

Integrated Models for Predicting the Effect of Manganese Fluctuation on the Phase Transformation of Industrial Hot-Forging Steels

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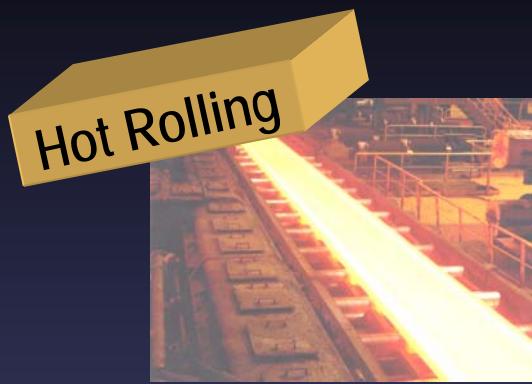
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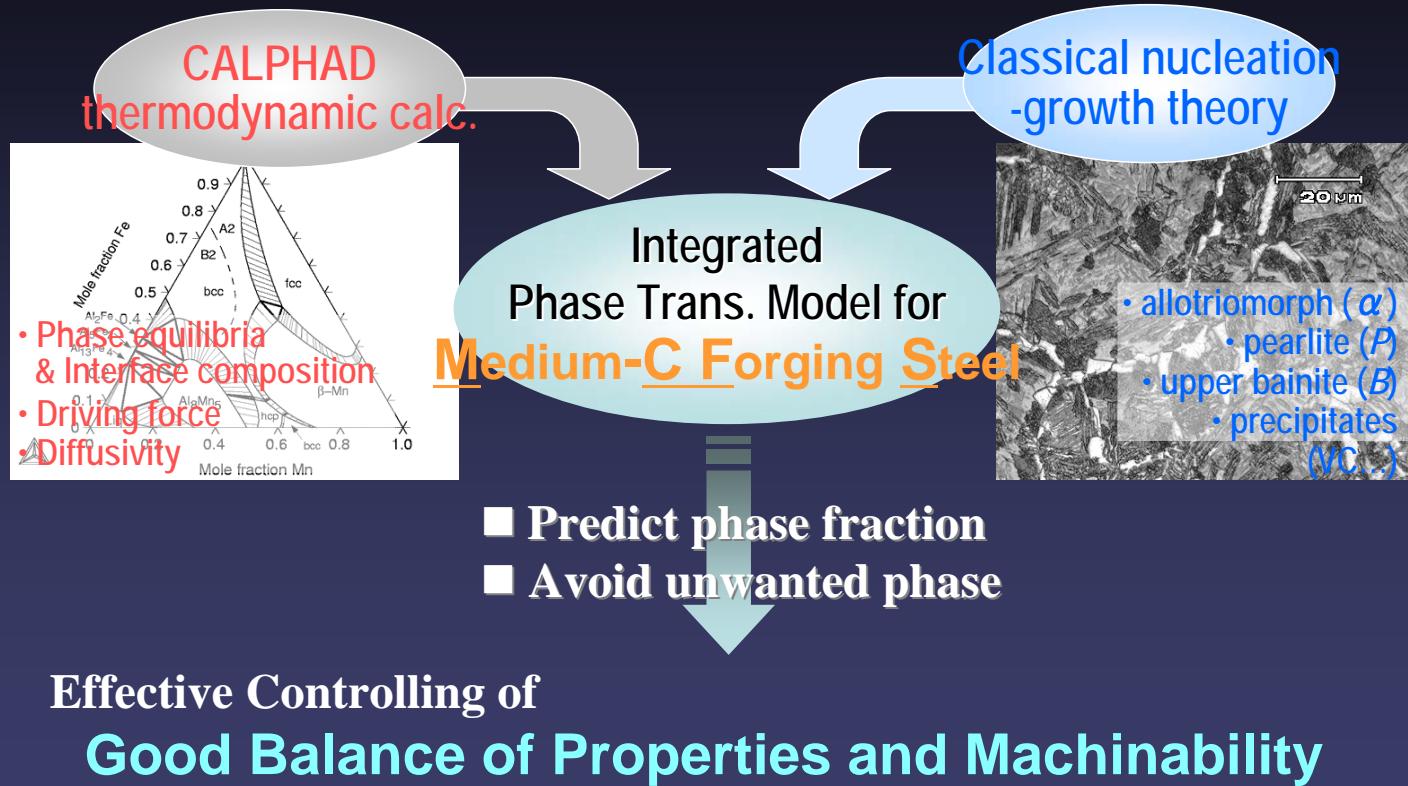


Phase Trans. Modeling in Steel Products



- '80s-'90s
Low-C sheet steel
→ for α grain size control
- Current
TRIP / DP steel
→ for γ_R distribution control
- '00s-Current
Medium C-Mn forging steel
→ Modelling required for the robust microstructure control of forging parts !

