
Decarburization behavior in steel sheet for hot stamping

**Retardation of decarburization
at higher temperatures**

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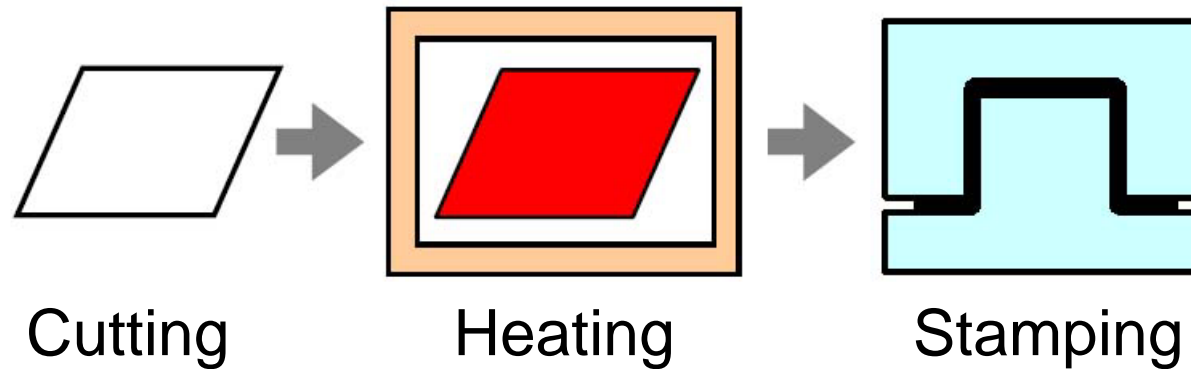
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Background

- Hot stamping

Automotive parts with tensile strength of 1500 MPa



- Heating conditions

Ambient or gas atmosphere (**Oxidation atmosphere**)

Austenitic temperature (**High temperature**)



Decarburization conditions

Experimental Procedure

- Sheet steel

Thickness: 1.6mm

Chemical composition in mass% and A_{e3} of steel :

