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	UTILITY			Attorney Docket	No.	CCLL-	1-002.0	
F	PATENT APPLICAT	ΓΙΟΝ		First Named Inv	entor	Liyu Li		
	TRANSMITTA	L	Ī	Title		METHODS AND SY	STEMS FOR PRODUC	CTION OF AN AQUEOUS HYPOCHLOROU
(Only fo	r new nonprovisional applications unde	er 37 CFR 1.53(b))		Priority Mail Expr Label No.	ess®	VIA EF	S	
See MPEP (APPLICATION ELEME chapter 600 concerning utility patent a			ADDRESS	то:		P.O. Bo	r for Patents x 1450 A 22313-1450
	ansmittal Form 8/17 or equivalent)			ACCOM	1PAN'	ING AP	PLICATIO	ON PAPERS
2. Applic	ant asserts small entity status. CFR 1.27					ocument(s))		
	ant certifies micro entity status. S				Name	e of Assignee	2	
4. Specific Both the (See Min See Min	ication [Total P e claims and abstract must start on a n PEP § 608.01(a) for information on the ng(s) (35 U.S.C. 113) [Total S Oath or Declaration [Total P bstitute statements under 37 CFR 1.64 n oath or declaration under 37 CFR 1.65 wly executed (original or copy) opy from a prior application (37 C ation Data Sheet *See note below CFR 1.76 (PTO/AIA/14 or equivalent) IM or CD-R icate, large table, or Computer Program ndscape Table on CD et and/or Amino Acid Sequence Su le, items a. – c. are required) mputer Readable Form (CRF) ecification Sequence Listing on: CD-ROM or CD-R (2 copies); or Paper tements verifying identity of above	ages 17 ew page. preferred arrangementets 2 ages 2 and assignments 8(e)) FR 1.63(d)) w. m (Appendix) bmission]	(when the life in	Translaticable) ation Discontinuous Copies	closure State 0-1449) of citations a endment Postcard uld be specific f Priority Do is claimed) Request 22(b)(2)(B)(i).	ement attached atty itemized) cument(s) Applicant must ic Filing;	Power of Attorney t attach form PTO/SB/35 itized Examination
(2) Fo	nefit claims under 37 CFR 1.78 an r applications filed under 35 U.S.C signee, person to whom the inven erest in the matter. See 37 CFR 1.	. 111, the applicat tor is under an obl 46(b).	tion m ligatio	nust contain an Al on to assign, or pe	OS specify rson who	ing the app	icant if the a	applicant is an
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Signature	/Bruce E. Black/				Date	- 41 12 P	Novemb	per 4, 2020
Name (Print/Type)	Bruce E. Black					ation No. ev/Agent)	41,622	

This collection of information is required by 37 CFR 1.53(b). The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.**

Privacy Act Statement

The **Privacy Act of 1974 (P.L. 93-579)** requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether disclosure of these records is required by the Freedom of Information Act.
- 2. A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patent.
- A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

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FEE	TRA	NSN	/IITTA	\L	Applic	ation Numb	per	Not Yet A	ssigned		
			, ,		Filing	Date		Novembe	r 4, 2020		
Applicant assert	ts small entit	ty status.	See 37 CFR 1	.27.	First N	lamed Inve	ntor	Liyu Li			
Applicant certifi		•			Exami	ner Name		Not Yet A	ssigned		
Form PTO/SB/15A been submitted p		alent must	either be encl	osed or have	Art Ur	it		Not Yet A	ssigned		
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Check Cred	dit Card	Money	Order 🔲 N	Ione 🗌 Ot	her (please i	dentify):					
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This collection of information is required by 37 CFR 1.136. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 30 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.**

Document Description: TrackOne Request

PTO/AIA/424 (04-14)

CERTIFICATION AND REQUEST FOR PF	RIORITIZED	EXAMINATION
UNDER 37 CFR 1.102(e)	(Page 1 of 1)	

	UNDER 37 CFI	(1.102(e) (Page 1 011)	
First Named Inventor:	Liyu Li	Nonprovisional Application Number (if known):	Not Yet Assigned
Title of Invention:	METHODS AND SYSTEMS FOR PROD	DUCTION OF AN AQUEOUS HYPOCH	ILOROUS ACID SOLUTION

APPLICANT HEREBY CERTIFIES THE FOLLOWING AND REQUESTS PRIORITIZED EXAMINATION FOR THE ABOVE-IDENTIFIED APPLICATION.

