

"Fatty Acid, Triacylglycerol, and Ketone Body Metabolism"

I. Overview

- Fatty acids exist:
 - Free in the body (i.e., nonesterified)
 - As fatty acyl esters in more complex molecules such as triacylglycerols (TAGs)
- Free fatty acids (FFA):
 - Occur in low levels in all tissues
 - Can be found in substantial amounts in plasma, particularly during fasting
- Plasma FFA:
 - Transported on serum albumin
 - In route from their point of origin:
 - TAG of adipose tissue
 - Circulating lipoproteins
 - Destination: site of consumption (most tissues)\
- FFA oxidation:
 - Can be oxidized by many tissues, particularly liver and muscle

- Purpose:
 - Provide energy
 - In the liver: provide substrate for ketone body synthesis
- Fatty acids are:
 - Structural components of membrane lipids:
 - Phospholipids
 - Glycolipids
 - Attached to certain proteins to enhance ability to associate with membranes
 - Precursors of hormone-like prostaglandins
- Esterified fatty acids (as TAG in white adipose tissue [WAT]):
 - Serve as the major energy reserve of the body
- Alterations in fatty acid metabolism:
 - Associated with obesity and diabetes

II. Fatty Acid Structure

- A fatty acid consists of:
 - Hydrophobic hydrocarbon chain
 - Terminal carboxyl group with a pKa of 4.8
- At physiologic pH:
 - Terminal carboxyl group ($-\text{COOH}$) ionizes to become $-\text{COO}^-$
- Note:
 - When pH is above the pK, the deprotonated form predominates
- Amphipathic nature of fatty acids:
 - Due to anionic group's affinity for water
 - Fatty acid has both:
 - Hydrophilic region ($-\text{COO}^-$)
 - Hydrophobic region (hydrocarbon chain)

- For long-chain-length fatty acids (LCFA):
 - Hydrophobic portion is predominant
 - These molecules are highly water insoluble
 - Must be transported in circulation in association with protein
- 90% of fatty acids in plasma:
 - Found in the form of fatty acid esters:
 - Primarily TAG
 - Cholesteryl esters
 - Phospholipids
 - Contained in circulating lipoprotein particles
- Free fatty acids (FFA):
 - Transported in circulation in association with albumin
 - Albumin: the most abundant protein in serum

A. Fatty Acid Saturation

- Fatty acid chains may:
 - Contain no double bonds → saturated
 - Contain one or more double bonds → mono- or polyunsaturated
- In humans:
 - The majority of fatty acids are saturated or monounsaturated
- When double bonds are present:
 - They are nearly always in the cis configuration
 - Cis double bonds cause the fatty acid to bend or kink at that position
- If a fatty acid has two or more double bonds:
 - They are always spaced at three-carbon intervals

- Note:
 - Addition of double bonds decreases the melting temperature (T_m) of a fatty acid
 - Increasing chain length increases the T_m
 - Membrane lipids typically contain long-chain fatty acids (LCFA)
 - Presence of double bonds helps maintain the fluid nature of those lipids
- In humans:
 - Fatty acids with even number of carbon atoms (16, 18, or 20) predominate
 - Longer fatty acids (>22 carbons) are found in the brain
- Carbon atom numbering:
 - Begins with the carbonyl carbon as carbon 1
- Notation system:
 - Number before the colon = number of carbon atoms in the chain
 - Numbers after the colon = positions of double bonds (relative to the carboxyl end)

- Example:

- Arachidonic acid \rightarrow 20:4(5,8,11,14)

- 20 carbons long

- 4 double bonds between:

- Carbon 5-6

- Carbon 8-9

- Carbon 11-12

- Carbon 14-15

- Note on carbon naming:

- Carbon 2 = α -carbon

- Carbon 3 = β -carbon

- Carbon 4 = γ -carbon

- Terminal methyl group carbon = ω -carbon, regardless of chain length

- Double bond position can also be referenced from the ω (methyl) end of the chain

- Arachidonic acid:

- Called an ω -6 fatty acid

- Terminal double bond is six bonds from the ω end

- Equivalent designation: n-6

- Other examples:
 - Linoleic acid → 18:2(9,12)
 - Essential ω -6 fatty acid
 - α -Linolenic acid → 18:3(9,12,15)
 - Essential ω -3 fatty acid

C. Essential Fatty Acids

- Linoleic acid:
 - Precursor of ω -6 arachidonic acid
 - Arachidonic acid is the substrate for prostaglandin synthesis
- α -Linolenic acid:
 - Precursor of ω -3 fatty acids
 - Important for growth and development
- Both linoleic acid and α -linolenic acid are dietary essentials in humans because:
 - Humans lack the enzymes that can form carbon-carbon double bonds after the 9th carbon from the methyl (ω) end of a fatty acid

- Plants provide humans with these essential fatty acids
- Note:
 - Arachidonic acid becomes essential if linoleic acid is deficient in the diet
- Essential fatty acid deficiency (rare):
 - Can result in dry, scaly dermatitis
 - Due to an inability to synthesize molecules that provide the water barrier in skin

III. Fatty Acid De Novo Synthesis

- Excess dietary carbohydrates and proteins (beyond the body's needs):
 - Can be converted to fatty acids
- In adults, de novo fatty acid synthesis occurs primarily in:
 - Liver
 - Lactating mammary glands
 - Adipose tissue (to a lesser extent)

- This process is:
 - Cytosolic
 - Endergonic
 - Reductive
- The process incorporates:
 - Carbons from acetyl coenzyme A (CoA) into the growing fatty acid chain
 - ATP
 - Reduced nicotinamide adenine dinucleotide phosphate (NADPH)
- Note:
 - Dietary TAG also supply fatty acids

A. Cytosolic Acetyl CoA Production

- First step in fatty acid synthesis:
 - Transfer of acetate units from mitochondrial acetyl CoA to the cytosol

- Mitochondrial acetyl CoA is produced by:
 - Oxidation of pyruvate
 - Catabolism of certain amino acids
- Problem:
 - The CoA portion of acetyl CoA cannot cross the inner mitochondrial membrane
- Solution:
 - Only the acetyl portion enters the cytosol, as part of citrate
- Citrate formation:
 - Formed by condensation of acetyl CoA with oxaloacetate (OAA)
 - Enzyme: Citrate synthase
- Note:
 - Citrate transport to cytosol occurs when mitochondrial citrate concentration is high
 - This happens when isocitrate dehydrogenase of the TCA cycle is inhibited by large amounts of ATP

- Result: Accumulation of citrate and isocitrate
- Therefore:
 - Cytosolic citrate is a high-energy signal
 - Since fatty acid synthesis needs a large amount of ATP, increased ATP and citrate enhance this pathway
- In the cytosol:
 - Citrate is cleaved to:
 - Oxaloacetate (OAA)
 - Acetyl CoA
 - Enzyme: ATP citrate lyase

B. Acetyl CoA Carboxylation to Malonyl CoA

- Energy source for carbon-to-carbon condensations in fatty acid synthesis:
 - Supplied by the carboxylation and then decarboxylation of acyl groups in the cytosol

- Carboxylation of acetyl CoA to malonyl CoA:
 - Catalyzed by acetyl CoA carboxylase (ACC)
- ACC reaction details:
 - Transfers CO_2 from bicarbonate (HCO_3^-) in an ATP-requiring reaction
 - Coenzyme: Biotin (vitamin B7)
 - Biotin is covalently bound to a lysyl residue of the carboxylase
- Mechanism:
 - ACC carboxylates the bound biotin
 - Biotin then transfers the activated carboxyl group to acetyl CoA

1. Acetyl CoA Carboxylase Short-Term Regulation

- This carboxylation step is:
 - The rate-limiting step
 - A regulated step in fatty acid synthesis

- Inactive ACC = protomer (complex of ≥ 2 polypeptides)
- Allosteric regulation:
 - Activated by citrate \rightarrow causes protomers to polymerize
 - Inhibited by palmitoyl CoA (end product) \rightarrow causes depolymerization
- Reversible phosphorylation:
 - Adenosine monophosphate-activated protein kinase (AMPK):
 - Phosphorylates and inactivates ACC
 - AMPK is:
 - Allosterically activated by AMP
 - Covalently activated by phosphorylation via several kinases
 - At least one of these AMPK kinases is activated by cAMP-dependent protein kinase A (PKA)

- Hormonal regulation:
 - In the presence of counterregulatory hormones (e.g., epinephrine and glucagon):
 - ACC is phosphorylated and inactive
 - In the presence of insulin:
 - ACC is dephosphorylated and active
- Note: This regulation is analogous to glycogen synthase regulation

2. Acetyl CoA Carboxylase Long-Term Regulation

- Prolonged consumption of excess-calorie diet (especially high-carbohydrate, low-fat):
 - Causes increase in ACC synthesis
 - Leads to increased fatty acid synthesis
- Low-calorie or high-fat, low-carbohydrate diet:
 - Has the opposite effect (decreases ACC synthesis)

- Regulatory pathways:
 - ACC synthesis is upregulated by:
 - Glucose → via carbohydrate response element-binding protein (ChREBP)
 - Insulin → via sterol regulatory element-binding protein-1c (SREBP-1c)
- Fatty acid synthase (FAS):
 - Is similarly regulated

Metformin and ACC Regulation

- Metformin (used for type 2 diabetes):
 - Lowers plasma TAG by:
 - Activating AMPK, which:
 - Inhibits ACC activity (via phosphorylation)
 - Decreases ACC and FAS expression (by reducing SREBP-1c)
 - Lowers blood glucose by:
 - Increasing AMPK-mediated glucose uptake by muscle

