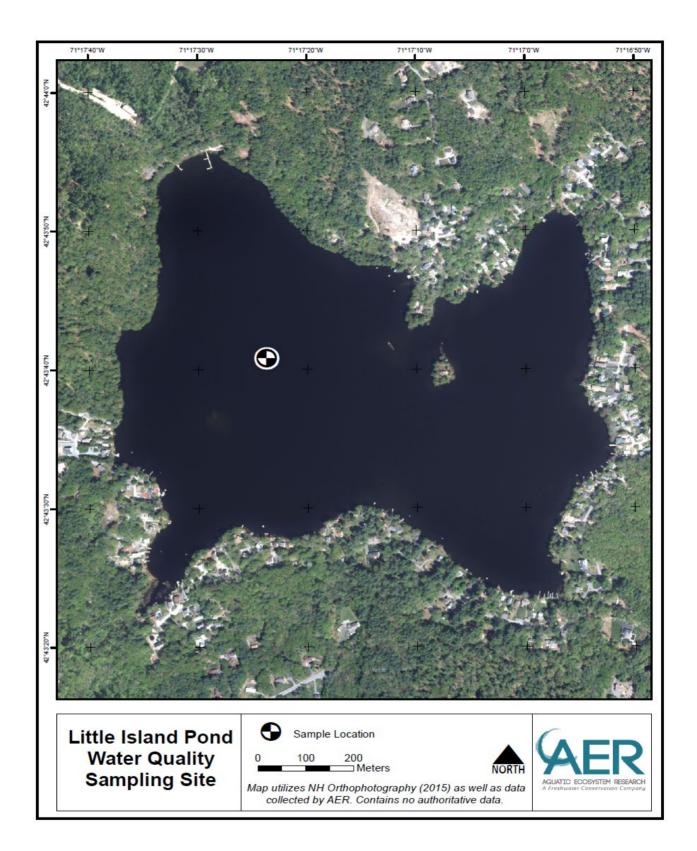
Table of Contents

Tab	Contents
Sampling Site	This tab contains a map of Little Island Pond and the location of the water quality sampling sites.
Database	This tab provides all the field and water chemistry data by depth. Future and past data can be added.
Field Data	This tab provides the data collected in the field (e.g. Secchi transparency, profile data). In addition, it provides density values for each depth and RTRM scores calculated for adjoining depths within the water column (See Temperature and Dissolved Oxygen tab).
Temperature and Dissolved Oxygen	This tab provides an explanation and graphical respresentations of temperature, RTRM, and oxygen profiles of the Little Island water column on July 30, 2020.
Specific Conductance ORP, pH	This tab provides an explanation and graphical respresentations of specific conductance, ORP, and pH of the Little Island water column on July 30, 2020.
Algal Communty	This tab contains a list of algal genera, results from analyses of algae concentrations, and a profile of relative Cyanobacteria biomass within the water column of the Little Island water column on July 30, 2020.
Trophic and Nutrient Data	This tab provides an explanation and graphical representation of the trophic and nutrient data determined for water samples collected at Little Island Pond on July 30, 2020. A table with trophic criteria is also provided.
Trophic Assessment	This tab provides an assessment of the trophic status of Little Island Pond based on analysis of data and water samples collected on July 30, 2020.
lon Data	This tab provides an explanation and graphical representation of the ion data determined for water samples collected at Little Island Pond on July 30, 2020.
Lab Reports	This tab provides the results of analyses performed by HydroTechologies, LLC on water samples collected at Little Island Pond on July 30, 2020.



