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A new heat treatment design for medium Mn steel: the austenitization from pearlite

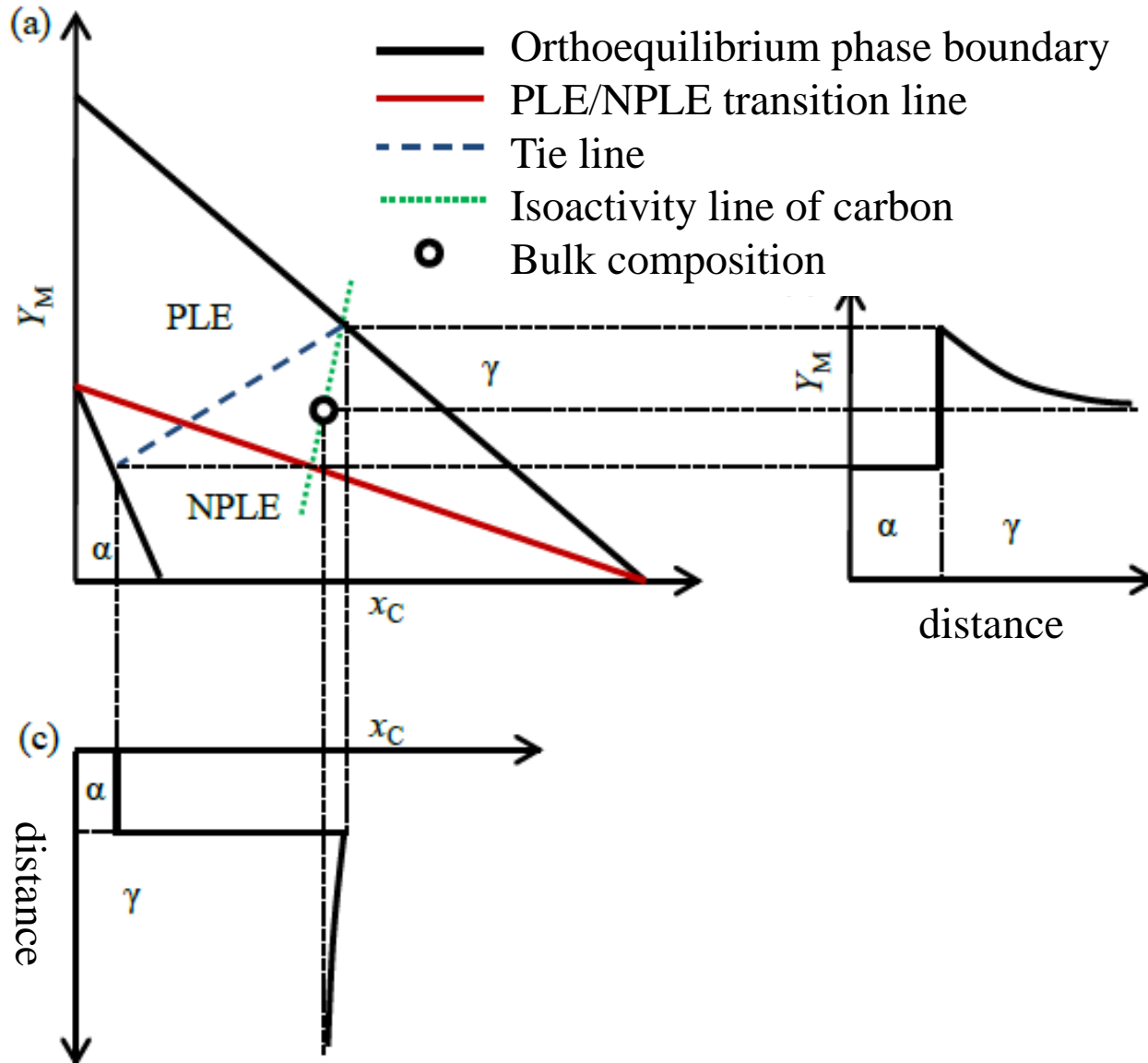
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2. *Emeritus Professor of Ibaraki University, Hitachi 316-8511, Japan*

2017. 6. 27

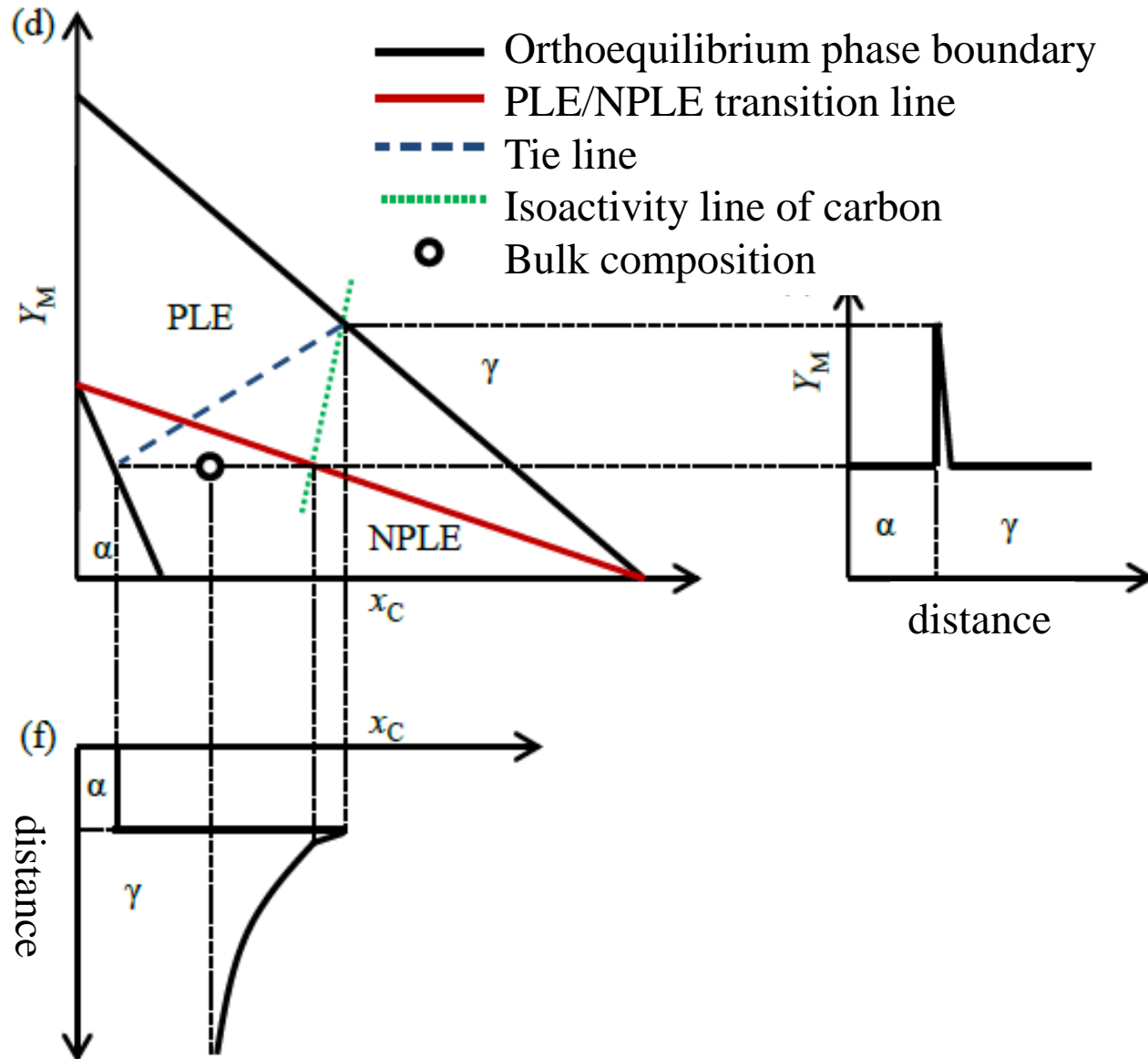


Ferrite Transformation-PLE mode





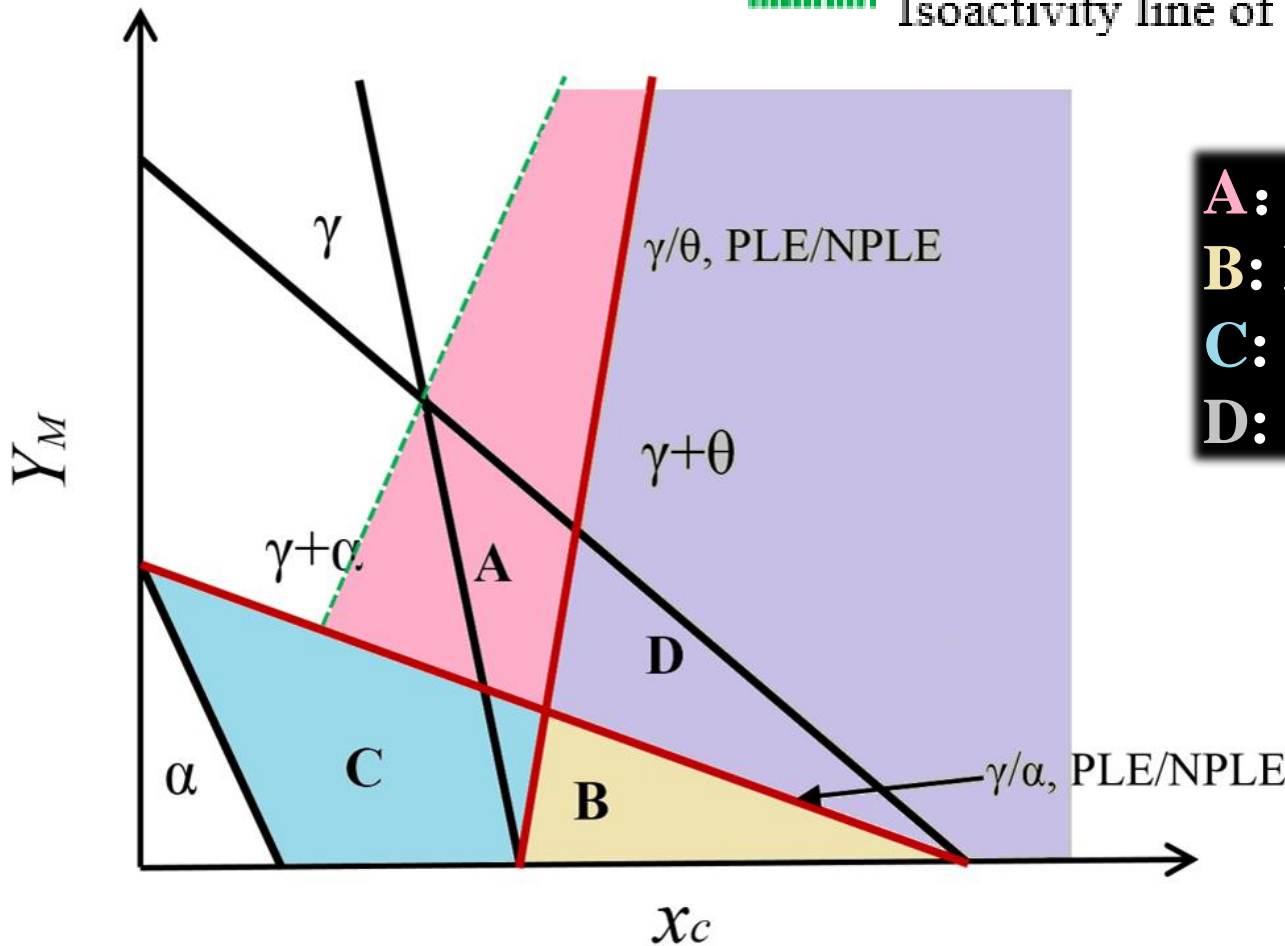
Ferrite Transformation-NPLE mode





Pearlite Transformation

- Orthoequilibrium phase boundary
- PLE/NPLE transition line
- Isoactivity line of carbon

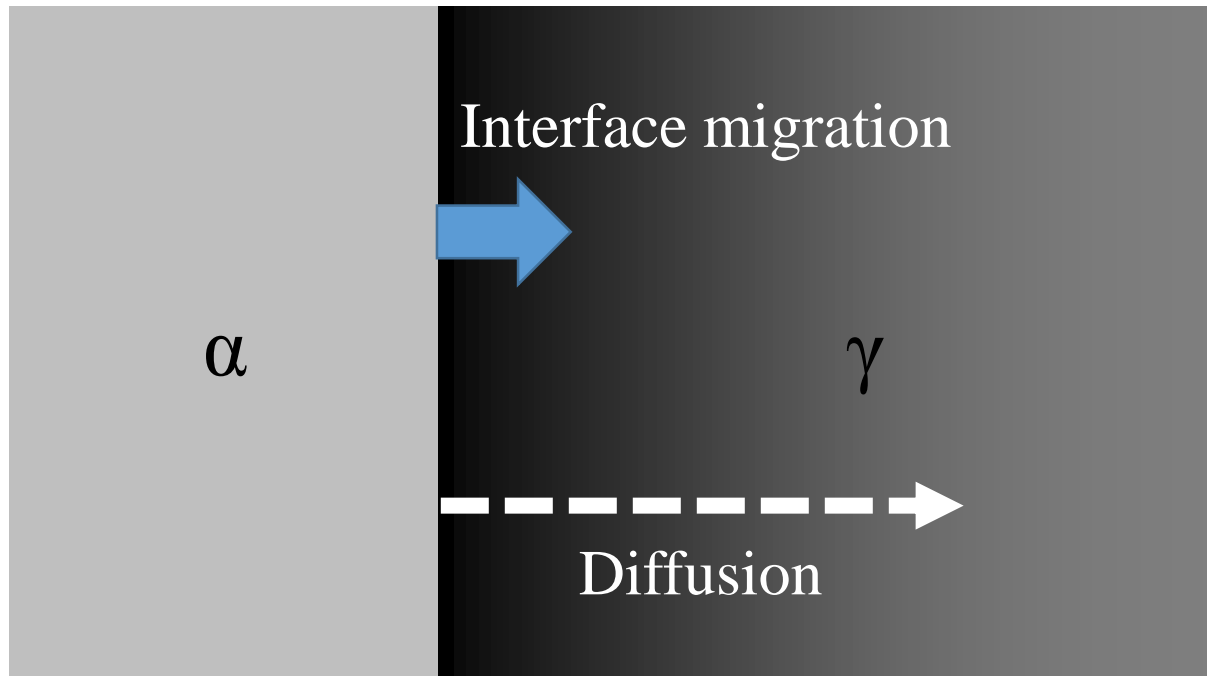


- A:** PLE- α + PLE- θ
- B:** NPLE- α + NPLE- θ
- C:** NPLE- α + PLE- θ
- D:** PLE- α + NPLE- θ

- A:** Ortho-pearlite
- B:** Para-pearlite
- C:** Proeutectoid α
- D:** Proeutectoid θ



Ferrite Transformation

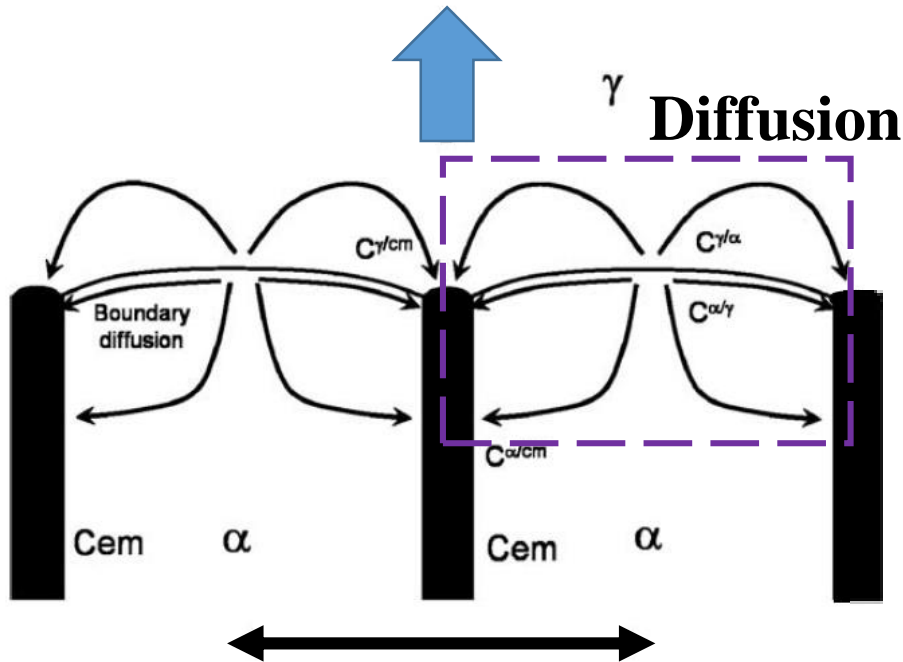


1D model



Pearlite Transformation

Interface migration



Lamellar Spacing

Diffusional path

$$\gamma \quad v = \frac{2D^\gamma S}{S^\alpha S^\theta} \left(\frac{c_e^{\gamma/\alpha} - c_e^{\gamma/\theta}}{c^{\theta/\gamma} - c^{\alpha/\gamma}} \right) \left(1 - \frac{S_C}{S} \right)$$

$$B \quad v = \frac{16D^B \delta}{S^\alpha S^\theta} \left(\frac{c_e^{\gamma/\alpha} - c_e^{\gamma/\theta}}{c^{\theta/\gamma} - c^{\alpha/\gamma}} \right) \left(1 - \frac{S_C}{S} \right)$$

α ?