- 1. The processing fee set forth in 37 CFR 1.17(i)(1) and the prioritized examination fee set forth in 37 CFR 1.17(c) have been filed with the request. The publication fee requirement is met because that fee, set forth in 37 CFR 1.18(d), is currently \$0. The basic filing fee, search fee, and examination fee are filed with the request or have been already been paid. I understand that any required excess claims fees or application size fee must be paid for the application.
- 2. I understand that the application may not contain, or be amended to contain, more than four independent claims, more than thirty total claims, or any multiple dependent claims, and that any request for an extension of time will cause an outstanding Track I request to be dismissed.
- 3. The applicable box is checked below:
 - I. Original Application (Track One) Prioritized Examination under § 1.102(e)(1)
- i. (a) The application is an original nonprovisional utility application filed under 35 U.S.C. 111(a).
 This certification and request is being filed with the utility application via EFS-Web.
 - (b) The application is an original nonprovisional plant application filed under 35 U.S.C. 111(a). This certification and request is being filed with the plant application in paper.
- ii. An executed inventor's oath or declaration under 37 CFR 1.63 or 37 CFR 1.64 for each inventor, <u>or</u> the application data sheet meeting the conditions specified in 37 CFR 1.53(f)(3)(i) is filed with the application.
 - II. Request for Continued Examination Prioritized Examination under § 1.102(e)(2)
- i. A request for continued examination has been filed with, or prior to, this form.
- ii. If the application is a utility application, this certification and request is being filed via EFS-Web.
- iii. The application is an original nonprovisional utility application filed under 35 U.S.C. 111(a), or is a national stage entry under 35 U.S.C. 371.
- iv. This certification and request is being filed prior to the mailing of a first Office action responsive to the request for continued examination.
- v. No prior request for continued examination has been granted prioritized examination status under 37 CFR 1.102(e)(2).

Signature / Bruce E. Black /	_{Date} November 4, 2020
Name (Print/Typed) Bruce E. Black	Practitioner 41622 Registration Number
Note: This form must be signed in accordance with 37 CFR 1.33. See 37 CFR 1.4(d) for Submit multiple forms if more than one signature is required.*	
*Total of forms are submitted.	

Privacy Act Statement

The **Privacy Act of 1974 (P.L. 93-579)** requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

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- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
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- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
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- 9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number.

Application Data	a Sheet 37 CFR 1.76	Attorney Docket Number	CCLL-1-002.0	
Application but		Application Number		
Title of Invention	METHODS AND SYSTEMS F	FOR PRODUCTION OF AN AQU	UEOUS HYPOCHLOROUS ACID SOLI	UTION
		provisional application for which it is ited States Patent and Trademark C	being submitted. The following form contain	s the
This document may be c	completed electronically and subr	mitted to the Office in electronic for	rmat using the Electronic Filing System (El	FS) or the
document may be printed	and included in a paper filed app	iicauon.		
Secrecy Order	37 CFR 5.2:			
			t may fall under a Secrecy Order pu er may not be filed electronically.)	rsuant to
		that fall under Secrecy Orde	er may not be liled electronically.)	
Inventor Inform	nation:			
Inventor 1			Remove	
Legal Name				
Prefix Given Name	e Mi	iddle Name	Family Name	Suffix
Liyu			Li	_
Residence Informa		Residency Non US Re		ce
City Bellevue	State/	Province WA Countr	y of Residence US	
Mailing Address of I	nventor:			
Address 1	7129 169th Ave SE			
Address 2			TT	
City Bellevu		State/Prov		
Postal Code	98006	Country	US	
	form by selecting the Add	ventor Information blocks lbutton.	May be Add	
Corresponden	ce Information:			
Enter either Custon	ner Number or complete	the Correspondence Inforn	nation section below.	
For further informa	tion see 37 CFR 1.33(a).			
☐ An Address is b	peing provided for the co	rrespondence Information	of this application.	
Customer Number	122997			
Email Address	patentdocketing@lov	vegrahamjones.com	Add Email Remov	re Email
Application In	formation:			
Title of the Inventio	n METHODS AND SY SOLUTION	STEMS FOR PRODUCTION O	F AN AQUEOUS HYPOCHLOROUS A	CID
Attorney Docket Nu		Small Ent	tity Status Claimed 🔀	
Application Type	Nonprovisional	l .		▼
Subject Matter	Utility			▼
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Under the Pap	erwork Reduction Act of	1995, no pers	sons are required	to respond to a collecti	on of information	unless it contains a valid OMB control number
Application Data	Sheet 37 CE	D 1 76	Attorney D	ocket Number	CCLL-1-002	2.0
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Title of Invention	METHODS AND S	YSTEMS F	OR PRODUC	CTION OF AN AQU	JEOUS HYPO	OCHLOROUS ACID SOLUTION
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application papers includi	ng a specification ar	nd any draw	ings are being	filed. Any domesti	c benefit or for	(a). Do not complete this section if eign priority information must be eign Priority Information").
For the purposes of a filing reference to the previously						plication are replaced by this
Application number of the filed application	ne previously	Filing dat	te (YYYY-MM-D	DD)	Intelle	i- ectual Property Authority or Country
Publication In	formation:				·	
Request Early P	ublication (Fee r	equired at	time of Req	uest 37 CFR 1.2	219)	
35 U.S.C. 122(b subject of an ap) and certify that	t the inven	ntion disclose	ed in the attache	d applicatior	not be published under n has not and will not be the al agreement, that requires
this information in the A	ation should be p pplication Data Sh Number or comple	rovided fo eet does nete the Rep	ot constitute a presentative N	power of attorney lame section belo	in the applic	ney in the application. Providing ation (see 37 CFR 1.32). Stions are completed the customer
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Application Da	ata Shoot 37 CED 1 76	Attorney Docket Number	CCLL-1-002.0
Application Data Sheet 37 CFR 1.76		Application Number	
Title of Invention	METHODS AND SYSTEMS F	FOR PRODUCTION OF AN AQU	JEOUS HYPOCHLOROUS ACID SOLUTION

