

TECHNICAL DIFFICULTIES

**PLEASE
STAND
BY**





Design and Operation of Moving Bed Processors

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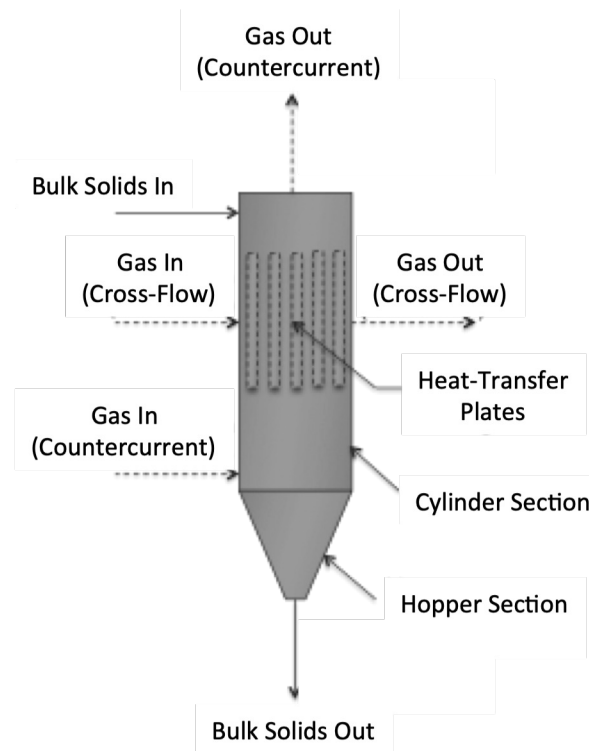
978-799-7311



Happiness Equals Reality Minus Expectations



Moving bed processors

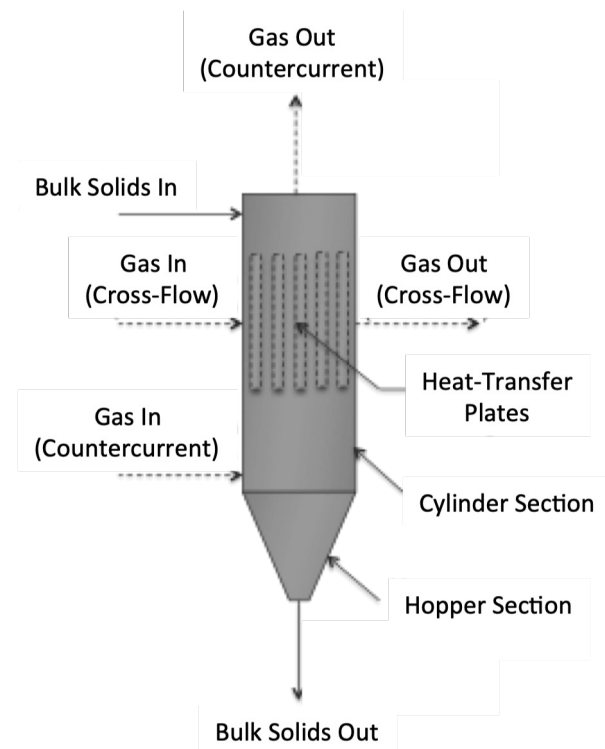


“GRUMP”



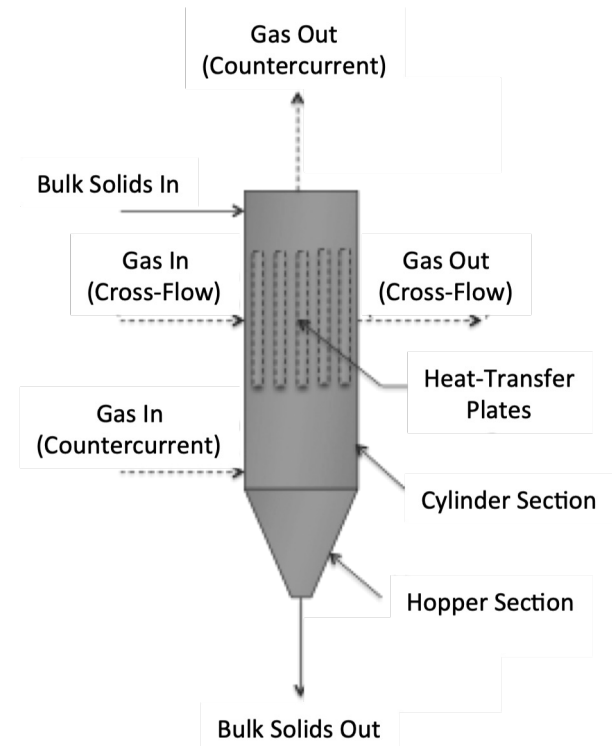
Advantages

- Lower capital, operating, and maintenance costs
 - No moving parts
 - Higher throughputs
- Offers surge or storage capacity
- Long residence times



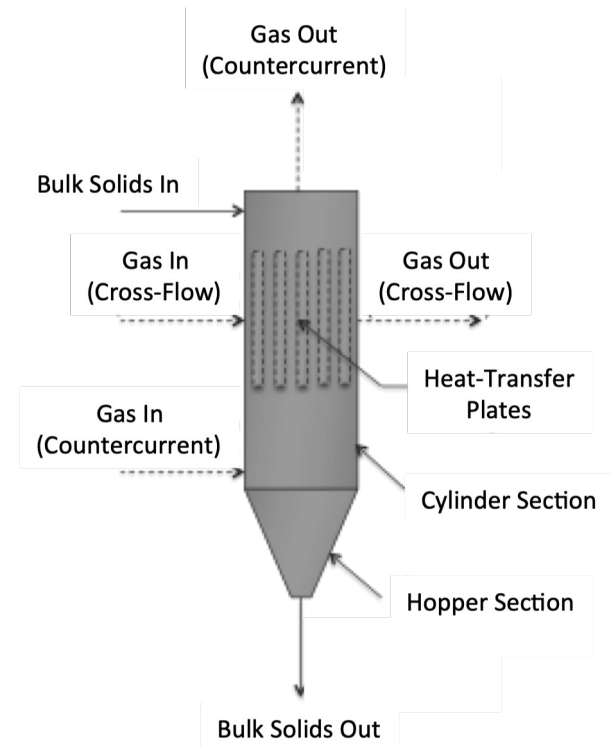
Disadvantages

- No mixing
- Not amenable to batch processes
- Fine powders may need large diameters or multiple vessels



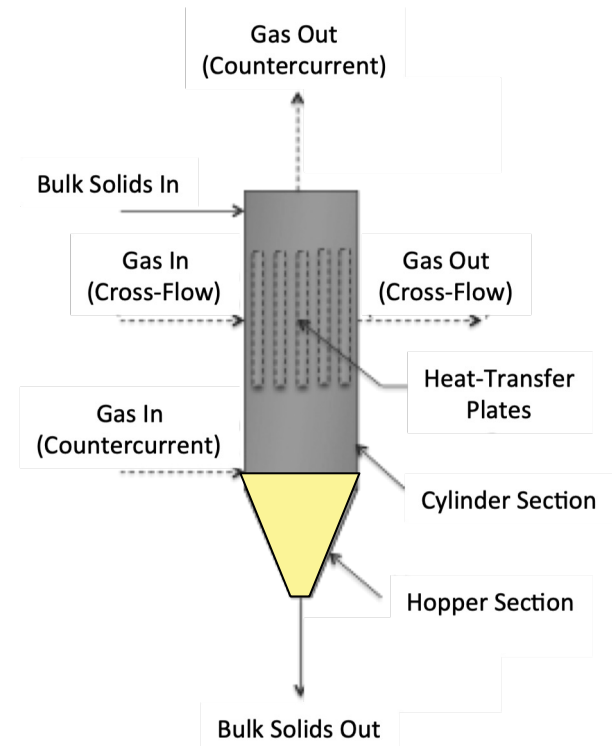
Requirements

- Plug flow of solids
- Proper outlet size
- Uniform gas flow
- Stable solids bed
- Necessary volume/residence time



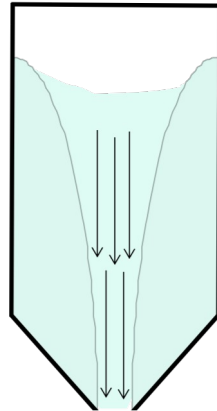
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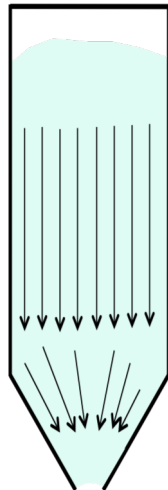


Flow patterns

- Funnel flow



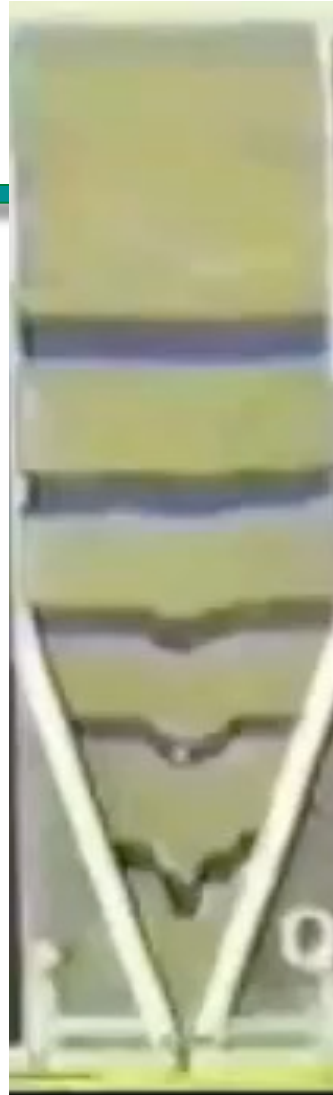
- Mass flow



Funnel flow



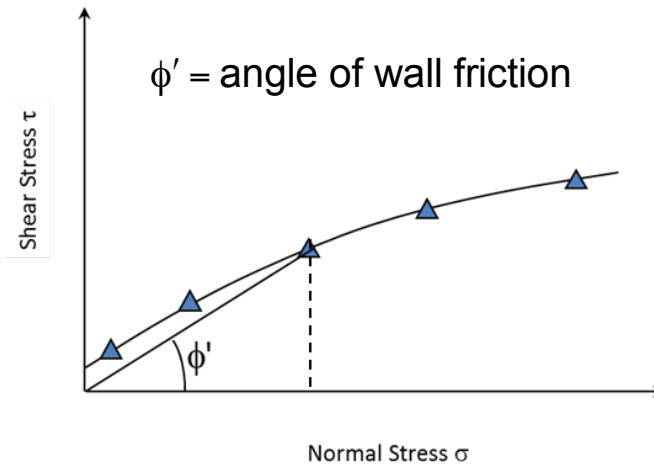
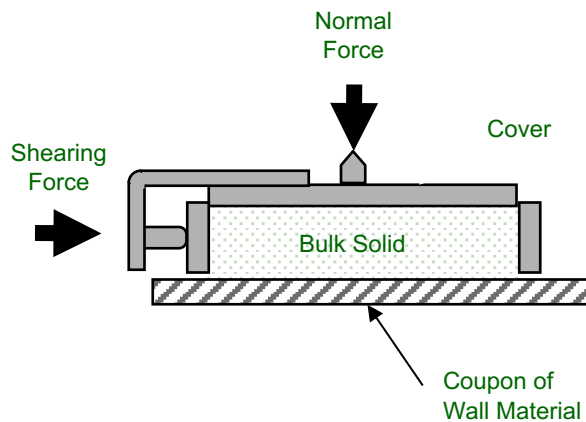
Funnel flow