1. Which path works for carbon? Which for alloying elements?
2. Is there transition between different modes?

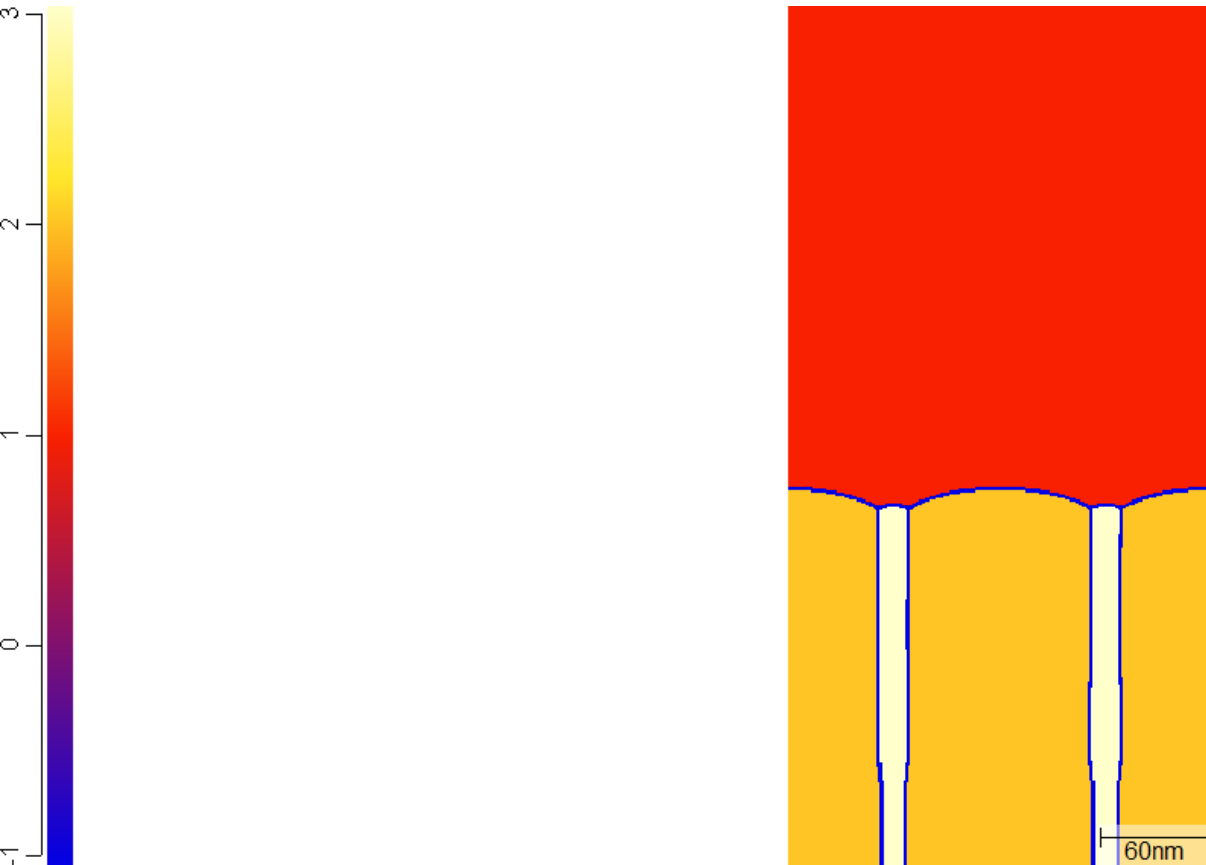


PFM_Phase Field

Fe-0.81C

T=675°C

t= 0s005s

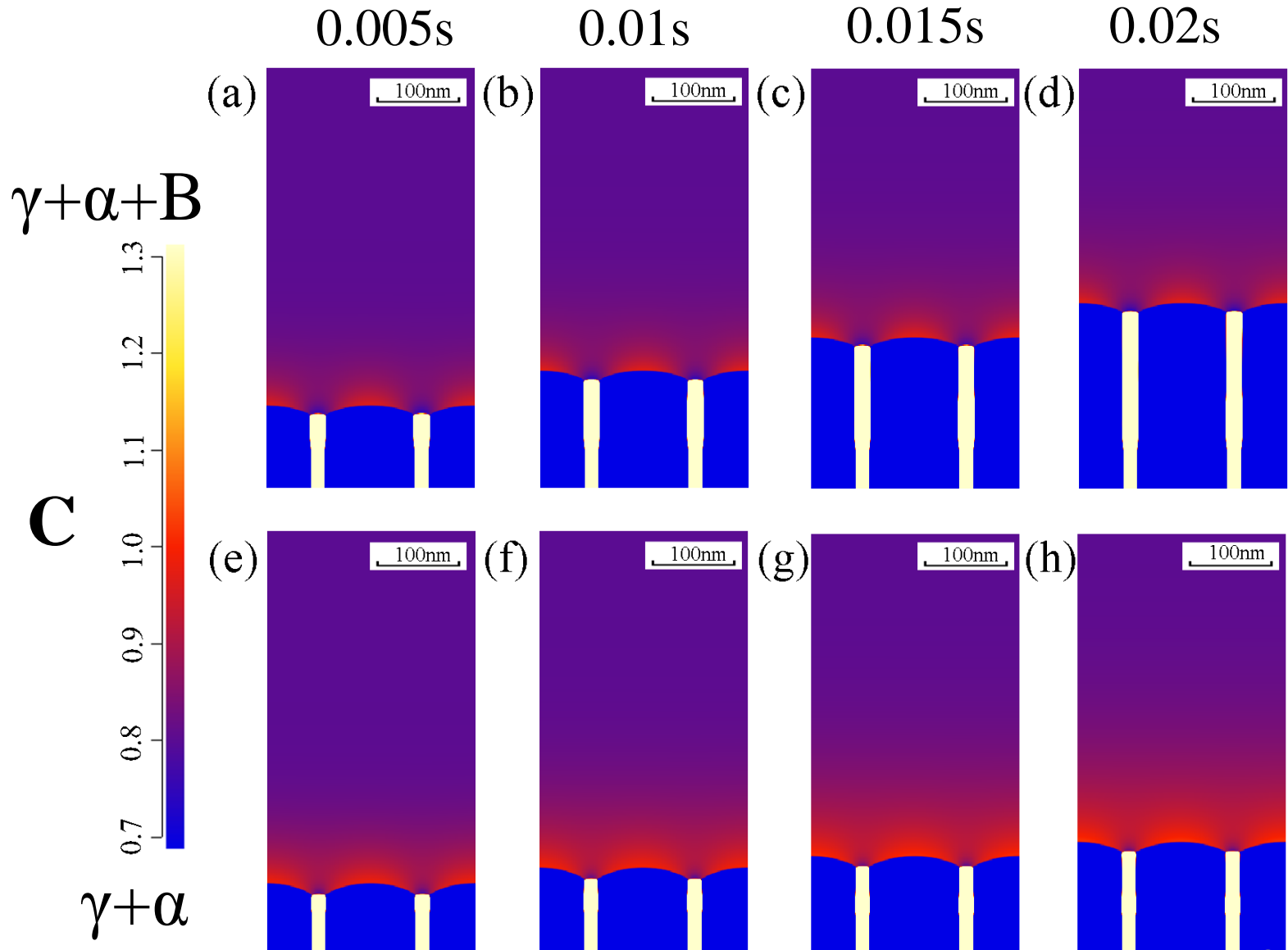


per_Fe-0.81C_675_GB_96851_phas.mcr, X: 1 to 238, Y: 1 of 1, Z: 1 to 480, Time: 0s



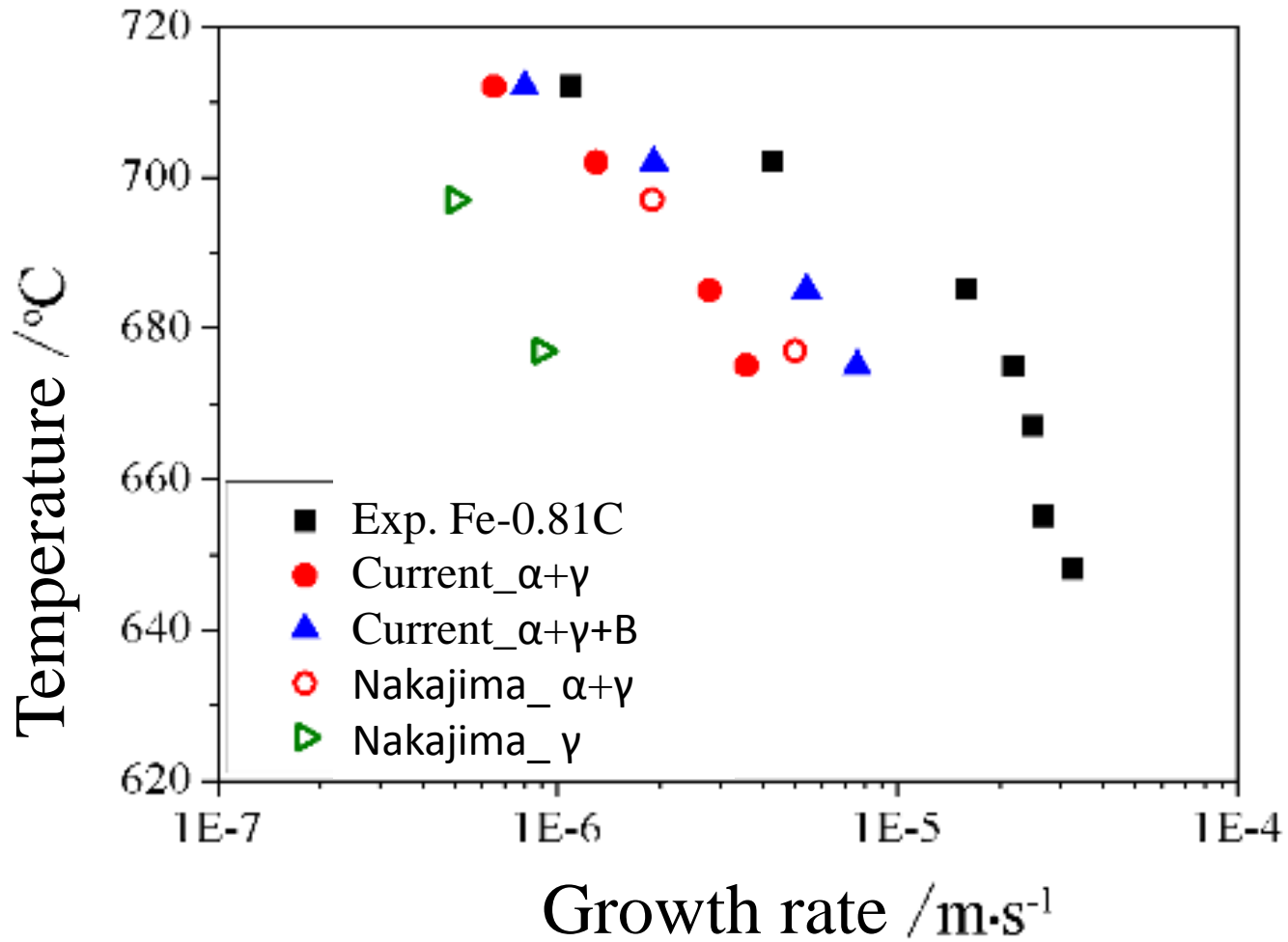


PFM_Conc. Field





PFM_Diffusion Path



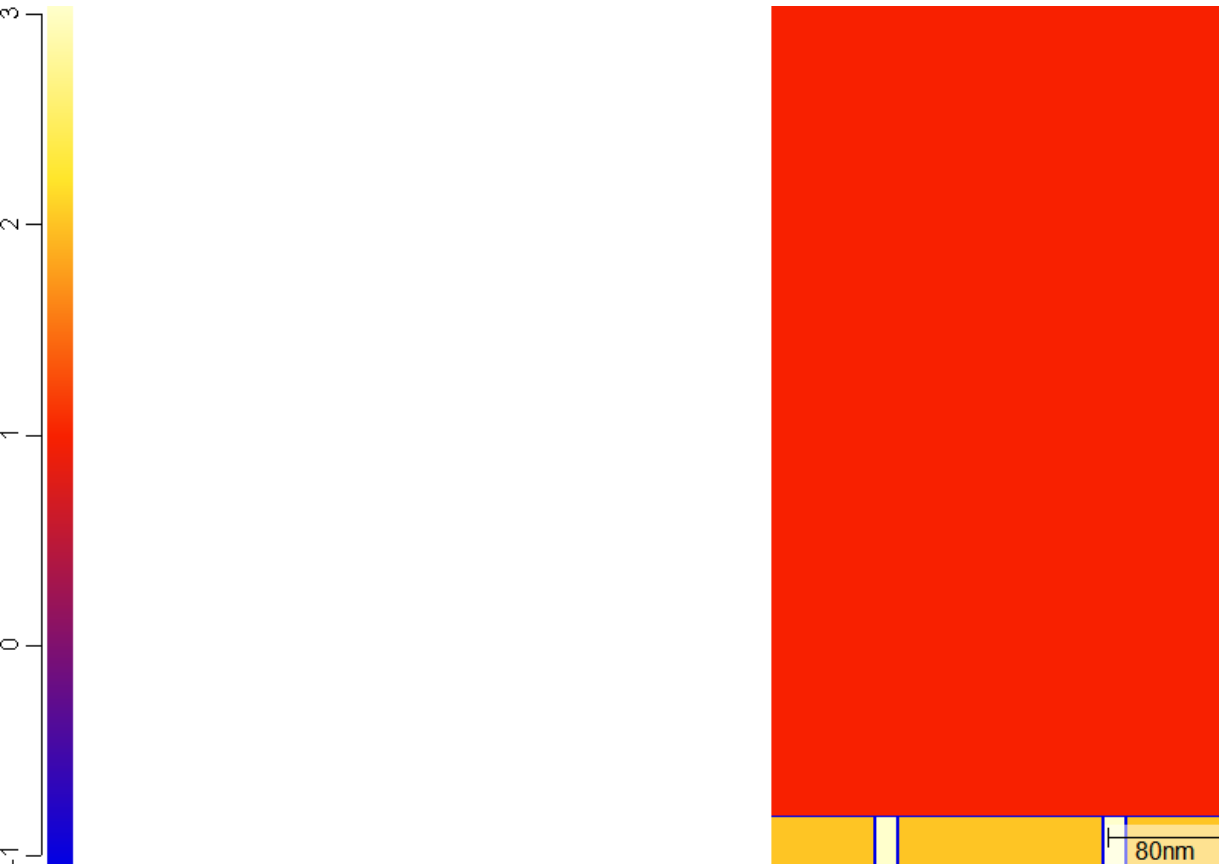


PFM_Phase Field

Fe-0.69C-1.80Mn

T=650°C

t=0s



per_923_Fe-0.69C-1.8Mn_Dgb-Mn155000-C96851_phas.mcr, X: 1 to 320, Y: 1 of 1, Z: 1 to 600, Time: 0s



