement with the values estimated through DLVO alone (by extrapolating the potential barrier from DLVO to be kinetic energy required to deposit)
- Macroscale simulations provide the velocity distribution of deposited particles



Coupled results shows that deposition is sensitive to the Stern potential



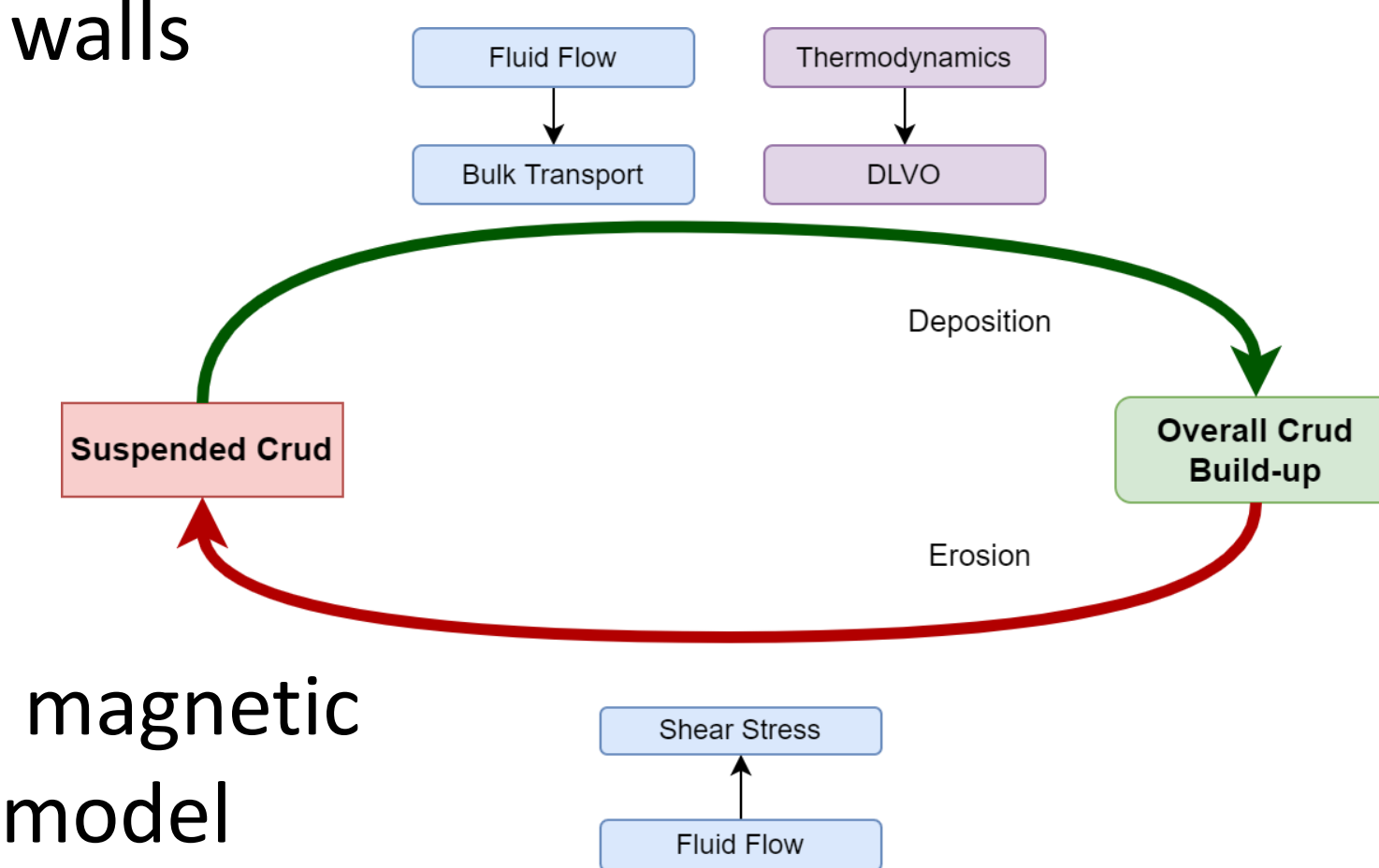
The coolant pH which would determine the Stern potential and affect the deposition significantly.

## Summary

- Formulated a Multiscale Model for the deposition of CRUD in primary coolant circuit.
- Quantitative and qualitative verification for fine-scale model: obtained values and trends match with DLVO theory predictions
- Preliminary investigations confirmed deposition is highly sensitive to Stern potential of particle, which is pH-dependent

## Future Work

- Investigate the effect of particle size on the deposition (drag forces, surface charge)
- Incorporate possible re-entrainment of particles from the walls



- Include magnetic forces in model

## Acknowledgments

This material is based upon work supported by the U.S. Nuclear Regulatory Commission, Office of Nuclear Regulatory Research under Award Number 31310020M0006, with additional support from the New York State Center of Excellence in Nanoelectronics and Nanotechnology (NYS CENN).

## References

- Terachi, T., Yamada, T., Miyamoto, T., Arioka, K., & Fukuya, K. (2012). Corrosion Behavior of Stainless Steels in Simulated PWR Primary Water—Effect of Chromium Content in Alloys and Dissolved Hydrogen—. <http://dx.doi.org/10.1080/18811248.2008.9711883>, 45(10), 975–984. <https://doi.org/10.1080/18811248.2008.9711883>
- Basset, M., McInerney, J., Arbeau, N., & Lister, D. H. (2000). The fouling of Alloy-800 heat exchange surfaces by magnetite particles. *Canadian Journal of Chemical Engineering*, 78(1), 40–52. <https://doi.org/10.1002/cjce.5450780108>
- Bindra, H., & Harbottle, D. (2016). Nuclear Fuels and Materials: Modeling Modeling crud attachment on fuel surfaces-Role of interaction forces. *Transactions of the American Nuclear Society*, 115, 453–455.
- 2.5: Zeta Potential Analysis - Chemistry LibreTexts. (n.d.). Retrieved September 23, 2023, from [https://chem.libretexts.org/Bookshelves/Analytical\\_Chemistry/Physical\\_Methods\\_in\\_Chemistry\\_and\\_Nano\\_Science\\_\(Barron\)/02%3A\\_Physical\\_and\\_Thermal\\_Analysis/2.05%3A\\_Zeta\\_Potential\\_Analysis](https://chem.libretexts.org/Bookshelves/Analytical_Chemistry/Physical_Methods_in_Chemistry_and_Nano_Science_(Barron)/02%3A_Physical_and_Thermal_Analysis/2.05%3A_Zeta_Potential_Analysis)
- Hogg, R., Healy, T. W., & Fuerstenau, D. W. (1966). Mutual coagulation of colloidal dispersions. *Transactions of the Faraday Society*, 62(615), 1638–1651. <https://doi.org/10.1039/tf966201638>
- Wiese, G. R., & Healy, T. W. (1970). Effect of particle size on colloid stability. *Transactions of the Faraday Society*, 66(0), 490–499. <https://doi.org/10.1039/tf9706600490>

Contact: kdunn1@albany.edu