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Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications

This application (1) claims priority to or the benefit of an application filed before March 16, 2013 and (2) also
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Application Da	ata Shaat 37 CED 1 76	Attorney Docket Number	CCLL-1-002.0
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Title of Invention	METHODS AND SYSTEMS F	FOR PRODUCTION OF AN AQU	JEOUS HYPOCHLOROUS ACID SOLUTION

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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	CCLL-1-002.0
		Application Number	
Title of Invention	METHODS AND SYSTEMS F	FOR PRODUCTION OF AN AQU	JEOUS HYPOCHLOROUS ACID SOLUTION

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Assignee		Legal Re	presentative un	der 35 U.S.	C. 117	Joint	Inventor	
Person to whom the inventor is obligated to assign. Person who shows sufficient proprietary interest								
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Application Data Sheet 37 CFR 1.76		Attorney Doo	Docket Number CCLL-1-002.0		002.0				
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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	CCLL-1-002.0	
		Application Number		
Title of Invention	METHODS AND SYSTEMS F	STEMS FOR PRODUCTION OF AN AQUEOUS HYPOCHLOROUS ACID SOLUTI		

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DECLARATION (37 CFR 1.63) FOR UTILITY OR DESIGN APPLICATION USING AN APPLICATION DATA SHEET (37 CFR 1.76)

Title of Invention	METHODS AND SYSTEMS FOR PRODUCTION OF AN AQUEOUS HYPOCHLOROUS ACID SOLUTION
As the below	w named inventor, I hereby declare that:
This declaration is directed to	\$ \$\$\$\$ \$ \$\$\$\$\\\\$\$\$\\\$\$\$\$\$\$\$\$\$\$\$\$\$\$\$\$\$\$\$
	United States application or PCT international application number
	filed on
The above-i	dentified application was made or authorized to be made by me.
I believe tha	t I am the original inventor or an original joint inventor of a claimed invention in the application.
	nowledge that any willful false statement made in this declaration is punishable under 18 U.S.C. 1001 prisonment of not more than five (5) years, or both.
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Inventor _	Liyu Li Date (Optional): 11-4-20
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- A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
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- A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

ABSTRACT

A method for making an aqueous hypochlorous acid (HClO) solution includes electrolyzing a solution of sodium chloride to produce a solution of sodium hypochlorite; and producing the aqueous hypochlorous acid solution by adjusting a pH of the solution of sodium hypochlorite to a value within a range of 3 to 8 by adding a selected weak acid to the solution of sodium hypochlorite to produce a buffer including the selected weak acid and a salt of the selected weak acid.

CLAIMS

What is claimed as new and desired to be protected by Letters Patent of the United States is:

1. A method for making an aqueous hypochlorous acid (HClO) solution, the method comprising

electrolyzing a solution of sodium chloride to produce a solution of sodium hypochlorite; and

producing the aqueous hypochlorous acid solution by adjusting a pH of the solution of sodium hypochlorite to a value within a range of 3 to 8 by adding a selected weak acid to the solution of sodium hypochlorite to produce a buffer comprising the selected weak acid and a salt of the selected weak acid.

- 2. The method of claim 1, wherein the aqueous hypochlorous acid solution has no more than 500 ppm hypochlorous acid.
- 3. The method of claim 1, further comprising adding a basic salt or a base to the solution of sodium chloride.
- 4. The method of claim 3, wherein the basic salt or base is selected from sodium bicarbonate, sodium carbonate, or sodium hydroxide.
- 5. The method of claim 1, wherein the selected weak acid is acetic acid and the buffer is a combination of acetic acid and sodium acetate.
- 6. The method of claim 5, wherein the buffer in the aqueous hypochlorous acid solution has a molar ratio of acetic acid to sodium acetate in a range from 1:100 to 100:1.