C. Eukaryotic Fatty Acid Synthase (FAS)

- The remaining series of reactions in fatty acid synthesis in eukaryotes:
 - Catalyzed by fatty acid synthase (FAS)
- FAS characteristics:
 - Multifunctional, homodimeric enzyme
 - Each FAS monomer:
 - Is a multicatalytic polypeptide
 - Contains six different enzymic domains
 - Contains an acyl carrier protein (ACP) domain with 4'-phosphopantetheine
- Function of FAS:
 - Adds two carbons from malonyl CoA to the carboxyl end of a series of acyl acceptors

4'-Phosphopantetheine (ACP Component)

- 4'-Phosphopantetheine:
 - A derivative of pantothenic acid (vitamin B5)
 - Carries acyl units on its terminal thiol (-SH) group
 - Presents the acyl units to the catalytic domains of FAS during fatty acid synthesis
 - Also a component of CoA

Sequential Steps of Fatty Acid Chain Elongation

Step 1:

- An acetyl group is transferred from acetyl CoA to the -SH group of the ACP domain

Step 2:

- This two-carbon fragment is then transferred to a temporary holding site

Step 3:

- The now-vacant ACP accepts a three-carbon malonyl group from malonyl CoA

Step 4:

- The acetyl group (on the cysteine residue) condenses with the malonyl group (on ACP)
- CO_2 is released — this CO_2 was originally added by acetyl CoA carboxylase (ACC)
- The result is a four-carbon unit attached to the ACP domain

Note: The next three reactions (steps 5-7) convert the 3-ketoacyl group into a saturated acyl group

Involve two NADPH-requiring reductions and one dehydration step

Step 5

- The keto group is reduced to an alcohol

Step 6

- A molecule of water is removed

- This creates a trans double bond between:
 - Carbon 2 (α -carbon)
 - Carbon 3 (β -carbon)

Step 7

- The double bond is reduced
- Produces a four-carbon group (butyryl):
 - Three terminal carbons are fully saturated
 - Group remains attached to the ACP domain

Cycle Repeats (Marked by Asterisk *)

- Repetition begins with:
 - Transfer of butyryl unit from ACP to cysteine residue [2*]
 - Attachment of a malonyl group to ACP [3*]
 - Condensation of the two groups \rightarrow CO_2 is released [4*]

- The newly formed carbon chain then undergoes:
 - Reduction of β -carbonyl [5*]
 - Dehydration [6*]
 - Reduction [7*]
- This sequence generates hexanoyl-ACP

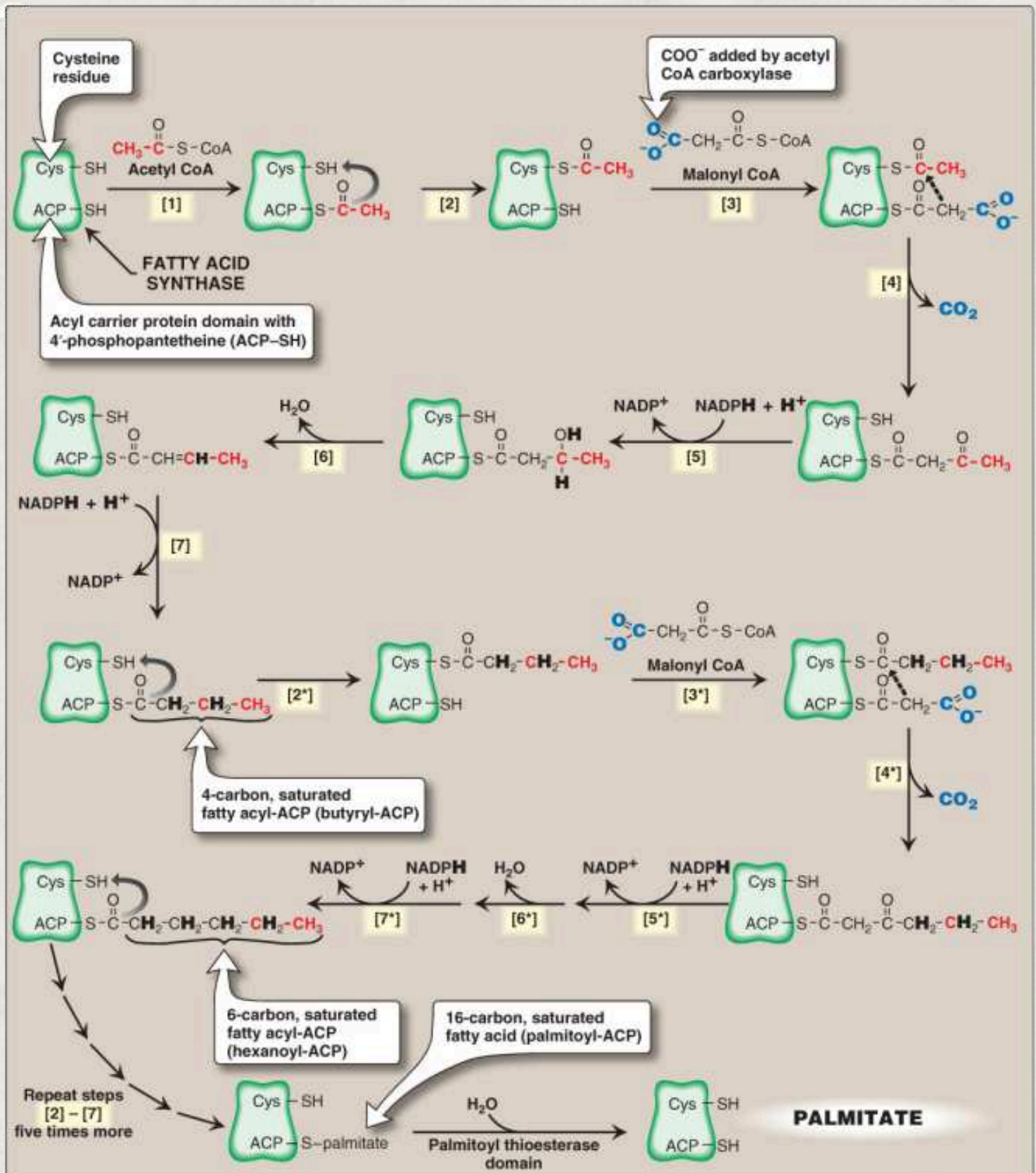
Elongation Process

- The cycle continues, repeating steps [2*] through [7*]
- Continues until the fatty acid chain reaches 16 carbons in length

Final Step of FAS

- The final catalytic activity of FAS:
 - Cleaves the thioester bond
 - Releases a fully saturated molecule of palmitate (16:0)

Synthesis of Palmitate (16:0) by Multifunctional Fatty Acid Synthase



Important Note:

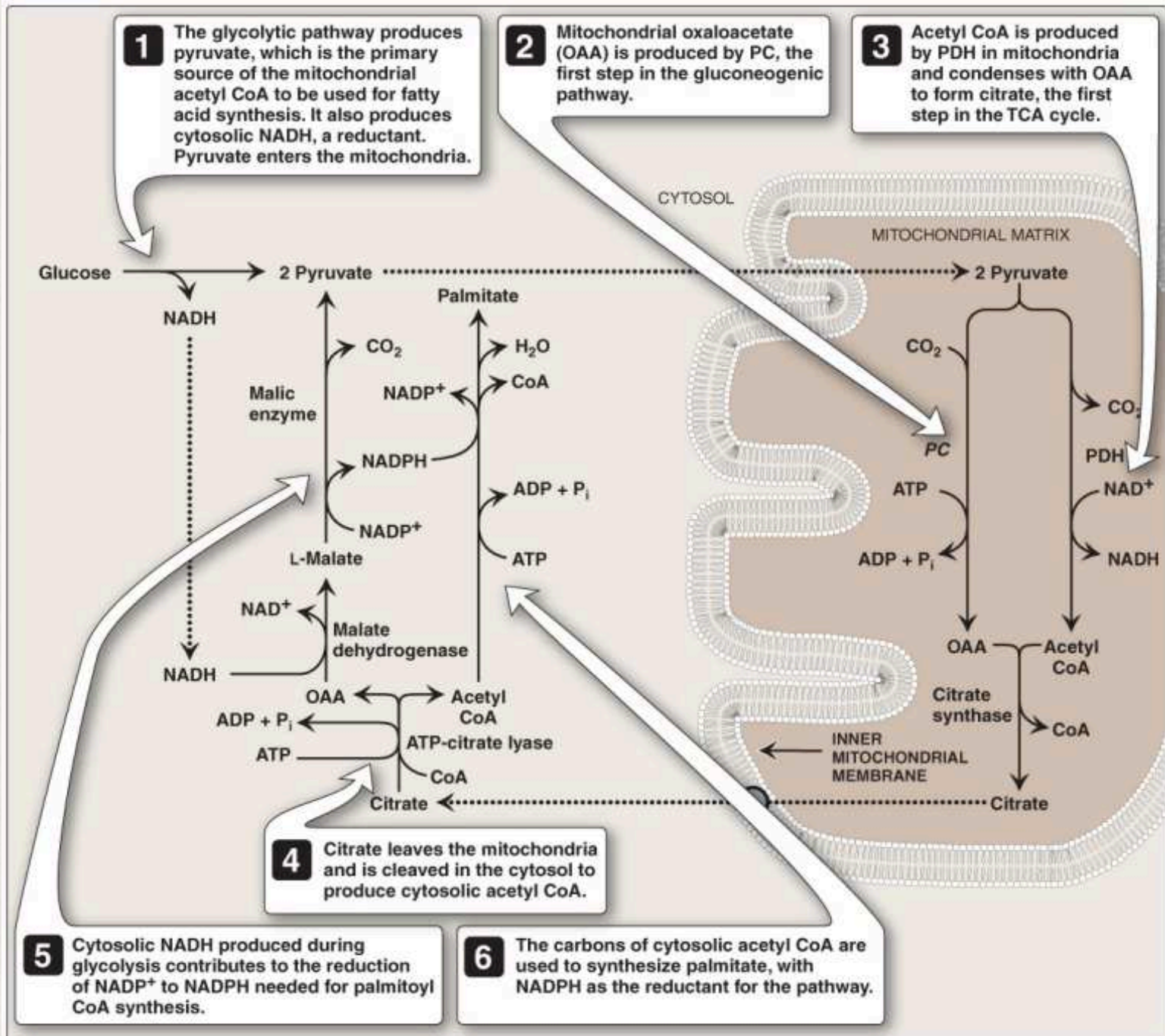
- All carbons in palmitic acid come from malonyl CoA
 - Except the two carbons donated by the original acetyl CoA (the first acyl acceptor)
 - These two are located at the methyl (ω) end of the fatty acid
- This highlights the rate-limiting nature of the ACC reaction
- Shorter-length fatty acids:
 - Are produced only in the lactating mammary gland