Objective of MCFS Phase Trans. Modeling



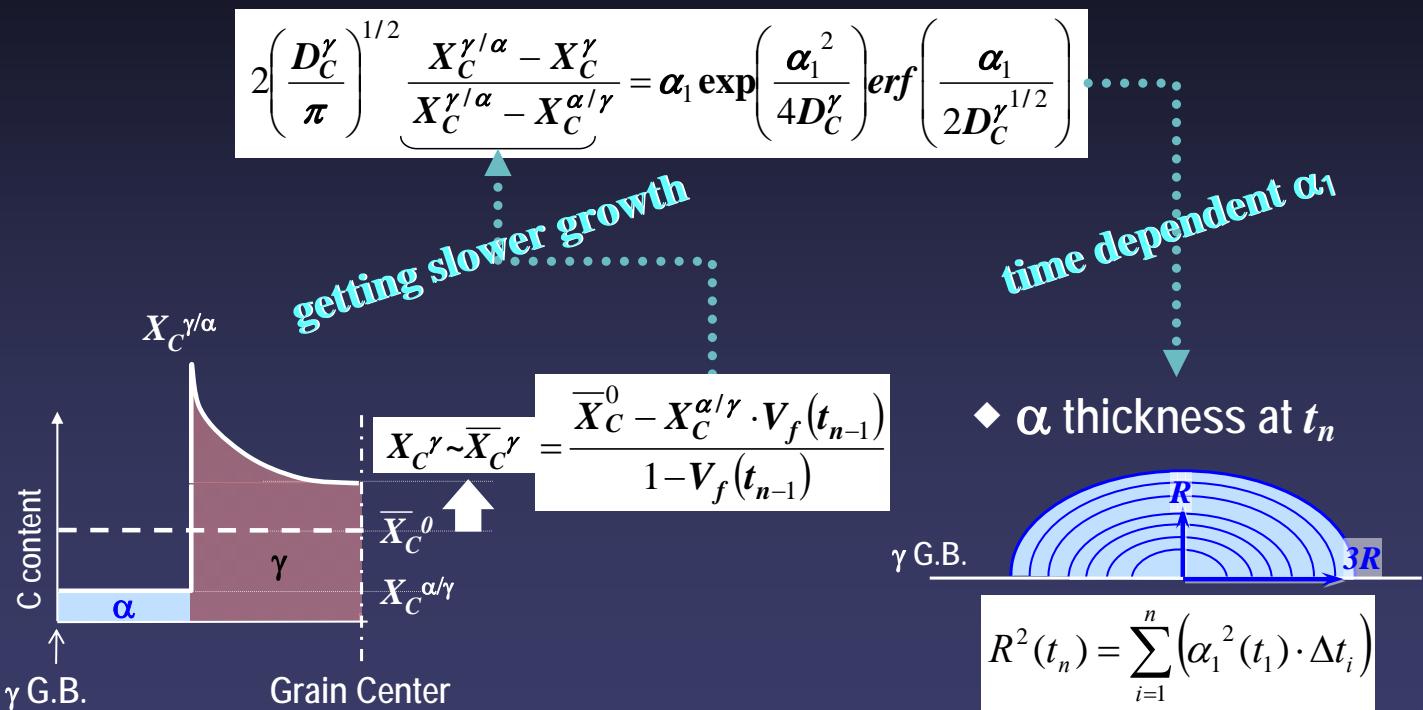
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Things to Consider in Phase Trans. of MCFS

- Compositions:
 - 0.2-0.5% C-Si-Mn-(Cr,V), mainly $\alpha + P$
- Growth kinetics of proeutectoid α
 - Soft impingement due to C enrichment in γ
 - Local interface equilibria - P.E. / N.P.L.E (Mn)
- Accuracy in predicting $\alpha \rightarrow P \rightarrow B$ transitions
 - Variation in C enrichment due to grain size distribution and preceding α

Soft Impingement Effect of Proeutectoid α

- ◆ Parabolic growth rate constant; α_1 (Christian '75)



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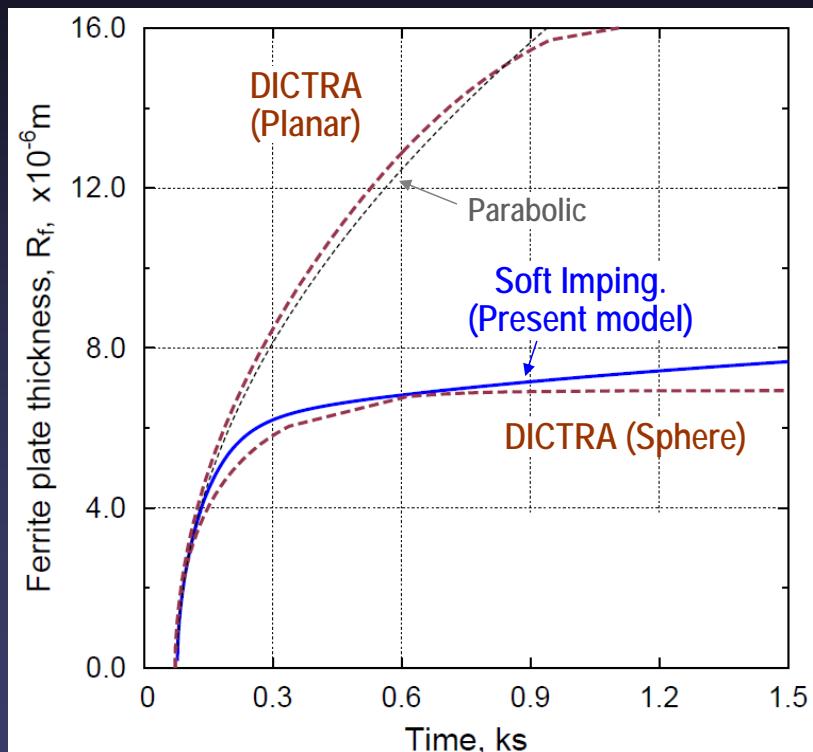
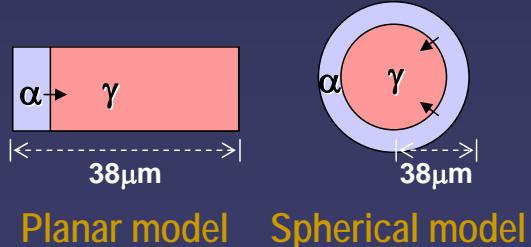
Comparison of α Growth under S.I.E with DICTRA Simulations

- ◆ α thickness: R_f

$$R_f = \sqrt{\sum_n \{\alpha_1(n\Delta t)\}^2 \Delta t}$$

$$= \sqrt{\sum_n \left\{ f\left(\bar{X}_C^{\gamma,n\Delta t}\right) \right\}^2 \Delta t}$$

- ◆ DICTRA simulation



Experimental & Calculation Procedure

- Steels investigated:

Steel	C	Si	Mn	Ni	Mo	Cr	V	Cu	Al
37V	0.37	0.56	1.45	-	0.03	0.02	0.11	0.14	0.024

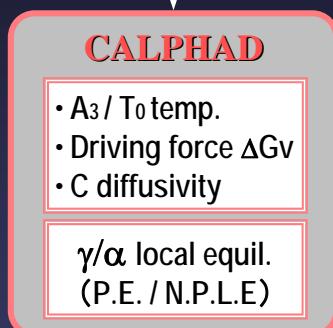
- Measurement of transformation kinetics:
 - Isothermal Test
 - 1523/1273K*60s ($d_0=11.76\mu\text{m}$) \rightarrow 973/913/873K*1500s \rightarrow He jet Q
(High resolution dilatometry, Quantitative metallography)
- Calculation:
 - Original program linked with Fortran Interface "TQ-i"
 - Thermodynamic & Mobility DB: TCFE3 & MOB2

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Calc. Flow of the Kinetics of Proeutectoid α

- Chemical compo.
- γ grain size: d_0
- Holding temperature

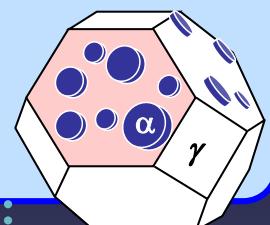


α nucleation?