- 7. The method of claim 1, further comprising diluting the solution of sodium hypochlorite.
- 8. The method of claim 1, wherein the solution of sodium hypochlorite after the electrolyzing has at least 500 ppm sodium hypochlorite.
- 9. The method of claim 1, wherein the solution of sodium hypochlorite after the electrolyzing has at least 1000 ppm sodium hypochlorite.
- 10. The method of claim 1, wherein producing the aqueous hypochlorous acid solution comprises producing the aqueous hypochlorous acid solution by adjusting a pH of the solution of sodium hypochlorite to a value within a range of 4 to 6.
- 11. The method of claim 1, wherein electrolyzing the solution of sodium chloride comprises electrolyzing the solution of sodium chloride in an electrolysis cell comprising at least one positive electrode and at least one negative electrode without a membrane or separator between the at least one positive electrode and the at least one negative electrode.
- 12. The method of claim 11, further comprising transferring the solution of sodium hypochlorite from the electrolysis cell after the electrolyzing and, after the transferring, receiving the selected weak acid in the electrolysis cell.
- 13. The method of claim 11, further comprising generating, storing or re-generating a solution of sodium hypochlorite in the electrolysis cell while the hypochlorous acid product in the product tank is being consumed.

- 14. The method of claim 13, further comprising regenerating the sodium hypochlorite solution by re-electrolyzing the sodium hypochlorite solution after storing for a predetermined time period, wherein the predetermined time period is at least twelve hours.
- 15. A system for making an aqueous hypochlorous acid (HClO) solution, the system comprising:

an electrolysis cell;

a water tank or a coupling arrangement configured for coupling to an external water source; an acid tank configured for receiving a selected weak acid;

a NaCl tank configured for receiving an aqueous sodium chloride solution;

a product tank;

conduits individually coupling the water tank, acid tank, NaCl tank, and product tank to the electrolysis cell; and

a controller configured and arranged to perform actions when the NaCl tank contains the sodium chloride solution, the acid tank contains the selected weak acid, water is in the water tank or the system is coupled to a water source using the coupling arrangement, the actions including:

directing a portion of the aqueous sodium chloride solution from the NaCl tank to the electrolysis cell;

electrolyzing the portion of the solution of sodium chloride to produce a solution of sodium hypochlorite in the electrolysis cell;

directing the solution of sodium hypochlorite into the product tank; and directing a portion of the selected weak acid in the acid tank into the solution of sodium hypochlorite to produce the aqueous hypochlorous acid solution by adjusting a pH to a value within a range of 3 to 8 by adding the selected weak acid to the solution of sodium hypochlorite to form a buffer using the selected weak acid and a salt of the selected weak acid.

- 16. The system of claim 15, wherein the system is configured to produce the aqueous hypochlorous acid solution by providing acetic acid in the acid tank and a NaCl solution in the NaCl tank
- 17. The system of claim 16, wherein the system is further configured to produce the aqueous hypochlorous acid when a basic salt or a base is provided in the sodium chloride solution to reduce production of chlorine gas, wherein the basic salt or base is selected from sodium carbonate, sodium bicarbonate, or sodium hydroxide.
- 18. The system of claim 15, wherein the electrolysis cell comprises at least one positive electrode and at least one negative electrode without a membrane or separator between the at least one positive electrode and the at least one negative electrode
- 19. The system of claim 18, wherein at least one of the at least one positive electrode or the at least one negative electrode comprises ruthenium and iridium.
- 20. The system of claim 18, wherein at least one of the at least one positive electrode or the at least one negative electrode comprises titanium.
- 21. The system of claim 15, further comprising a housing, wherein the electrolysis cell, water tank or coupling arrangement, acid tank, NaCl tank, product tank, and controller are disposed in the housing.
- 22. The system of claim 15, further comprising at least one level indicator in at least one of the acid tank, NaCl tank, or product tank and coupled to the controller.

METHODS AND SYSTEMS FOR PRODUCTION OF AN AQUEOUS HYPOCHLOROUS ACID SOLUTION

5 <u>FIELD</u>

The present invention is directed to the area of methods and systems for the production of an aqueous hypochlorous acid (HClO) solution. The present invention is also directed to methods and systems for the production of hypochlorous acid by consumers.

10 <u>BACKGROUND</u>

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Hypochlorous acid has been found to have bactericidal and virucidal properties. Compared to sodium hypochlorite, which is often used as sterilizing agent, hypochlorous acid is said to be over 80 times more effective. Hypochlorous acid can keep hands smooth, is eco-friendly, and can be used directly to wash vegetables, fruits, and tableware. It can also effectively eliminate many pesticide residues on the surface of vegetables. Hypochlorous acid is approved by the US Food and Drug Administration (FDA) and recommended by Japan's Ministry of Health, Labour and Welfare as a food-grade germicidal solution.