D. Reductant Sources

- Synthesis of one palmitate requires:
 - 14 NADPH (reductant/reducing agent)
- Major source of NADPH:
 - Pentose phosphate pathway
 - Two NADPH are produced per molecule of glucose 6-phosphate entering the pathway

- Additional cytosolic NADPH production:
 - From the conversion of malate to pyruvate
 - Malate is oxidized and decarboxylated by:
 - Cytosolic malic enzyme
 - (NADP⁺-dependent malate dehydrogenase)
 - Produces:
 - Cytosolic NADPH
 - CO₂
- Note:
 - Malate can arise from reduction of oxaloacetate (OAA) by:
 - Cytosolic NADH-dependent malate dehydrogenase
- Source of cytosolic NADH:
 - Glycolysis
- Source of OAA:
 - From citrate cleavage by ATP citrate lyase

Interrelationship between Glucose Metabolism and Palmitate Synthesis



E. Further Elongation

- Palmitate (16:0):
 - A 16-carbon, fully saturated long-chain fatty acid (LCFA)
 - It is the primary end product of FAS activity
- Further elongation of palmitate:
 - Occurs by addition of two-carbon units to the carboxylate end
 - Takes place primarily in the smooth endoplasmic reticulum (SER)
- Elongation process:
 - Requires a system of separate enzymes
 - Not a multifunctional enzyme like FAS
 - Malonyl CoA serves as the two-carbon donor
 - NADPH supplies the electrons

- Brain-specific elongation:
 - The brain has additional elongation capabilities
 - Can produce very long-chain fatty acids (VLCFA):
 - VLCFA = fatty acids over 22 carbons
 - Required for synthesis of brain lipids

F. Chain Desaturation

- Enzymes involved:
 - Fatty acyl CoA desaturases, located in the smooth endoplasmic reticulum (SER)
- Function:
 - Desaturate long-chain fatty acids (LCFA) by adding cis double bonds
- Desaturation reaction requirements:
 - Oxygen (O_2)
 - NADH
 - Cytochrome b_5
 - FAD-linked reductase (flavin adenine dinucleotide-linked)

- Reaction mechanism:
 - Both the fatty acid and NADH are oxidized
 - O_2 is reduced to H_2O
- Typical desaturation site:
 - Between carbons 9 and 10
 - Produces:
 - Oleic acid, 18:1(9)
 - Palmitoleic acid, 16:1(9) (in smaller amounts)
- Polyunsaturated fatty acid (PUFA) synthesis:
 - Achieved through additional desaturation combined with elongation
- Human desaturase enzymes present:
 - Carbon 9, 6, 5, and 4 desaturases
- Limitation:
 - Humans lack the ability to insert double bonds between carbon 10 and the ω end of the chain

- Nutritional consequence:
 - Explains the essentiality of:
 - ω -6 linoleic acid
 - ω -3 linolenic acid

G. Storage as TAG Components

- Mono-, di-, and triacylglycerols:
 - Comprise one, two, or three fatty acids esterified to a molecule of glycerol
- Esterification:
 - Fatty acids are esterified through their carboxyl groups
 - Results in:
 - Loss of negative charge
 - Formation of neutral fat
- Note:
 - An acylglycerol that is:
 - Solid at room temperature = fat
 - Liquid at room temperature = oil

1. Arrangement of Fatty Acids in TAG

- Triacylglycerol (TAG) = Three fatty acids esterified to a glycerol molecule
- Fatty acids are usually not of the same type:
 - Carbon 1 → typically saturated
 - Carbon 2 → typically unsaturated
 - Carbon 3 → can be either saturated or unsaturated
- Note:
 - Presence of unsaturated fatty acid(s) lowers the T_m (melting temperature) of the lipid

2. Triacylglycerol Storage and Function

- Solubility:
 - TAGs are only slightly soluble in water
 - Cannot form stable micelles on their own

- Storage in white adipocytes:
 - TAGs coalesce to form large oily, nearly anhydrous droplets
 - Stored in cytosolic lipid droplets
 - These droplets are the major energy reserve of the body
- Note:
 - TAGs in brown adipocytes are used for heat production through nonshivering thermogenesis

3. Glycerol 3-Phosphate Synthesis

- Glycerol 3-phosphate:
 - Is the initial acceptor of fatty acids during TAG synthesis
- Two major pathways for glycerol 3-phosphate production:

A. From Glucose (in Liver and Adipose Tissue)

- Glucose \rightarrow DHAP via glycolytic pathway

- DHAP is reduced by glycerol 3-phosphate dehydrogenase → Glycerol 3-phosphate

B. From Free Glycerol (in Liver only)

- Free glycerol is converted to glycerol 3-phosphate
 - Enzyme: Glycerol kinase
- Note:
 - This pathway is absent in adipose tissue

Additional Notes

- Glyceroneogenesis:
 - A third process for glycerol 3-phosphate synthesis
- Insulin Dependence in Adipocytes:
 - GLUT-4 (glucose transporter in adipocytes) is insulin dependent

- When plasma glucose is low, adipocytes:
 - Have limited ability to synthesize glycerol phosphate
 - Therefore, cannot produce TAG de novo

4. Fatty Acid Activation

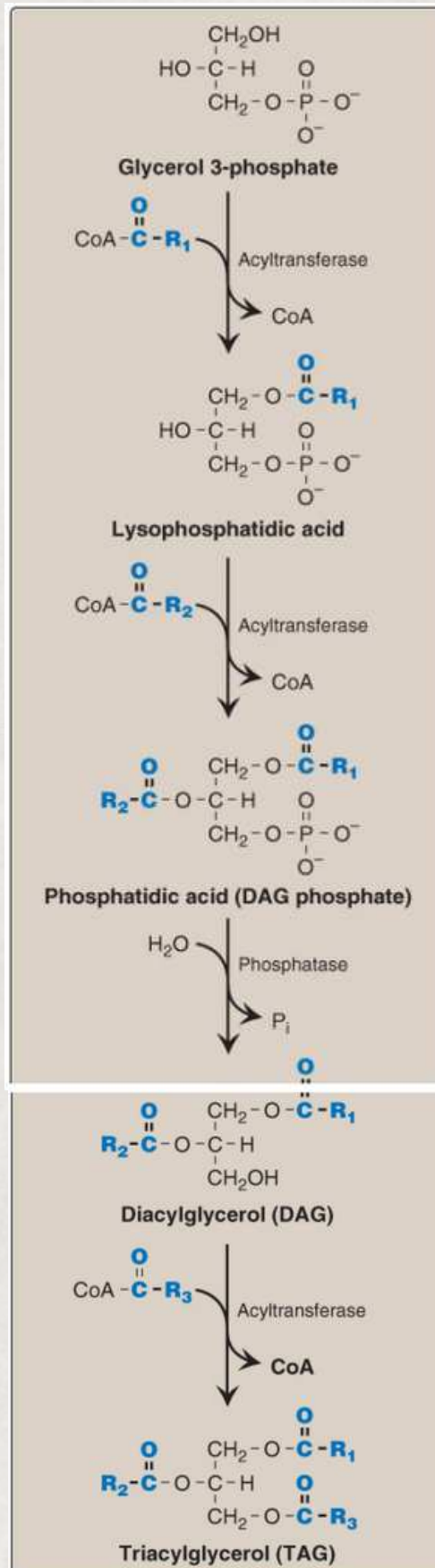
- A free fatty acid (FFA) must be converted to its activated form before participating in metabolic processes such as TAG synthesis
- Activated form:
 - Fatty acid bound to Coenzyme A (CoA) via a thioester bond
- Catalyzing enzyme:
 - A family of fatty acyl CoA synthetases (also called thiokinases)

5. Triacylglycerol (TAG) Synthesis

- Pathway begins from glycerol 3-phosphate

- Involves four reactions:
 - Addition of 1st fatty acid from fatty acyl CoA
 - Addition of 2nd fatty acid from fatty acyl CoA
 - Removal of phosphate group
 - Addition of third fatty acid