Mass flow



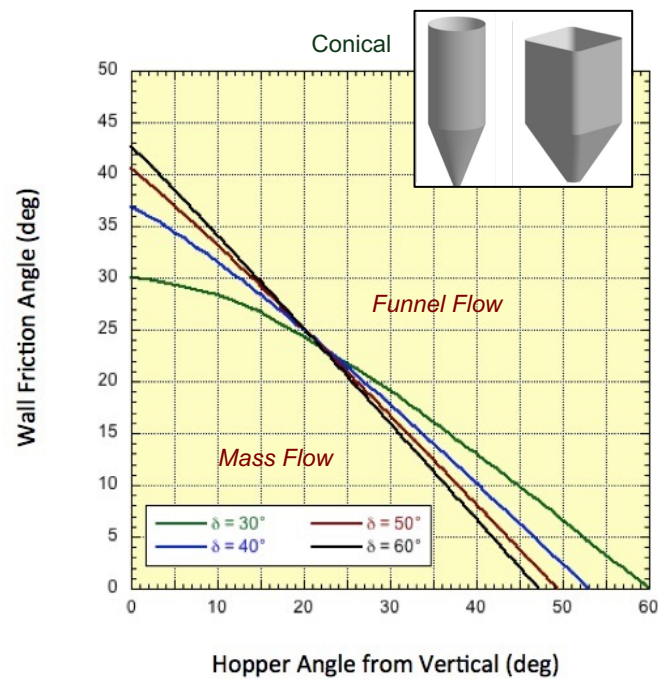
Wall friction



$$\tan \phi' = \frac{\text{Shear Stress}}{\text{Normal Stress}} = \mu$$

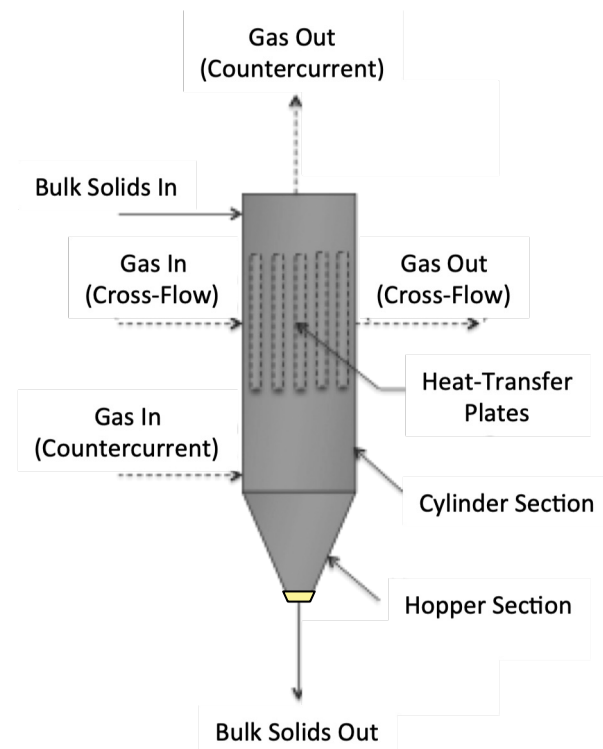


Mass flow hopper angle

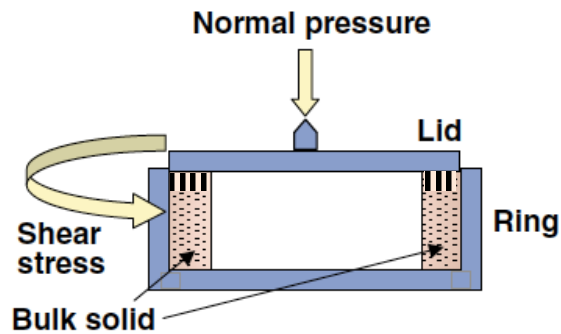


Requirements

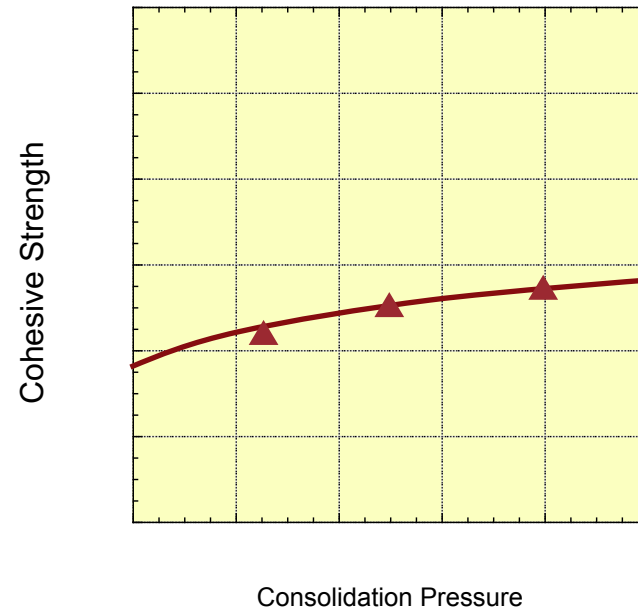
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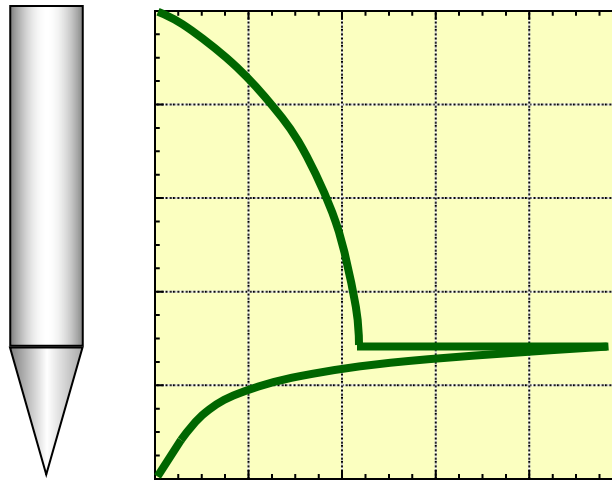
Cohesive strength