BRIEF SUMMARY

One embodiment is a method for making an aqueous hypochlorous acid (HClO) solution that includes electrolyzing a solution of sodium chloride to produce a solution of sodium hypochlorite; and producing the aqueous hypochlorous acid solution by adjusting a pH of the solution of sodium hypochlorite to a value within a range of 3 to 8 by adding a selected weak acid to the solution of sodium hypochlorite to produce a buffer including the selected weak acid and a salt of the selected weak acid.

In at least some embodiments, the aqueous hypochlorous acid solution has no more than 500 ppm hypochlorous acid. In at least some embodiments, the method

further includes adding a basic salt or a base to the solution of sodium chloride. In at least some embodiments, the basic salt or base reduces or absorbs chlorine gas generated during the electrolysis. In at least some embodiments, the basic salt or base is selected from sodium bicarbonate, sodium carbonate, or sodium hydroxide.

In at least some embodiments, the selected weak acid is acetic acid and the buffer is a combination of acetic acid and sodium acetate. In at least some embodiments, the buffer in the aqueous hypochlorous acid solution as a molar ratio of acetic acid to sodium acetate in a range from 1:100 to 100:1.

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In at least some embodiments, the method further includes diluting the solution of sodium hypochlorite. In at least some embodiments, the solution of sodium hypochlorite after the electrolyzing has at least 500 ppm sodium hypochlorite. In at least some embodiments, the solution of sodium hypochlorite after the electrolyzing has at least 1000 ppm sodium hypochlorite.

In at least some embodiments, producing the aqueous hypochlorous acid solution includes producing the aqueous hypochlorous acid solution by adjusting a pH of the solution of sodium hypochlorite to a value within a range of 4 to 6. In at least some embodiments, electrolyzing the solution of sodium chloride in an electrolysis cell including at least one positive electrode and at least one negative electrode without a membrane or separator between the at least one positive electrode and the at least one negative electrode. In at least some embodiments, the method further includes transferring the solution of sodium hypochlorite from the electrolysis cell after the electrolyzing and, after the transferring, receiving the selected weak acid in the electrolysis cell. In at least some embodiments, the selected weak acid removes calcium or magnesium deposits one the electrode surface. In at least some embodiments, un-softened water is used in the system for solution preparation and for dilution.

In at least some embodiments, the method further includes generating, storing or re-generating a solution of sodium hypochlorite in the electrolysis cell while the

hypochlorous acid product in the product tank is being consumed. In at least some embodiments, the method further includes regenerating the sodium hypochlorite solution by re-electrolyzing the sodium hypochlorite solution after storing for a predetermined time period, wherein the predetermined time period is at least twelve hours.

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Another embodiment is a system for making an aqueous hypochlorous acid (HClO) solution. The system includes an electrolysis cell; a water tank or a coupling arrangement configured for coupling to an external water source; an acid tank configured for receiving a selected weak acid; a NaCl tank configured for receiving an aqueous sodium chloride solution; a product tank; conduits individually coupling the water tank, acid tank, NaCl tank, and product tank to the electrolysis cell; and a controller configured and arranged to perform actions when the NaCl tank contains the sodium chloride solution, the acid tank contains the selected weak acid, water is in the water tank or the system is coupled to a water source using the coupling arrangement, the actions including: directing a portion of the aqueous sodium chloride solution from the NaCl tank to the electrolysis cell; electrolyzing the portion of the solution of sodium chloride to produce a solution of sodium hypochlorite in the electrolysis cell; directing the solution of sodium hypochlorite into the product tank; and directing a portion of the selected weak acid in the acid tank into the solution of sodium hypochlorite to produce the aqueous hypochlorous acid solution by adjusting a pH to a value within a range of 3 to 8 by adding the selected weak acid to the solution of sodium hypochlorite to form a buffer using the selected weak acid and a salt of the selected weak acid.

In at least some embodiments, the system is configured to produce the aqueous hypochlorous acid solution by providing acetic acid in the acid tank and a NaCl solution in the NaCl tank. In at least some embodiments, the system is further configured to produce the aqueous hypochlorous acid when a base or a basic salt is provided in the sodium chloride solution to reduce production of chlorine gas, wherein the base or basic salt is selected from sodium hydroxide, sodium carbonate, or sodium bicarbonate.

In at least some embodiments, the electrolysis cell includes at least one positive electrode and at least one negative electrode without a membrane or separator between the at least one positive electrode and the at least one negative electrode. In at least some embodiments, at least one of the at least one positive electrode or the at least one negative electrode includes ruthenium and iridium. In at least some embodiments, at least one of the at least one positive electrode includes titanium.