C	Mn	Others	A_{e3}
0.21	1.2	Si, B, Ti, Cr	1073K



- Heating conditions

Rapid heating

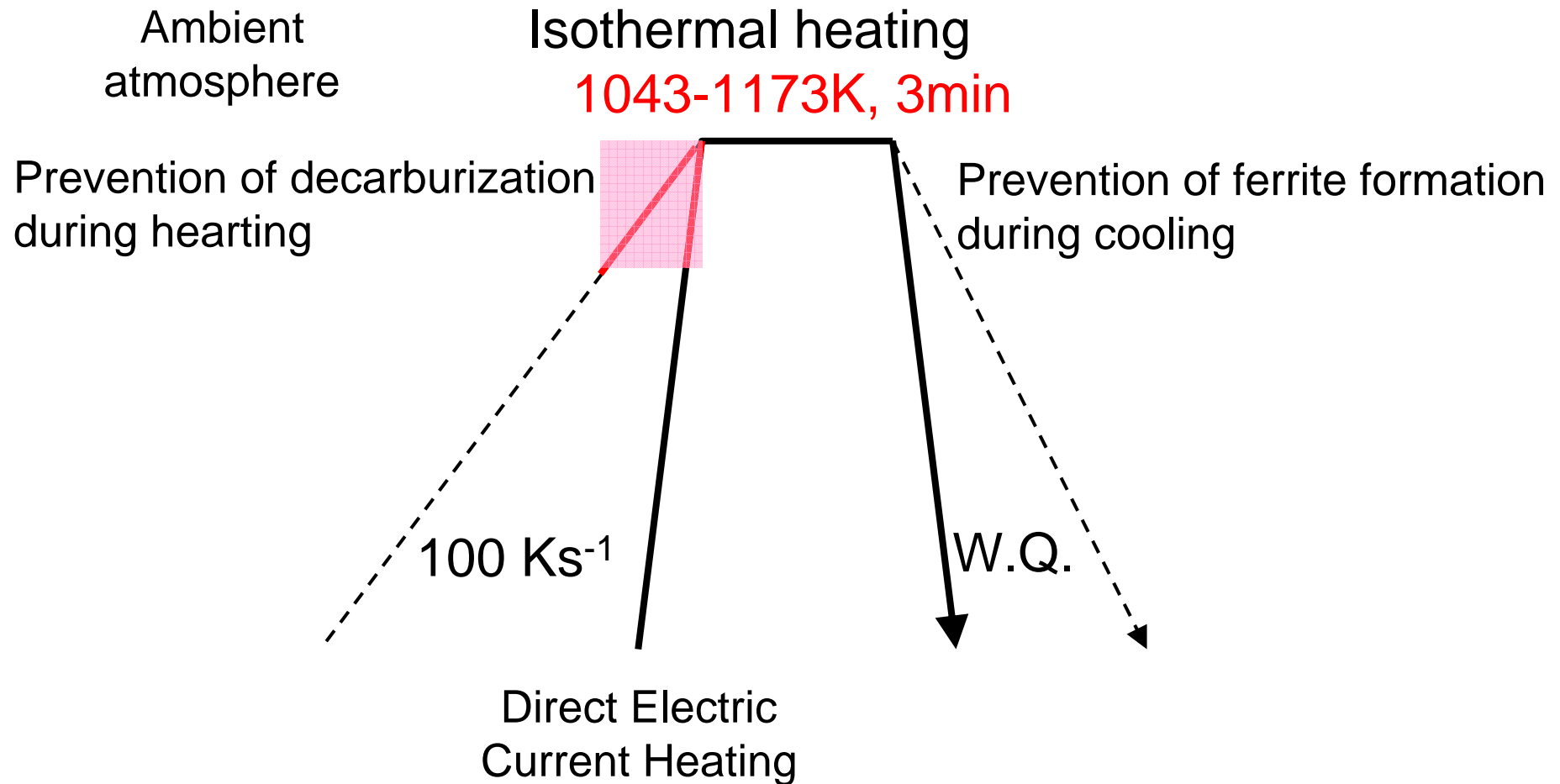
Water quenching

- Microstructure observation

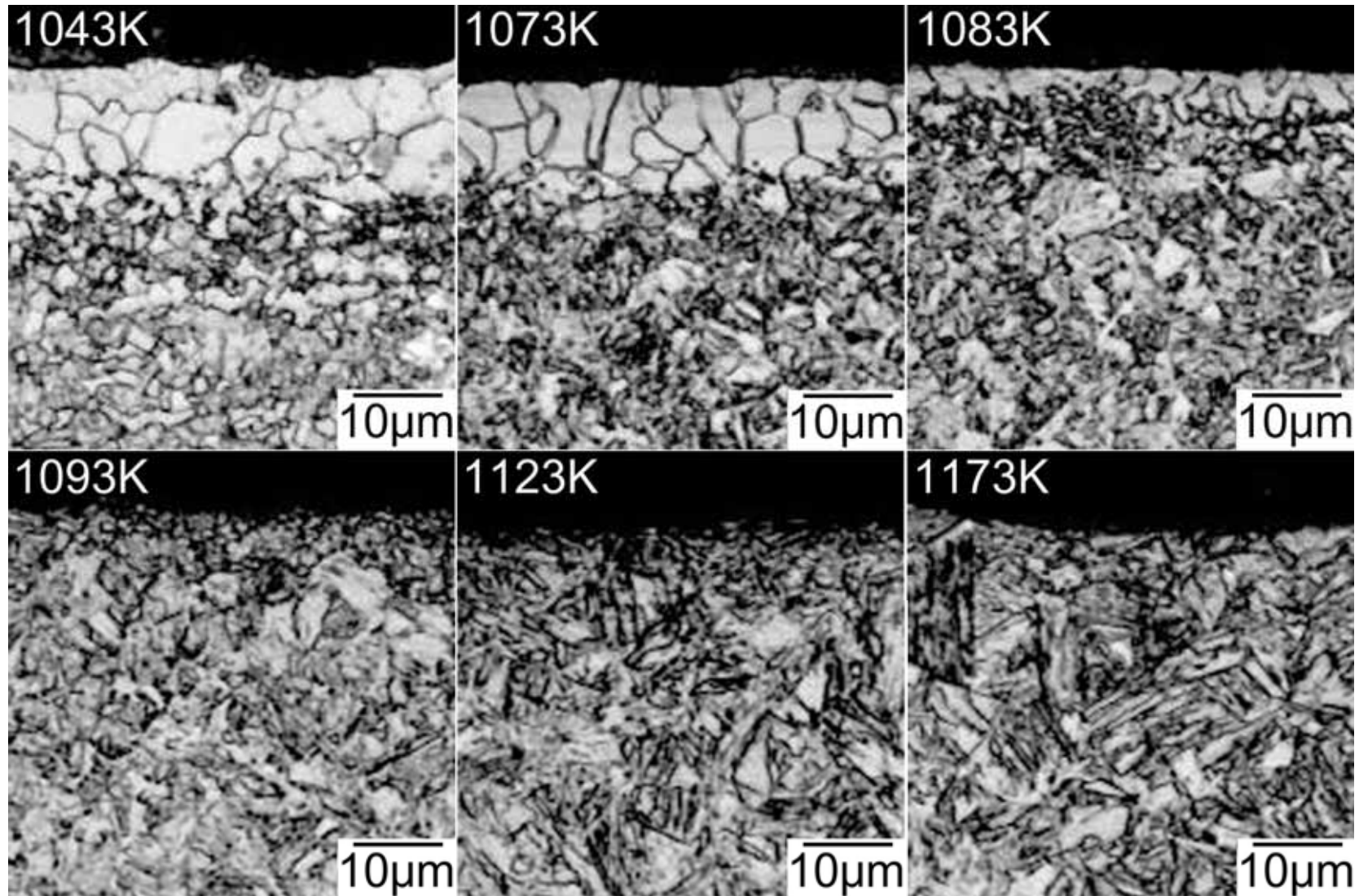
Laser microscope