Flow Function

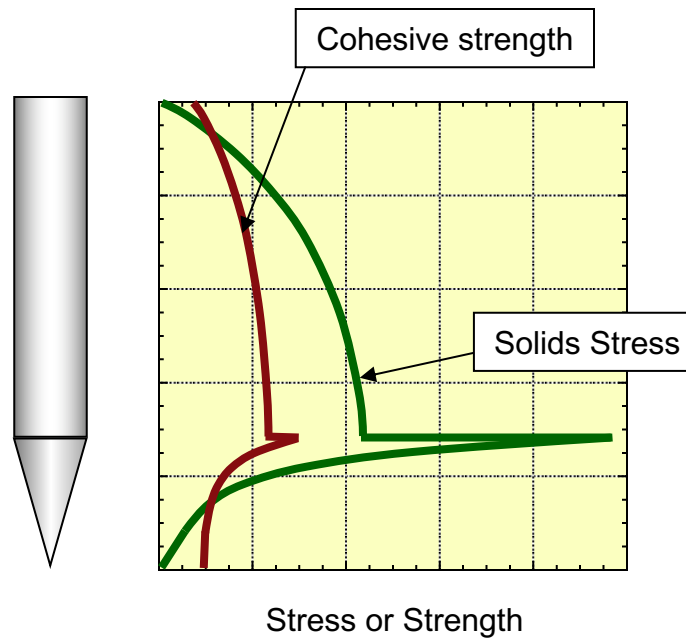


Solids stress profile

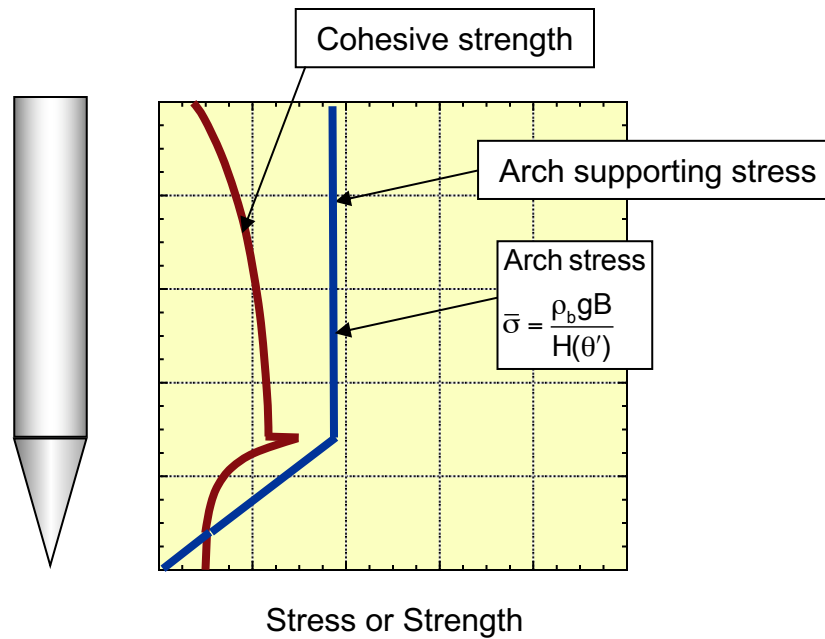


Consolidation Stress

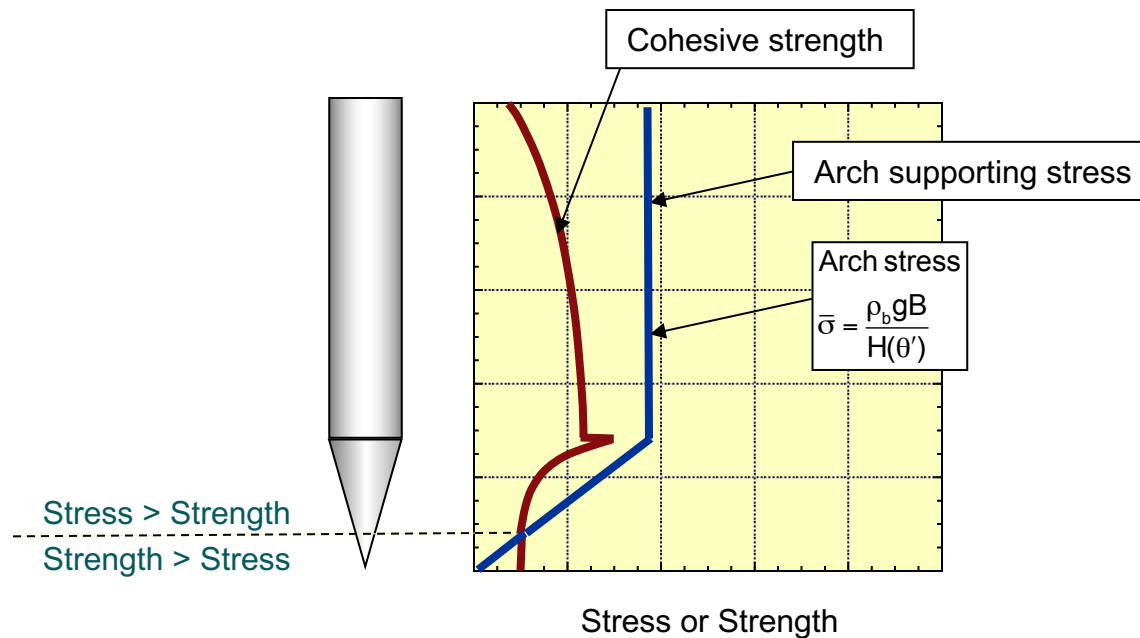
Solids stress profile



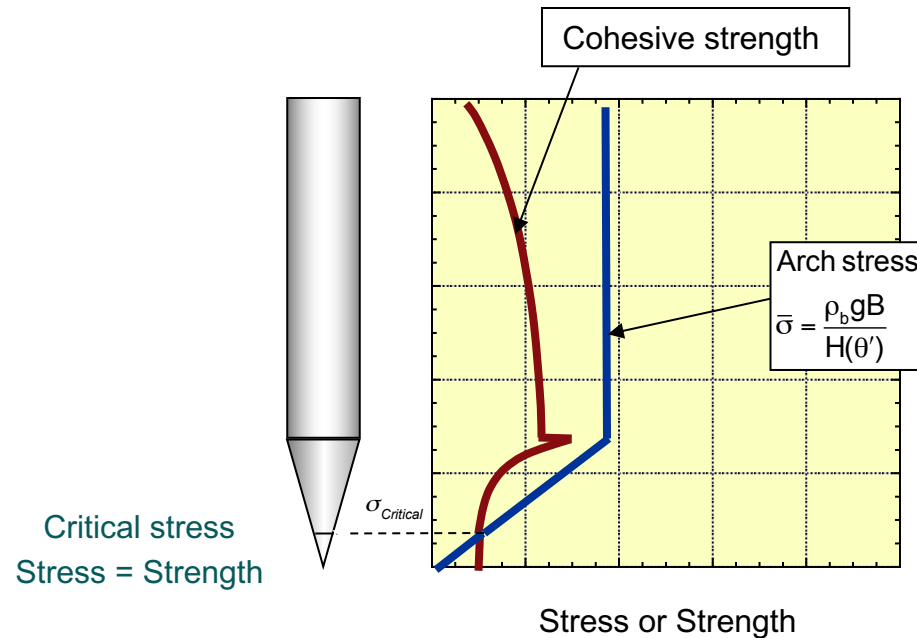
Arch stress



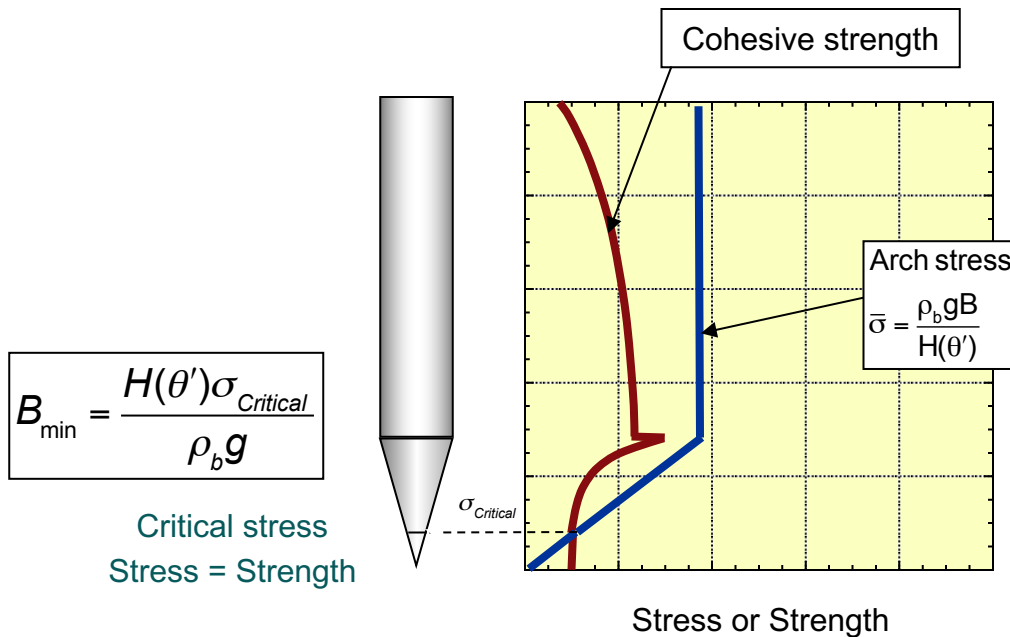
Arch stress



Critical arching diameter



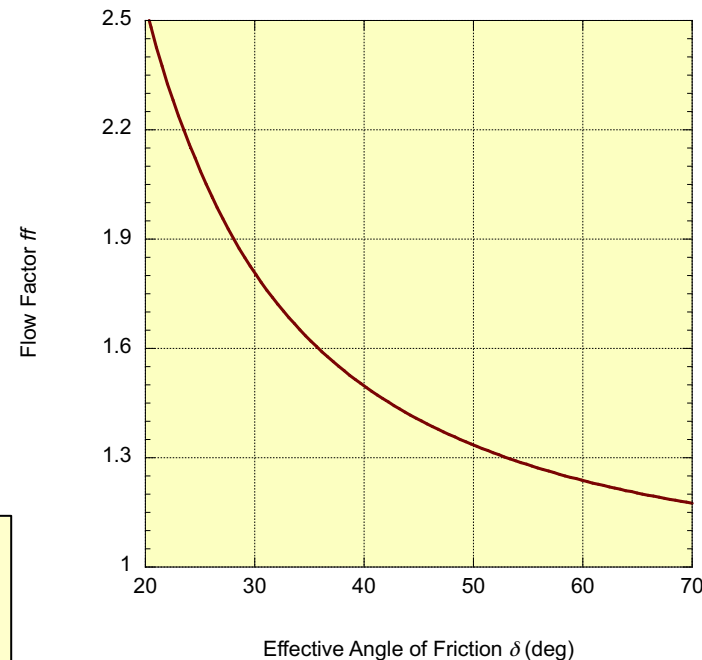
Critical arching diameter



Flow factor

- Flow function (FF)
Relationship between major principal stress and cohesive strength
- Flow factor (ff)
Ratio of major principal stress and arch supporting stress

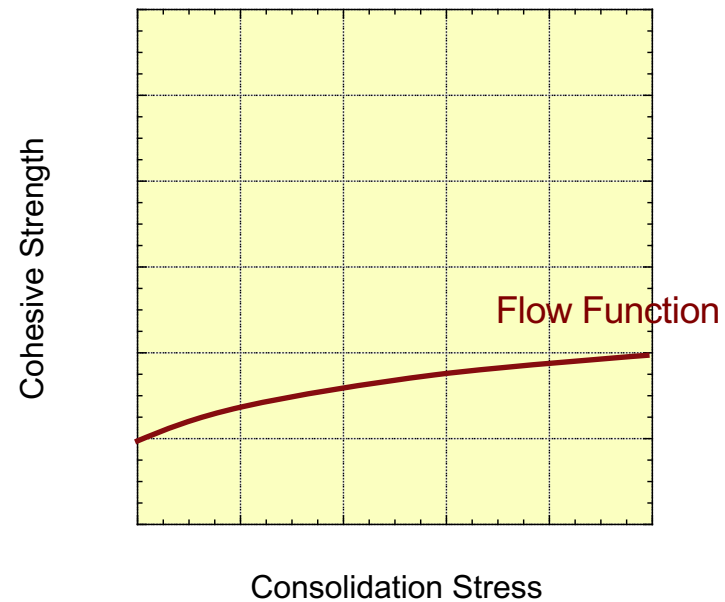
$$ff = \frac{\sigma_1}{\bar{\sigma}_1} = \frac{\text{major principle stress}}{\text{arch stress}}$$



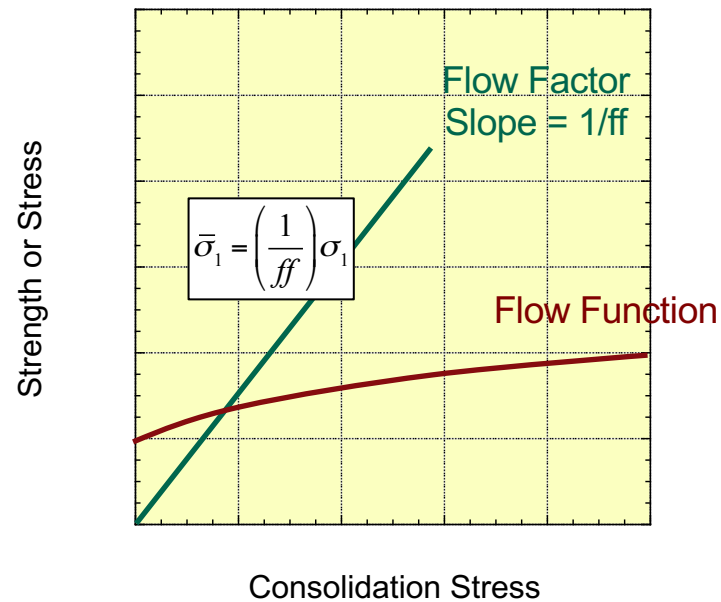
$$ff = 1.118 + \frac{0.285}{(\tan \delta)^2}$$



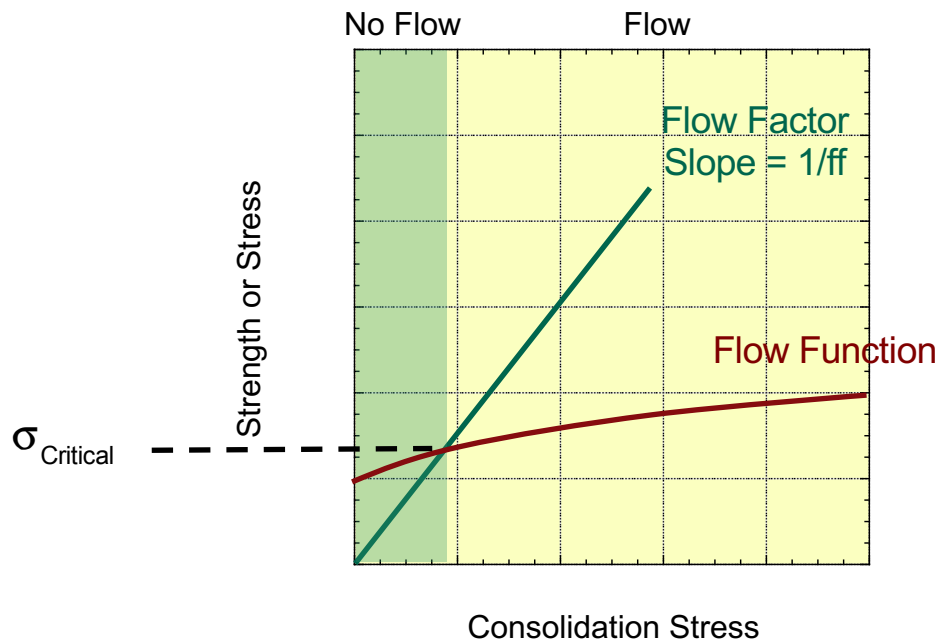
Flow function



Flow factor



Critical arching diameter



$$B_{\min} = \frac{H(\theta')\sigma_{\text{Critical}}}{\rho_b g}$$

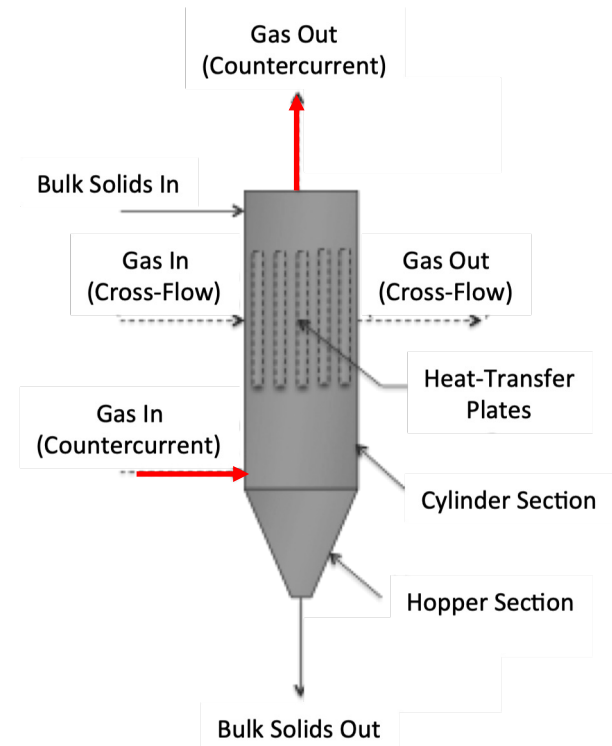


Critical arching diameter



Requirements

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- **Uniform gas flow**
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residence time



Gas Injection

- Nozzles
- Screens
- Sparge rings
- Inserts
- Annulus
- Crossbeams
- Gas permeable walls



Gas Injection

- Nozzles
- Screens
- ~~Sparge rings~~
- Inserts
- Annulus
- Crossbeams
- Gas permeable walls



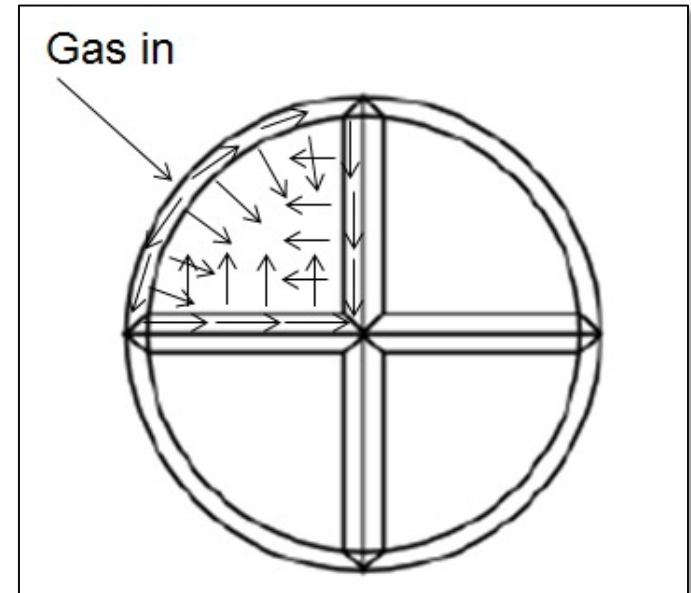
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Gas Injection