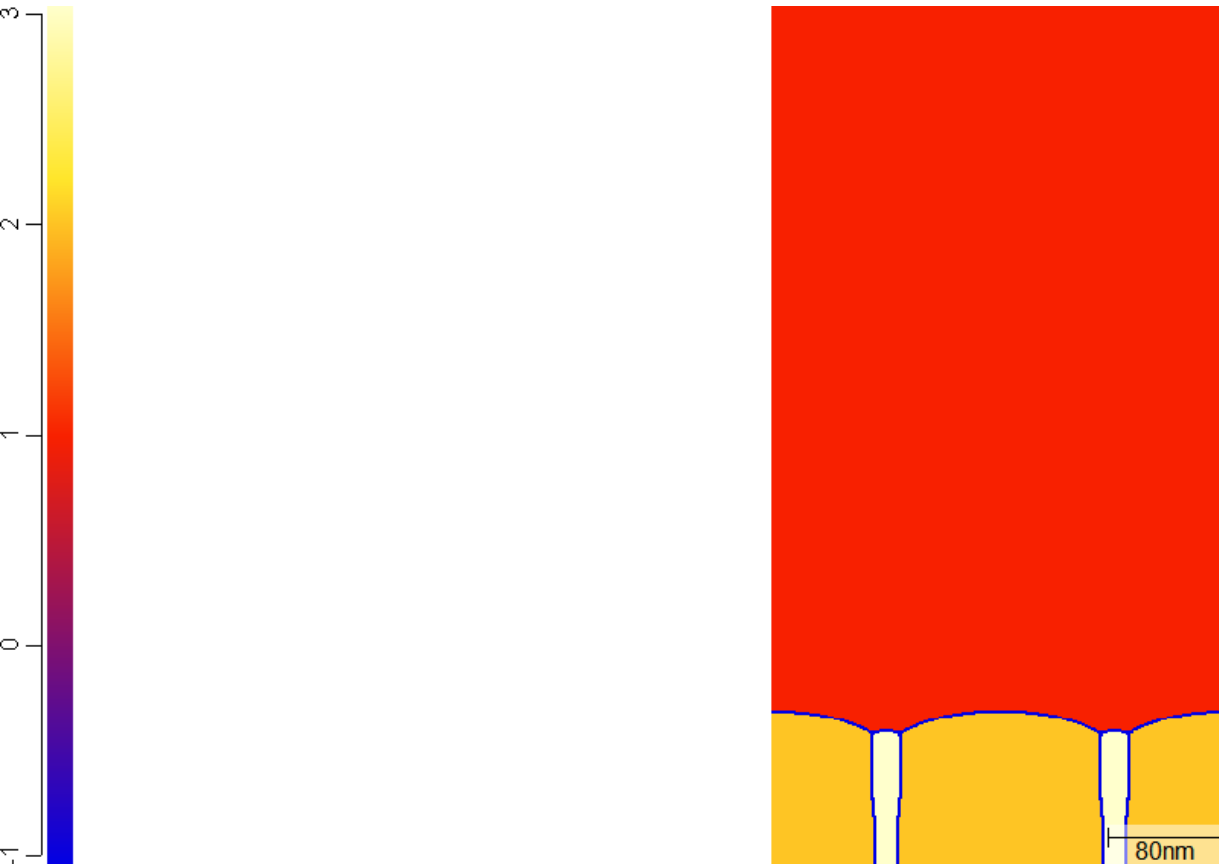


PFM_Phase Field

Fe-0.69C-1.80Mn

T=650°C

t=0.02s



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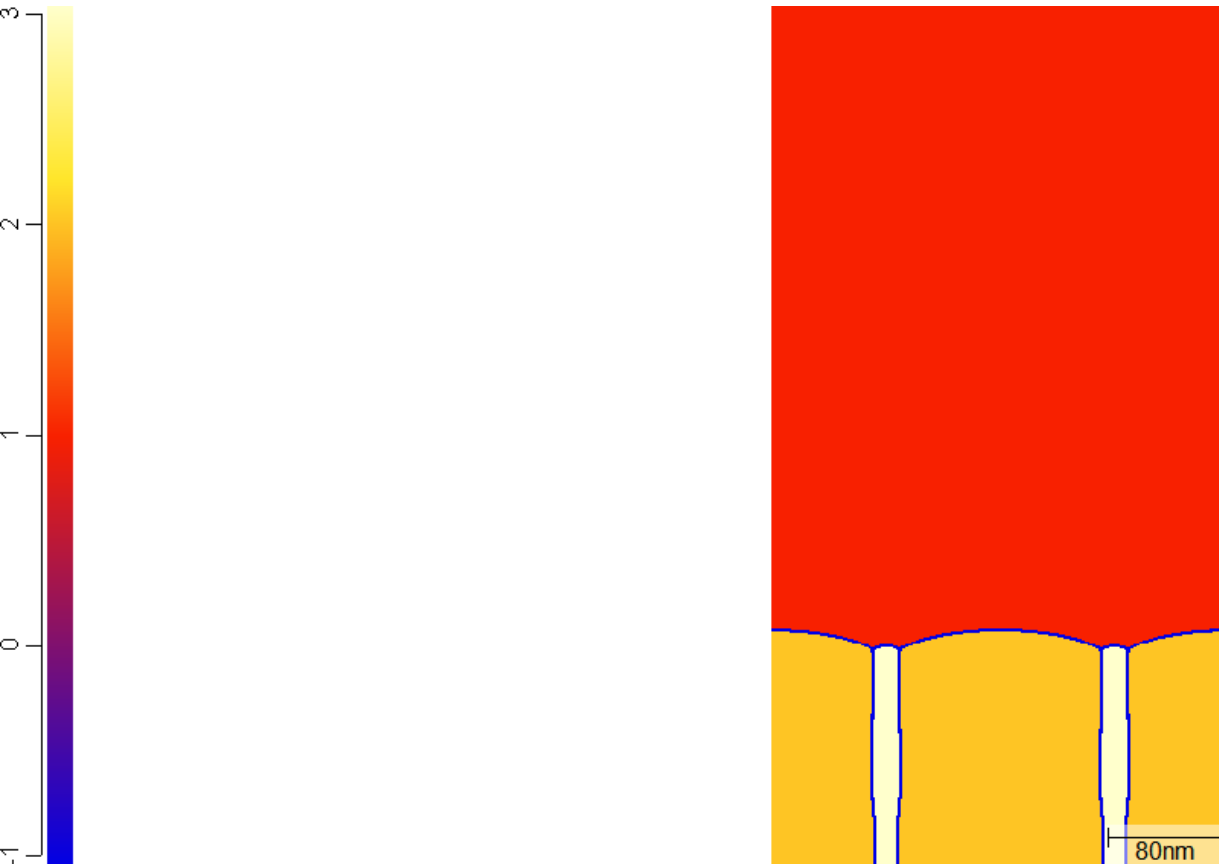


PFM_Phase Field

Fe-0.69C-1.80Mn

T=650°C

t=0.04s



per_923_Fe-0.69C-1.8Mn_Dgb-Mn155000-C96851_phas.mcr, X: 1 to 320, Y: 1 of 1, Z: 1 to 600, Time: 0s



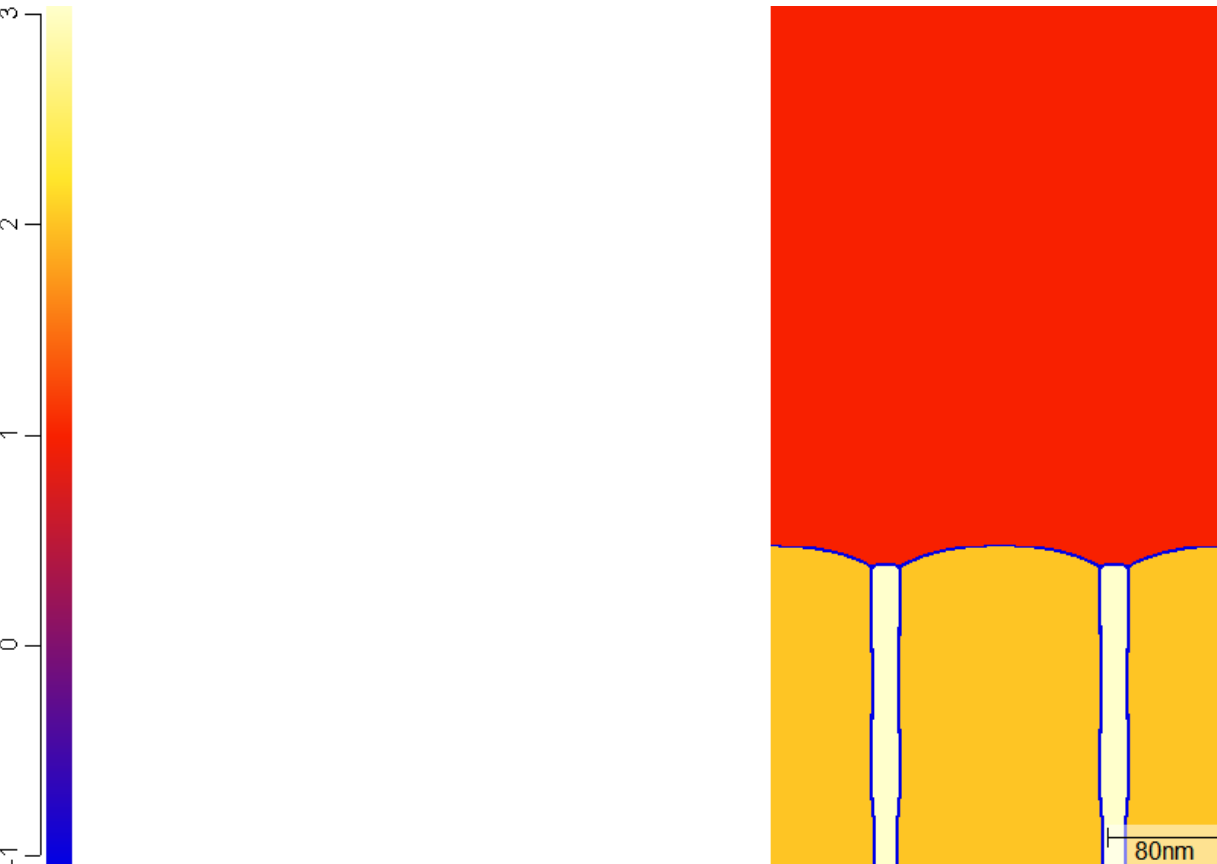


PFM_Phase Field

Fe-0.69C-1.80Mn

T=650°C

t=0.06s



per_923_Fe-0.69C-1.8Mn_Dgb-Mn155000-C96851_phas.mcr, X: 1 to 320, Y: 1 of 1, Z: 1 to 600, Time: 0s





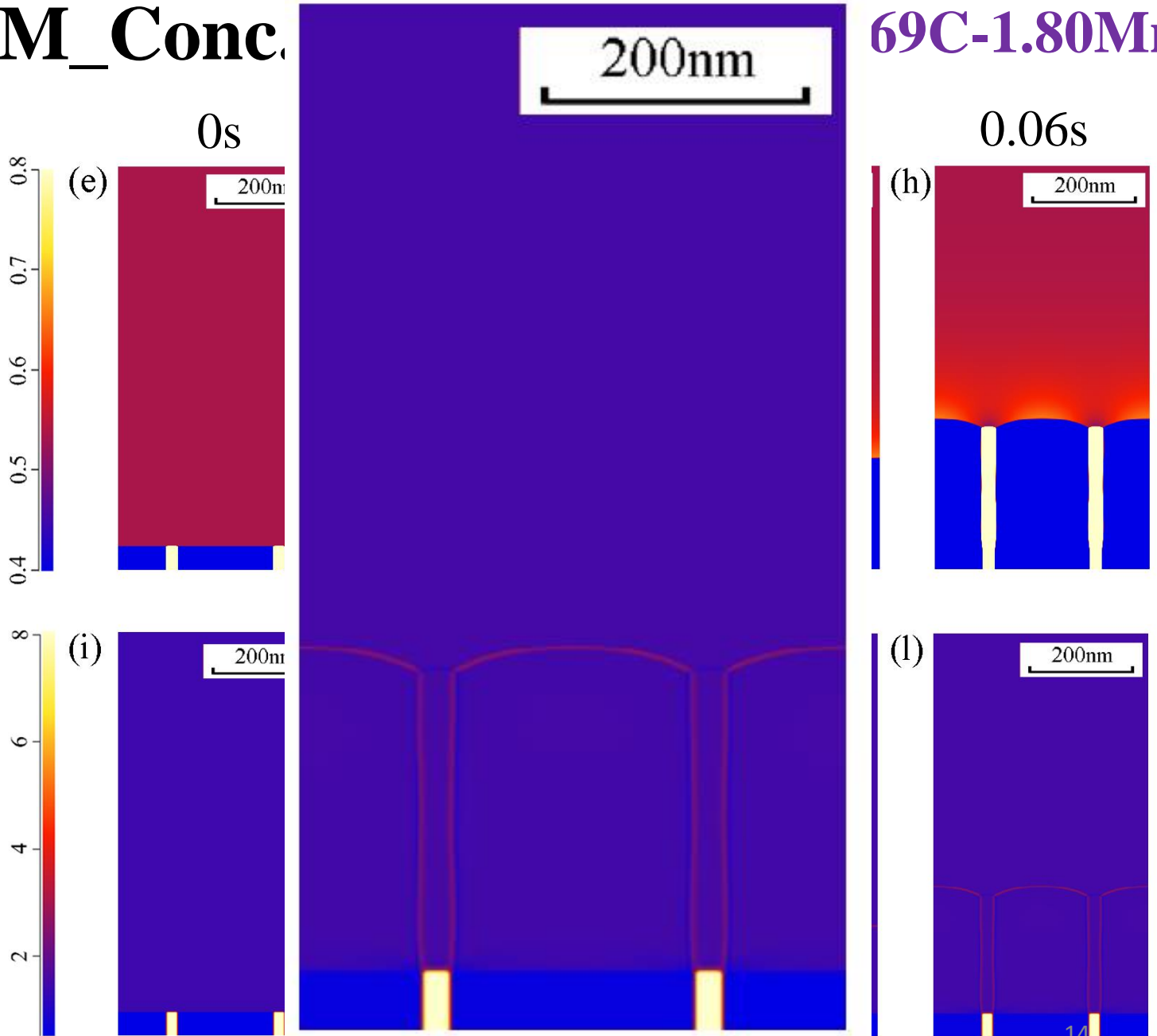
PFM_Conc.