Date	Depth	Temp. (°C)	Oxygen (mg/L)	DO (%)	Rel. Cyano.	Spec Cond. (µS/cm)	ORP (mV)	рН	Chl-a (µg/L)	Secchi (m)	Ammonia (µg/L)	Nitrate (µg/L)	Nitrite (µg/L)	TKN (µg/L)	Tot. Nit. (µg/L)	Tot. Phos (µg/L)	Alkalinity (mg/L)	DOC (mg/L)	Na⁺ (mg/L)	K⁺ (mg/L)	Ca ²⁺ (mg/L)	Mg ²⁺ (mg/L)	CI [.] (mg/L)
30-Jul-20	0.5	29.1	7.7	104	22.4	206	219	8.5															
30-Jul-20	1	29.1	7.7	105	24.6	206	230	8.3	0.59	4.94	0	0	0	288	288	0	18	4.6	28	0	7.1	0	45
30-Jul-20	2	28.9	7.8	105	14.2	206	239	8.2															
30-Jul-20	3	28.4	7.8	104	1.3	205	246	8.0															
30-Jul-20	4	27.0	6.3	82	1.2	204	279	7.4															
30-Jul-20	5	23.0	2.6	31	1.0	197	307	6.9															
30-Jul-20	6	17.7	5.4	58	1.5	192	316	7.0															
30-Jul-20	7	14.9	5.7	59	1.2	192	317	7.0															
30-Jul-20	8	12.5	3.6	35	1.0	192	328	6.9															
30-Jul-20	9	11.6	2.2	21	1.2	191	336	6.7															
30-Jul-20	10	10.8	2.1	20	1.0	191	338	6.7															
30-Jul-20	11	10.1	1.0	9	0.9	192	341	6.7															
30-Jul-20	12	9.7	0.4	4	0.9	194	343	6.7															
30-Jul-20	13	9.5	0.2	1	1.1	196	343	6.7															
30-Jul-20	14	9.4	0.1	1	1.3	197	343	6.7															
30-Jul-20	15	9.4	0.1	1	1.1	198	343	6.7															
30-Jul-20	16	9.3	0.0	0	1.0	199	343	6.7															
30-Jul-20	16.5	9.3	0.0	0	1.1	199	341	6.7			167	64	0	424	488	45	20	3.9					

30-Jul-20

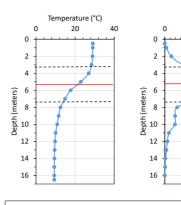
Depth	Temp	DO	DO	Rel.	Spec Cond.	ORP			
(m)	(°C)	(mg/L)	(%)	Cyano.	(µS/cm)	(mV)	рН	Water Density	RTRM
0.5	29.05	7.66	103.5	22.41	205.7	219.3	8.49	996.3965223	0
1	29.05	7.74	104.5	24.61	205.8	230.1	8.29	996.3965223	2.98386
2	28.93	7.8	105.1	14.21	205.7	239.2	8.15	996.4203932	12.4343
3	28.43	7.77	103.8	1.27	205.2	246.3	8.03	996.5198675	36.0735
4	26.98	6.29	81.9	1.19	203.9	279	7.43	996.8084554	100.12
5	22.96	2.56	30.9	0.97	197.3	306.5	6.87	997.6094152	131.246
6	17.7	5.36	58.3	1.52	192.4	316.1	6.95	998.6593833	70.9778
7	14.86	5.73	58.7	1.16	191.8	317.3	6.99	999.2272056	58.0418
8	12.54	3.64	35.4	0.99	191.5	327.8	6.85	999.6915398	23.5323
9	11.6	2.23	21.2	1.18	191.4	336.4	6.71	999.8797981	19.5335
10	10.82	2.1	19.6	1.03	191.1	337.7	6.7	1000.036066	19.2891
11	10.05	0.96	8.8	0.85	192.1	341.4	6.65	1000.190379	7.76741
12	9.74	0.42	3.8	0.86	194.2	342.6	6.65	1000.252518	5.01175
13	9.54	0.15	1.3	1.12	195.5	342.9	6.66	1000.292612	3.00724
14	9.42	0.08	0.7	1.25	197.3	342.8	6.67	1000.31667	1.25306
15	9.37	0.06	0.5	1.1	198.3	342.8	6.67	1000.326695	0.75185
16	9.34	0.04	0.4	0.96	199.2	342.7	6.68	1000.332709	0
16.5	9.34	0.02	0.2	1.06	199.4	340.5	6.72	1000.332709	

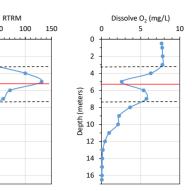
Secchi Disk Transparency	4.94

Temperature profile data allows for the determination of thermal characteristics by affording the ability to calculate where the water column is and is not mixing due to temperature/density differences. In shallow New England lakes or sites in the same lake, stratification can occur but it may be short in duration as energy from wind can mix the water column. In deeper lakes or sites, a middle transitional layer (aka metalimnion) separates the upper warmer layer (aka epilimnion) from lower colder waters below (aka hypolimnion). Within the metalimnion boundaries resides the thermocline, which is the layer between strata where temperature/density changes are greatest with increasing depth. Stratified conditions will often persist in deeper lakes or sites for the entire summer and into the fall until turnover mixes the water column.