Heating Conditions

Rapid heating & Water quenching

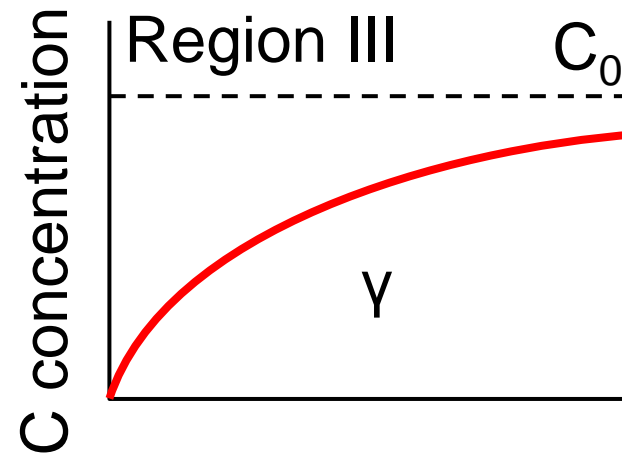
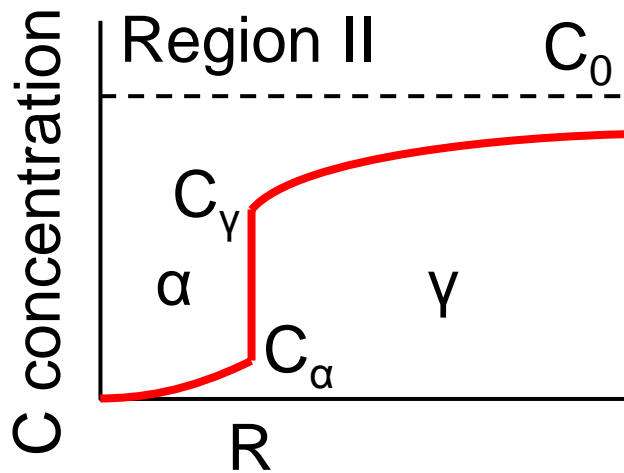
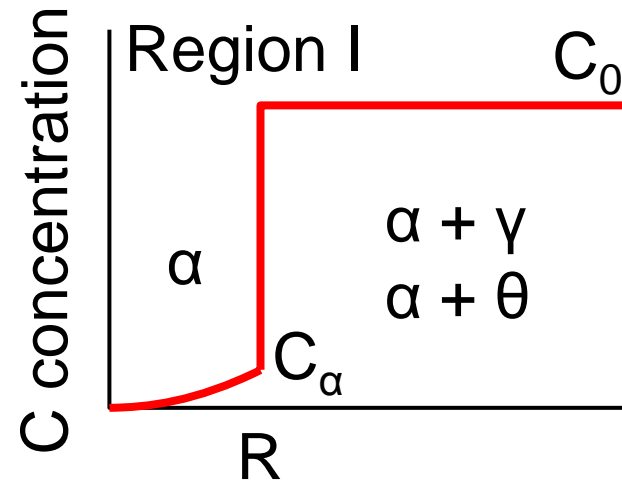
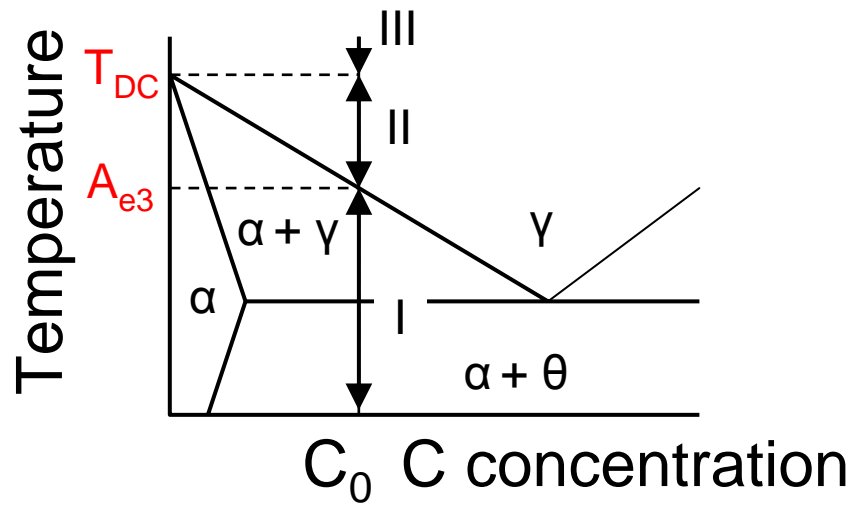