69C-1.80Mn

T=650°C

C

Mn

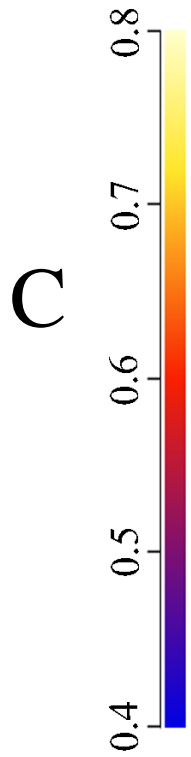




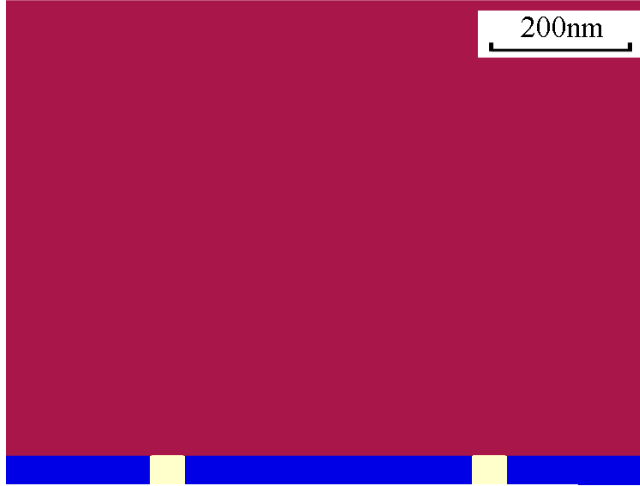
PFM_Conc. Field

Fe-0.69C-1.80Mn

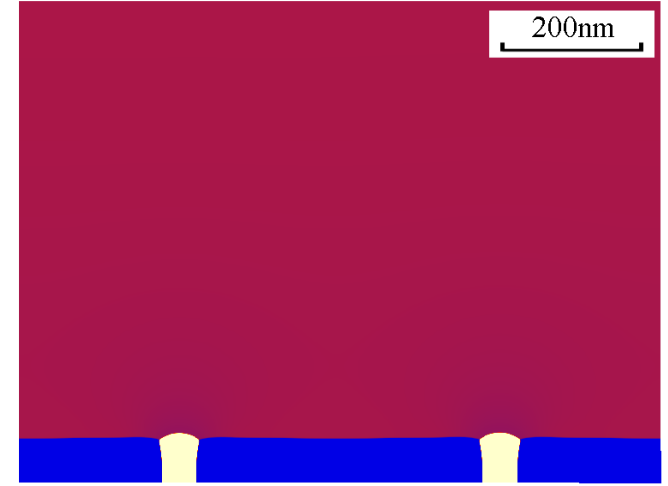
$T=670^{\circ}\text{C}$



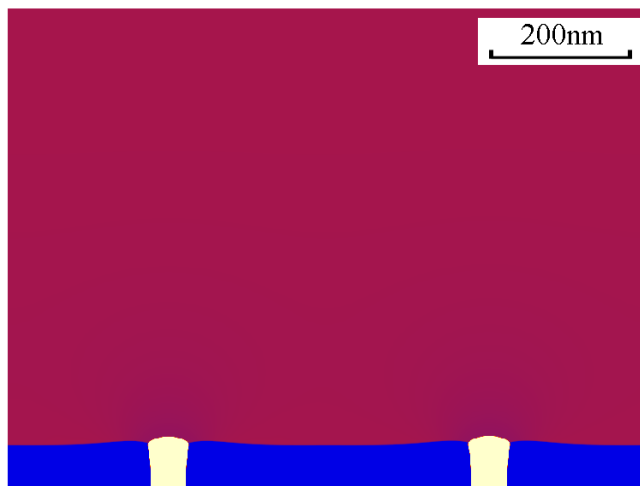
0s



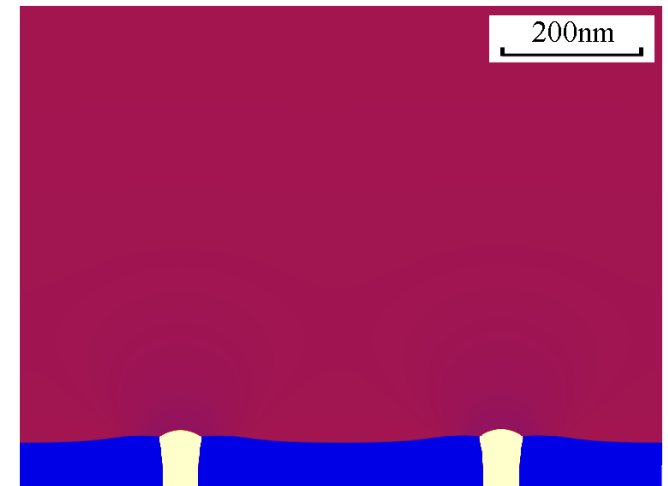
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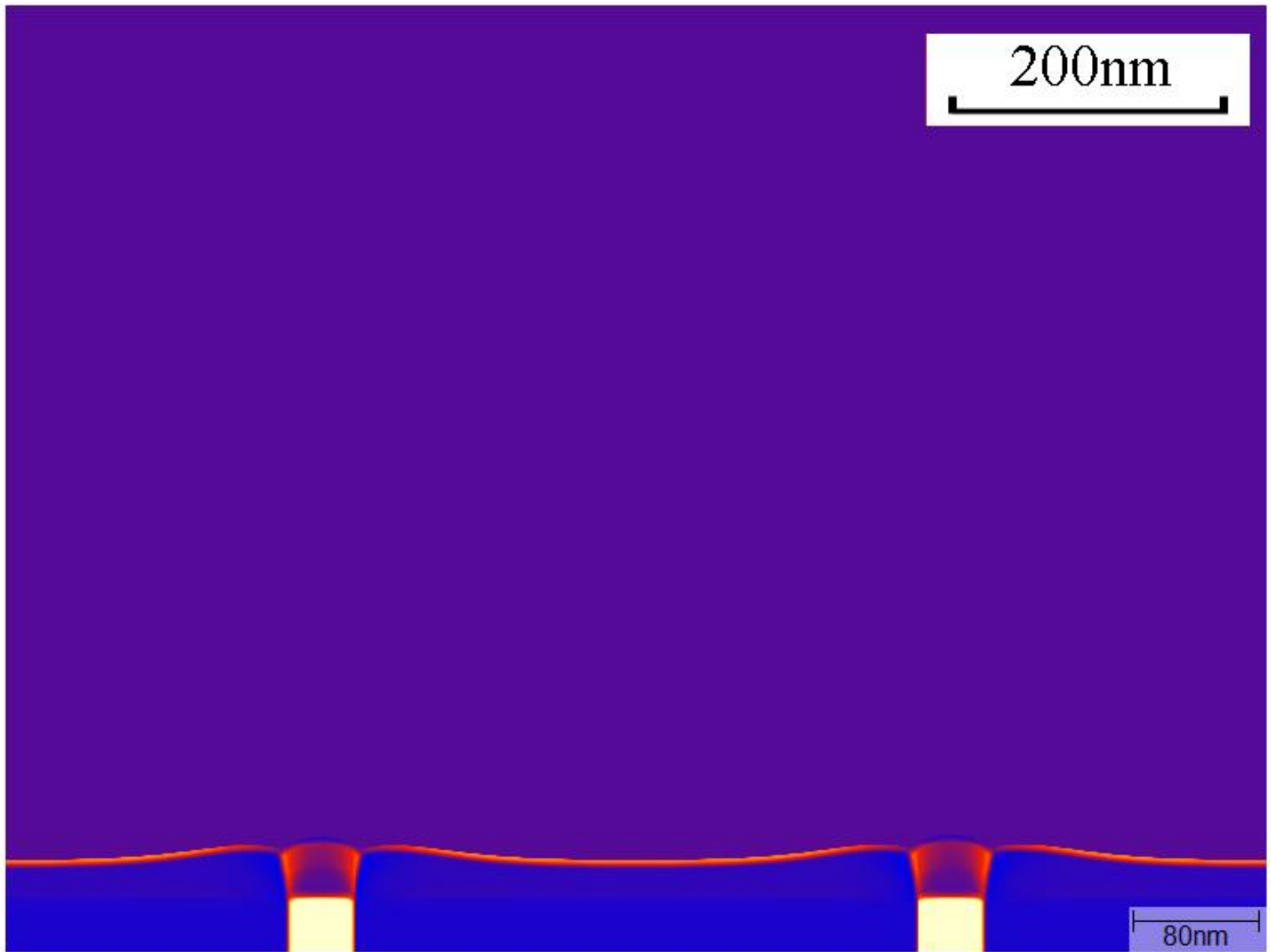


0.04s



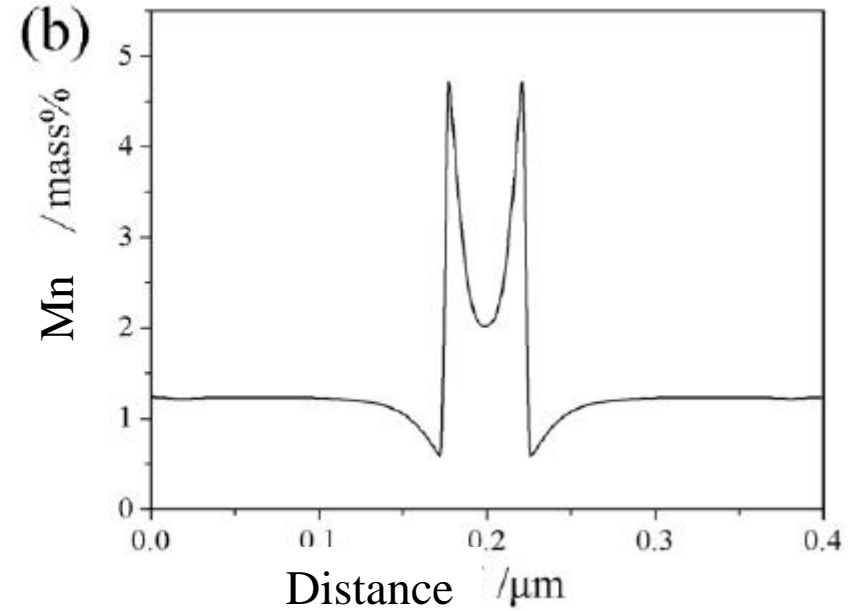
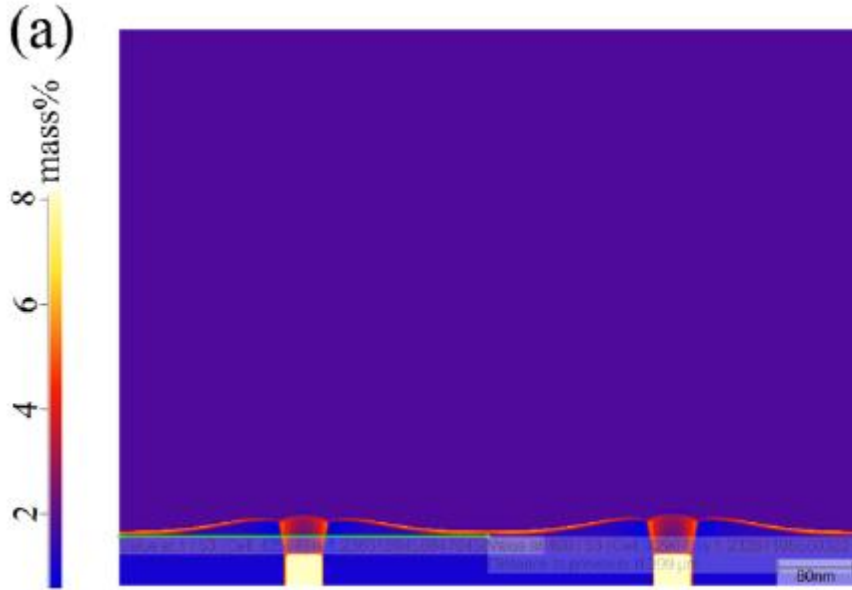
0.06s







PFM_Conc.



Temperature / °C	Growth rate / $\mu\text{m}\cdot\text{s}^{-1}$		Partition coefficient $k^{\theta/\alpha}$	
	Exp.	Sim.	Exp.	Sim.
650	0.15	2.95	1.3	1.1
660	0.042	0.11	2	2.36
670	0.01	0.024	3.4	3