Nucleation rate

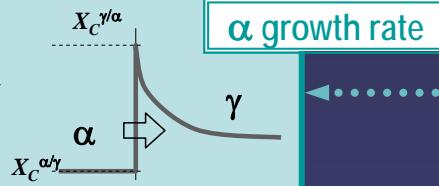
Geometry Model

α vol. fraction

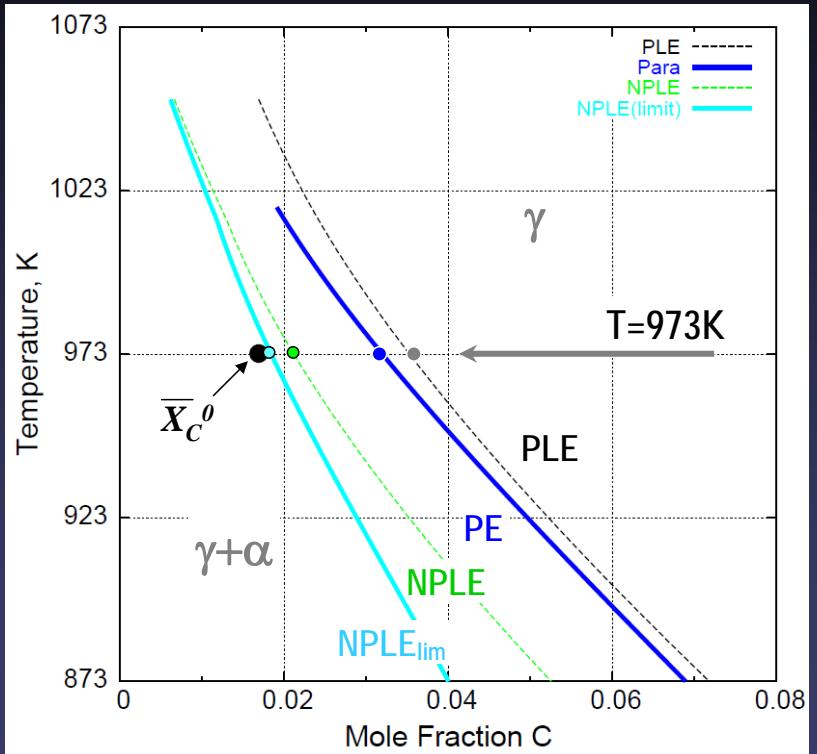
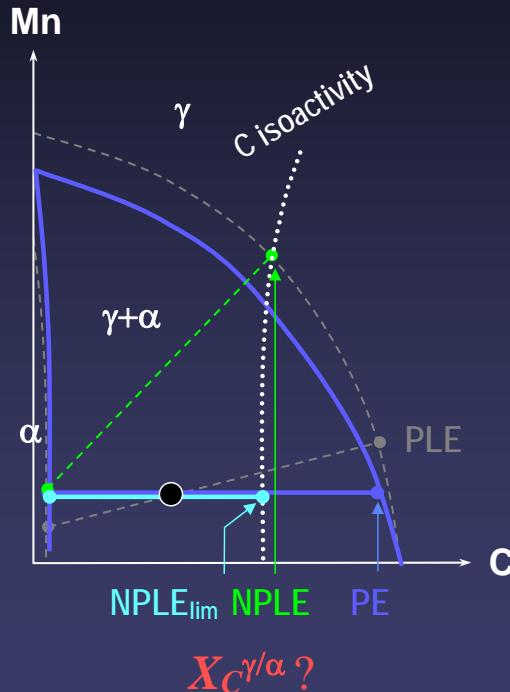


- C enrichment in γ
- Velocity of interface

**Diffusion-
Controlled
Growth
Model**



CALPHAD Calculation: γ/α Equilibria (37V)



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Comparison with the Detailed Kinetics (37V)

973K, $d_0 = 76\mu\text{m}$

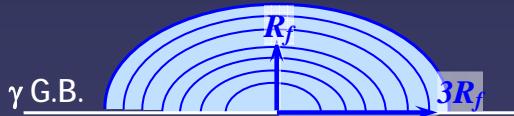
- ◆ α incubation time: F_s
(Lange, Enomoto)

$$\tau = \frac{12kTa^4 \cdot \sigma^{\alpha/\gamma}}{D_C^\gamma \overline{X}_C \cdot v_\alpha^2 \cdot (\Delta G_{\max} / V_m^\alpha)^2}$$

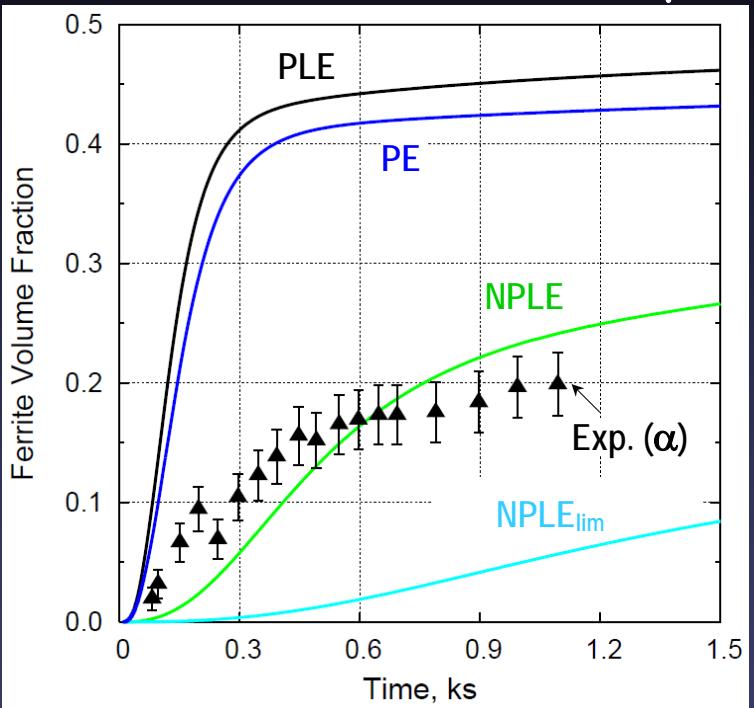
- ◆ α nucleation rate: I
(Aaronson, Umemoto, Yoshie)

$$I = \frac{K_1(1 - \overline{X}_C^\gamma) D_C^\gamma}{\sqrt{T}} \exp\left(-\frac{K_2}{RT \Delta G_V^2}\right)$$