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In at least some embodiments, the system further includes a housing, wherein the electrolysis cell, water tank or coupling arrangement, acid tank, NaCl tank, product tank, and controller are disposed in the housing. In at least some embodiments, the system further includes at least one level indicator in at least one of the acid tank, NaCl tank, or product tank and coupled to the controller.

BRIEF DESCRIPTION OF THE DRAWINGS

Non-limiting and non-exhaustive embodiments of the present invention are described with reference to the following drawings. In the drawings, like reference numerals refer to like parts throughout the various figures unless otherwise specified.

For a better understanding of the present invention, reference will be made to the following Detailed Description, which is to be read in association with the accompanying drawings, wherein:

- FIG. 1 is a schematic diagram of one embodiment of system for making an aqueous HClO solution, according to the invention; and
 - FIG. 2 is a flowchart of a one embodiment of method for making an aqueous HClO solution, according to the invention.

DETAILED DESCRIPTION

The present invention is directed to the area of methods and systems for the production of hypochlorous acid (HClO). The present invention is also directed to methods and systems for the production of hypochlorous acid by consumers.

Many commercial methods of HClO production involve brine electrolysis using a cell with a membrane. These commercial methods may be relatively complicated and only practical for commercial-scale applications. Production of hypochlorous acid has also been achieved on a commercial scale by adding an acid, such as hydrochloric acid, to sodium hypochlorite (NaClO) via a precious pH control process. For small-scale, onsite, and on-demand applications, NaClO and HCl are not always available. In addition, the precious pH control during the mixing process is required to prevent or reduce the generation of toxic chlorine gas and may be difficult in non-industrial arrangements. A solution of sodium hypochlorite (NaClO) has been produced using a small, low cost electrochemical cell and a brine solution. However, sodium hypochlorite is more toxic to the human body.

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Methods and systems for generating an aqueous hypochlorous acid solution are described herein. In at least some embodiments, the methods and systems can be used by consumers to produce the hypochlorous acid solution on a small-scale, on-site, or on-demand basis. In at least some embodiments, these methods and systems utilize readily available components, such as water, sodium chloride (NaCl), acetic acid (e.g., vinegar), and either sodium bicarbonate (e.g., baking soda), sodium carbonate (e.g., soda ash or washing soda), or sodium hydroxide (e.g., caustic soda or lye) to generate the hypochlorous acid. In at least some embodiments, the water is unsoftened.

The methods and systems described herein utilize the equilibrium of HClO and NaClO in aqueous solution where HClO is the dominate species in solutions with a pH between approximately 3 and approximately 7. In at least some embodiments, the methods and systems are configured to produce an aqueous hypochlorous acid solution with a pH in a range from 3 to 8, a range from 3.5 to 7, a range from 4 to 6.5, or a range from 4 to 6. HClO and NaClO typically reach equilibrium very quickly in solution. HClO exists stably at a pH range of at least 4 to 6.5.

To maintain the pH in the desired range, the aqueous hypochlorous acid solution has a buffer that includes a weak acid and a salt of the weak acid and which maintains

the pH in the desired range. In at least some embodiments, the buffer includes acetic acid and a salt of acetic acid such as sodium acetate, aluminum acetate, ammonium acetate, or potassium acetate.

In at least some embodiments, the buffer can be formed by addition of a weak acid, such as acetic acid or the like, to the NaClO solution. The addition of acetic acid to the NaClO solution results in the formation of sodium acetate (which, in water, exists primarily in the form of sodium and acetate ions). Thus, in at least some embodiments, the buffer can be prepared using household chemicals, such as acetic acid (e.g., vinegar).

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$$CH_3COOH \rightarrow CH_3COO^- + H^+$$
 pKa = 4.76.

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In water, CH₃COONa: CH₃COOH in a 1:1 molar ratio gives a solution with pH of approximately 4.76. CH₃COONa: CH₃COOH in a 10:1 molar ratio gives a solution with pH of approximately 5.76.

In at least some embodiments, the resulting aqueous hypochlorous acid solution is stable for at least 5, 10, 15, 30, 60, or more days.

In at least some embodiments, a buffer of acetic acid and an acetate salt, such as sodium acetate, can reliably keep the pH within a range of 4 to 6.5 with a molar ratio of acetic acid to acetate salt in a range from 1:100 to 100:1. In at least some embodiments, no accurate volumetric control is needed for pH control due to the buffer.