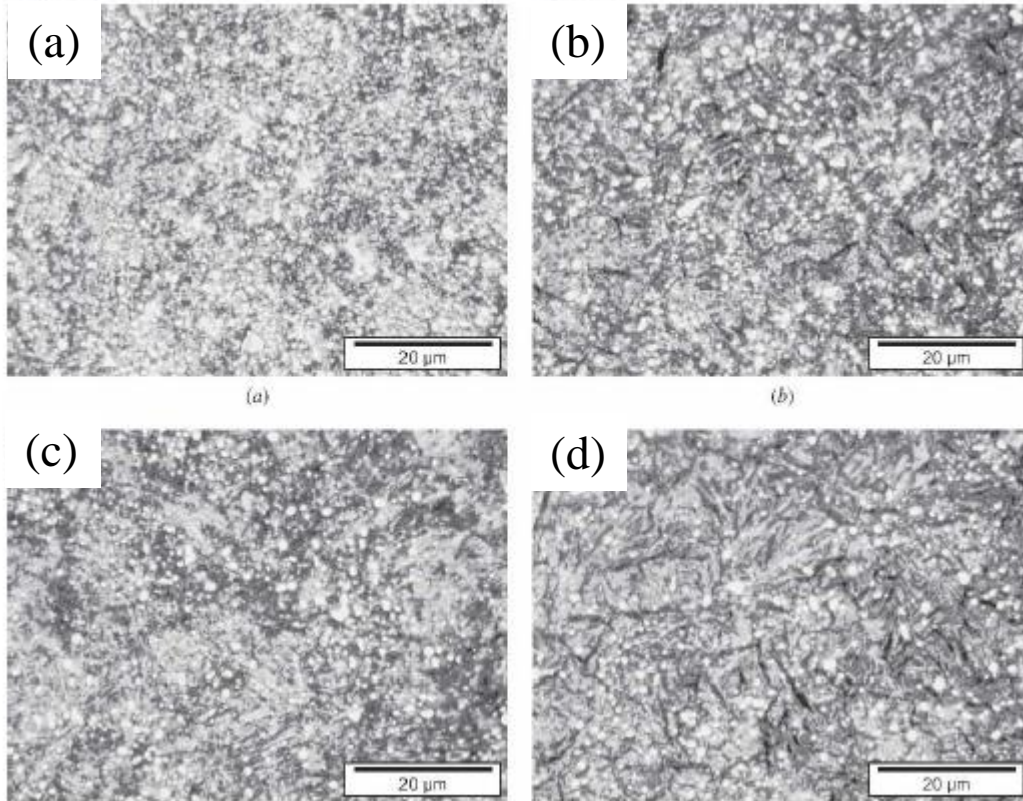


PFM_Summary

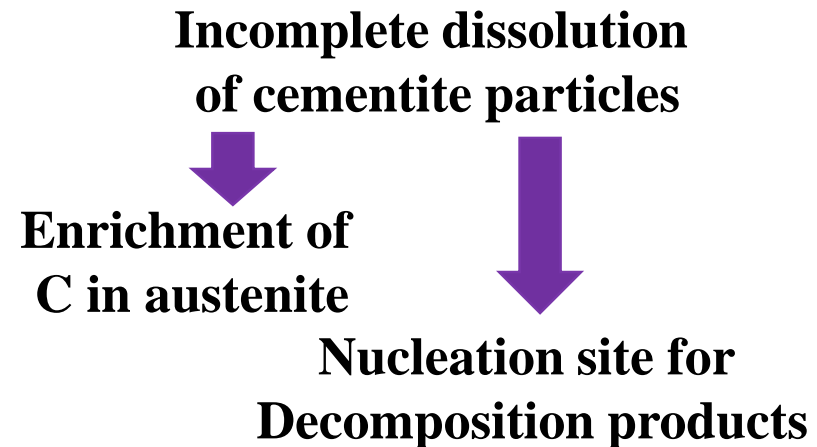
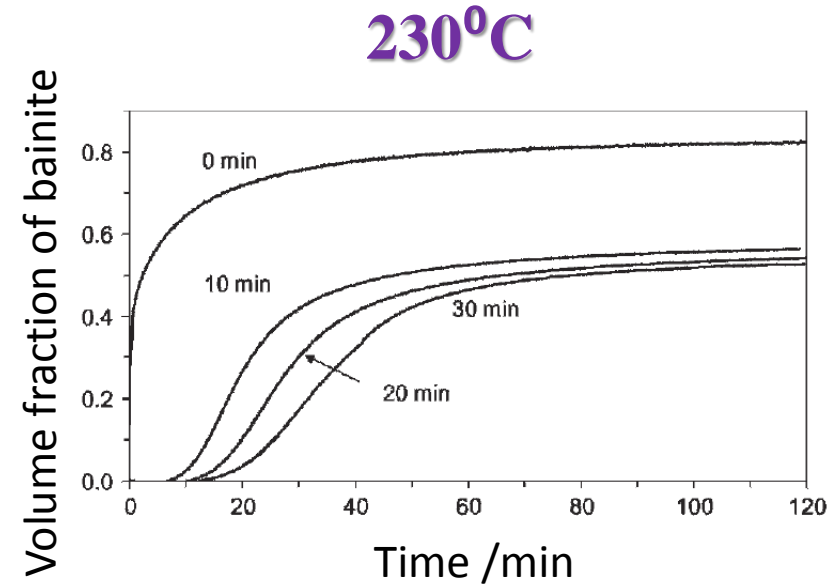
1. In Fe-C binary system, the diffusion path of carbon includes austenite, ferrite and γ/p interface. Bulk diffusion dominates the transformation at higher temperature, while boundary diffusion becomes important at lower temperature.
2. The pearlite transformation of Fe-0.69C-1.80Mn ternary system is controlled by C diffusion at 650°C , while by Mn (boundary) diffusion at 670°C .



Austenitization from pearlite

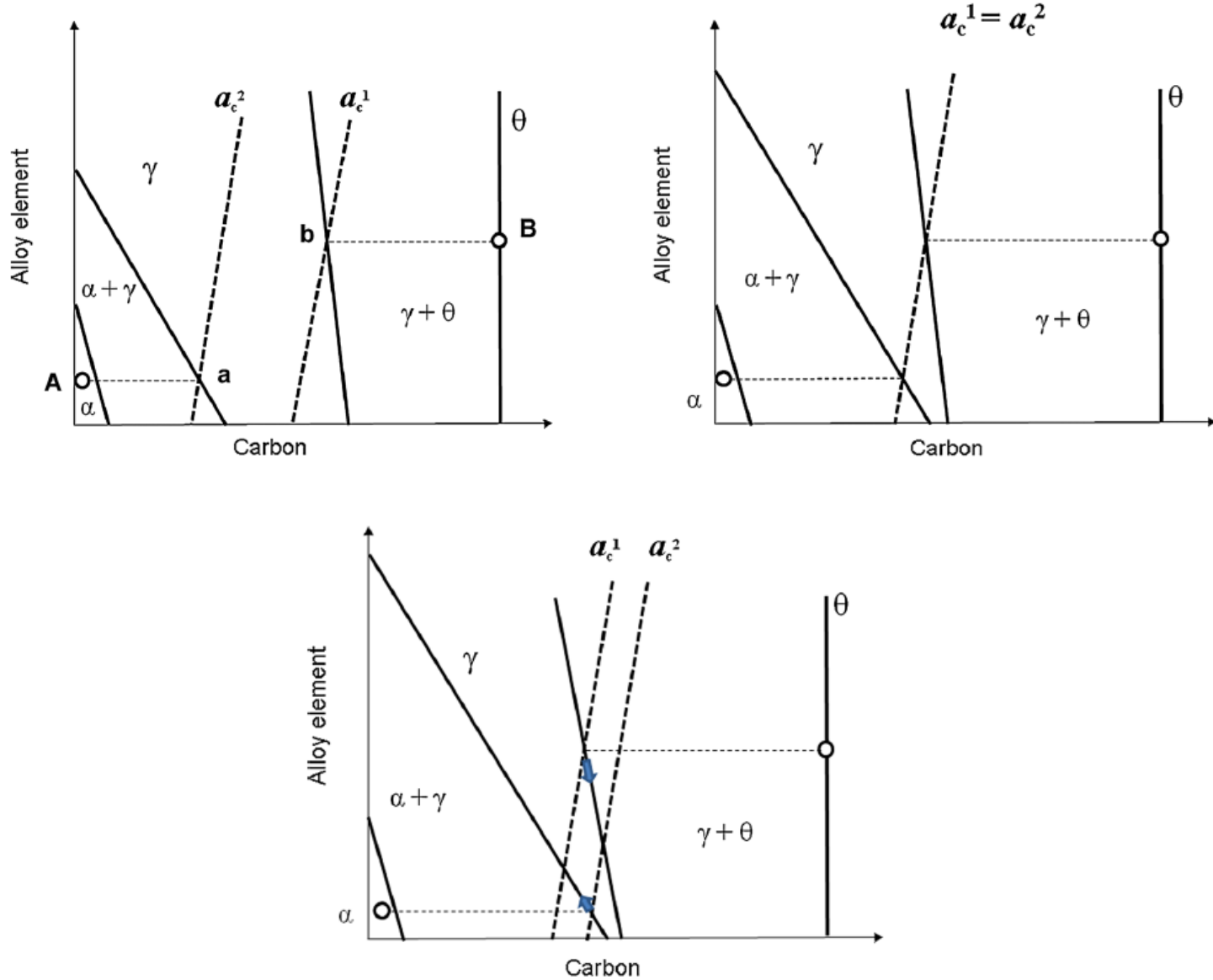


Fe-1C-1.36Cr alloy austenitized at 860 °C for
(a)0min (b)10min (c)20min (d)30min





Austenitization_PNTT-I





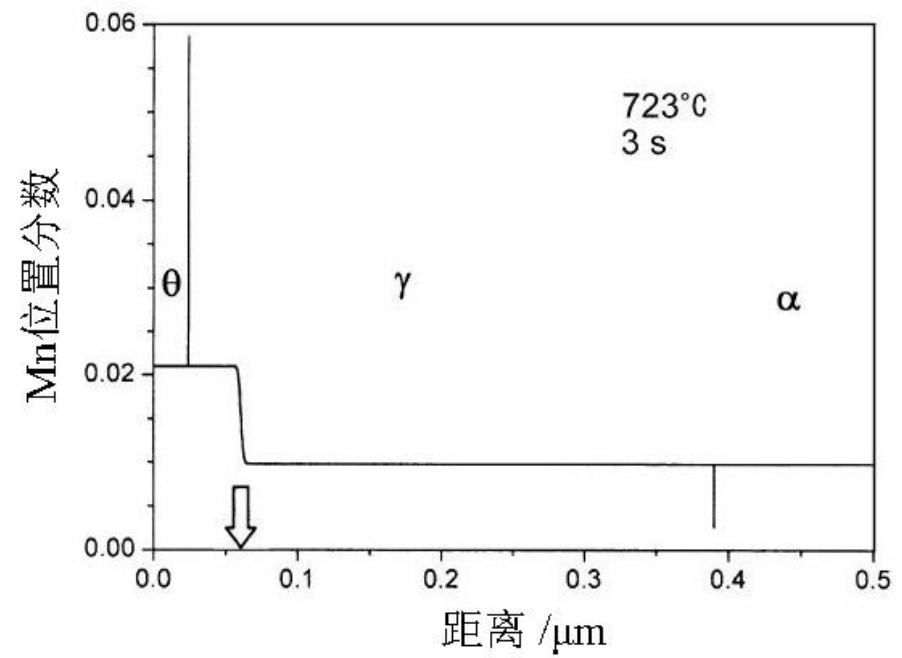
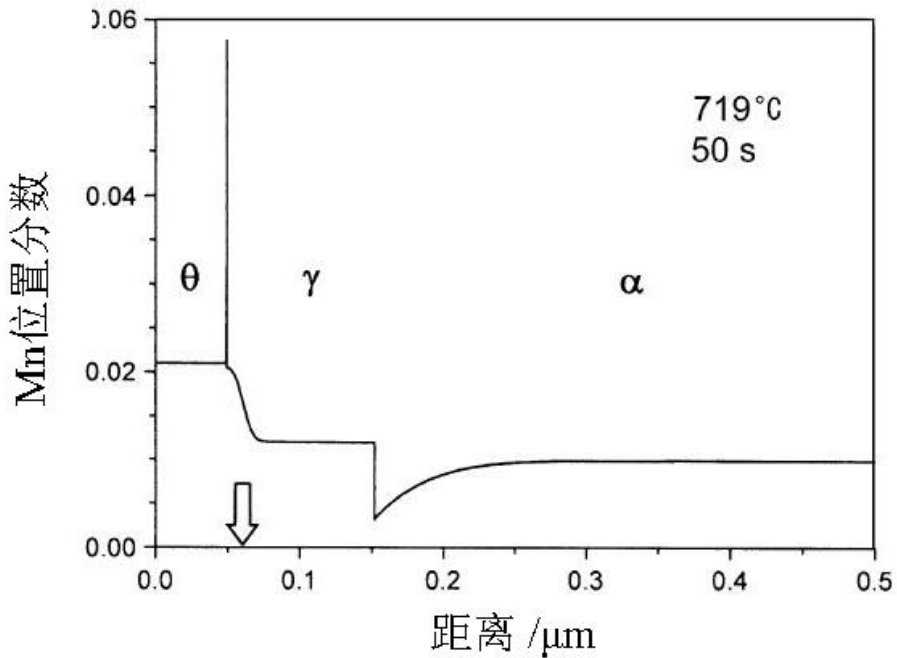
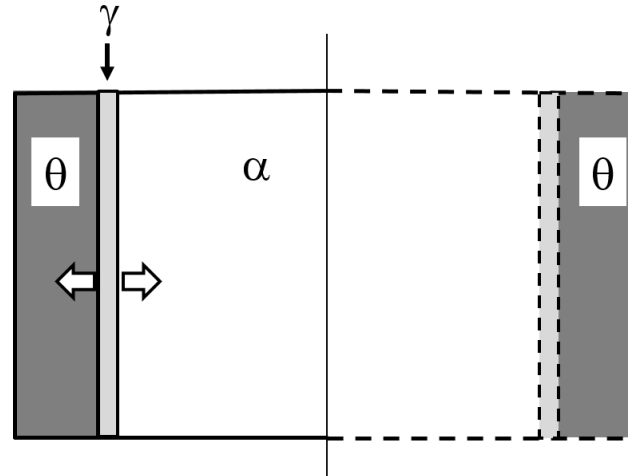
Austenitization_Dictra