- ◆ α volume fraction: V_f (Umemoto)



$$V_f(t_n)/V_f^{eq} = 1 - \exp\left[-2S_V \int_0^{R(t_n, t_1)} 1 - \exp\left(-\sum_{i=1}^k I \cdot \Delta t \cdot \pi \{R_f^2(t_n, t_i) - y^2\}\right) dy\right]$$



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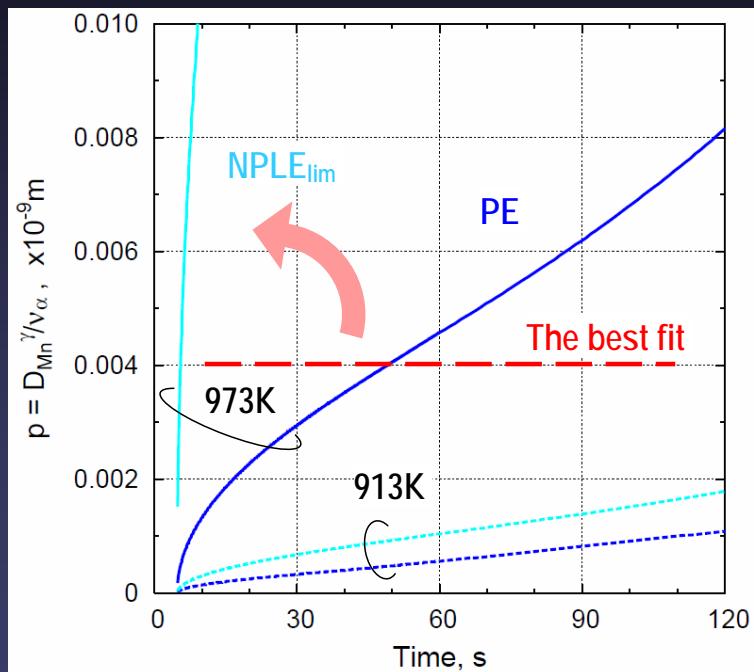
D/v criterion for PE to NPLE Transition

- ◆ γ/α Interface velocity: v^α

$$v^\alpha = \frac{dR_f(t)}{dt} = \frac{d(\alpha_1 \sqrt{t})}{dt} \approx \frac{\alpha_1}{2\sqrt{t}}$$

- ◆ D/v criterion: p
(Bradley and H.I.Aaronson)

$$p \equiv \frac{D_{Mn}^{Bound}}{v^\alpha} = \frac{n \cdot D_{Mn}^r \sqrt{t}}{\alpha_1}$$

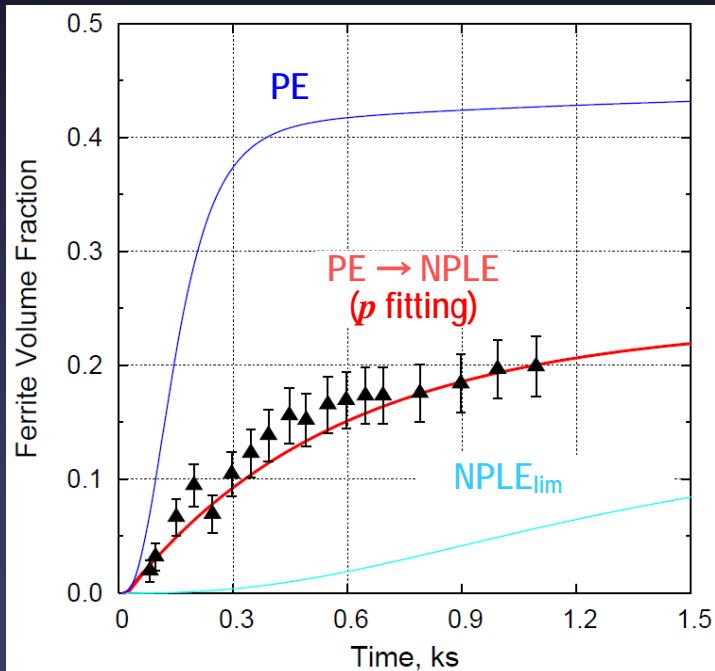


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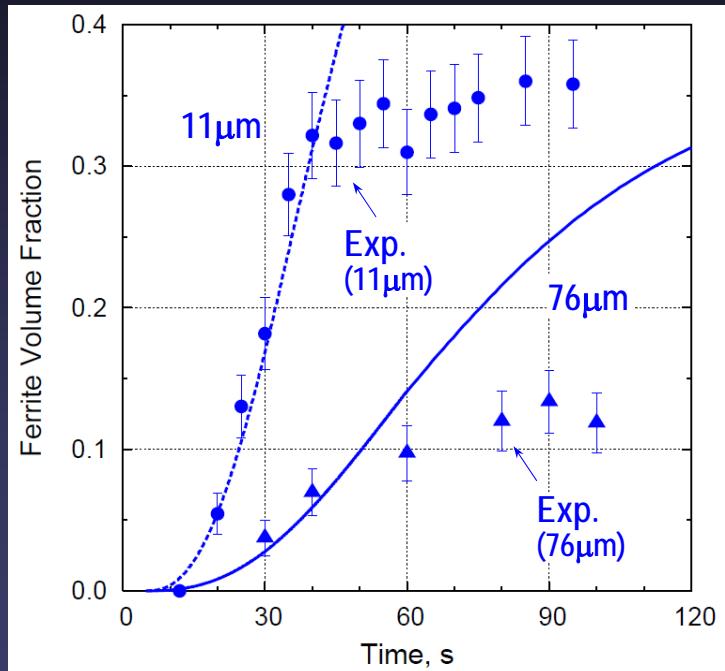
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Isothermal Kinetics with D/v criterion (37V)

- ◆ 973K, $d_0 = 76\mu\text{m}$



- ◆ 913K, $d_0 = 11, 76\mu\text{m}$



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Validation on the Effect of Mn Fluctuation

- Industrial Forging Steels (ASTM1538 grade):

	C	Si	Mn	Ni	Mo	Cr	Cu	Al
38M-H	0.39	0.51	1.69	0.07	0.01	0.15	0.12	0.009
38M-M	0.39	0.50	1.51	0.05	0.01	0.14	0.03	0.009
38M-L	0.39	0.51	1.29	0.06	0.01	0.15	0.11	0.008

- Measurement of transformation kinetics:

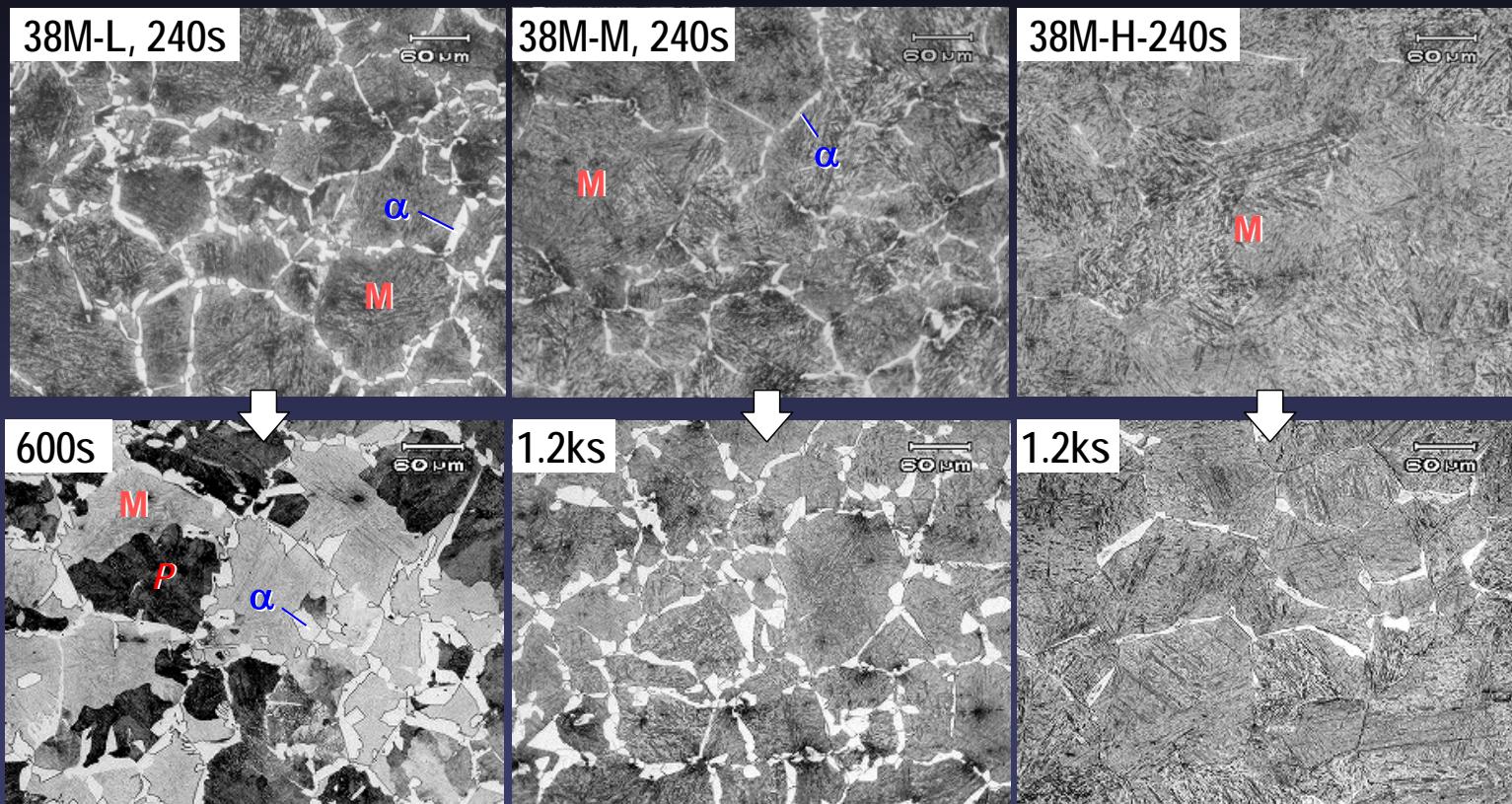
- Isothermal Test

- 1173K*70s ($d_0=57\mu\text{m}$) \rightarrow 928K*1200s \rightarrow He jet Q

- Continues Cooling Test

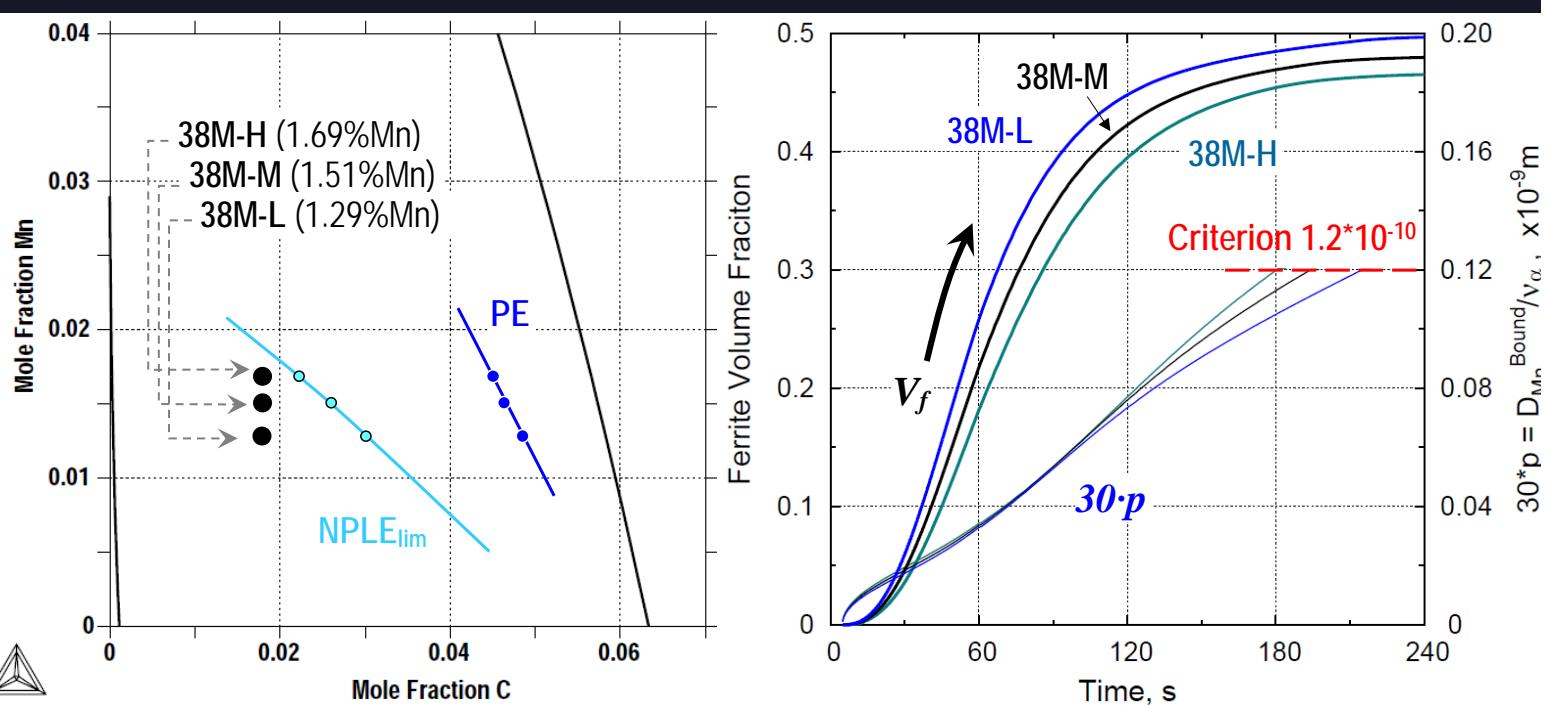
- 1173K*70s \rightarrow from 1063K (Ae3) at 10~100K/min. to 373K
(Dilatometry, Quantitative metallography)