Synthesis of TAG



H. Triacylglycerol Fate in Liver and Adipose Tissue

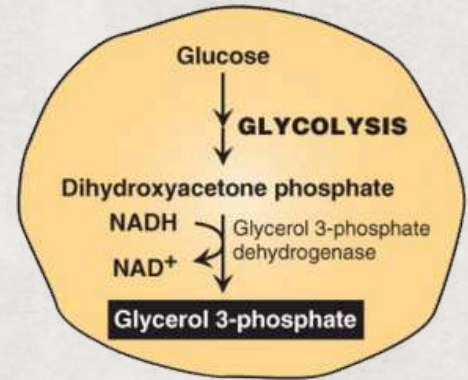
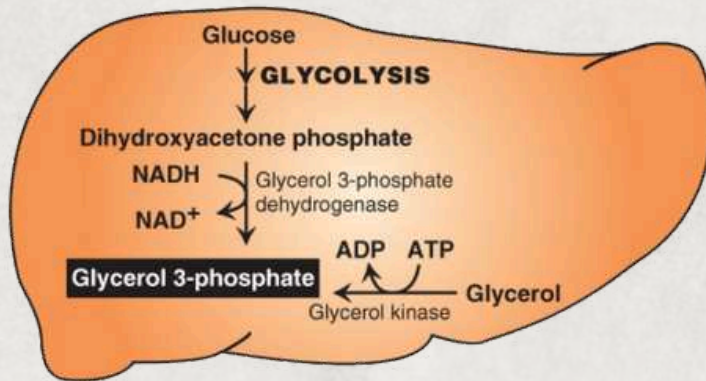
In White Adipose Tissue (WAT)

- TAG storage:
 - Stored as nearly anhydrous fat droplets in the cytosol
- Fat droplet structure:
 - Coated with perilipins:
 - A family of proteins that sequester and protect TAG from lipolysis
 - Activated only when the body requires fatty acids for fuel
- Clinical relevance:
 - Perilipins may play a role in:
 - Type 2 diabetes
 - Atherosclerosis
 - Cardiovascular disease

In Liver

- Healthy liver:
 - Stores little TAG
- Most TAG in liver is:
 - Exported, packaged with other lipids and apolipoproteins
 - Forms very-low-density lipoproteins (VLDL)
- VLDL characteristics:
 - Nascent VLDL are secreted directly into the blood
 - Mature in blood and deliver endogenously derived lipids to peripheral tissues
- Note:
 - Chylomicrons carry dietary (exogenous) lipids

Pathways for Production of Glycerol 3-phosphate in Liver and Adipose Tissue



IV. Fat Mobilization And Fatty Acid Oxidation

- Stored fatty acids in white adipose tissue (WAT):
 - Stored as neutral triacylglycerol (TAG)
 - Serve as the body's major fuel storage reserve
- Why TAGs are efficient energy stores:
 - Highly reduced
 - Largely anhydrous

- Energy yield from complete oxidation:
 - Fatty acids $\rightarrow \text{CO}_2 + \text{H}_2\text{O} = 9 \text{ kcal/g fat}$
 - Compared to:
 - 4 kcal/g for protein or carbohydrate

A. Fatty Acid Release from Fat

- Lipolysis:
 - Required to release free fatty acids (FFA) and glycerol from TAG
 - Carried out by perilipins and lipases
- Steps of lipolysis:
 - a. Initiated by adipose triglyceride lipase (ATGL):
 - Produces diacylglycerol (DAG)
 - DAG is the preferred substrate for:
 - a. Hormone-sensitive lipase (HSL):
 - Produces monoacylglycerol (MAG)
 - a. MAG lipase acts on MAG to complete hydrolysis

1. Regulation of Perilipins and HSL

- PKA-mediated phosphorylation:
 - Both perilipins and HSL are phosphorylated by PKA
 - PKA is activated by cAMP, which is produced when:
 - Catecholamines (e.g., epinephrine) bind to β -adrenergic receptors
 - Activate adenylyl cyclase
- Effects of phosphorylation:
 - Phosphorylated perilipin allows translocation and binding of phosphorylated (active) HSL to lipid droplet
 - Process similar to glycogen phosphorylase activation
- cAMP cascade effects:
 - Inhibits ACC (acetyl CoA carboxylase)
 - Turns off fatty acid synthesis
 - Turns on TAG degradation

- Insulin effects:
 - High plasma insulin:
 - Causes dephosphorylation and inactivation of HSL
 - Suppresses ATGL expression

2. Fate of Glycerol

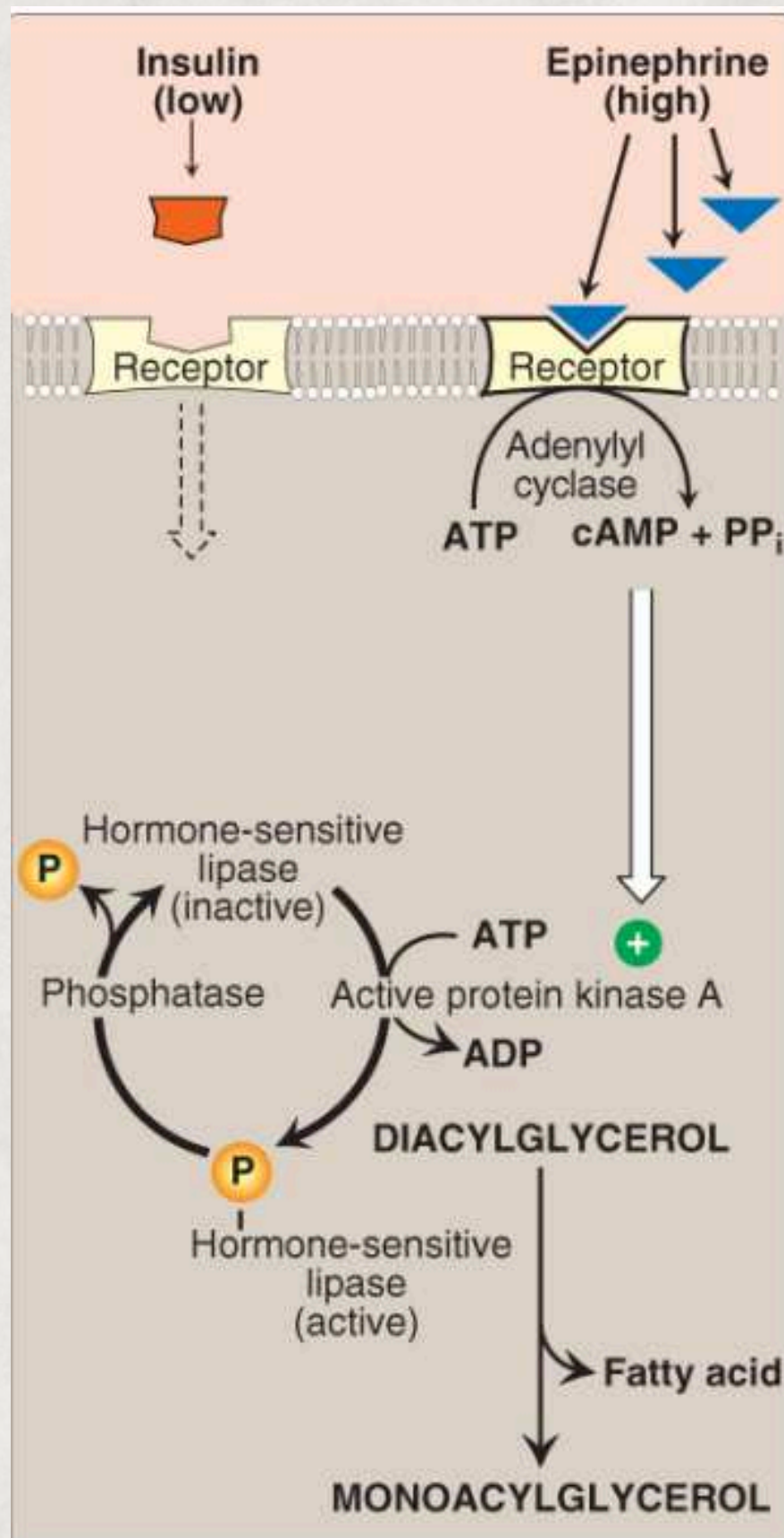
- Adipocytes lack glycerol kinase:
 - Cannot metabolize glycerol
- Glycerol is transported to the liver:
 - Liver contains glycerol kinase
 - Converts glycerol → glycerol 3-phosphate
- Fate of glycerol 3-phosphate in liver:
 - Used to form TAG
 - Or converted to DHAP by glycerol 3-phosphate dehydrogenase (reverse reaction)
 - DHAP can enter:
 - Glycolysis
 - Gluconeogenesis

3. Fate of Fatty Acids

- Free fatty acids (FFA):
 - Exit adipocyte → enter blood → bind serum albumin
- FFA transported to:
 - Tissues like muscle
 - Inside cells:
 - Activated to CoA derivatives
 - Oxidized for energy in mitochondria
- Tissues that do not utilize plasma FFA:
 - Red blood cells (RBCs):
 - No mitochondria
 - Brain:
 - Uses little/no fatty acids
 - Reason not well understood
- Reesterification of fatty acids:
 - >50% of fatty acids released from adipose TAG are reesterified to glycerol 3-phosphate

- Problem: WAT does not express glycerol kinase
- Solution: Glycerol 3-phosphate is produced by glyceroneogenesis aka incomplete gluconeogenesis:
 - Incomplete gluconeogenesis:
 - Pyruvate → oxaloacetate (OAA) via pyruvate carboxylase
 - OAA → phosphoenolpyruvate (PEP) via PEP carboxykinase
 - PEP → DHAP (via shared glycolytic/gluconeogenic steps)
 - DHAP → glycerol 3-phosphate
- Clinical relevance:
 - Reesterification reduces plasma FFA levels
 - Elevated FFA levels are associated with:
 - Insulin resistance
 - Type 2 diabetes
 - Obesity