Fe-0.80C-1.08Mn

$T_{ptr}=692^{\circ}\text{C}$

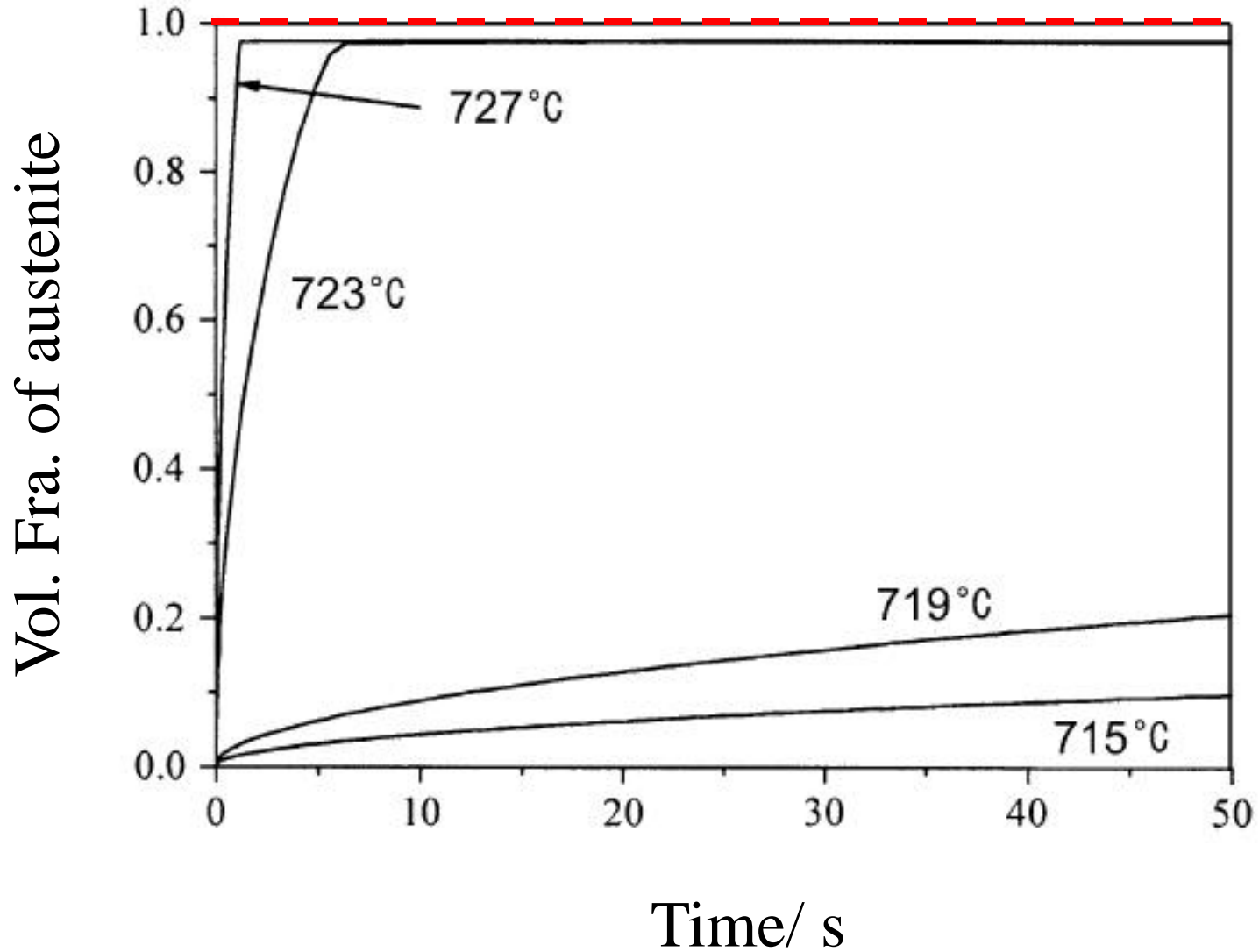
Lamellar pearlite

$PNTT-I=721^{\circ}\text{C}$





Austenitization_Dictra





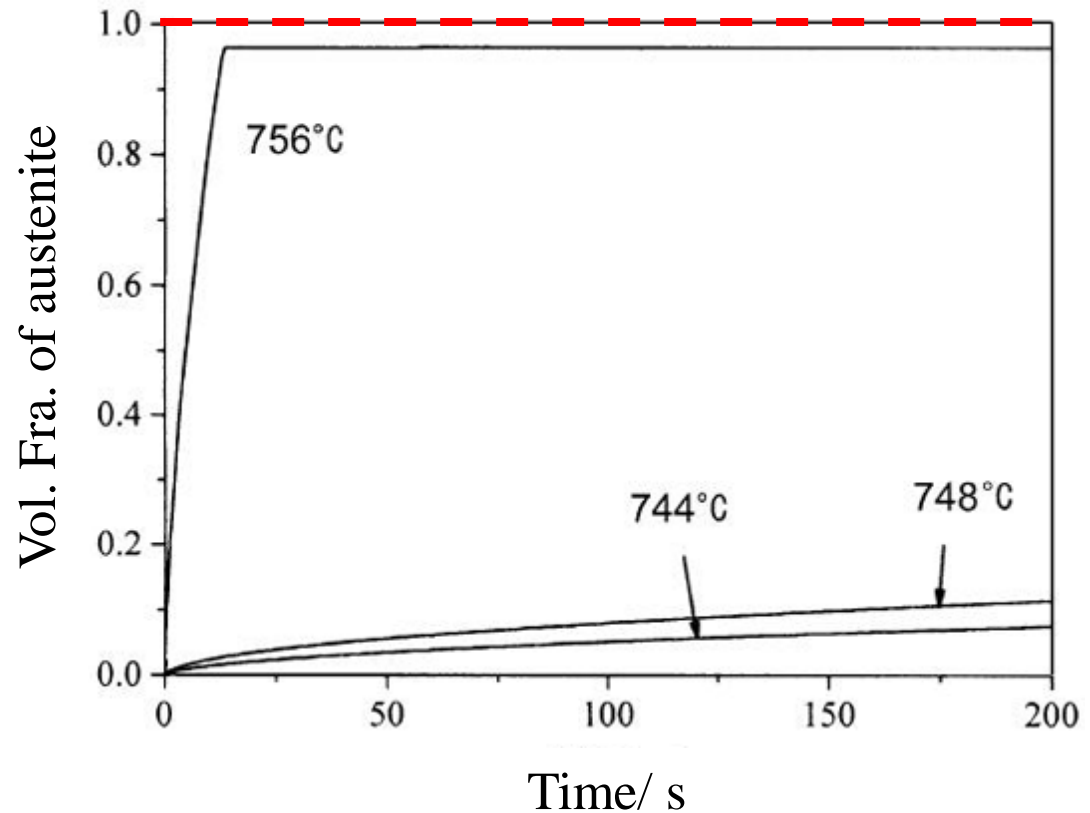
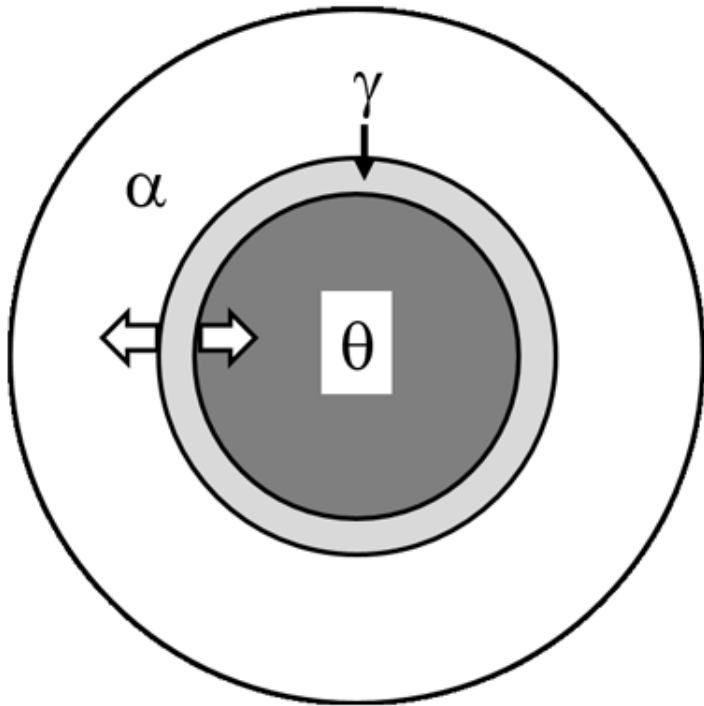
Austenitization_Dictra

Fe-0.69C-1.80Mn

Spherodized pearlite

$T_{ptr}=660^{\circ}\text{C}$

$PNTT-I=750^{\circ}\text{C}$

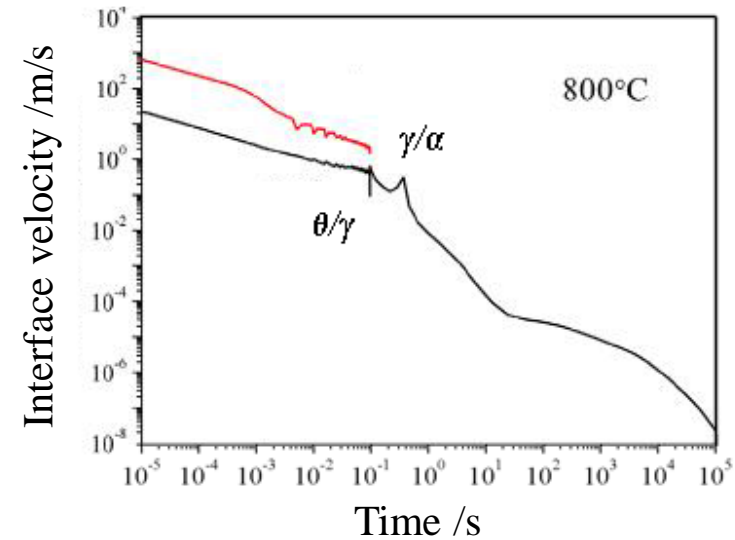
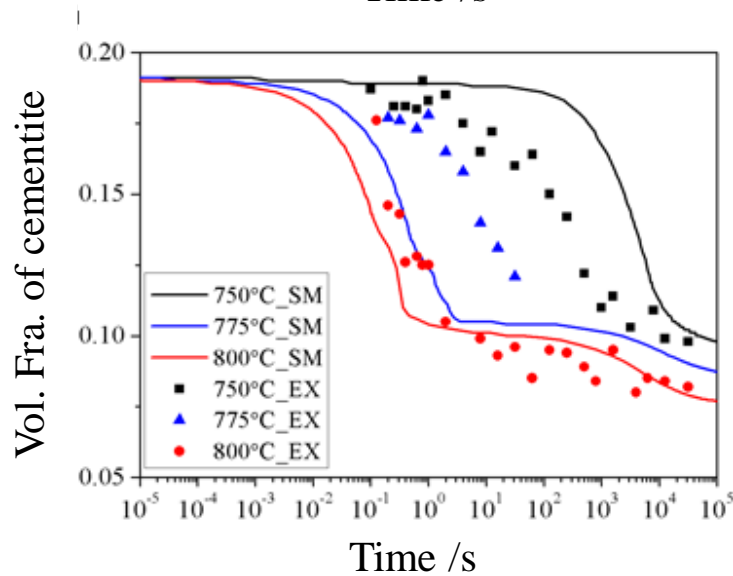
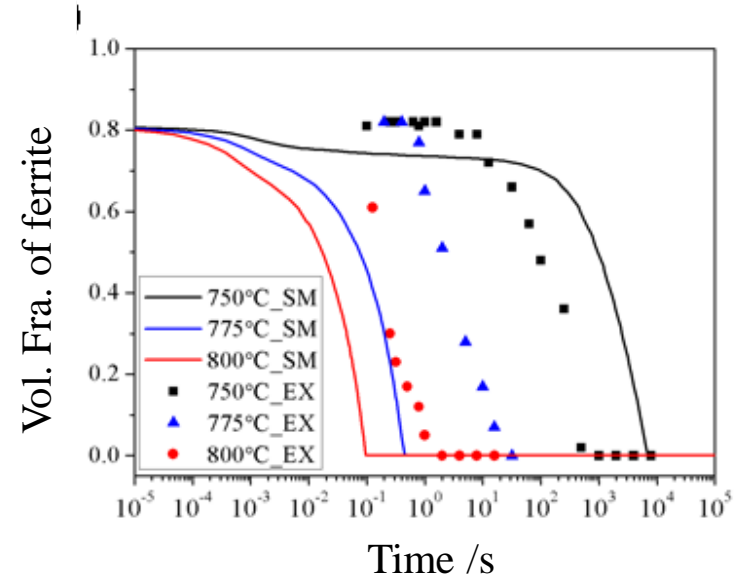
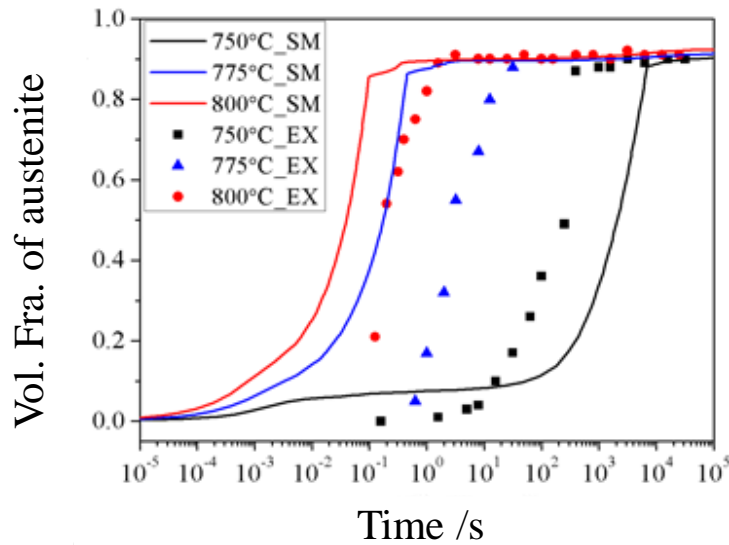




Austenitization_Exp.

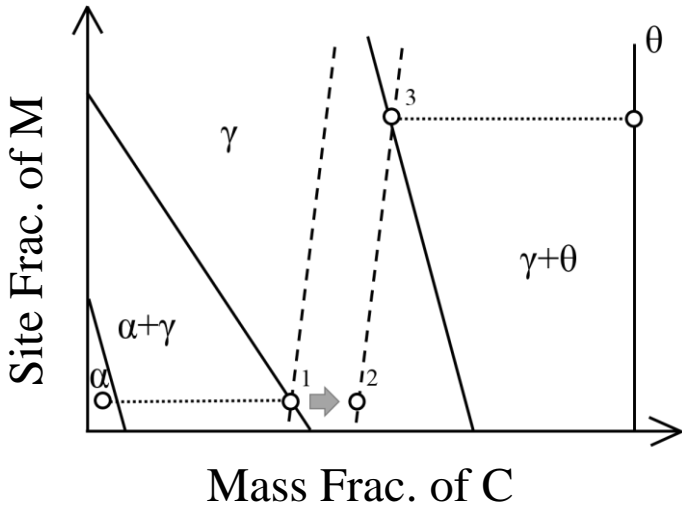
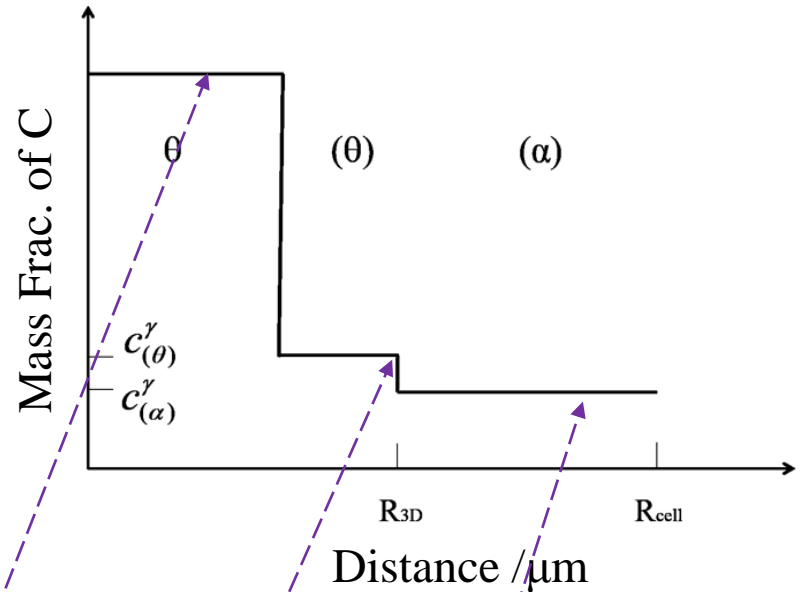
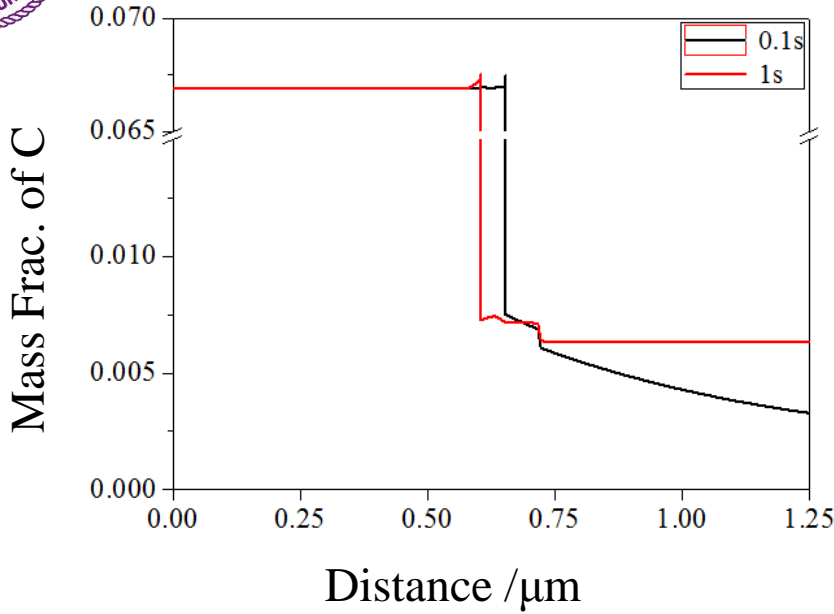
Fe-1.27 C-0.36Mn-0.19Cr