Observed Isothermal Trans. Kinetics (38M)



CALPHAD: γ/α Equilibria and D/v Criterion

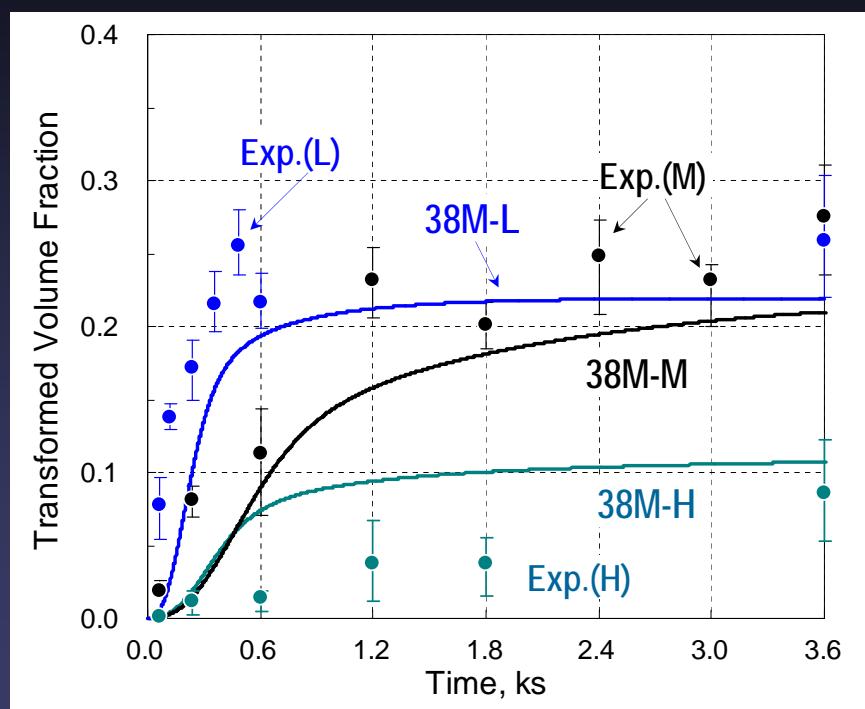
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Isothermal Kinetics in γ/α NPLE (38M)



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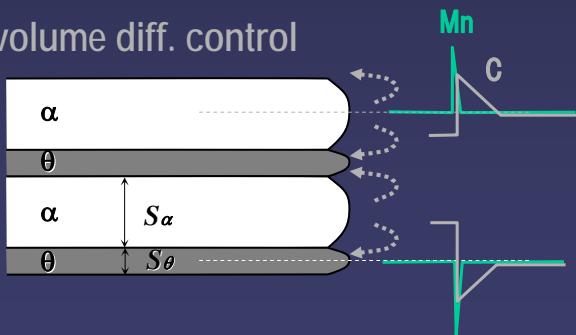
Growth Rate of Pearlite in Fe-C-Mn System

(Takahashi, Capdevila)

◆ LE mode: G_{LE} **Mn boundary diff. control**

$$G_{LE} = 12 \underbrace{K^p D_{Mn}^{Bound}}_{\delta} \frac{X_{Mn}^{\gamma/\alpha} - X_{Mn}^{\gamma/\theta}}{X_{Mn}} \frac{1}{S_\alpha S_\theta} \left(1 - \frac{S_{crit}}{S_0} \right)$$

$$K^p D_{Mn}^B \delta = 2.223 \cdot 10^{-7} \exp\left(-\frac{14940}{RT}\right)$$

◆ NPLE mode: G_{NPLE} **C volume diff. control**

$$S = 1.43 \cdot 10^{-7} - 6.40 \cdot 10^{-10} \cdot (Ae1 - T) \leq 2S_{crit}$$

$$S_{crit} = \frac{2\sigma^{\alpha/\theta}}{6.09 \cdot 10^{-8}} \left(\frac{Ae1}{Ae1 - T} \right)$$

$$\sigma^{\alpha/\theta} = 0.6 J/m^2, S_\alpha = 7/8 \cdot S, S_\theta = 1/8 \cdot S$$

$$G_{NPLE} = \frac{D_C^\gamma}{0.72} \frac{X_{C,NPLE}^{\gamma/\alpha} - X_{C,NPLE}^{\gamma/\theta}}{X_C^{\theta/\gamma} - X_{C,NPLE}^{\alpha/\gamma}} \frac{S}{S_\alpha S_\theta} \left(1 - \frac{S_{crit}}{S_0} \right)$$