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Jenike & Johanson J-Purge™

Gas Injection

- Nozzles
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Zeppelin Systems



Gas Injection

- Nozzles
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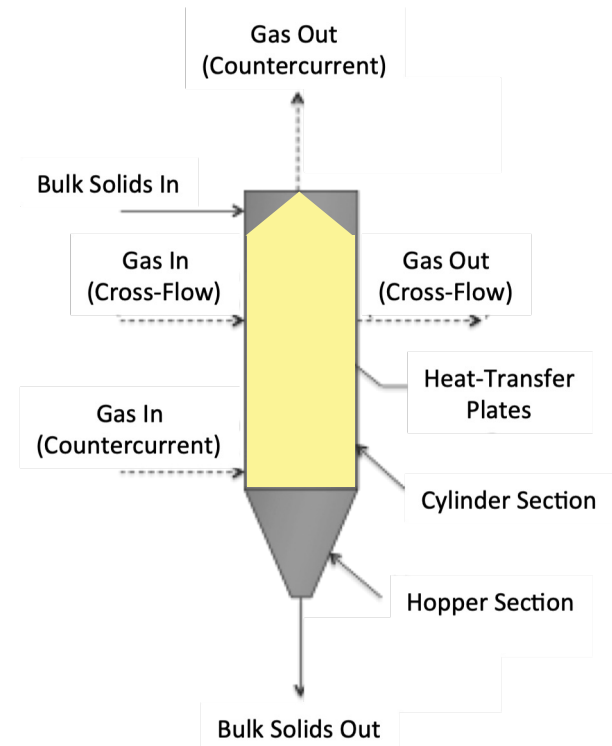


Young Industries
SilentFlow

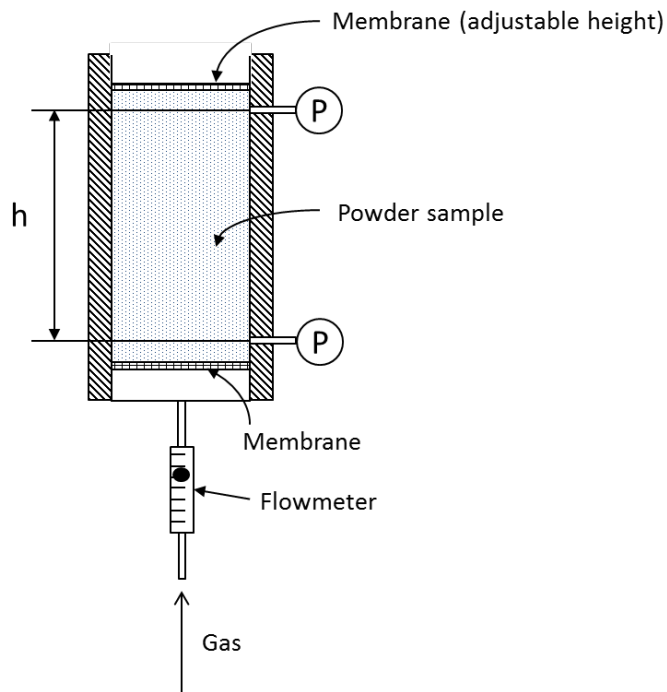


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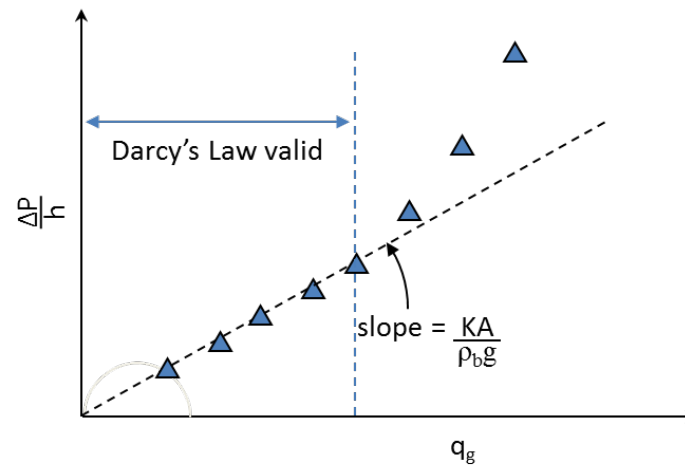


Permeability

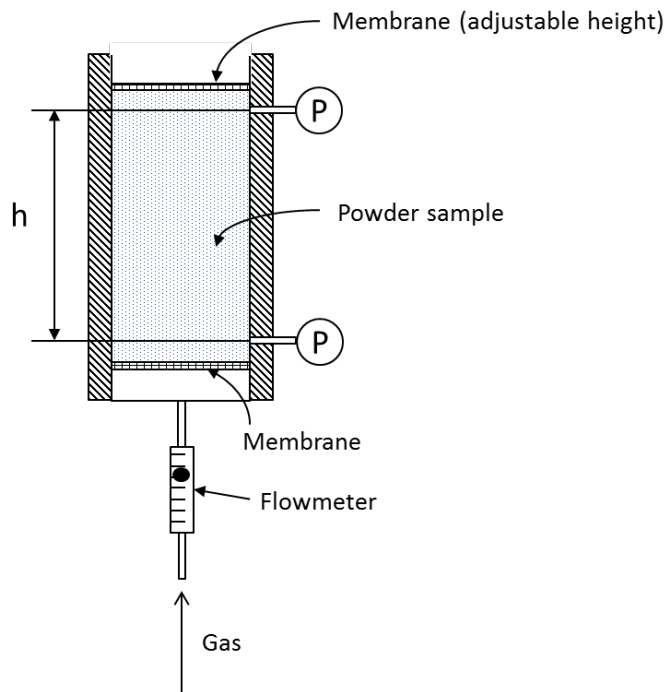


Darcy's Law:

$$q_g = \frac{KA}{\rho_b g} \frac{\Delta P}{h} = \frac{kA}{\eta} \frac{\Delta P}{h}$$



Permeability



Darcy's Law:

$$q_g = \frac{KA}{\rho_b g} \frac{\Delta P}{h} = \frac{kA}{\eta} \frac{\Delta P}{h}$$

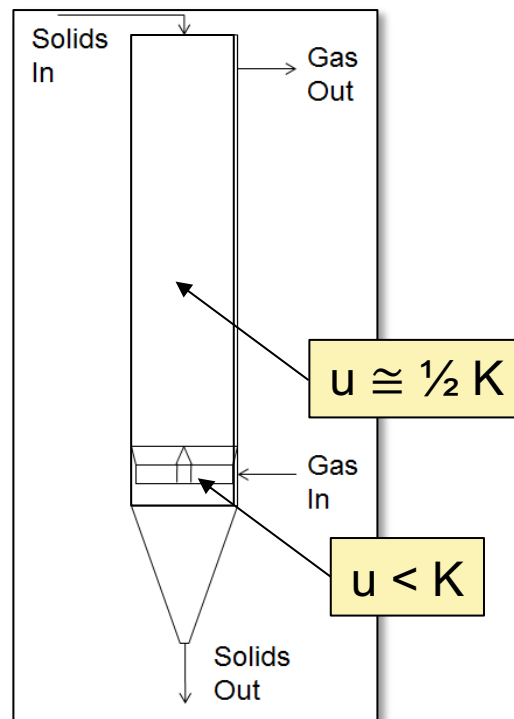
$$u = \frac{K}{\rho_b g} \frac{dP}{dz}$$

$$\text{when } \rho_b g = \frac{dP}{dz}$$

$$u = K = u_{mf}$$

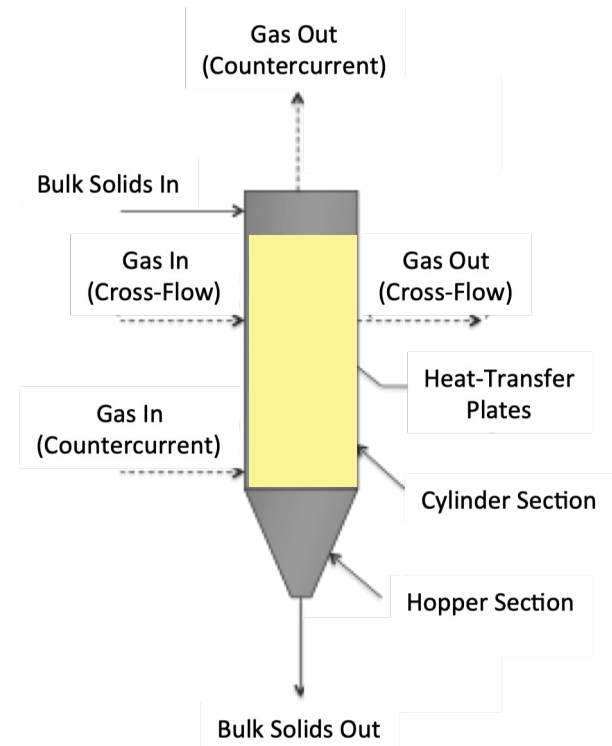


Allowable gas velocity



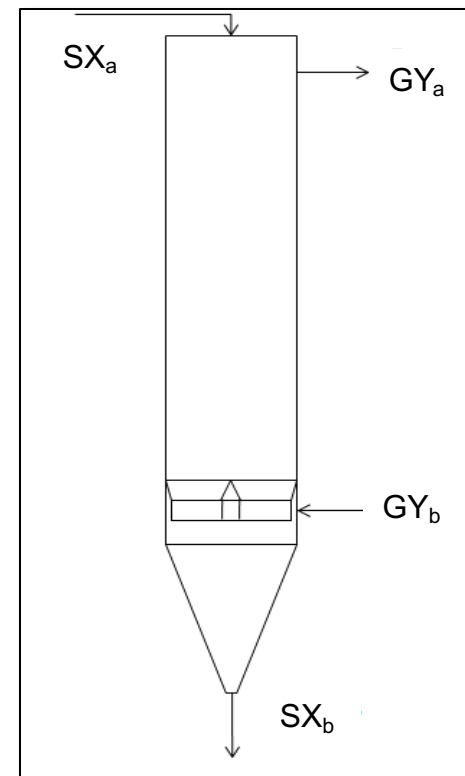
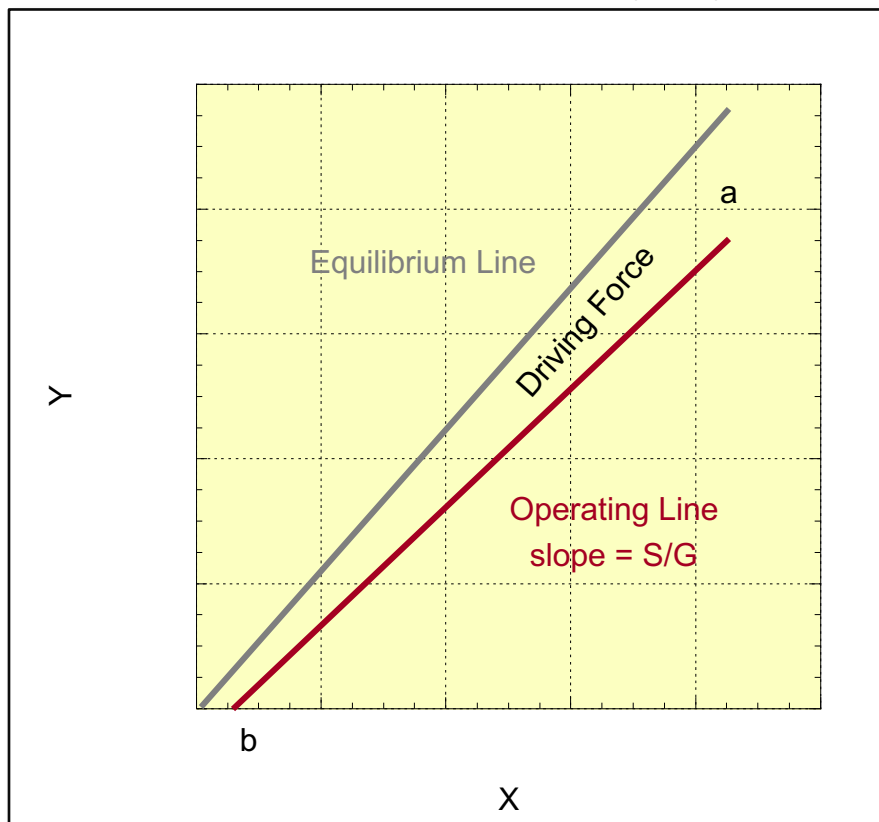
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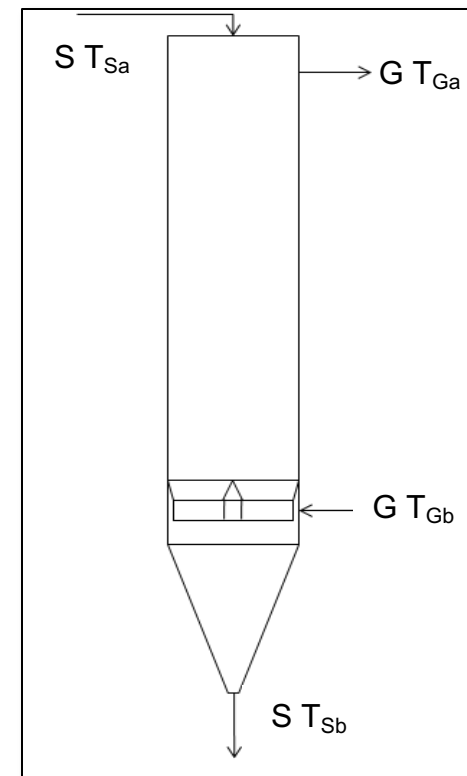
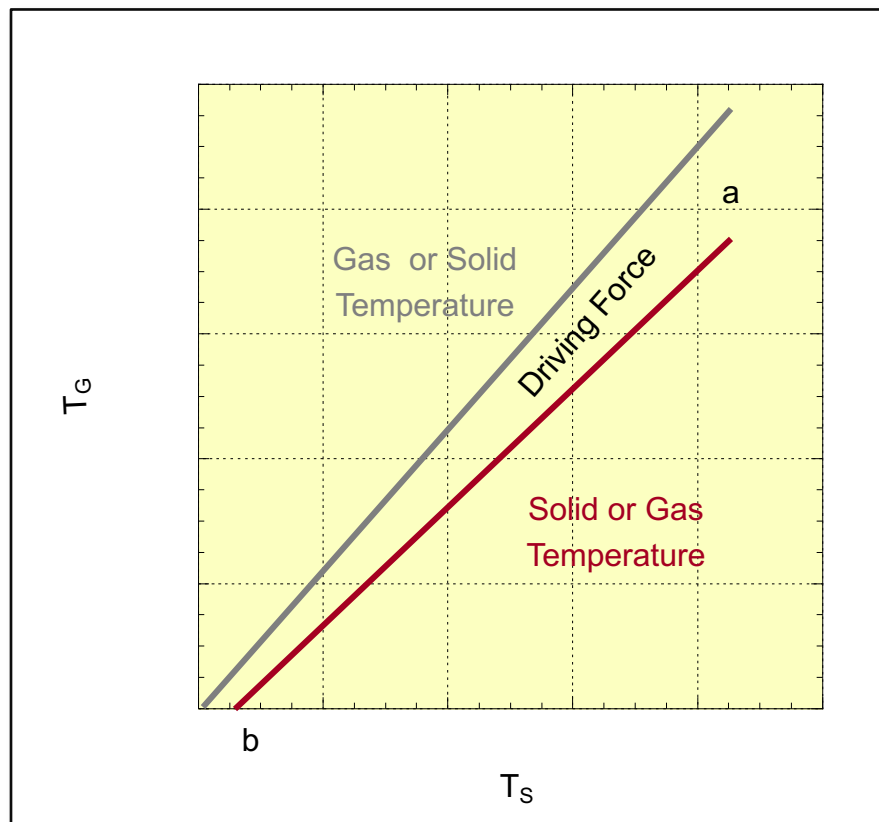
Minimum gas rate