Microstructure



Decarburization Model

Birks and Jackson (1970)



Partition of alloying elements

Phillion(2004)

Hutchinson(2004)

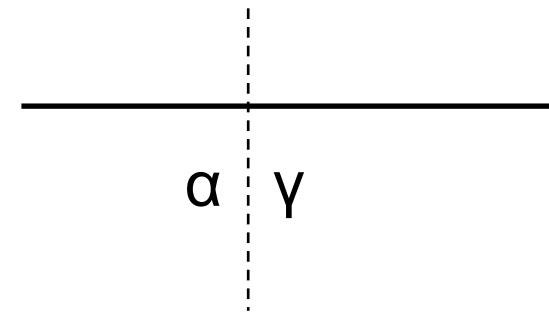
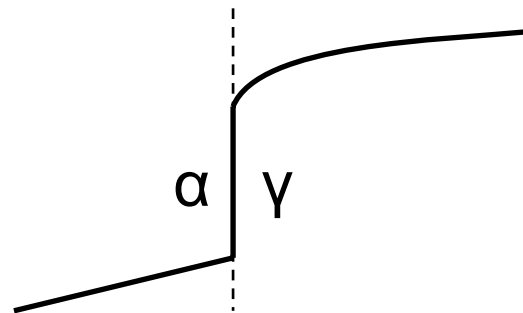
Beche(2007)

Zurob(2008)

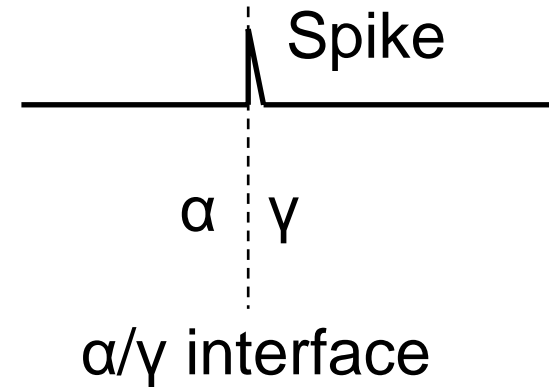
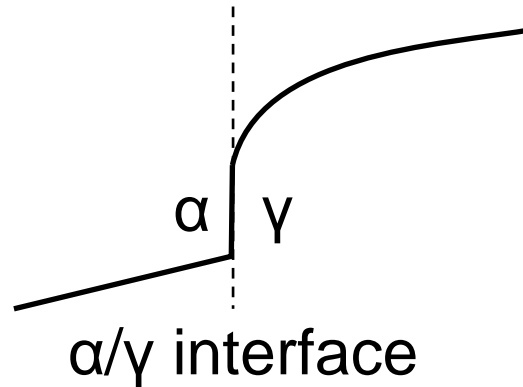
Para Equilibrium (PE)