Resistance to mixing, which is an assessment of the ability of two different water volumes – that differ in temperature and density – to mix, was calculated from temperature profile data using the Relative Thermal Resistance to Mixing (RTRM) formula: $(D_1 - D_2)/(b^2 - D^3)$, where D_1 is the density of upper water volume, D_2 is the density of the lower water volume, D' is the density of water at S^5 C, and D^6 is the density of water at 4^4 C. The thermocline in the water column, where resistance to mix is greatest, is that stratum with the highest RTRM value that is 320. RTRM value of 280 is indicative of strong resistance to mixing.

An oxygen concentration of Smg/L is generally thought to be the threshold that delineates favorable conditions for most aerobic organisms in freshwater systems. As concentrations decrease below that threshold, conditions become stressful for different forms of life. Minimum oxygen requirements for fisheries in Connecticut's lakes and ponds range from 4 to Smg/L for cold-water fish (e.g. trout), 2mg/L for cold-water fish (e.g. bass and panfish; Jacobs and O'Donnell 2002).





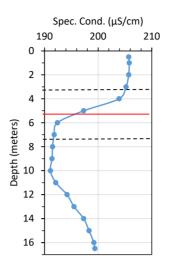
Profiles of temperature (left), Relative Thermal Resistance to Mixing (RTRM, middle) and dissolved oxygen (right) at Little Island Pond on July 30, 2020. The dashed lines represent the upper and lower boundary of

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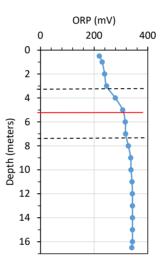
Conductivity is a surrogate measure for the ion concentration in water; simply put, it is a measure of water's ability to transmit an electrical current. Specific conductance is conductivity standardized to a set water temperature (25°C) because, in the field, temperature varies with depths and/or dates. Specific conductance (i.e. conductivity) is an important metric in Limnological studies due to its ability to detect pollutants and/or nutrient loadings. Specific conductance can also have an influence on organisms that inhabit a lake or pond; particularly, algae. The composition of algal communities has been shown to be related, in part, to conductivity levels in lakes (e.g. Siver 1993, McMaster &

The oxidation-reduction potential (otherwise known as redox potential or ORP) in lakes refers to the oxidative or reductive state in a particular stratum of the water column and can aid in understanding phosphorus dynamics in the lentic system. In general, when OPR is ≥200 millivolts (mV) phosphate remains bound to iron; at ORP values of <200mV, iron starts to becomes reduced and phosphate starts to be released (Søndergaard 2009). Under the latter scenario, a sudden mixing of phosphateladen waters to the upper reaches of the water column during a storm event could trigger a harmful algal bloom.

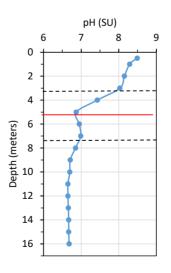
The pH of lake water is important for several reasons. Firstly, very low (<5) or very high (>9) pH levels will not support diverse lentic flora and fauna. Algal communities are influenced by pH due in part - to its influence on the form of dissolved carbon in the water column. For example, at pH greater than 8.3, bicarbonate is the dominant form of carbon available to the pelagic algal community. Cyanobacteria (aka bluegreen algae) have an adaptive advantage over other algal groups because they can more efficiently utilize that form of carbon. Other algal groups are dependent upon carbon dioxide, which is much less available in water with a pH above 8.3.



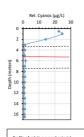
Profile of specific conductance (Spec. Cond.) at Little Island Pond on July 30, 2020. The dashed lines represent the upper and lower boundary of the metalimnion and the solid red line represents the approximate location of the thermocline.



Profile of oxidation-reduction potenial (ORP) at Little Island Pond on July 30, 2020. The dashed lines represent the upper and lower boundary of the metalimnion and the solid red line represents the approximate location of the thermocline.