Hormonal Regulation of Diacylglycerol Degradation in the Adipocyte



B. Fatty Acid β -Oxidation

- Definition:

- The major pathway for catabolism of fatty acids
- Occurs in the mitochondria
- Process: Two-carbon fragments are successively removed from the carboxyl end of fatty acyl CoA

- Products:

- Acetyl CoA
- NADH
- FADH₂

I. Long-Chain Fatty Acid Transport into Cytosol and Mitochondria

A. Cellular Uptake of Fatty Acids

- Mechanisms:

- Passive diffusion
- Lipid transport proteins:
 - Fatty acid translocase (FAT)
 - Fatty acid-binding protein (FABP)
 - Fatty acid transport protein (FATP)

- Long-chain fatty acids (LCFA):
 - Taken up specifically by FATP

B. Activation of Fatty Acids

- Once LCFA enter the cell:
 - Converted in the cytosol to fatty acyl CoA
- Enzyme: Long-chain fatty acyl CoA synthetase (thiokinase)
 - Located on the outer mitochondrial membrane

C. Mitochondrial Transport Requirement

- β -Oxidation occurs in the mitochondrial matrix
- Inner mitochondrial membrane is impermeable to CoA
- Therefore, the acyl group must be transported via a specialized carrier

D. Carnitine Shuttle (Rate-limiting Transport Process)

- Carrier: Carnitine
- Name of process: Carnitine shuttle

a. Translocation Steps (Three Key Steps)

1. Transfer of acyl group to carnitine:

- Enzyme: Carnitine palmitoyltransferase I (CPT-I)
 - Also known as carnitine acyltransferase I (CAT-I)
 - Located on the outer mitochondrial membrane
- Reaction:
 - Acyl group from CoA \rightarrow Carnitine
 - Forms acylcarnitine
 - Regenerates free CoA

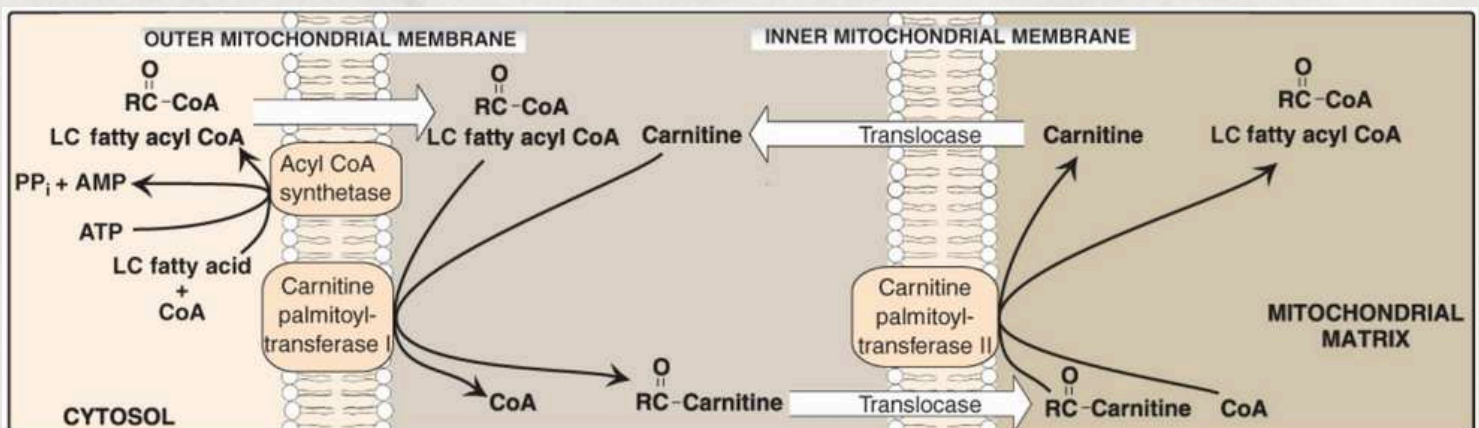
2. Transport across inner mitochondrial membrane:

- Transporter: Carnitine-acylcarnitine translocase
- Exchanges acylcarnitine into matrix and free carnitine out

3. Transfer of acyl group back to CoA:

- Enzyme: Carnitine palmitoyltransferase II (CPT-II)
 - Also known as carnitine acyltransferase II (CAT-II)
 - Located on the inner mitochondrial membrane
- Reaction:
 - Acyl group from carnitine \rightarrow CoA in matrix
 - Regenerates free carnitine

Carnitine Shuttle



b. Carnitine Shuttle Inhibitor

- Inhibitor: Malonyl CoA
- Inhibits CPT-I (Carnitine Palmitoyltransferase I)
- Prevents entry of long-chain acyl groups into the mitochondrial matrix

- Functional Significance:
 - When fatty acid synthesis is occurring in the cytosol (indicated by presence of malonyl CoA):
 - The newly made palmitate cannot be transferred into mitochondria
 - Prevents degradation of newly synthesized fatty acids
- Tissue-Specific Notes:
 - Muscle tissue:
 - Does not synthesize fatty acids
 - But contains mitochondrial isozyme of ACC (ACC2)
 - Allows regulation of β -oxidation
 - Liver:
 - Contains both isozymes of ACC
- Additional Regulation:
 - Regulated by the acetyl CoA / CoA ratio:
 - As the ratio increases, the CoA-requiring thiolase reaction decreases

c. Carnitine Sources

- Dietary source:
 - Carnitine is found primarily in meat products
- Endogenous synthesis:
 - Synthesized from lysine and methionine
 - Enzymatic pathway is present in:
 - Liver
 - Kidneys
 - Not synthesized in:
 - Skeletal muscle
 - Cardiac muscle
- Tissue dependency:
 - Skeletal and cardiac muscles are:
 - Totally dependent on carnitine uptake from:
 - Endogenous synthesis (via blood)
 - Diet
- Distribution note:
 - Skeletal muscle contains 97% of all carnitine in the body

Carnitine Transport into Cells

- Transport mechanism: Carnitine enters cells via carnitine transporters
- High-affinity transporter (in heart, muscle, kidney):
 - Organic cation transporter novel 2 (OCTN2)
- Liver transporter:
 - Different, low-affinity, high-capacity carnitine transporter
- Primary carnitine deficiency:
 - Due to defect in OCTN2
 - Leads to:
 - Urinary loss of carnitine
 - Low serum and cellular carnitine levels

d. Carnitine Deficiencies

1. Primary Carnitine Deficiency

- Cause: Defects in membrane transporter

- Prevents carnitine uptake by:
 - Cardiac muscle
 - Skeletal muscle
 - Kidneys
- Effect: Carnitine is excreted in urine
- Treatment: Carnitine supplementation

2. Secondary Carnitine Deficiency

- Primary cause: Defects in fatty acid oxidation
 - Leads to accumulation of acylcarnitines
 - Acylcarnitines are excreted in urine
 - Result: Decreased carnitine availability
- Acquired causes:
 - Liver disease → Decreased carnitine synthesis
 - Valproic acid (antiseizure drug) → Decreased renal reabsorption

3. CPT-I and CPT-II Deficiencies (Defects in mitochondrial oxidation)

- CPT-I deficiency:

- Affects the liver
- Impaired LCFA usage for fuel
- Severe effect on hepatic glucose synthesis during fasting (an endergonic process)
- Leads to:
 - Severe hypoglycemia
 - Coma
 - Death

- CPT-II deficiency:

- Affects:
 - Liver
 - Cardiac muscle
 - Skeletal muscle
- Most common form: Affects skeletal muscle (least severe)
 - Presents as:
 - Muscle weakness
 - Myoglobinemia after prolonged exercise

- Treatment:
 - Avoid fasting
 - High-carbohydrate, low-fat diet
 - Supplement with medium-chain TAG

2. Entry of Shorter-Chain Fatty Acids into Mitochondria

- Fatty acids ≤ 12 carbons:
 - Can cross inner mitochondrial membrane without carnitine or CPT system
- Once inside mitochondria:
 - Activated to CoA derivatives by matrix enzymes
 - Undergo oxidation
- Note:
 - Medium-chain fatty acids:
 - Plentiful in human milk
 - Oxidation is independent of CPT-I
 - Therefore, malonyl CoA does not inhibit their oxidation

3. β -Oxidation Reactions

- Overview:

- Involves four sequential reactions at the β -carbon (carbon 3)
- Result: Shortens the fatty acid by two carbons at the carboxylate end

Steps of β -Oxidation:

1. Oxidation:

- Produces FADH_2

2. Hydration

3. Second oxidation:

- Produces NADH

4. CoA-dependent thiolytic cleavage:

- Releases one molecule of acetyl CoA

- Enzyme Specificity:
 - Each step is catalyzed by enzymes with chain-length specificity
- Note:
 - For long-chain fatty acids (LCFA), the last three steps are catalyzed by a trifunctional protein

Cycle Repetition:

- For saturated fatty acids with even-numbered carbon chains:
 - Number of cycles = $(n/2) - 1$ (where n = number of carbons)
- Each cycle yields:
 - 1 acetyl CoA
 - 1 NADH
 - 1 FADH_2

- Final cycle produces:

- 2 acetyl CoA

Fates of Acetyl CoA and Reduced Coenzymes:

- Acetyl CoA can be:

- Oxidized
 - Used in hepatic ketogenesis

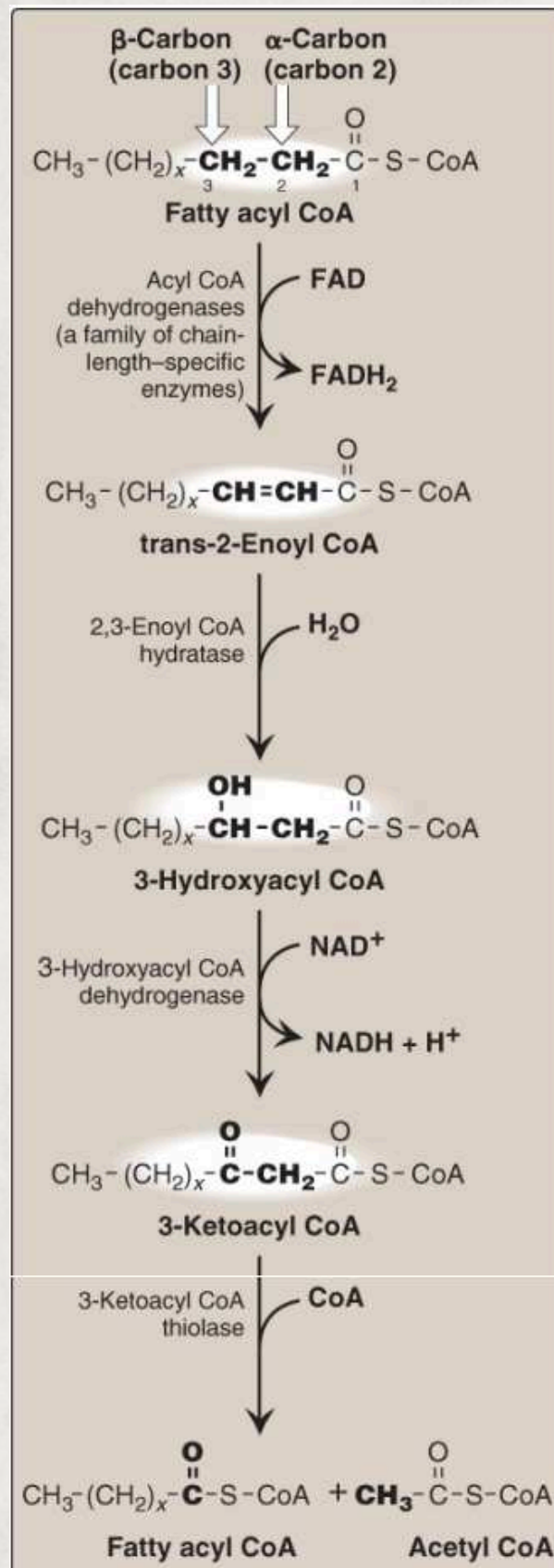
- Reduced coenzymes:

- NADH oxidized by ETC Complex I
 - FADH₂ oxidized by coenzyme Q

- Note:

- Acetyl CoA is a positive allosteric effector of pyruvate carboxylase (see Chapter 10)
 - This links fatty acid oxidation with gluconeogenesis

Enzymes involved in the β -oxidation of fatty acyl coenzyme A (CoA)



4. β -Oxidation Energy Yield

- Fatty Acid β -Oxidation Produces High Energy:
 - The energy yield from fatty acid β -oxidation is very high.

Example: Complete Oxidation of Palmitoyl CoA ($C_{16:0}$)

- Yields the following:
 - 8 acetyl CoA
 - 7 NADH
 - 7 $FADH_2$
- These products can generate a total of 131 ATP.
- Activation of the fatty acid consumes 2 ATP, so:
 - Net ATP yield = 129 ATP.

Energy Calculation for even-numbered saturated fatty acid

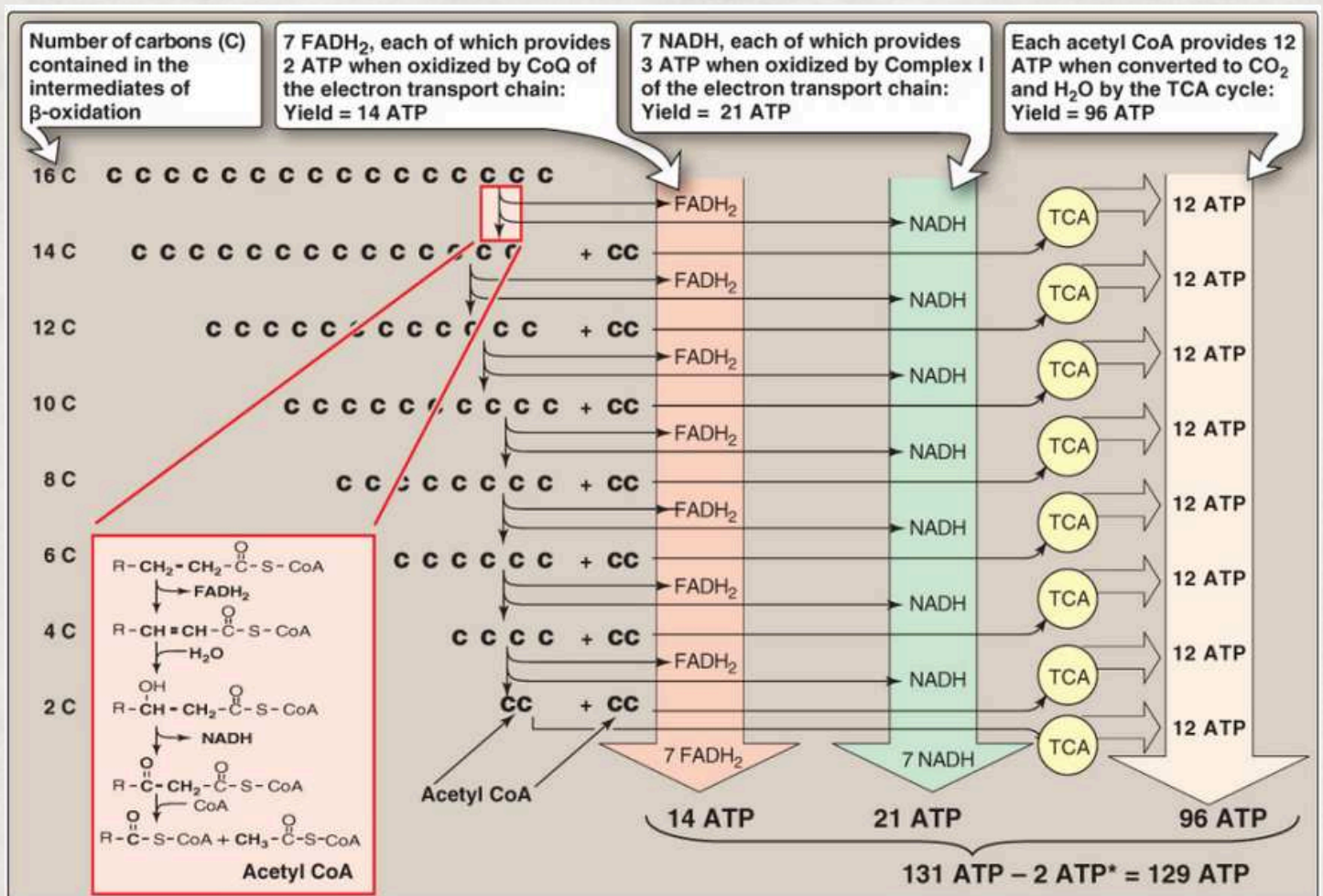
- Let n = number of carbon atoms in the saturated fatty acid (even-numbered).

- Use the following steps:
 - Number of β -oxidation cycles = $(n / 2) - 1$
 - Each cycle removes 2 carbons as 1 acetyl CoA.
 - So for C_{16} , $(16 / 2) - 1 = 7$ cycles.
 - Acetyl CoA produced = $(n / 2)$
 - One additional acetyl CoA is produced in the final cycle, when a 4-carbon fatty acyl CoA is cleaved into two acetyl CoA.
 - So for palmitate (C_{16}), you get 8 acetyl CoA.
 - NADH and $FADH_2$ = number of β -oxidation cycles
 - Each of the 7 cycles produces:
 - 1 NADH = 7 NADH (total)
 - 1 $FADH_2$ = 7 $FADH_2$ (total)

ATP Yield Summary for Palmitate (C₁₆)

Molecule	Quantity	ATP per molecule	Total ATP
Acetyl CoA	8	12	96
NADH	7	3	21
FADH ₂	7	2	14
Subtotal	—	—	131 ATP
Activation cost	—	—	-2 ATP
Grand total	—	—	Net: 129 ATP

Summary of the Energy Yield from the Oxidation of Palmitoyl Coenzyme A (CoA) (16 carbons)



Note: Activation of palmitate to palmitoyl CoA requires the equivalent of 2 ATP [$\text{ATP} \rightarrow \text{AMP} + \text{PPi}$]

S. Medium-Chain Fatty Acyl CoA Dehydrogenase (MCAD) Deficiency

- Fatty Acyl CoA Dehydrogenases in Mitochondria:
 - There are four species of fatty acyl CoA dehydrogenases.
 - Each has distinct but overlapping specificity for:
 - Short-chain fatty acids
 - Medium-chain fatty acids
 - Long-chain fatty acids
 - Very-long-chain fatty acids

MCAD Deficiency

- Most common inborn error of β -oxidation
- Genetic basis:
 - Autosomal recessive disorder
- Prevalence:
 - Found in 1:14,000 births worldwide
 - Higher incidence in Caucasians of Northern European descent

Pathophysiology

- Impaired oxidation of fatty acids with 6 to 10 carbons
- Results in:
 - Decreased production of acetyl CoA
 - Increased reliance on glucose for energy
 - Leads to hypoketotic hypoglycemia

Laboratory Findings

- Urine studies show:
 - Accumulation of medium-chain acyl carnitines
 - Presence of medium-chain dicarboxylic acids

Treatment:

- Avoidance of fasting

6. Oxidation of Fatty Acids with an Odd Number of Carbons

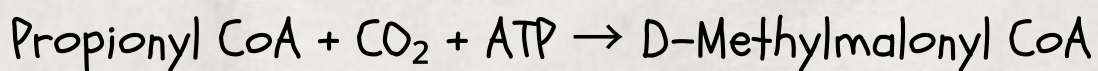
- Follows same β -oxidation steps as even-numbered fatty acids.