Purge column or gravity dryer

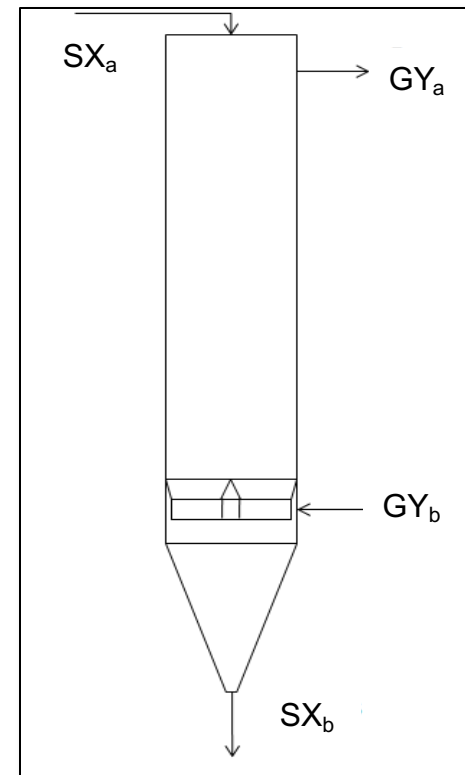
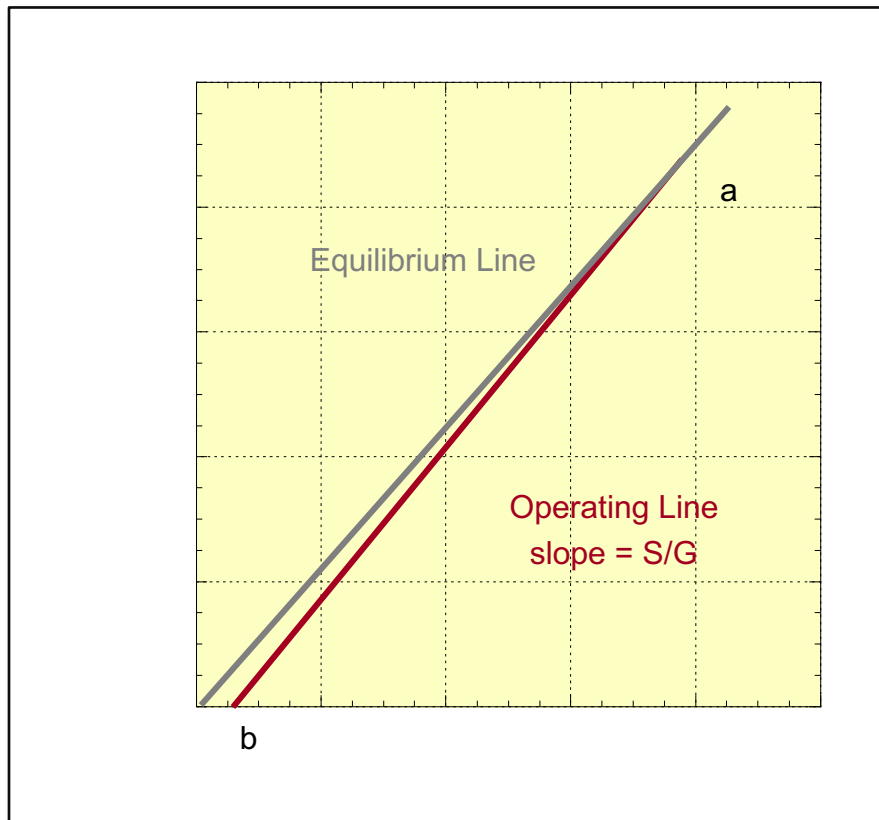


Minimum gas rate

Direct Heat Exchanger



Minimum gas rate



Kinetics

- Purge columns/gravity dryers

$$N_A = k_y a_v (y_A^* - y_A)$$

- Direct contact heat exchangers

$$q = h a_v (T_s - T_g)$$

Flux

Coefficient

Driving force



Kinetics

- Purge columns/gravity dryers

$$N_A = k_y a_v (y_A^* - y_A)$$

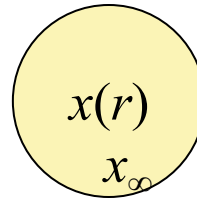
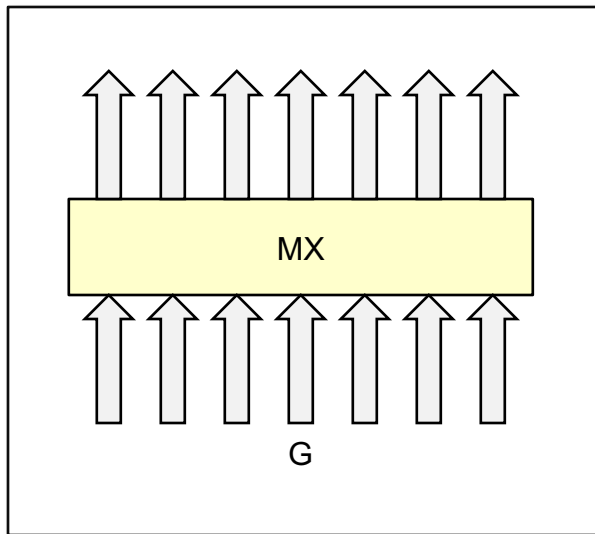
$$\frac{\partial X_A}{\partial t} = \frac{\mathcal{D}}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial X_A}{\partial r} \right)$$

- Direct contact heat exchangers

$$q = h a_v (T_s - T_g)$$



Kinetics



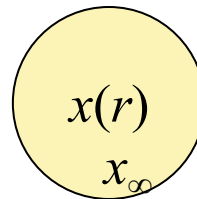
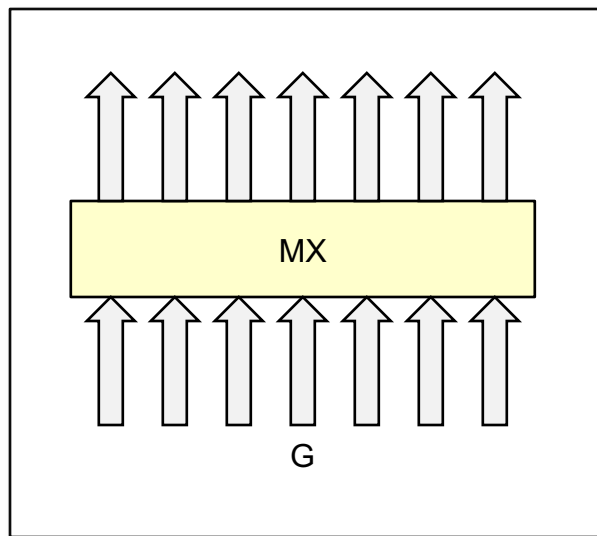
$$\frac{\partial x}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(D_{eff} r^2 \frac{\partial x}{\partial r} \right)$$

$$x(r, 0) = x_{init}$$

$$\frac{\partial x(0, t)}{\partial r} = 0$$

$$x(r_o, t) = x_\infty$$

Kinetics



$$\frac{\partial x}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(D_{eff} r^2 \frac{\partial x}{\partial r} \right)$$

$$x(r, 0) = x_{init}$$

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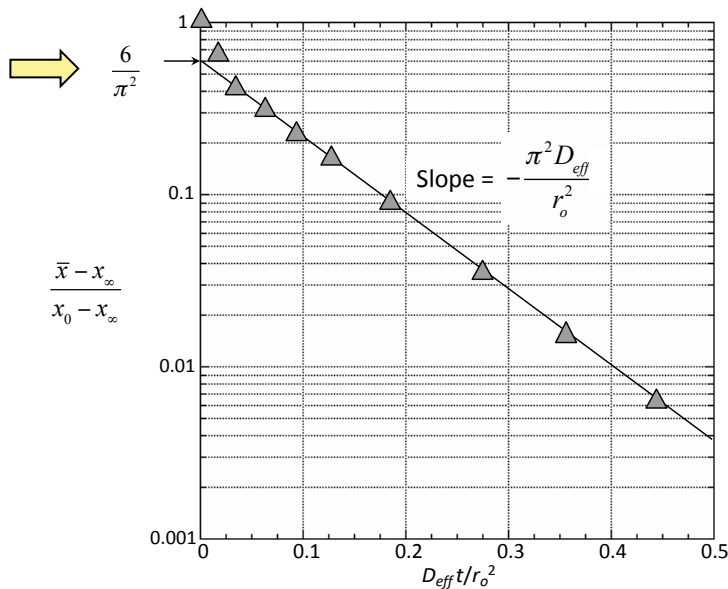
$$x(r_o, t) = x_\infty$$

$$\frac{\bar{x} - x_\infty}{x_{init} - x_\infty} = \frac{6}{\pi^2} \sum_{n=1}^{\infty} \frac{1}{n^2} \exp \left(\frac{-n^2 \pi^2 D_{eff} t}{r_o^2} \right)$$