PNTT-I=756°C





Austenitization_PNTT-II

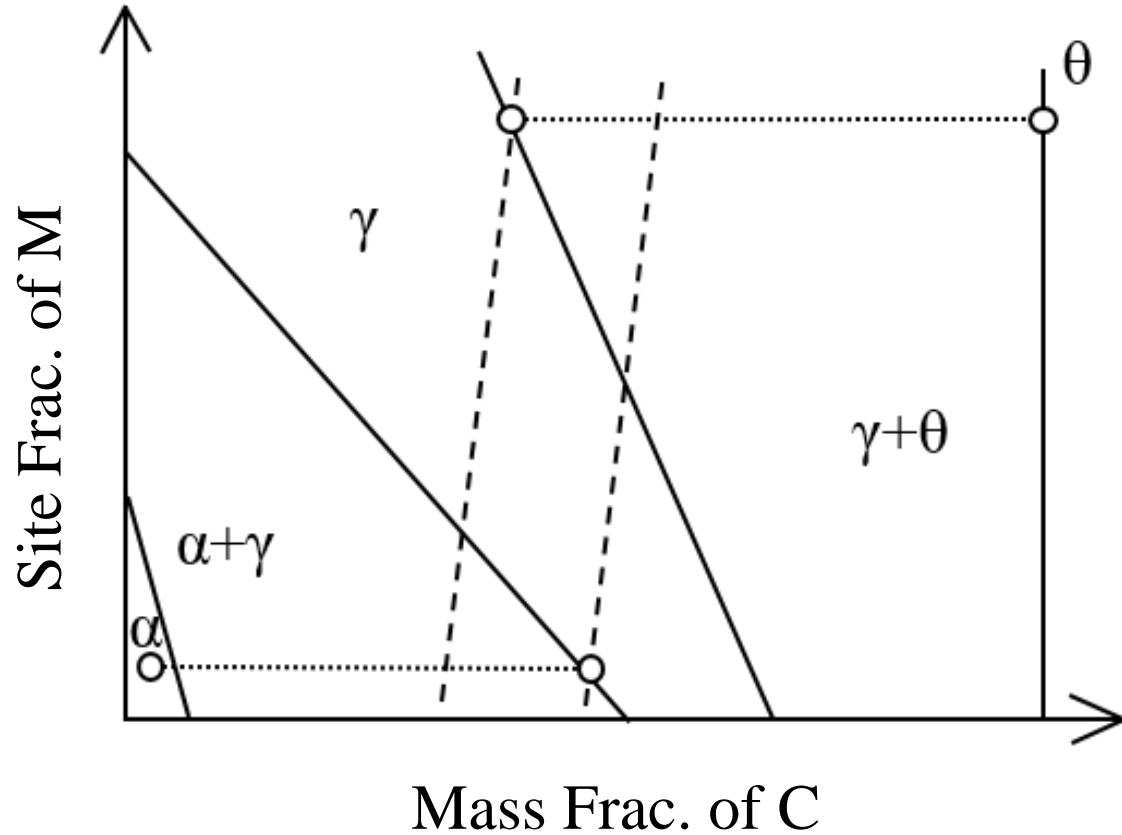
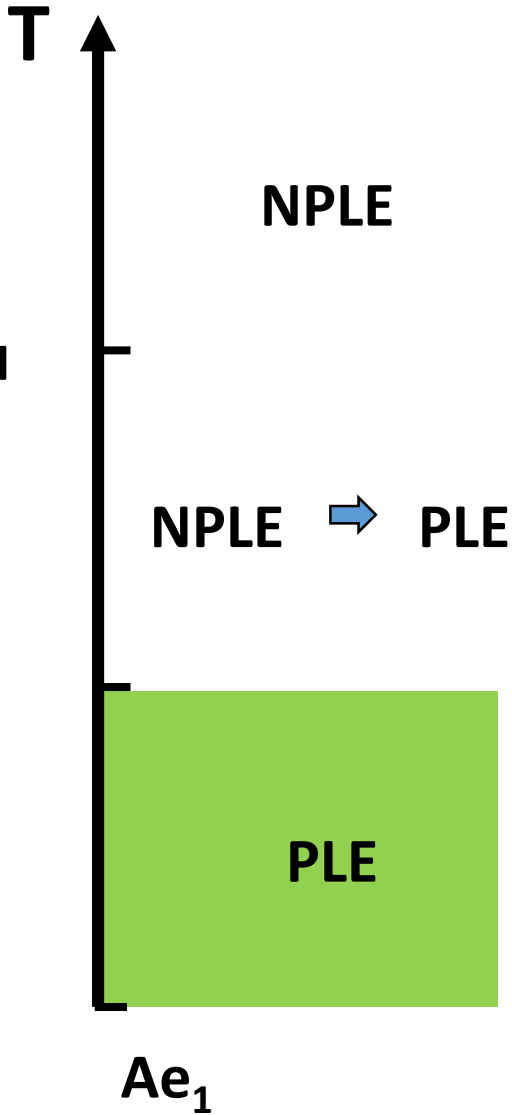


$$C_C^\theta \cdot f^\theta + C_C^{(\theta)} \cdot f^{(\theta)} + C_C^{(\alpha)} \cdot f^{(\alpha)} = C_C^0$$

$$= 0$$

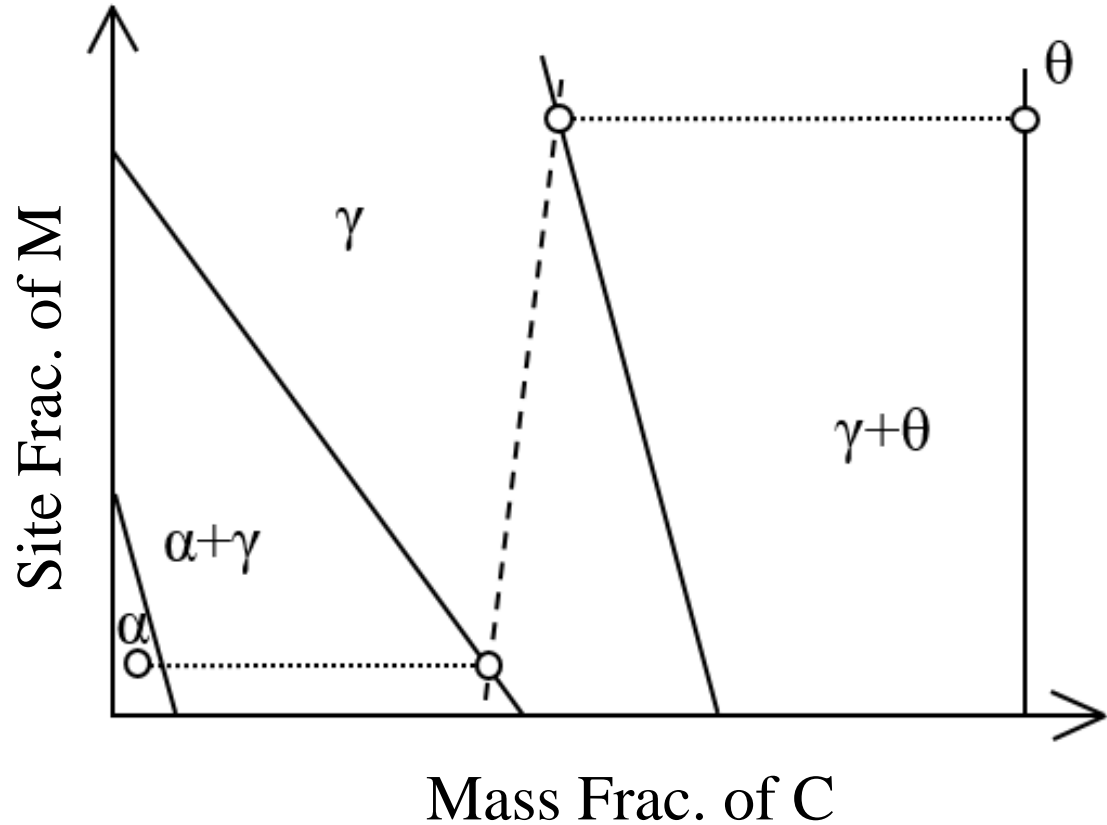
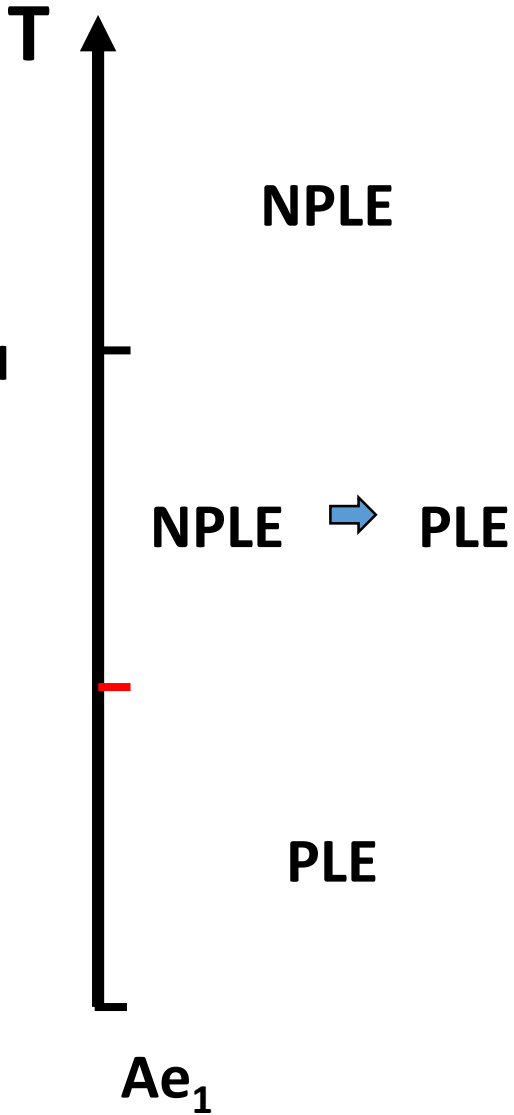


Austenitization_PNTTs



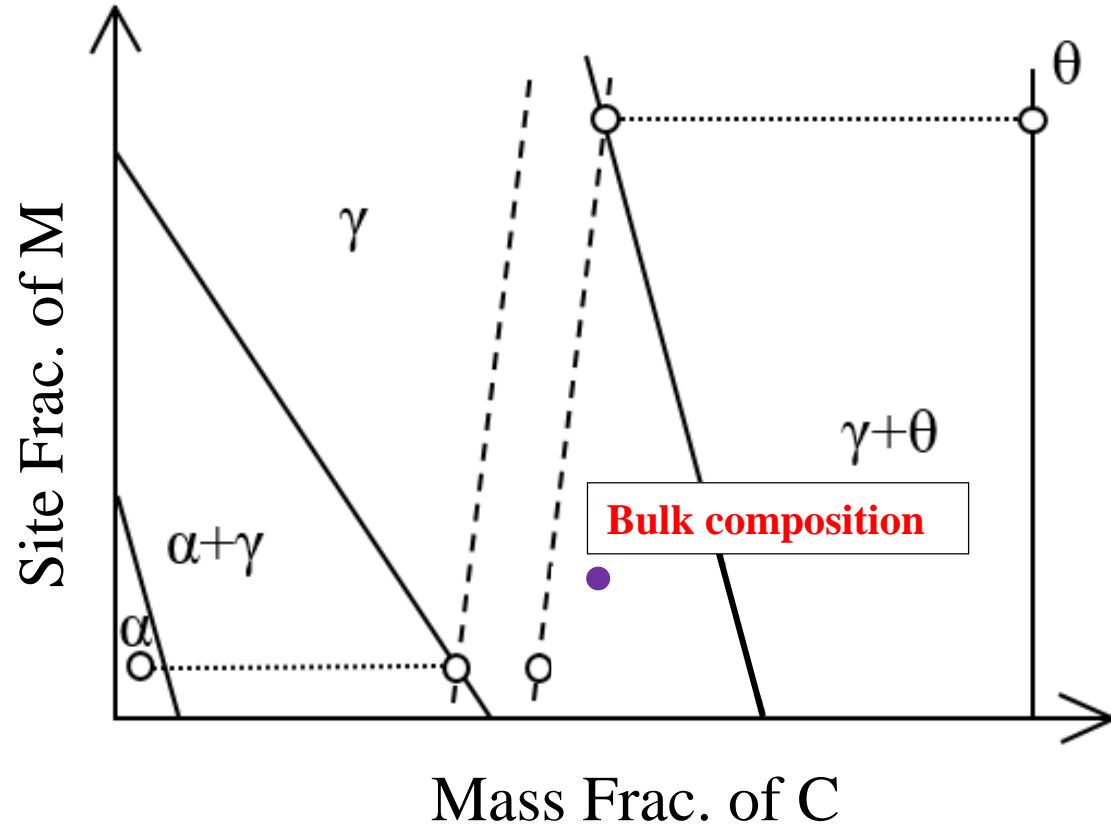
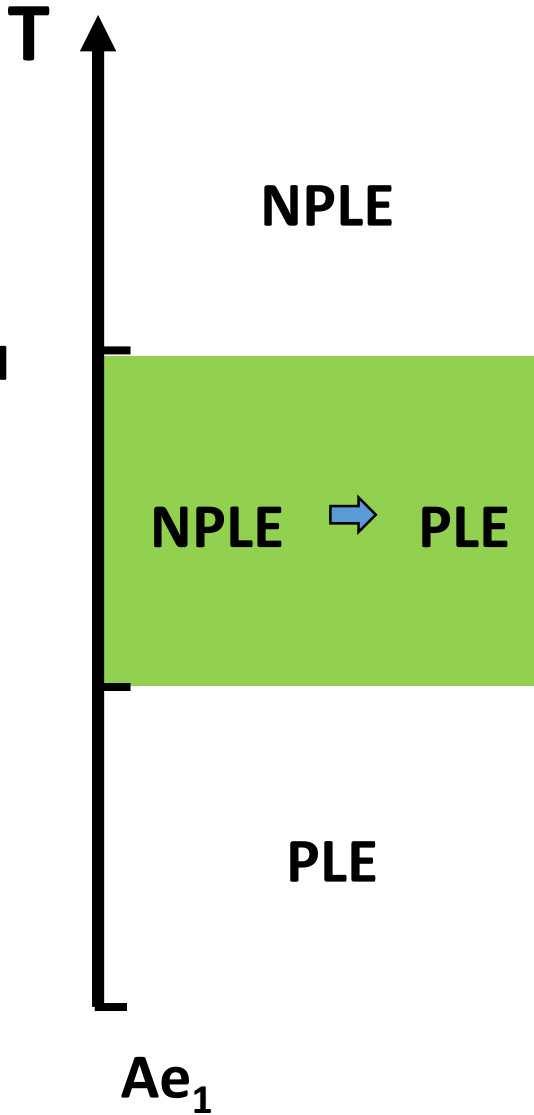