- Continues until final three-carbon fragment: Propionyl CoA.
 - Also produced during the metabolism of some amino acids

Three-Step Metabolism of Propionyl CoA:

a. D-Methylmalonyl CoA Synthesis

- Enzyme: Propionyl CoA carboxylase
- Reaction:



- Cofactors:
 - Biotin (Vitamin B7) — same as in acetyl CoA carboxylase (ACC)
 - ATP

b. L-Methylmalonyl CoA Formation

- Enzyme: Methylmalonyl CoA racemase
- Converts D-isomer to L-isomer

c. Succinyl CoA Synthesis

- Enzyme: Methylmalonyl CoA mutase
- Reaction:



- Cofactor: Vitamin B12 (deoxyadenosylcobalamin)
 - One of only two B12-dependent enzymes in humans

Clinical Correlation: Gluconeogenesis

- Succinyl CoA enters the TCA cycle
- → Only known glucogenic product of fatty acid oxidation

Vitamin B12-Related Notes

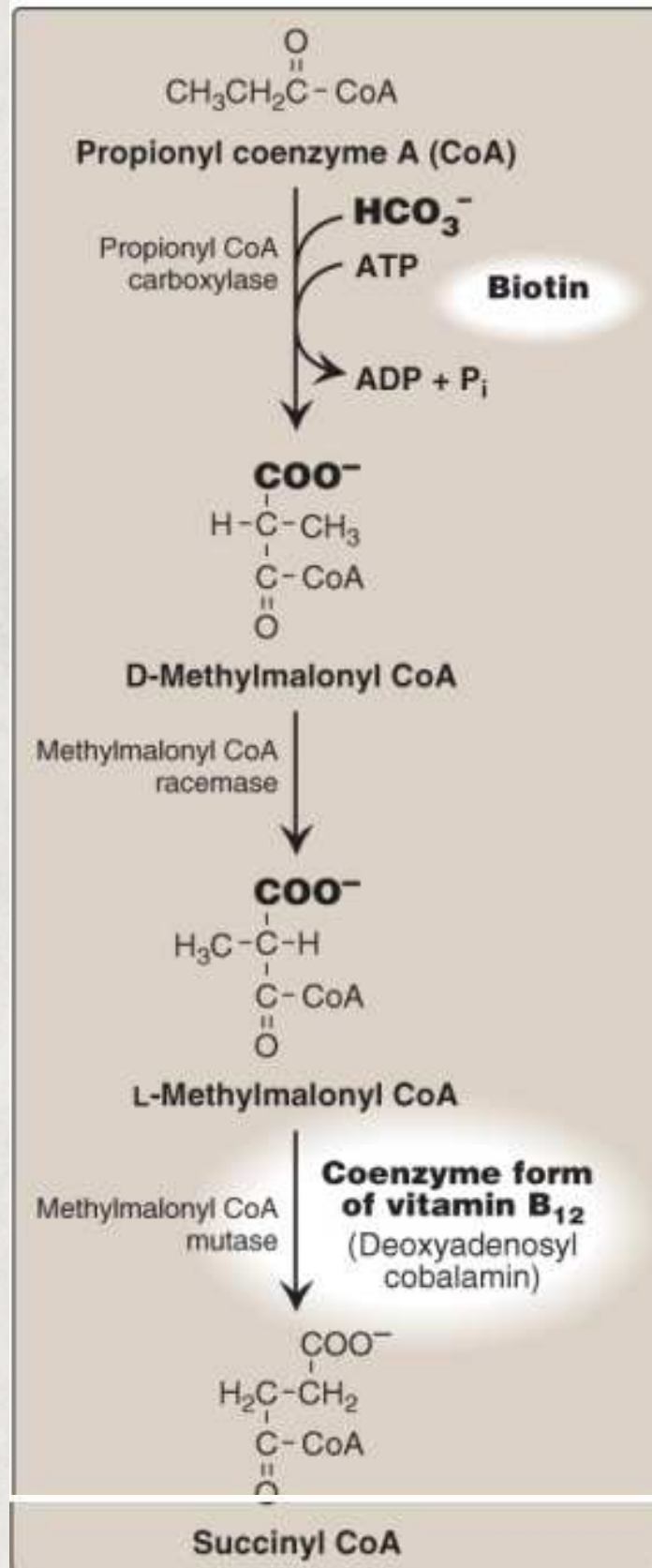
- Second B12-dependent enzyme: Methionine synthase
 - Converts homocysteine → methionine
 - Essential for:
 - Folate recycling
 - Conversion of B12 to coenzyme form

- Deficiency consequences:
 - Early: Hematologic abnormalities (similar to folate deficiency)
 - Late: Neurologic symptoms: paresthesias, numbness, ataxia
- Diagnostic markers:
 - ↑ Methylmalonic acid (MMA) in serum = B12 deficiency
 - Both propionic acid and MMA appear in urine
 - MMA helps differentiate B12 deficiency from folate deficiency

Inherited Disorders

- Methylmalonic acidemia/aciduria:
 - Caused by:
 - Defects in methylmalonyl CoA mutase
 - Deficiency of methionine synthase

Metabolism of propionyl CoA



7. β -Oxidation of Unsaturated Fatty Acids

- Challenge: Unsaturated FA oxidation forms intermediates that cannot be acted on by 2,3-enoyl CoA hydratase (step 2 of β -oxidation).

a. Odd-Numbered Double Bonds

- Example: Oleic acid = 18:1(9)
- After 3 β -oxidation rounds \rightarrow Intermediate: 3-cis-enoyl CoA
- Solution:
 - Enzyme: 3,2-enoyl CoA isomerase
 - Converts 3-cis \rightarrow 2-trans, the correct substrate for hydratase

b. Even-Numbered Double Bonds

- Example: Linoleic acid = 18:2(9,12)
- Problem: Produces 2,4-dienoyl CoA intermediate, which is not a substrate

- Solution:
 - Enzymes required:
 - 3,2-enoyl CoA isomerase
 - 2,4-dienoyl CoA reductase (requires NADPH)
- Note: Unsaturated FAs are less reduced → fewer NADH/FADH₂ → less ATP yield

8. Peroxisomal β -Oxidation (for VLCFA ≥ 22 carbons)

- VLCFAs are first oxidized in peroxisomes, not mitochondria.

a. Activation

- Peroxisomes contain acyl CoA synthetases for VLCFA.
- VLCFA-CoA is formed and enters peroxisomal β -oxidation.

b. Comparison: Mitochondrial vs. Peroxisomal β -Oxidation

Feature	Mitochondrial β -Oxidation	Peroxisomal β -Oxidation
Primary Substrate	Short-, medium-, and long-chain fatty acids (≤ 22 C)	Very-long-chain fatty acids (VLCFA ≥ 22 C)
Site of Activation	Cytosol/outer mitochondrial membrane	Peroxisome
Initial Dehydrogenation Enzyme	Acyl CoA dehydrogenase (produces FADH_2)	Acyl CoA oxidase (produces FADH_2 used to reduce $\text{O}_2 \rightarrow \text{H}_2\text{O}_2$)
Electron Acceptor	$\text{ETF} \rightarrow \text{ETC} \rightarrow \text{ATP}$	O_2 (no ATP generated directly)
Hydrogen Peroxide Formation	Not formed	H_2O_2 formed; detoxified by catalase to H_2O
End Products	Acetyl CoA, NADH, FADH_2	Shortened fatty acyl-CoA (e.g., octanoyl-CoA) + H_2O_2
Further Oxidation Site	Complete within mitochondria	Requires transfer to mitochondria (via carnitine) for completion
Clinical Relevance	Impaired in mitochondrial disorders	Defective in Zellweger syndrome, X-linked ALD

(ETF = Electron Transfer Flavoprotein)

End Product: Shortened FA (linked to carnitine) → sent to mitochondria for completion of oxidation

c. Clinical Correlations

- Zellweger Syndrome:
 - Cause: Peroxisomal biogenesis defect
 - Result: Impaired import of matrix proteins
 - Consequence: ↑ VLCFA in blood and tissues
- X-linked Adrenoleukodystrophy (ALD):
 - Cause: Defective VLCFA transport into peroxisomes
 - Consequence: Accumulation of VLCFA
 - Affects: Primarily brain and adrenal cortex

C. Peroxisomal α -Oxidation

- Substrate: Phytanic acid
 - A branched-chain fatty acid derived from chlorophyll metabolism
 - Problem: Methyl group on β -carbon prevents β -oxidation

1. α -Oxidation Mechanism

- Step 1: Hydroxylation at α -carbon (carbon-2)
 - Enzyme: Phytanoyl CoA α -hydroxylase (PhyH)
 - Requires peroxisomes
- Step 2: Decarboxylation — carbon 1 released as CO_2
- Step 3: Product: Pristanal (ISC) \rightarrow oxidized to pristanic acid
- Step 4: Pristanic acid-CoA formed \rightarrow undergoes β -oxidation (in peroxisomes)

2. Refsum Disease

- Cause: Autosomal recessive deficiency of PhyH
- Effect: Accumulation of phytanic acid in plasma and tissues
- Symptoms: Neurologic, e.g., peripheral neuropathy, cerebellar ataxia, retinitis pigmentosa

- Treatment:
 - Strict dietary restriction of phytanic acid (avoidance of green vegetables, ruminant fats, and dairy)

3. ω -Oxidation

- Site: Smooth Endoplasmic Reticulum (SER)
- Process: Oxidation at ω -carbon (methyl end)
 - Produces dicarboxylic acids
- Significance: Normally minor \rightarrow upregulated in conditions like MCAD deficiency
 - Acts as backup when β -oxidation is impaired

V. Ketone Bodies: Alternative Fuel For Cells

- Liver Mitochondrial Role:
 - Converts acetyl CoA (from fatty acid oxidation) into ketone bodies.