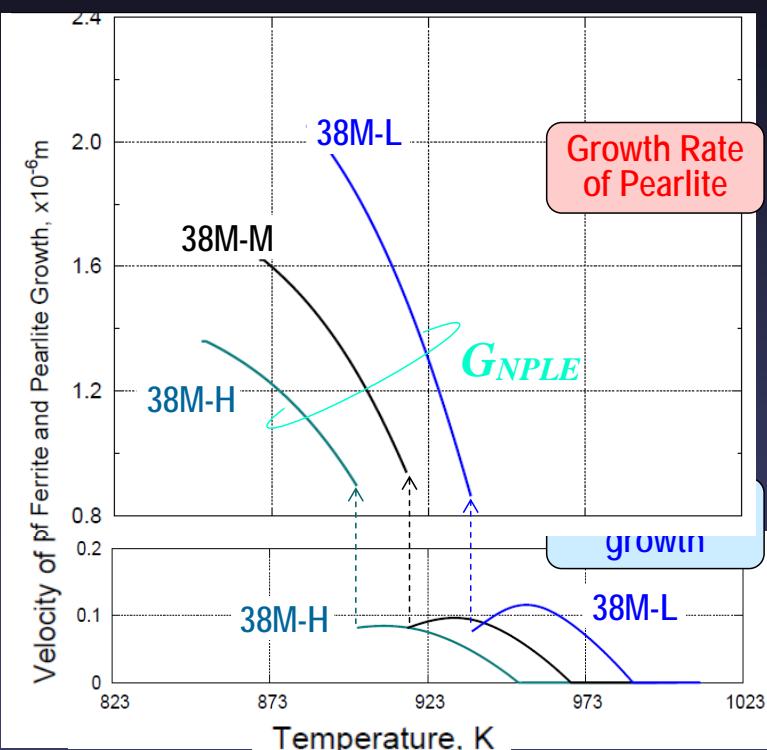
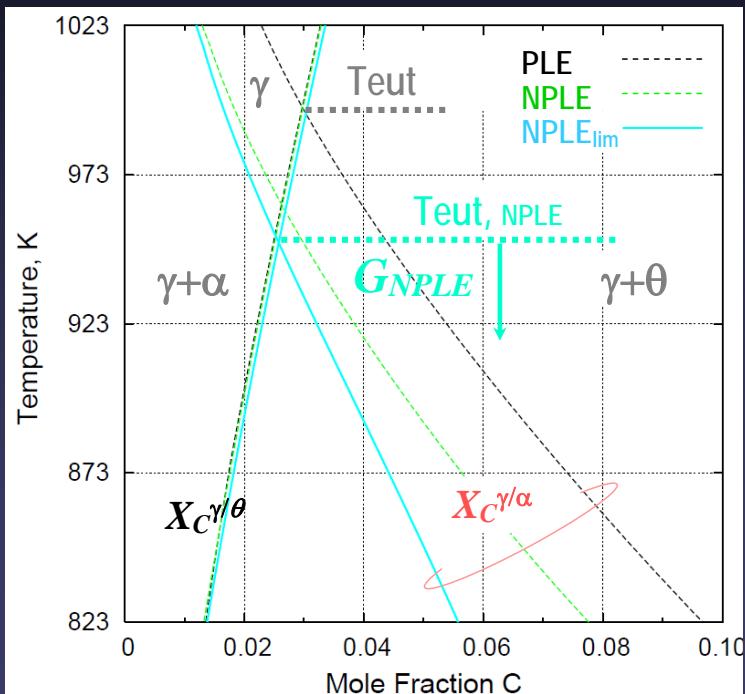
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γ/α and γ/θ Equilibria and Pearlite Growth Rate

ex. 38M-M



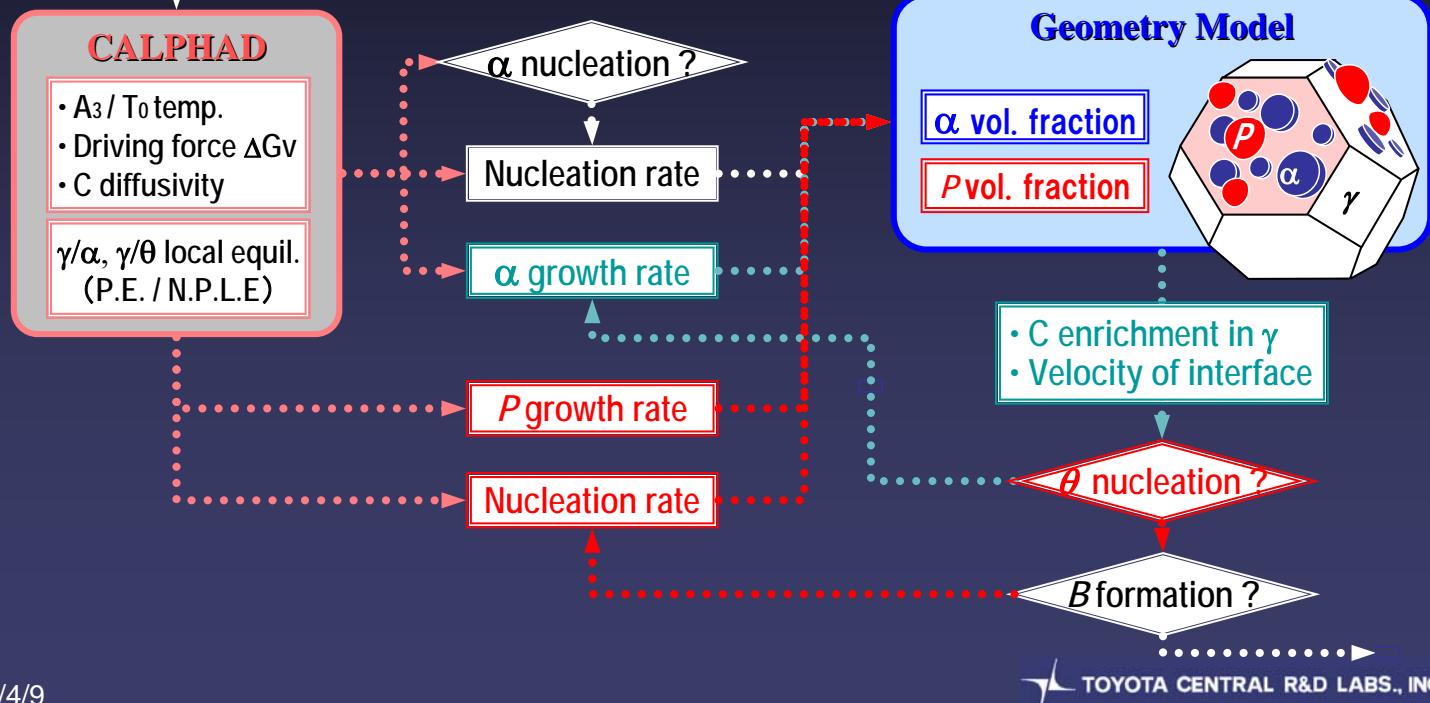
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Calc. Flow in The $\alpha/P/B$ Integrated Model