Profile of pH at Little Island Pond on July 30, 2020. The dashed lines represent the upper and lower boundary of the metalimnion and the solid red line represents the approximate location of the thermocline. Relative cyanobacteria (aka blue green daga) biomas profiles were collected with a fluorimeter incorporated into the annumeter. Fluorimeters work on the principal that a particular substance fluoresces at a specific wavelength when light of another suscelength a fluorimeter in ARS-field instrumentation emits a wavelength that interacts with a partocyathetic proposanity. This amount is at a proposanity therefore, measurements are not quantitative, instead the measurements in the water column or at other sites.



Pr	ofile of relative cyanobacterial
bi	omass (Rel. Cyanos.) at Little Island
Pc	ind on July 30, 2020. The dashed
lin	es represent the upper and lower
bc	undary of
th	e metalimnion and the solid red line
re	presents the approximate location

Таха	Genus / species	Cells / mL	%	Taxa cells / mL	Taxa %
Cyanophyta	Chroococcus sp.	15	1.8	303	37.4
	Dolichospermum sp.	13	1.6		
	Pseudoanabaena sp.	84	10.4		
	Woronichinia sp.	191	23.6		
Chlorophyta	Cosmanium sp.	4	0.5	436	53.8
	Coelastrum sp.	64	7.9		
	Gloeocystis sp.	288	35.6		
	Nephrocytium sp.	41	5.1		
	Oocystis sp.	13	1.6		
	Scenedesumus sp.	19	2.3		
	Sphaerocystis sp.	7	0.9		
Chrysophyta	Mallomonas sp.	2	0.2	15	1.8
	Uroglenopsis sp.	13	1.6		
Bacillariophyta	Tabellaria sp.	laria sp. 2 0.2 13	1.6		
	Epiphytic Diatom	11	1.4		
Pyrrohophyta	Ceratium sp.	0	0.0	6	0.7
	Glenodinium sp.	6	0.7		
	Gymnodinium sp.	0	0.0		
	Peridinium sp.	0	0.0		
Cryptophyta	Cryptomonas sp.	13	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1.6	
	Rhodomonas sp.	0	0.0		
Euglenophyta	Euglena sp.	0	0.0	4	0.5
	Phacus sp.	ō	0.0		
	Trachelomonas sp.	4	0.5		
Ochrophyta	Stichogloea sp.	7	0.9	7	0.9
	Unidentified	13	1.6	13	1.6
	Totals	810	100	810	100.0

CYANOPHYTA	Chroococcus sp.					
	Dolichospermum sp.					
	Microcystis sp.					
	Pseudoanabaena sp.					
	Woronichinia sp.					
CHLOROPHYTA	Coelastrum sp.					
	Cosmanium sp.					
	Elakatothrix sp.					
	Gloeocystis sp.					
	Nephrocytium sp.					
	Oocystis					
	Quadrigula sp.					
	Scenedesmus sp.					
	Sorastrum sp.					
	Sphaerocystis sp.					
CHRYSOPHYTA	Mallomonas sp.					
	Uroglenopsis sp.					
BACILLARIOPHYTA	Tabellaria sp.					
PYRRHOPHYTA	Glenodinium sp.					
CRYPTOPHYTA	Cryptomonas sp.					
EUGLENOPHYTA	Trachelomonas sp.					
OCHROPHYTA	Stichogloea sp.					

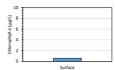
Analyses	Units	Surface	Bottom
Chlorophyll-a	µg/L	0.59	N/A
Secch Transparency	meter	4.94	N/A
Ammonia	µg/L	0	167
Nitrate	µg/L	0	64
Nitrite	µg/L	0	0
TKN	µg/L	288	424
TN	µg/L	288	488
TP	µg/L	0	45
Alklalinity	mg/L	18	20
Dissolved Organic Carbon (DOC)	mg/L	4.6	3.9

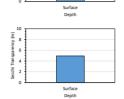
Chlorophyll-a is a photosynthetic pigment common to all algae, including blue-green algae (aka cyanobacteria); it is used by the algae and plants in the production of sugars and can be used as a proxy for algal biomass.