- Ketone bodies include:
 - Acetoacetate
 - 3-Hydroxybutyrate (β -hydroxybutyrate)
 - Acetone (nonmetabolized)
 - (Note: Acetoacetate and 3-hydroxybutyrate are functional organic acids.)

- Peripheral Use:

- Transported via blood to peripheral tissues.
- Reconverted into acetyl CoA \rightarrow enters TCA cycle for energy.
- Tissues that use ketone bodies:
 - Skeletal muscle
 - Cardiac muscle
 - Intestinal mucosa
 - Renal cortex
 - Brain (during prolonged fasting or high ketone levels)

- Advantages of Ketone Bodies:

a. Water-soluble — no need for lipoproteins or albumin.

b. Formed when hepatic acetyl CoA exceeds liver's oxidative capacity.

c. Utilized in proportion to their plasma concentration by extrahepatic tissues.

- Glucose-Sparing Role:

- Especially important during prolonged fasting.
- Reduces glucose requirement of brain and peripheral tissues.

- Clinical Note:

- Fatty acid oxidation disorders:
 - Lead to hypoketosis (↓ acetyl CoA availability).
 - Lead to hypoglycemia (↑ glucose reliance).

A. Ketone Body Synthesis by the Liver: Ketogenesis

- Trigger:

- Occurs during fasting when liver receives excess FFAs from adipose tissue.

- Resulting Hepatic Changes:
 - \uparrow Acetyl CoA (from β -oxidation):
 - Inhibits pyruvate dehydrogenase.
 - Activates pyruvate carboxylase (PC) \rightarrow OAA \rightarrow used in gluconeogenesis (not TCA).
 - NADH $\uparrow \rightarrow$ OAA \rightarrow malate (see p. 124).
 - Low OAA = acetyl CoA diverted to ketogenesis.
 - Other acetyl CoA sources: Catabolism of ketogenic amino acids.

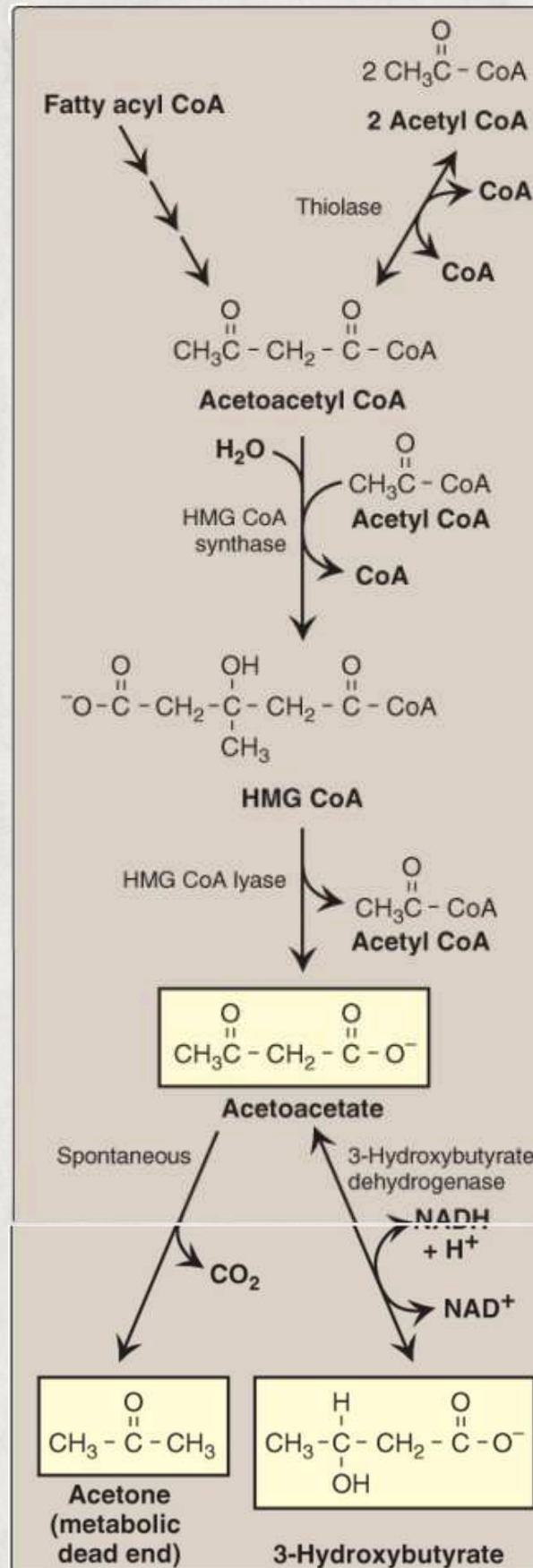
1. HMG CoA Synthesis (Rate-Limiting Step):

- Step 1: Two acetyl CoA molecules \rightarrow acetoacetyl CoA (via reversal of fatty acid thiolase step).
- Step 2: Third acetyl CoA + acetoacetyl CoA \rightarrow HMG CoA (via HMG CoA synthase).
- HMG CoA synthase:
 - Rate-limiting enzyme of ketogenesis.
 - Found in significant levels only in liver mitochondria.
 - (Note: HMG CoA also appears in cytosolic cholesterol synthesis — different cellular location & context.)

2. Ketone Body Formation:

- HMG CoA \rightarrow acetoacetate + acetyl CoA (via HMG CoA lyase).
- Acetoacetate can:
 - Be reduced to 3-hydroxybutyrate (using NADH).
 - Spontaneously decarboxylate to acetone in blood.
- Transport:
 - Not CoA-bound \rightarrow can freely cross mitochondrial membrane.
- Equilibrium between ketones:
 - Controlled by NAD^+/NADH ratio.
 - Fatty acid oxidation \rightarrow low NAD^+/NADH ratio \rightarrow favors 3-hydroxybutyrate formation.

Synthesis of Ketone Bodies



B. Ketone Body Use by the Peripheral Tissues: Ketolysis

- Liver constantly produces low levels of ketone bodies.
- During fasting, production increases to supply energy to peripheral tissues.
- Step 1: Oxidation of 3-hydroxybutyrate
 - 3-Hydroxybutyrate \rightarrow Acetoacetate (via 3-hydroxybutyrate dehydrogenase).
 - NADH is produced in the process.
- Step 2: Activation of acetoacetate
 - Acetoacetate + CoA (from succinyl CoA) \rightarrow Acetoacetyl CoA
 - Enzyme: Succinyl CoA:acetoacetate CoA transferase (thiophorase)
 - This reaction is reversible, but:
 - Acetoacetyl CoA is rapidly cleaved to two acetyl CoA by thiolase, pulling the reaction forward.

- BUtilization Sites:

- Extrahepatic tissues, including:
 - Skeletal muscle
 - Cardiac muscle
 - Brain (during prolonged fasting)
- Excluded tissues:
 - RBCs (lack mitochondria)
 - Liver: despite producing ketone bodies, lacks thiophorase, so cannot use them as fuel.

C. Excessive Ketone Body Production in Diabetes Mellitus

- Overproduction of ketone bodies leads to:
 - Ketonemia (elevated blood ketone levels)
 - Ketonuria (presence in urine)

C. Excessive Ketone Body Production in Diabetes Mellitus

- Overproduction of ketone bodies leads to:
 - Ketonemia (elevated blood ketone levels)
 - Ketonuria (presence in urine)

- Common in uncontrolled Type 1 Diabetes Mellitus (T1D):
 - Blood ketones: can reach 90 mg/dL (normal: <3 mg/dL)
 - Urinary excretion: up to 5,000 mg/24 hrs
- Acidemia Mechanism:
 - Ketone bodies (acetoacetate, 3-hydroxybutyrate) have acidic carboxyl groups ($pK_a \sim 4$)
 - In blood, they lose H^+ , leading to decreased blood pH
 - Termed diabetic ketoacidosis (DKA)
- Compounding Factors:
 - Glucose and ketone loss in urine \rightarrow dehydration
 - Loss of water \rightarrow concentration of H^+ in plasma \rightarrow worsens acidosis

- Signs and Symptoms:

- Fruity odor of breath (from acetone, a volatile ketone)
- May occur in:
 - Uncontrolled T1D
 - Prolonged fasting
 - Excessive ethanol intake

Ketone Body Synthesis in the Liver and Use in Peripheral Tissues