Austenitization_PNTTs



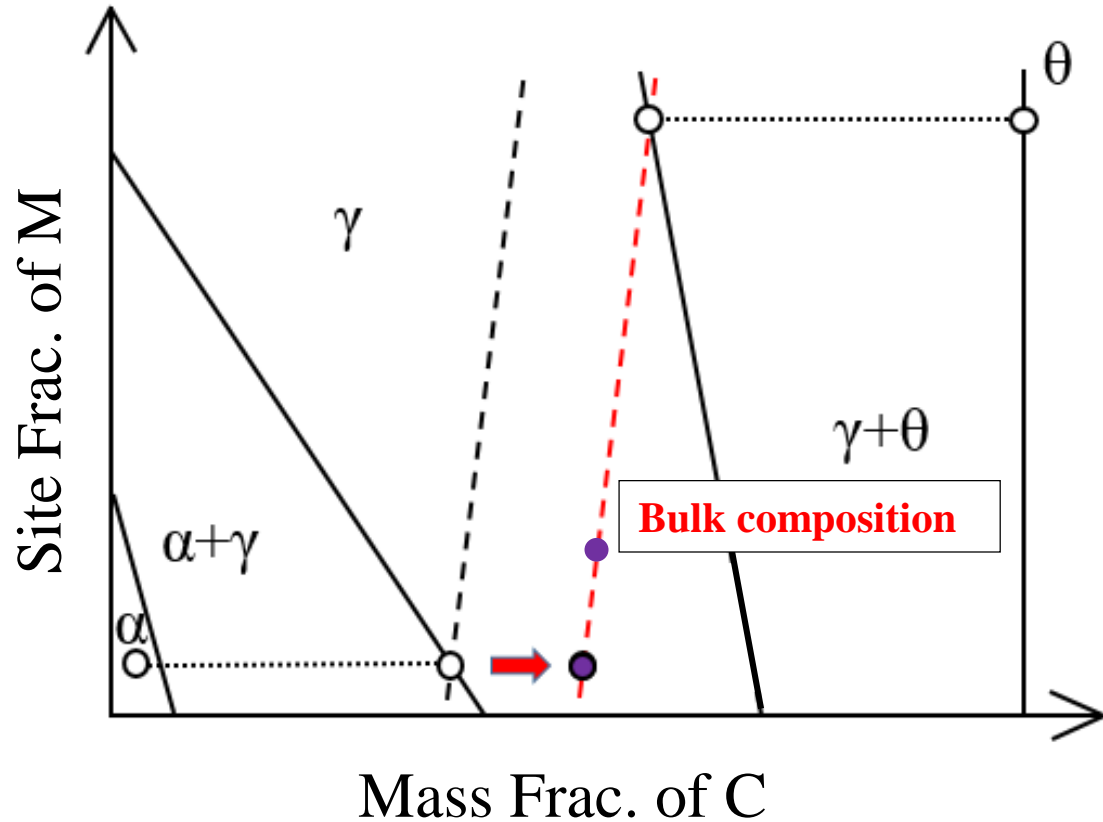
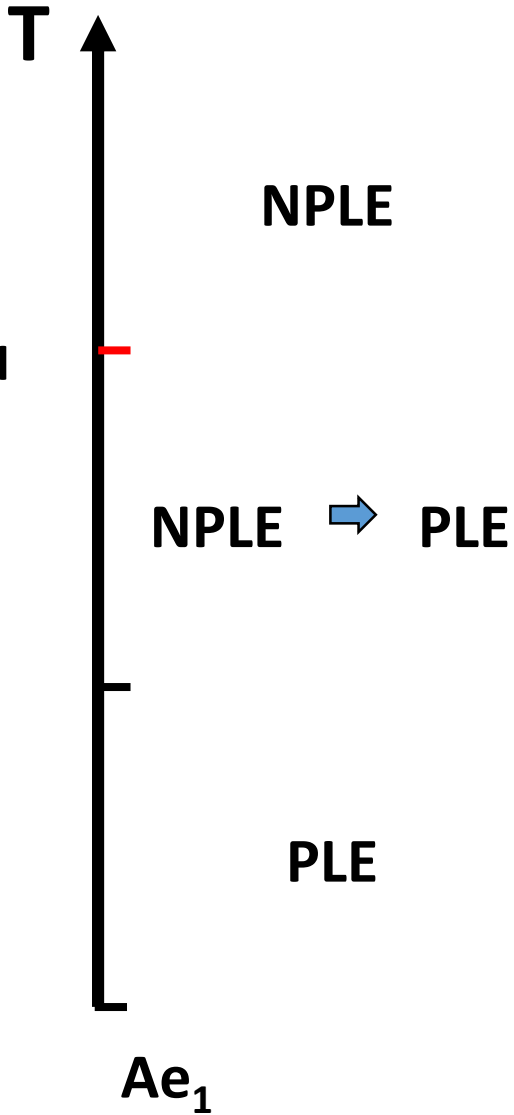


Austenitization_PNTTs



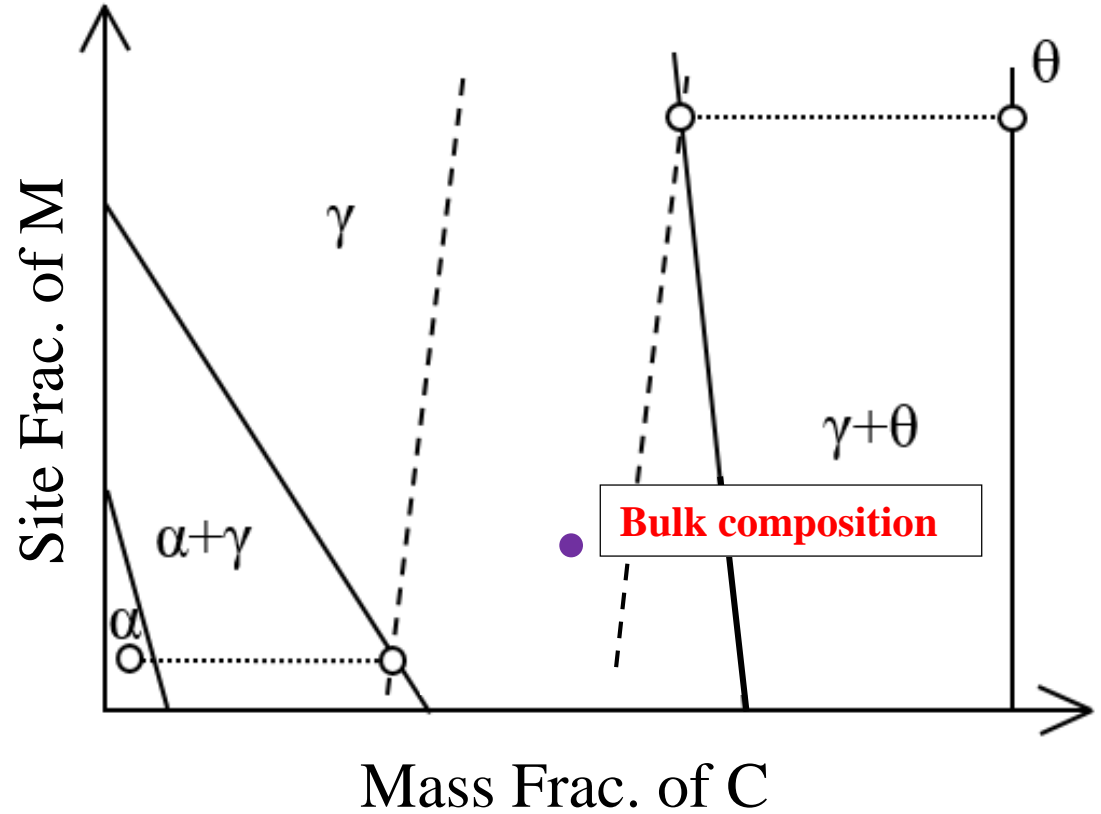
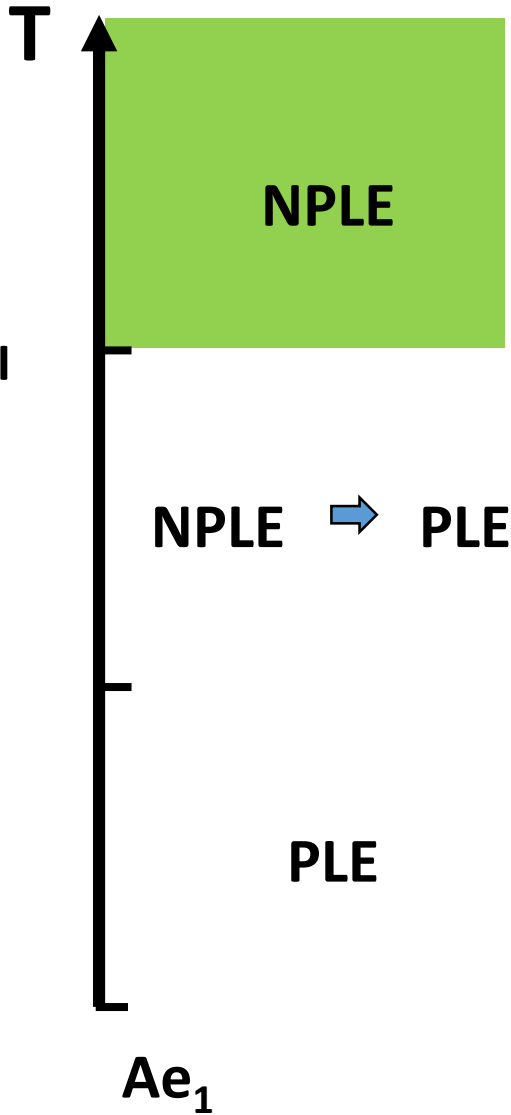


Austenitization_PNTTs



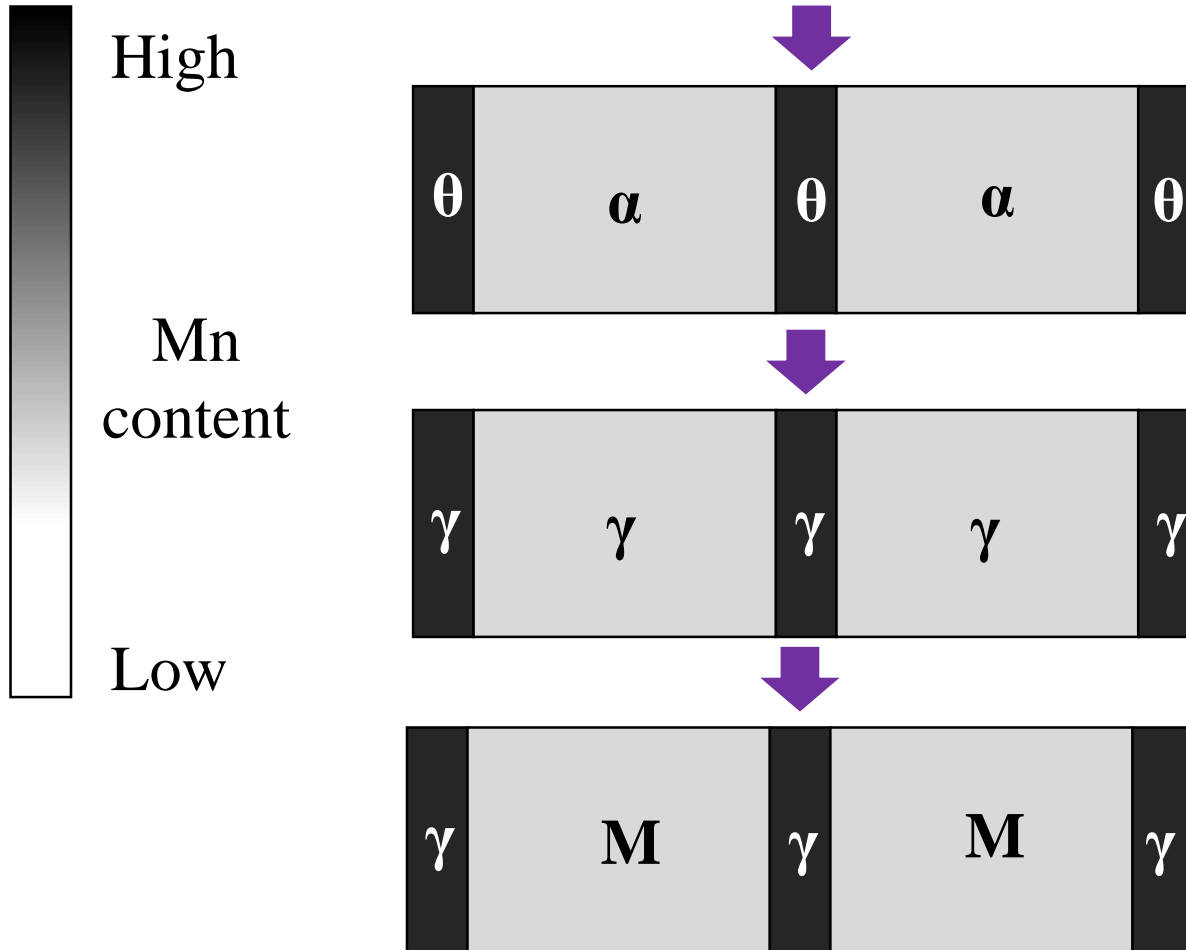


Austenitization_PNTTs





Medium Mn steel



Pearlite transformation

Partition of Mn

Austenitization
T > PNTT-II

Non-uniform distribution

Quench

MA dual phase



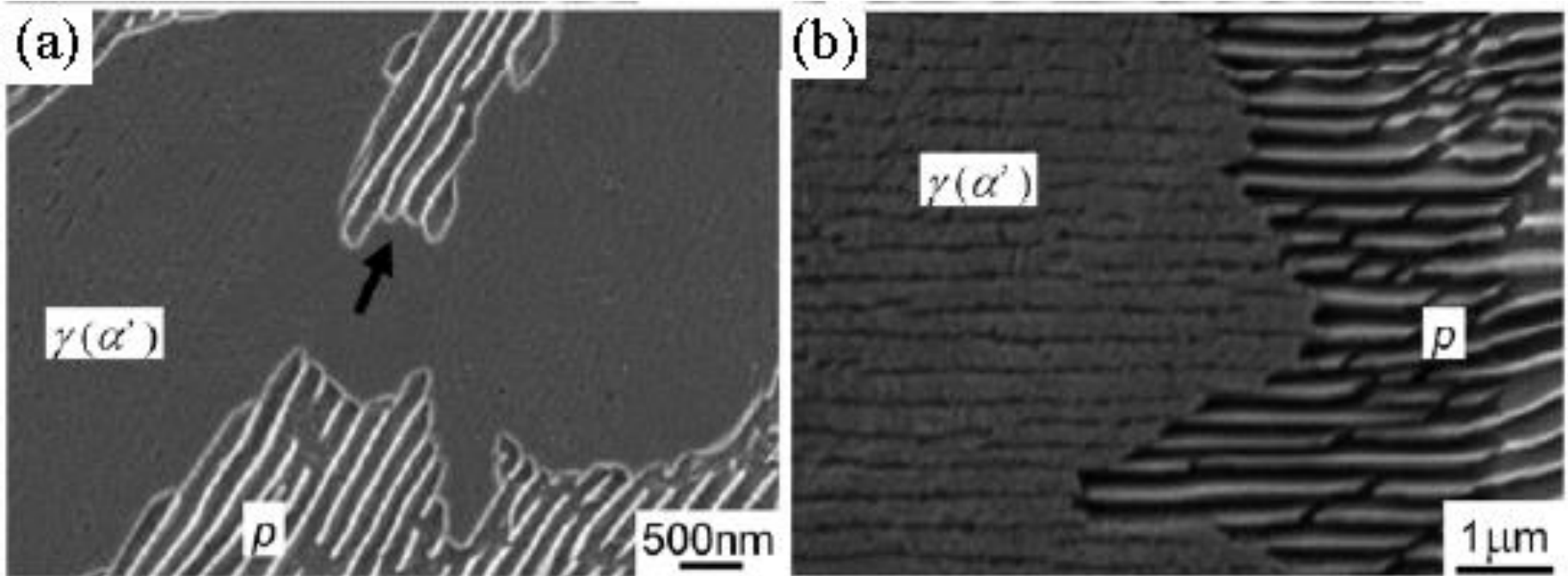
Medium Mn steel

Fe-0.6C-1Mn

800°C-1.1s

Fe-0.6C-2Mn

800°C-1.4s





Conclusion

1. Pearlite transformation is simulated via PFM. Compared with bulk diffusion, boundary diffusion of C and Mn plays an important role in the kinetics of transformation and partition of Mn between ferrite and cementite
2. A new heat treatment design for medium Mn steel is proposed. A non-uniform distribution of Mn is created by austenitization from pearlite, and the subsequent quenching may lead to a M&A dual phase microstructure.



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Thanks for your attention