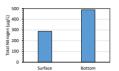
proxy for algal biomast. Seccial disk transparency is a standard Limological measurement that assesses how multiplik a transmitted through the water column. Now far light penetrates is invessivy related to the amount particulate matter suppend on the water column that abouts ar scatters light. The particulate matter can be inorgalic (e.g. sit), starting the start scatter in the inorgalic (e.g. sit), starting the start scatter in the inorgalic (e.g. sit), starting the start scatter in the inorgalic (e.g. sit), scatter in the water column, the measurement has historically been used to assess algal productivity and trophic level (Table 1). Seccid disk transparency is also used to determine the strata in the water column where enough light penetrates to support a condition where more organic is produced from aligh choisynthesis thin is compensation Depth and is estimated by doubling the Seccid transparency. Phosphorus in the thewater systems is commonly the nutrient in the

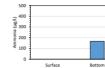
Compensation Depth and is estimated by doubling the Secchi transparency. Phosphorus in freshwater systems is commonly the nutrient in the shortest supply and in greatest demand by the alger, therefore, it dents the nutrient limiting alger phosticity. Sources of phosphorus can be imported from the watersheld (i.e. allochthonous) for devi-denternally from anous definents (i.e. autokthonous). Total phosphorus is the analysis frequently conducted; it represents all from of phosphorus in a sample, i.e. greaticulate and soluble form. Nitrogen is commonly the second most limiting nutrient for algein extensive strainers. It can be present in a number of forms in lake water. Ammonia – a reduced from of nitrogen – is important because it can affect the productively, diversity, and dynamics of the algait and plant communities. Ammonia can be indicated or literation (e.g. nitrite and nitrate) in lieu of oxygen for cellular respiration dured ranoic condition, resulting in annone and/ment of the hypolimon.

hypoliminon. Total kijdala linitagen (ka. TMA) is a messum of the reduced forms of nitrogen (including amonola) and total organic proteins in the water colume. Since TOX incounts for biologically derived, nitrogen-rich protection is the water colume. It is useful in assessing the productivity of the line kircly setter. Nitrates and nitrite are of ento below detectable levels in natural systems because they are quickly cycled by bacteria and against plants. Total nitrogen is the sum total of TNA, nitrate, and nitrite. Since the latter two are often below detectable limits, TNA levels are often similar or equal to total nitrogen levels.







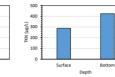


Depth

Trophic and No categor

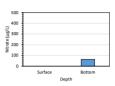
Trophic Category

Oligotrophic Early Mesotrophic Mesotrophic Late Mesotrophic Eutrophic Highly Eutrophic

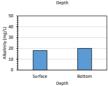


Classification orientia used by the Connecticut Experimental Applicational Station (Print, real), B4(a) and the CT DEP (B51) to assess the trophic status of Connecticut Idaes. The rearrange from objectorphic or least productive to highly userphic or most productive.

Category
Total Nitrogen Characterization (Up (L))
Summer Sacchi (Up (L))
Summer Sac



Depth 50 (1/8rf) snuodsoyd 01 gt n Surface Bottom Denth



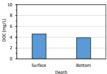


Table 7. Trophic classification criteria used by the Connecticut Experimental Agricultural Station (Frink and Norvell, 1984) and the CT DEP (1991) to assess the trophic status of Connecticut lakes. The categories range from oligotrophic or least productive to highly eutrophic or most productive.

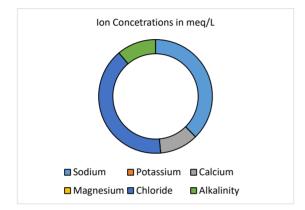
	Total	Total	Summer	Summer Secchi
Trophic Category	Phosphorus	Nitrogen	Chlorophyll-a	Disk Transparency
	(µg / L)	(µg / L)	(µg / L)	(m)
Oligotrophic	0 - 10	0 - 200	0 - 2	>6
Early Mesotrophic	10 - 15	200 - 300	2 - 5	4-6
Mesotrophic	15 - 25	300 - 500	5 - 10	3 - 4
Late Mesotrophic	25 - 30	500 - 600	10 - 15	2 3
Eutrophic	30 - 50	600 - 1000	15 - 30	1 - 2
Highly Eutrophic	> 50	> 1000	> 30	0 - 1

Based on these trophic classification criteria and conditions on July 30, 2020, Little Island Pond is an oligotrophic - early mesotrophic lake.