Kinetics

Diffusion controlled



$$\frac{\partial x}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(D_{eff} r^2 \frac{\partial x}{\partial r} \right)$$

$$x(r_o, t) = x_{\infty}$$

$$x(r, 0) = x_{init}$$

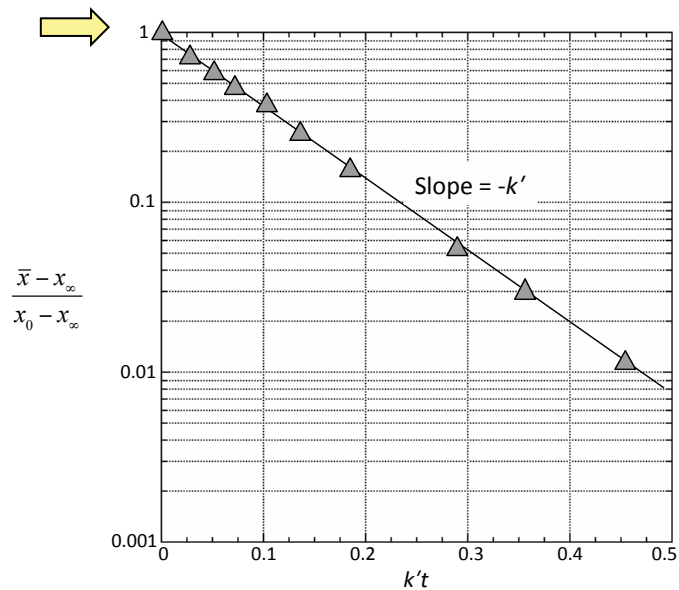
$$\frac{\partial x(0, t)}{\partial r} = 0$$

$$\frac{\bar{x} - x_{\infty}}{x_{init} - x_{\infty}} = \frac{6}{\pi^2} \sum_{n=1}^{\infty} \frac{1}{n^2} \exp \left(\frac{-n^2 \pi^2 D_{eff} t}{r_o^2} \right)$$



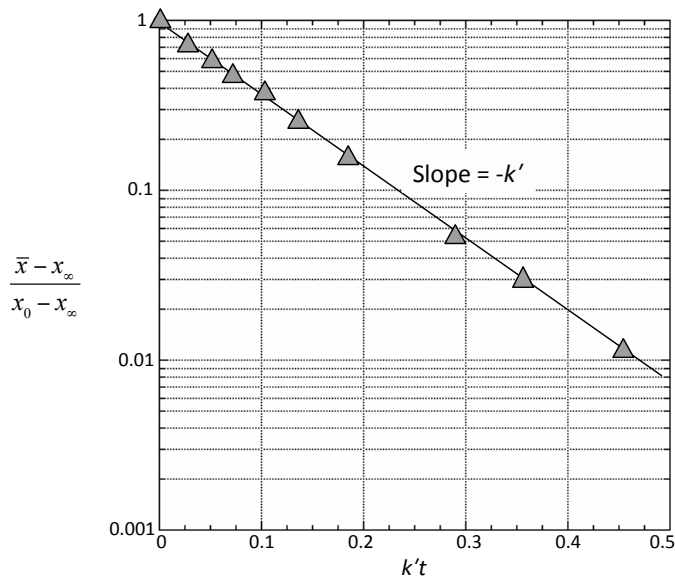
Kinetics

Controlled by convection in the gas phase



Kinetics

Controlled by convection in the gas phase



$$\frac{d\bar{x}}{dt} = -k'(\bar{x} - x_{\infty})$$

$$x(r_o, t) = x_{\infty}$$

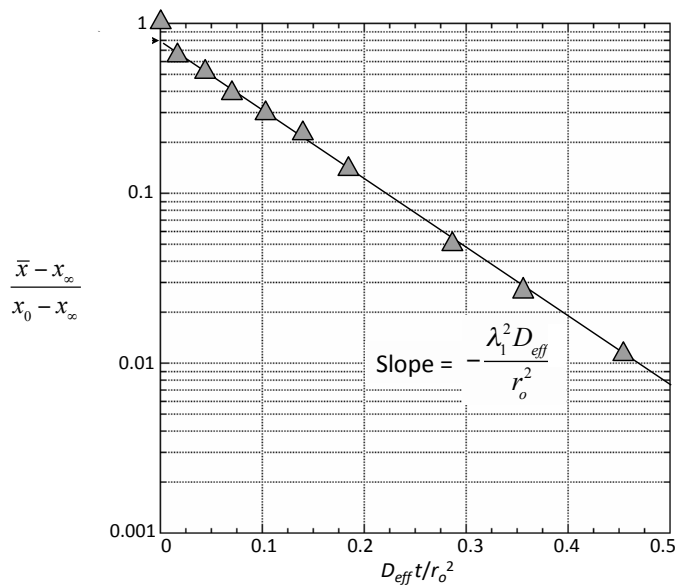
$$x(r, 0) = x_{init}$$

$$\frac{\bar{x} - x_{\infty}}{x_{init} - x_{\infty}} = \exp(-k't)$$



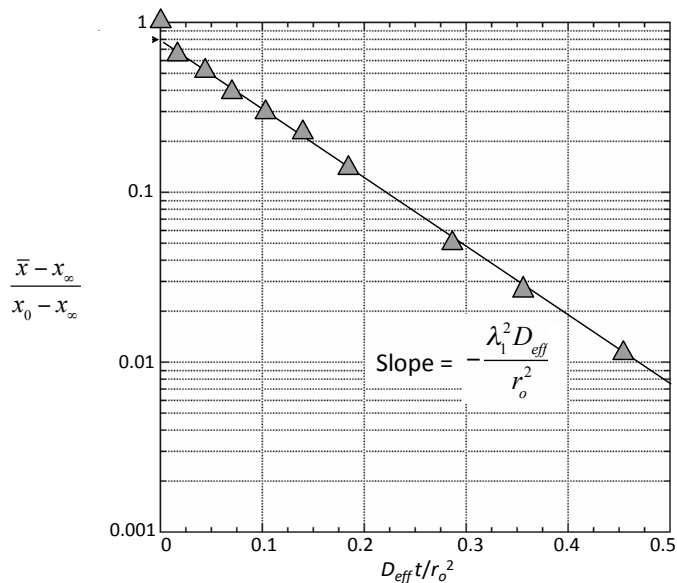
Kinetics

Controlled by both convection and diffusion



Kinetics

Controlled by both convection and diffusion



$$\frac{\partial x}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(D_{eff} r^2 \frac{\partial x}{\partial r} \right)$$

$$x(r_o, t) = x_\infty$$

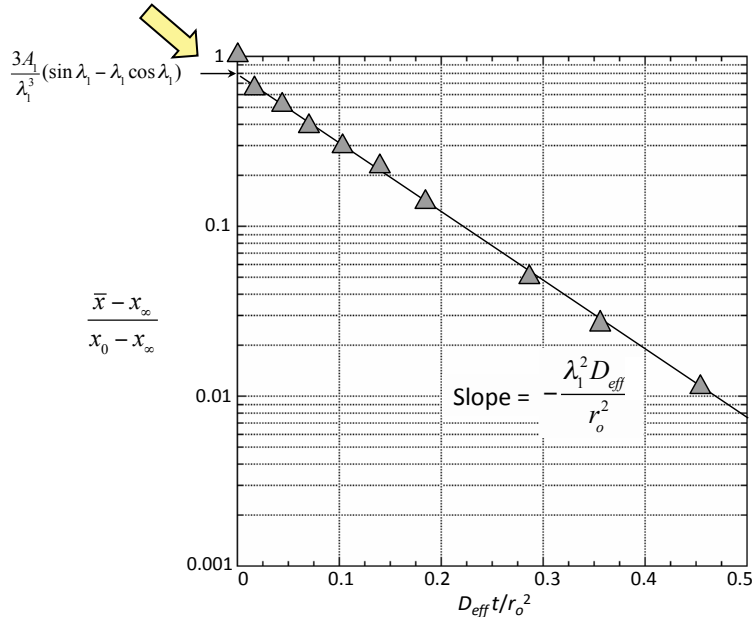
$$\frac{\partial x(0, t)}{\partial r} = 0$$

$$-D_{eff} \frac{\partial [x(r_o, t)]}{\partial r} = k_x (x - x_\infty)$$



Kinetics

Controlled by both convection and diffusion



$$\frac{\partial x}{\partial t} = \frac{1}{r^2} \frac{\partial}{\partial r} \left(D_{eff} r^2 \frac{\partial x}{\partial r} \right)$$

$$x(r_o, t) = x_{\infty}$$

$$\frac{\partial x(0, t)}{\partial r} = 0$$

$$-D_{eff} \frac{\partial [x(r_o, t)]}{\partial r} = k_x (x - x_{\infty})$$

$$\frac{\bar{x}(t) - x_{\infty}}{x_{init} - x_{\infty}} = \frac{3A_1}{\lambda_1^3} (\sin \lambda_1 - \lambda_1 \cos \lambda_1) \exp \left(-\lambda_1^2 \frac{D_{eff} t}{r_o^2} \right)$$

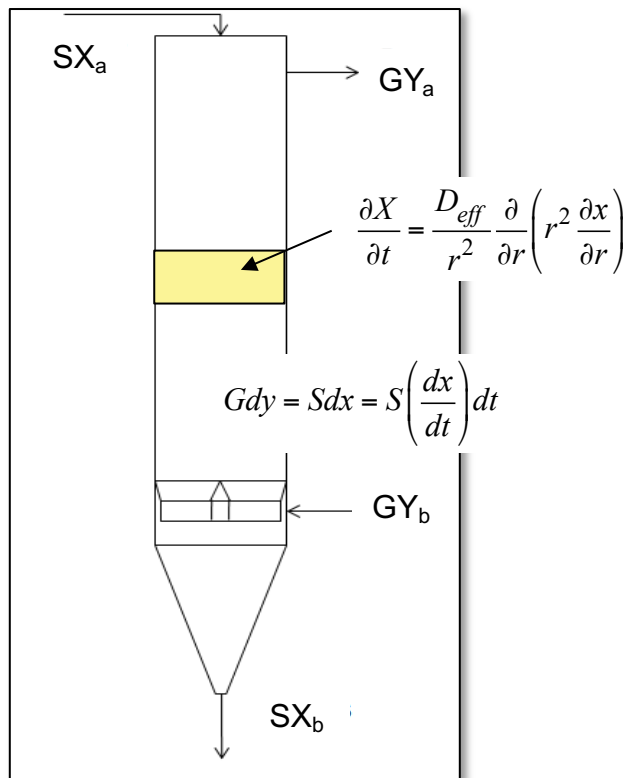
$$1 - \frac{\lambda_1}{\tan \lambda_1} = \mathbf{Bi}$$

$$\mathbf{Bi} = \frac{k_x r_o}{D_{eff}}$$

$$A_1 = \frac{4 \sin(\lambda_1) - \lambda_1 \cos(\lambda_1)}{2\lambda_1 - \sin(2\lambda_1)}$$



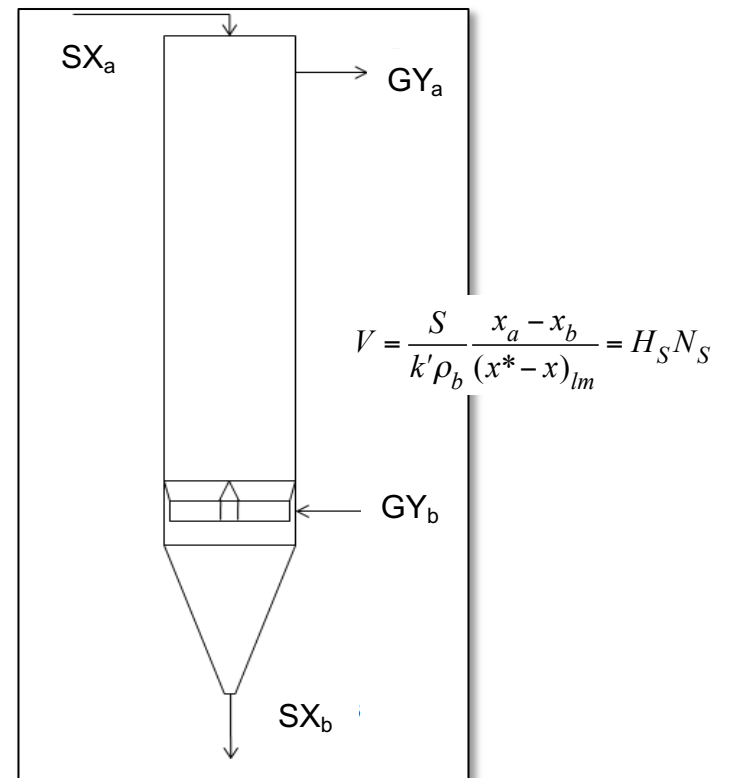
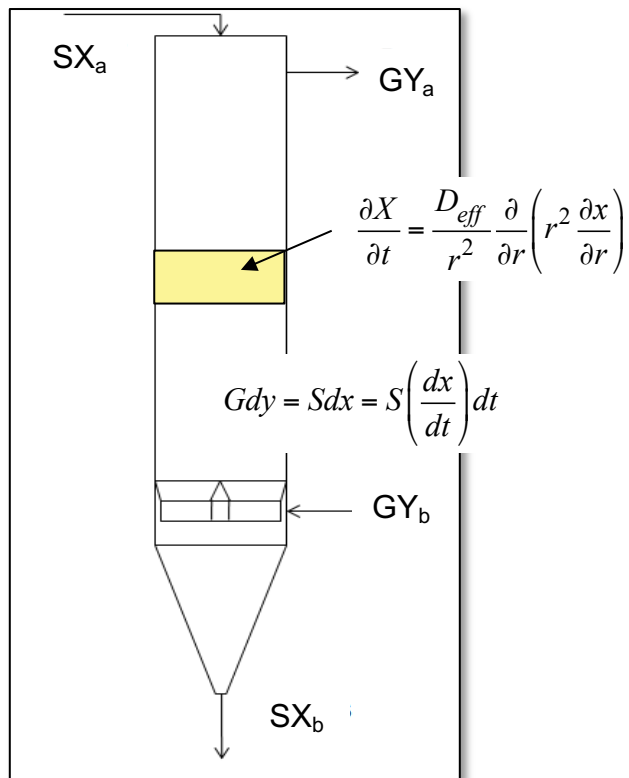
Design