In at least some embodiments, the addition of a soluble basic salt or base may further facilitate formation of the buffer. For example, the addition of soluble sodium bicarbonate, sodium carbonate, or sodium hydroxide to the acetic acid may further the formation of sodium acetate (which, in water, exists primarily in the form of sodium and acetate ions). In at least some embodiments, the soluble basic salt or base can be household chemicals, such as sodium bicarbonate (NaHCO₃), commonly known as baking soda, sodium carbonate (Na₂CO₃), commonly known as soda ash or washing

soda, or sodium hydroxide (NaOH), commonly known as caustic soda or lye. For example,

$$CH_3COOH + NaHCO_3 \rightarrow CH_3COONa + H_2O + CO_2$$

In at least some embodiments, adding the basic salt or base, such as, for example, NaHCO₃, Na₂CO₃, or NaOH, for the buffer into the NaCl solution can reduce Cl₂ generation during the electrolysis of NaCl.

In at least some embodiments, the components of the buffer are introduced separately during the production of the HClO solution.

Methods and systems for preparing an aqueous HClO solution include generating a concentrated NaClO solution in an electrochemical device from a solution containing NaCl, adjusting the pH of the NaClO solution using a buffer to produce the HClO solution, and diluting the as-produced NaClO or HClO to a desired concentration using water (preferably, unsoftened water).

The methods and systems include the electrochemical production of an aqueous NaClO solution from an aqueous NaCl solution according to the following equation:

$$NaCl + H_2O \rightarrow NaClO + H_2$$

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In at least some embodiments, the electrochemical production of aqueous NaClO is thought to proceed according to the following equations (although the invention does not rely on any particular mechanism or sequence of reactions):

Figure 1 illustrates one embodiment of a system 100 for production of an aqueous HClO (hypochlorous acid) solution. The system 100 includes a water source,

such as water tank 102 or a coupling arrangement that can be coupled to a stream of water from an external source; a sodium chloride source, such as NaCl tank 104 containing an aqueous solution of sodium chloride and, optionally, the basic salt or base for reducing the generation of Cl₂ gas; a weak acid source, such as acid tank 106 containing the weak acid that forms the buffer; an electrolysis cell 108; and a product tank 110. Any of the tanks can be replaced by any other suitable reservoir or a coupling arrangement that can be coupled to a continuous or intermittent stream source.

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The system 100 also includes a number of pumps 112 and various conduits 116, such as tubing or the like to carry the reactants and other components of the HClO solution, as well as the solution itself. Any other suitable mechanisms, methods, or techniques for flowing the components from the various sources to the electrolysis cell 108 and the product tank 110 can be used. For example, optional check valves may be used to prevent flow in the wrong directions. The HClO solution can be obtained at an outlet 118.

The system 100 also includes a control unit 122 that operates the system in general including the pumps 112, as needed. In at least some embodiments, the control unit 122 can include one or more user operable components, such as switches, buttons, a touchscreen, or the like to permit user control of the system 100.

The system 100 may include an optional filter 120 to filter the water from the water tank 102. In at least some embodiments with a filter 120, there may also be a conduit from the water tank 102 to the valve 114 that bypasses the filter 120. The pumps 112 can be individually any suitable type of pump including, but not limited to, peristaltic pumps, diaphragm pumps, centrifugal pumps, or the like. The pumps 112 can be all the same type of pump or different types of pumps.

In at least some embodiments, one or more of the water tank 102, NaCl tank 104, acid tank 106, or product tank 110 can include a level gauge 124 to monitor the level of the respective solution or component in that tank. In at least some embodiments, the control unit 122 monitors the level gauge(s) 124 and, preferably,

alerts a user if any level gauge drops below a predetermined level or rises above a predetermined level. Level gauges 124 in the water tank 102, NaCl tank 104, or acid tank 106 may indicate when additional source materials (e.g., water, NaCl, or weak acid) are needed. A level gauge 124 in the product tank 110 can indicate how much hypochlorous acid solution is presently available. In at least some embodiments, the product tank 110 may include one level gauge to monitor or warn for low fluid level and another level gauge to monitor or warn for high fluid level.

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In at least some embodiments, the product tank 110 (or any other tank) may include a pH gauge 126. In at least some embodiments, the control unit 122 monitors the pH gauge 126. In at least some embodiments, the control unit 122 may alert a user if any pH gauge is outside a desired pH range and the control unit 122 may direct the user to dispose of the contents of the product tank 110. In at least some embodiments, the control unit 122 may automatically (or under user direction) pump the weak acid from the acid tank 106 or the solution containing the basic salt from the NaCl tank 104 to adjust the pH.