 $\langle \rangle$

Denotes levels measured at 1 meter of depth at Little Island Pond on July 30, 2020..

lons at 1m of Depth	mg/L	meq/L
Sodium	28	1.22
Potassium	0	0
Calcium	7.1	0.36
Magnesium	0	0
Chloride	45	1.31
Alkalinity	18	0.36
Total Cations	35.1	1.57
Total Anions	63	1.67



Base cation and anion concentrations are important in understanding natural influences (e.g. dissolved salts from bedrock geology) as well as anthropogenic influences from the watershed (e.g. road salts). In lakes of the Northeast, the dominant base cations in lake waters are calcium (Ca^{2+}), magnesium (Mg^{2+}), sodium (Na^{+}) and potassium (K^{+}). Dominant anions include chloride (Cl-), sulfate (SO_4^{-2-}), carbonate (CO^{2-}_3), and bicarbonate (HCO_3^{-}). In this assessment we examined the base cations, chloride, and alkalinity anions (carbonate and bicarbonate).

62 Bank St. New Milford, CT 06 860/355 8773 TEL 860/350 2258 FAX www.gohydro.com		ECHNOLOGIES	uc	CT DPH LAB #PH-0627 EPA LAB #CT00051 Water, Wastewater Soil and Air T esting Sampling and Consulting	62 Bank St. New Milford, CT 0 860/355 8773 TEL 860/350 2258 FAX www.gohydro.com		HYDROTEC		S 110	CT DPH LAB #PH-062 EPA LAB #CT0005 Water, Wastewate Soil and Air T esting Sampling and Consulting
	Repo	rt of Analysis					Report of	f Analysis		
Name: Sample Date: Receipt Date: Report Date: Site:	Aquatic Ecosystem Research 1204 Main Street Suite 161 Branford, CT 06405 77302020 4:30 PM 77312020 1:14 PM 82772020 10:2725 AM Little Island Pond	Work ID#: H243; Sample ID#: 33733 Sample Type: Surfac Sample Source: EPI Sampler: LM C	o e water		Name: Sample Date: Receipt Date: Report Date: Site:	1204 M Suite 1 Branfo 7/30/20 7/31/20 8/27/20	Aain Street Sar 61 Sar ord, CT 06405 Sar	ork ID#: H243 nple ID#: 3373 nple Type: Surfa nple Source: Hypa npler: LM C	31 ce water	
	Parameter	Sample Result	Units			[Parameter	Sample Result	Units	
	Biological Chlorophyll a	0.59	ug/L				Chemical Dissolved Organic Carbon	3.9	mg/L	
	Chemical Dissolved Organic Carbon	4.6	mg/L				Minerals Alkalinity	20	mg/L	
	Metals	4,0	mg/L				Nutrient	20	ing/L	-
	Potassium	<5.0	mg/L				Ammonia as N	0.167	mg/L	
	Sodium	28	mg/L				Kjeldahl Nitrogen as N	0.424	mg/L	
	Minerals						Nitrate as N	0.064	mg/L	
	Alkalinity	18	mg/L				Nitrite as N	ND	mg/L	
	Calcium	7.1	mg/L				Phosphorus-T as P	0.045	mg/L	
	Chloride	45	mg/L							
	Magnesium	<2.5	mg/L							
	Nutrient									
	Ammonia as N	ND	mg/L							
	Kjeldahl Nitrogen as N	0.288	mg/L							
	Nitrate as N	ND	mg/L							
	Nitrite as N	ND	mg/L							
	Phosphorus-T as P	ND	mg/L							
Comments: Ca,] Pti⊣	Mg, Na, K., Dissolved Organic Carbon by Envi	roTest Labs, Newburgh, NY -	CT DPH Lab#	ND − Not Detected * α Above Specified Limit **C = Data Qualifier	Comments: Dis Lab	solved O # PH-05	rganic Carbon results by EnviroTest Laborat S4.	ories, LLC Newburgh, 1	NY - CT DPH	ND - Not Detected • - Above Specified Limit •*< - Data Qualifier
		Du						DIII		
Note: The test res	by Hydro Technologies, LLC: ults are only valid for the date sample was taken. We do n CT DPH DWD reg. 19-13-B103 (c.1). All work perform	ot accept any liability for the use of the red is subject to Terms and Condition	rese results. Note: Unle	ss indicated otherwise, Odor dro.com (Terms)results.	Note: The test re	sults are or	Technologies, LLC; nly valid for the date sample was taken. We do not acce 1 DWD reg. 19-13-B103 (c.1). All work performed is st	pt any liability for the use of f abject to Terms and Condition	hese results. Note: Unly	ess indicated otherwise, Odor ydro.com (Terms)results.
		Page 1 of 1					Page 1	of 1		