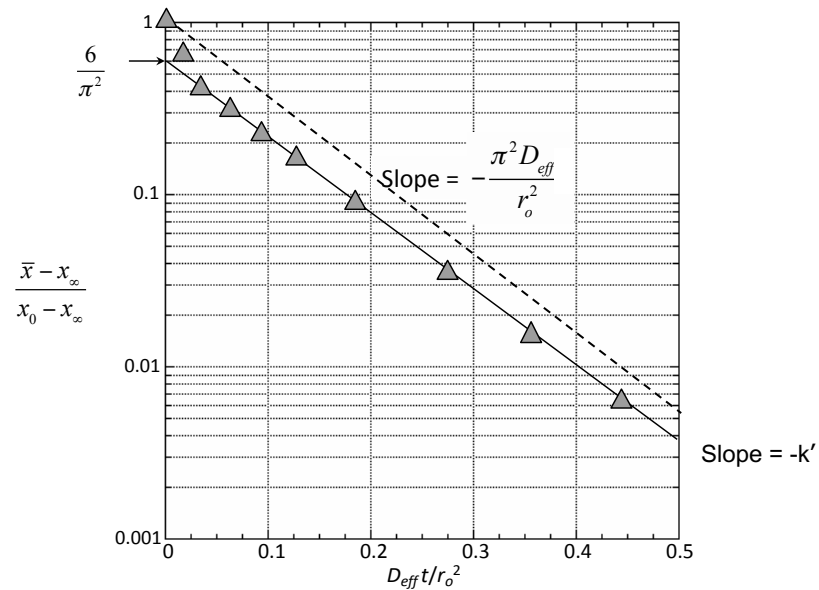
- Estimate Y_A
- Solve diffusion equation for time spent in short section in column
- Determine volatiles content entering section from mass balance
- Continue until bottom of column is reached
- Iterate until $Y_{ab} = 0$



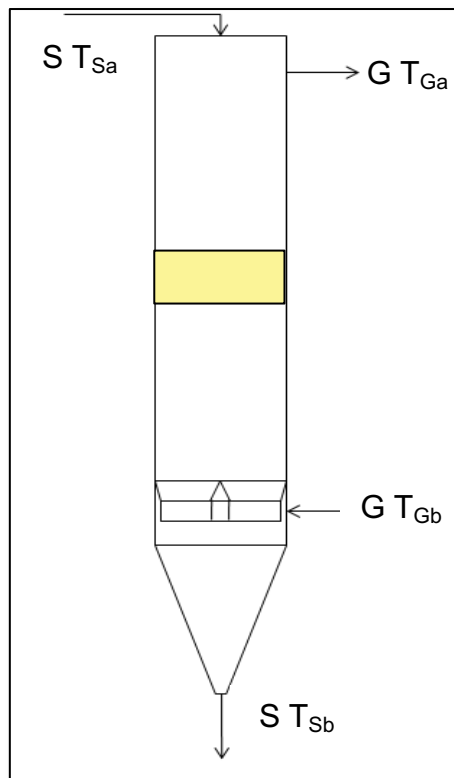
Design



Design



Direct bulk solids heat exchangers

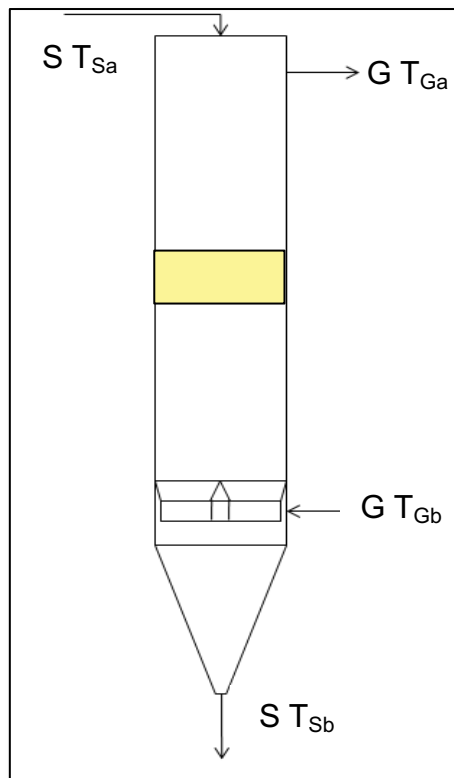


$$SC_{PS} \frac{dT_s}{dz} = ha_v (T_G - T_S)$$

$$SC_{GS} \frac{dT_s}{dz} = ha_v (T_S - T_G)$$



Direct bulk solids heat exchangers



$$SC_{PS} \frac{dT_s}{dz} = ha_v (T_G - T_S)$$

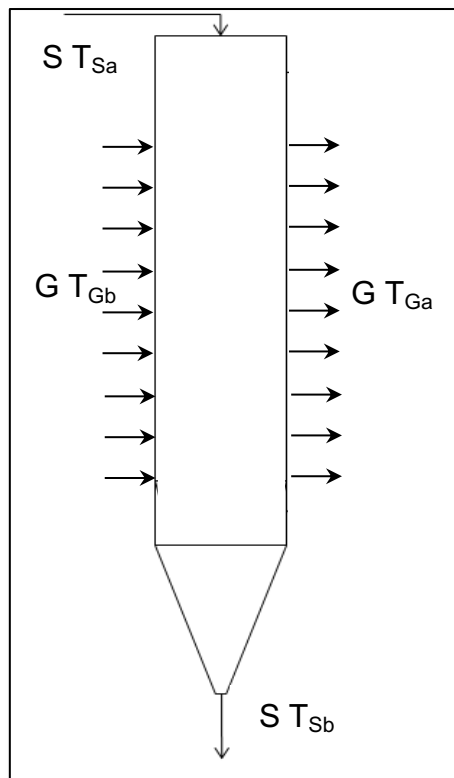
$$SC_{GS} \frac{dT_s}{dz} = ha_v (T_S - T_G)$$

$$\ln \left(\frac{T_{Ga} - T_{Sa}}{T_{Gb} - T_{Sb}} \right) = -ha_v \left(\frac{1}{GC_{PG}} - \frac{1}{SC_{PS}} \right) V$$



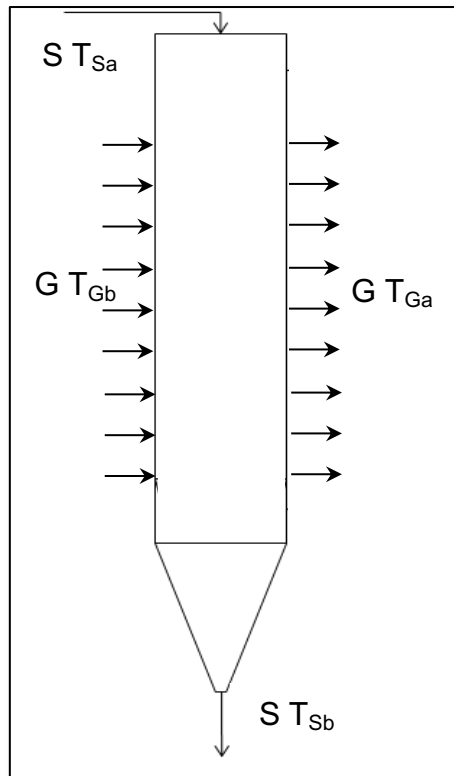
Direct bulk solids heat exchangers

Cross-flow



Direct bulk solids heat exchangers

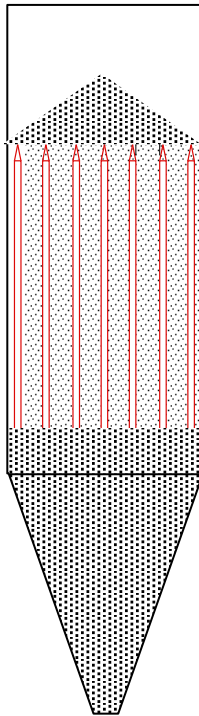
Cross-flow



$$\frac{T_s - T_{sin}}{T_{gin} - T_{sin}} = 1 - \exp\left(-\frac{A_z H h a_s}{\dot{m}_s C_{Ps}} - \frac{A_x x h a_s}{\dot{m}_g C_{Pg}}\right) \sum_{j=0}^{\infty} \frac{\left(\frac{A_x x h a_s}{\dot{m}_g C_{Pg}}\right)^j}{j!} \sum_{k=0}^j \frac{\left(\frac{A_z H h a_s}{\dot{m}_s C_{Ps}}\right)^k}{k!}$$



Indirect moving bed heat exchangers



$$\rho_b u C_{ps} \frac{\partial T}{\partial z} = -k_{eff} \frac{\partial^2 T}{\partial x^2}$$

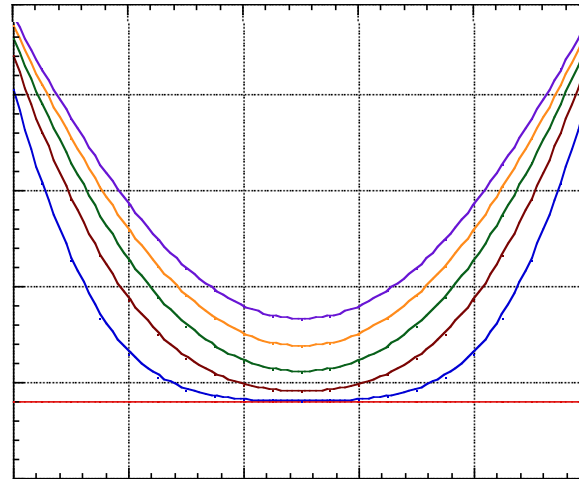
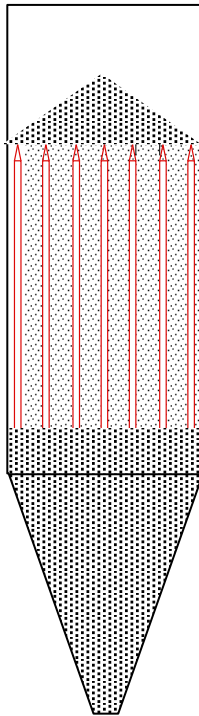
$$T(x, 0) = T_0(x, 0)$$

$$-k_{eff} \frac{\partial T(-\frac{b}{2}, z)}{\partial x} = h[T(-\frac{b}{2}, z) - T_f(z)]$$