- Chemical composition
- Hot working condition
- Cooling schedule

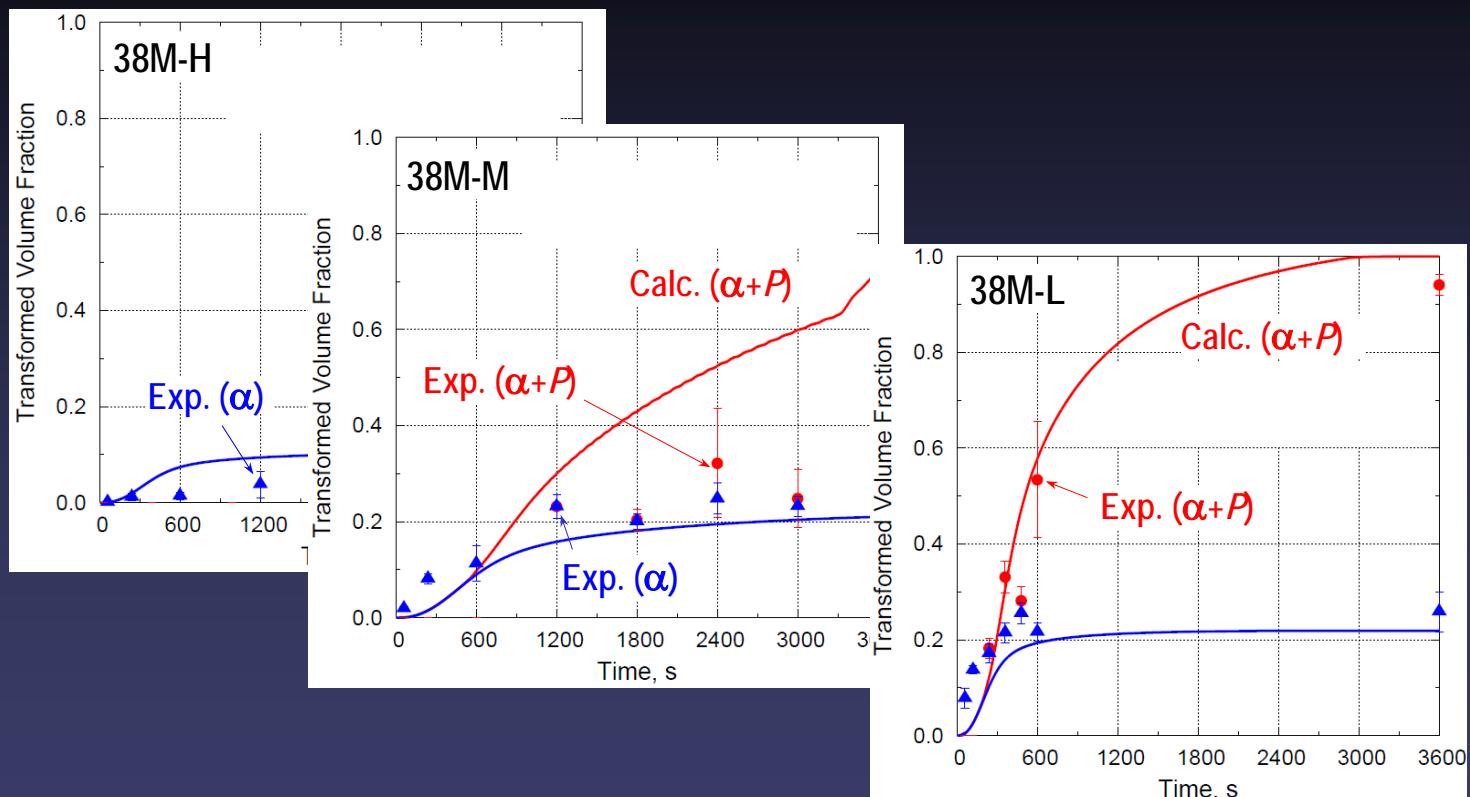
γ grain size: d_0
G.S. distribution: $d_0(n)$



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Comparison of Exp. & Calc. α/P Kinetics

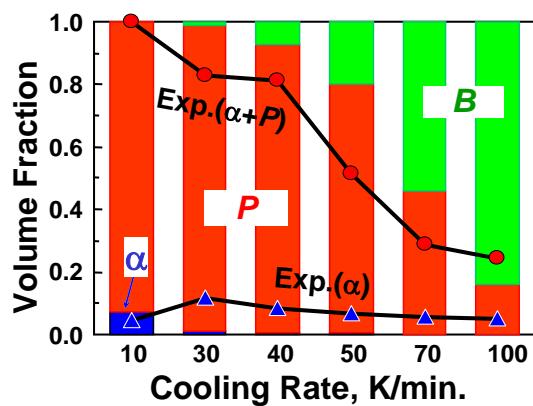


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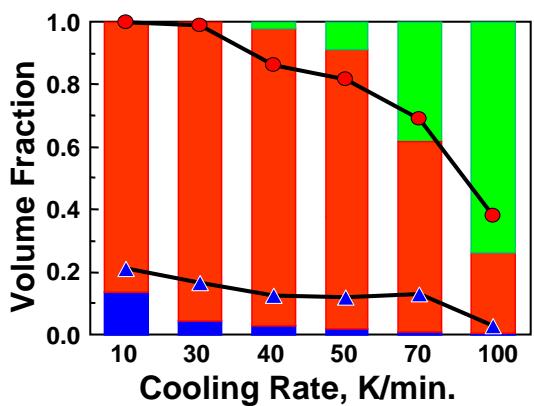
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Prediction of α /P/B Fractions after Cont. Cool.

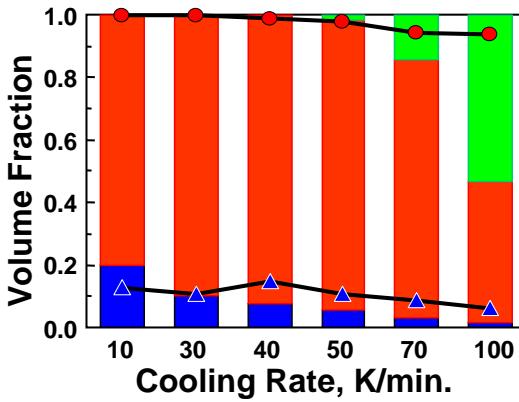
38M-H



38M-M



38M-L



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Summary

Features in the integrated phase transformation model for MCFS include:

1. Time-dependent α_1 of proeutectoid α (S. I. E.)
2. γ/α interface equilibrium transition to NPLE
3. Integration of P kinetics and B prediction from thermodynamics of remaining γ

The calculation reproduced well the slow α growth kinetics, but the problem of inconsistent D/v criteria remains.

The program showed an ability of predicting the difference in isothermal / continues cooling transformation kinetics caused by a minor Mn fluctuation.

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