Interstitial element
C profile

Substitutional element
Mn profile

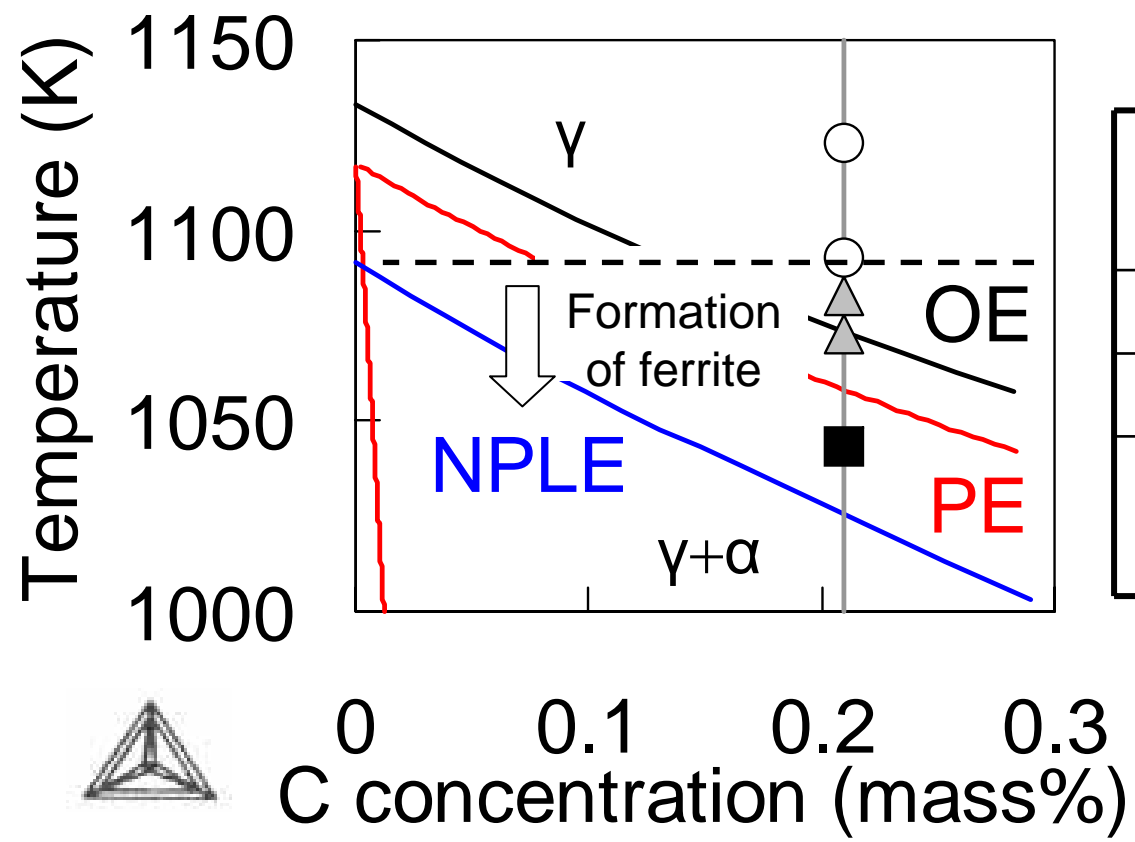


No Partition Local Equilibrium
(NPLE)



Phase diagram and Microstructure

Fe-C-1.2Mn



Isothermal heating

	Surface region	Inside region
	Martensite	Martensite
	Ferrite	Martensite
	Ferrite	F+M Dual-phase

Conclusions

Fe–0.21C–1.2Mn, $T = 1043–1173\text{K}$, $t = 180\text{s}$

- Formation of ferrite on surface due to decarburization

$$T = 1043–1083\text{K}$$

- Growth rate of ferrite: v

$$v_{1043\text{K}} > v_{1083\text{K}}$$

- Heating at $T > 1093\text{K} \Rightarrow$ Prevention of ferrite formation

- Relationship

$$A_{3(\text{OE}) \text{ at } 0\% \text{C}} > A_{3(\text{PE})} \quad T_{\text{DC}} \quad A_{3(\text{NPLE})}$$