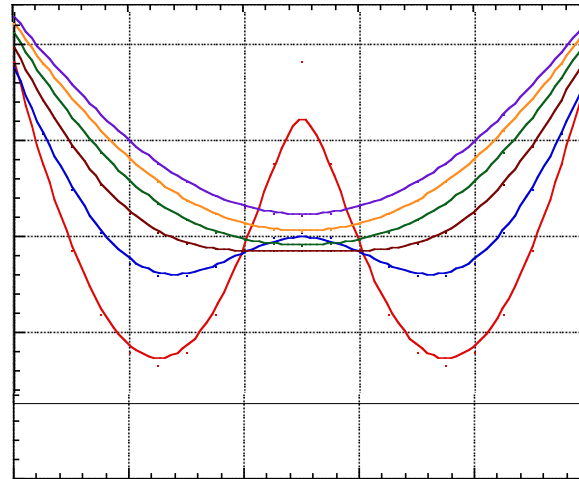
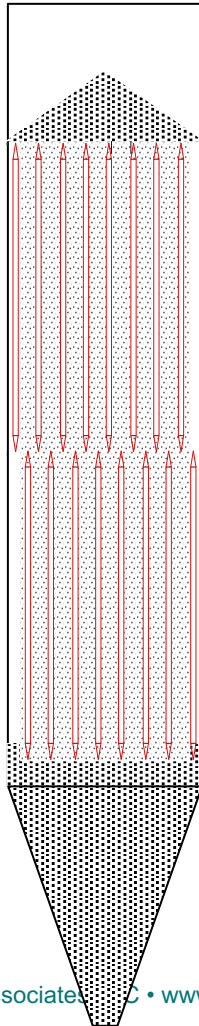
$$-k_{eff} \frac{\partial T(\frac{b}{2}, z)}{\partial x} = h[T(\frac{b}{2}, z) - T_f(z)]$$

$$T_f(z) = T_{fb} + C[\bar{T}(z) - \bar{T}_a]$$

Indirect moving bed heat exchangers



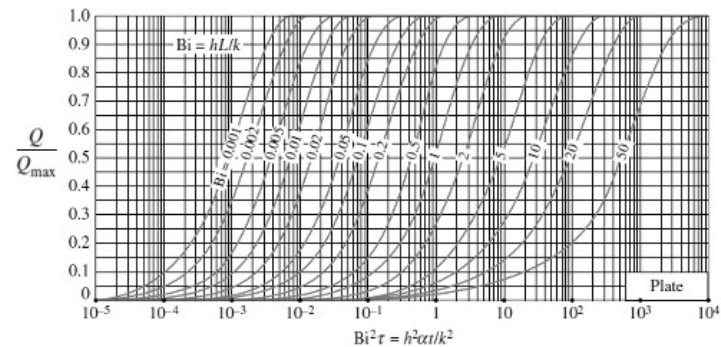
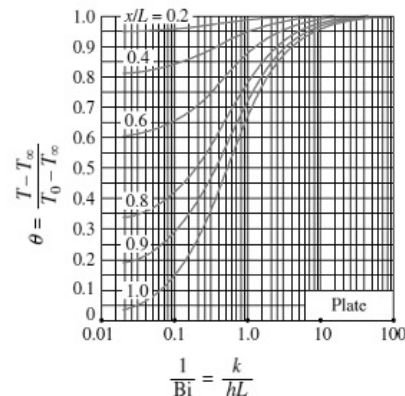
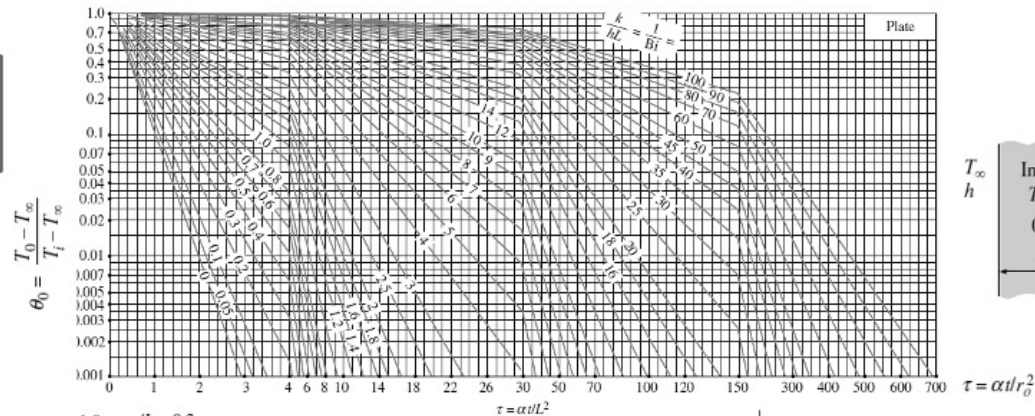
Indirect moving bed heat exchangers



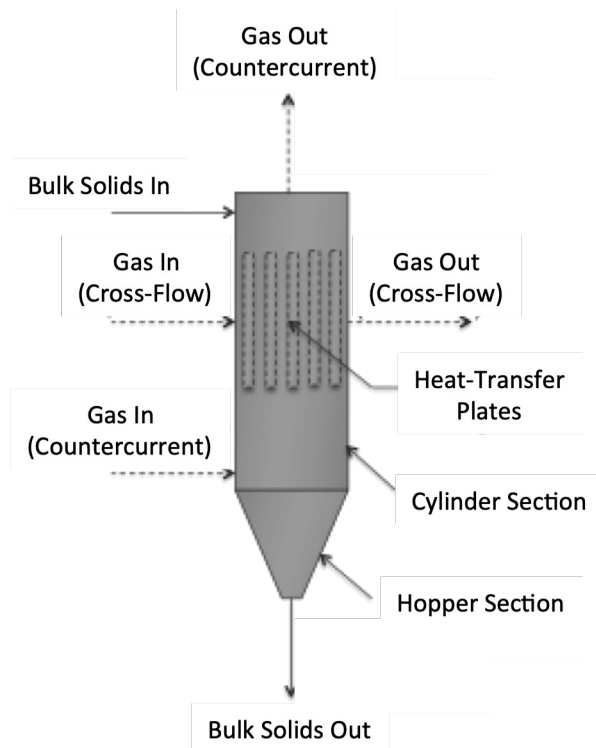
Indirect moving bed heat exchangers

Transient temperature and heat transfer charts for a **plane wall of thickness $2L$** initially at a uniform temperature T_i subjected to convection from all sides to an environment at temperature T_∞ with a convection coefficient of h .

$$\frac{\partial^2 T(x, t)}{\partial x^2} = \frac{1}{\alpha} \frac{\partial T(x, t)}{\partial t}$$



Moving bed reactors

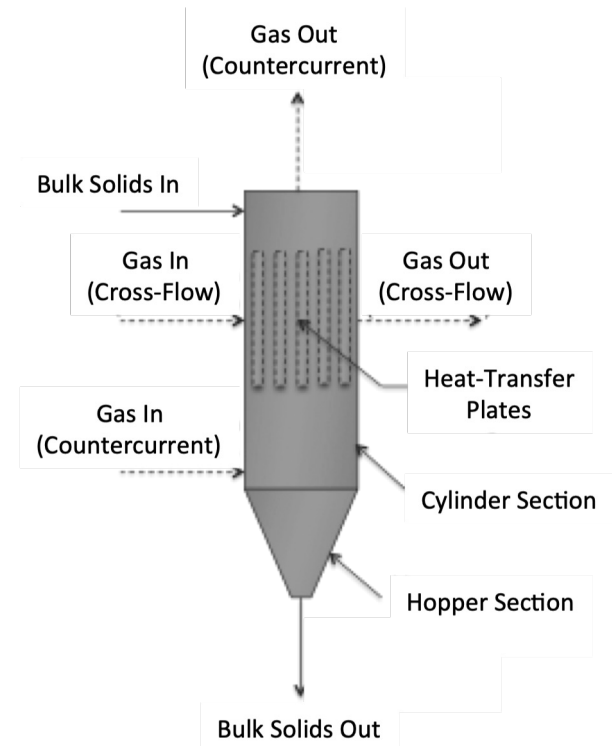


- Batch → Continuous
- Eqn's are the same!

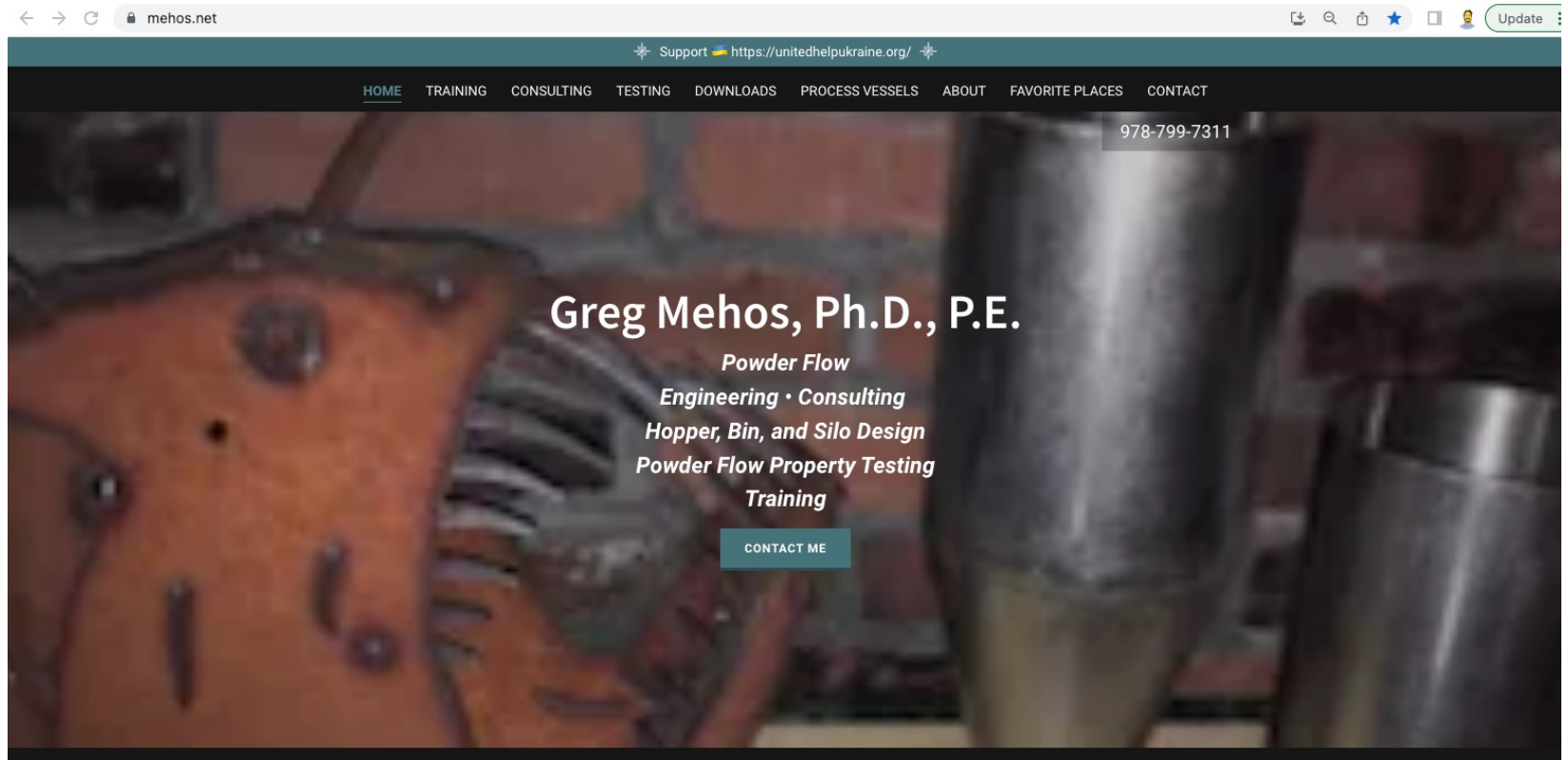


Requirements

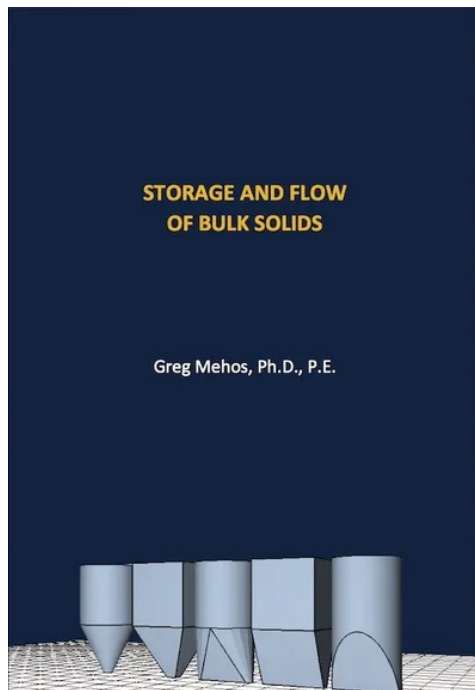
- Plug flow of solids
- Proper outlet size
- Uniform gas flow
- Stable solids bed
- Necessary volume/residence time



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8. BETTER REFERENCES THAN THIS ONE

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