Any suitable electrolysis cell 108 with two or more electrodes 109a, 109b can be used. In at least some embodiments, the electrolysis cell 108 does not include a membrane or separator between the electrodes 109a, 109b. In at least some embodiments, the surface of the positive electrode(s) 109a or negative electrode(s) 109b contains ruthenium, iridium, or any combination thereof. In at least some embodiments, the surface of the negative electrode(s) 109b contains platinum. In at least some embodiments, a bulk material of the positive and negative electrode(s) 109a, 109b is titanium, although any other suitable metal, alloy, or combination thereof can be used.

In at least some embodiments, the system 100 can be disposed in a single housing 130. In at least some embodiments, the system 100 and housing 130 can be portable. In at least some embodiments, as indicated above, instead of a water tank 102 (or other tank), the system 100 may include a coupling arrangement to couple to a streaming source of water (or other component). In at least some embodiments, one or

more of the water tank 102, acid tank 106, NaCl tank 104, or product tank 110 can be disposed outside the housing 130.

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Figure 2 is a flowchart of operation of the system to produce an aqueous solution of HClO. In step 202, a portion of the NaCl solution from the NaCl tank 104 is pumped into the electrolysis cell 108. In at least some embodiments, the NaCl solution includes a basic salt or base (for example, sodium bicarbonate, sodium carbonate, or sodium hydroxide) to reduce the generation of Cl₂ gas during the electrolysis. In at least some embodiments, the NaCl solution includes at least 5, 10, 20, 30, 50, 100, 200, or 300 grams (or more) of NaCl per liter of water. In at least some embodiments, the NaCl solution includes, for example, 0.1, 0.25, 0.5, or 1 gram of the basic salt or base per liter of the NaCl solution. In at least some embodiments, pre-prepared concentrated NaCl solution and a weak acid solution are used to prepare the NaCl solution and the weak acid solution in tanks 104 and 106.

In step 204, the NaCl solution is electrolyzed in the electrolysis cell 108 to produce an aqueous NaClO solution as described above. In at least some embodiments, the concentration of NaClO in the aqueous NaClO solution produced in the electrolysis cell 108, prior to dilution, is at least 500, 1000, or 5000 ppm. In at least some embodiments, the on-site generation of high concentration NaClO via electrolysis of high concentration NaCl is safe and efficient due to high concentration of the NaCl reactant and high solution conductivity. In at least some embodiments, the consumption of NaCl is no more than 0.4, 0.5, or 1 gram of NaCl per liter of aqueous HClO solution.

In step 206, the NaClO solution is pumped into the product tank 110 and, in step 208, the NaClO solution is diluted using water from the water tank 102. Steps 206 and 208 can be performed in any order so that the NaClO solution is diluted before or after being pumped into the product tank 110. In some embodiments, step 208 is skipped, and dilution occurs during later steps.

In step 210, after the electrolysis cell 108 is emptied, the weak acid from the acid tank 106 flows into the electrolysis cell 108 and, optionally, at least partially cleans

the electrolysis cell of deposits such as calcium carbonate or magnesium carbonate residue. In at least some embodiments, the weak acid in the acid tank 106 is acetic acid and has a molarity of at least 0.1 or 0.3 M and may be in the range of 0.1 to 16 M. In optional step 212, the weak acid is diluted in the electrolysis cell 108 using water from the water tank 102.

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In step 214, the acid, after optional dilution, is pumped into the product tank 110 and combined with the NaClO solution. The weak acid forms a buffer to adjust the pH to a range from 3 to 8, a range from 3.5 to 7, a range from 4 to 6.5, or a range from 4 to 6 to produce the aqueous HClO solution. In at least some embodiments, the dilution of the NaClO or HClO solution during steps 208 and 214 is at least a factor of 25, 50, 75, 100, 150, or 200 or more.

The user can remove the aqueous HClO solution through the outlet 118. In at least some embodiments, the concentration of HClO in the aqueous HClO solution, after dilution in the product tank 110, is in a range from 1 to 500 ppm.

In at least some embodiments, the operation of the system 100 can be continuous with the steps in Figure 2 repeated on a continuous cycle. In at least some embodiments, the system 100 can be programmed so that the operation of the system 100 is repeated on a regular or periodic basis. In at least some embodiments, the system 100 can be programmed so that the operation of the system 100 is repeated based on measurements from the level gauge 124 in the product tank 110.

In at least some embodiments, the system 100 can store the concentrated NaClO solution or the diluted NaClO solution for periods of time (minutes, hours, or days) prior to introduction of the weak acid. In at least some embodiments, the concentrated NaClO solution or the diluted NaClO solution can be stored in the electrolysis cell 108, the product tank 110, or another storage tank (not shown) or any combination thereof.

In at least some embodiments, the aqueous hypochlorous acid solution is kept in the product tank. When the aqueous hypochlorous acid solution is consumed, a NaClO solution for the next batch of the aqueous hypochlorous solution is generated in the electrolysis cell 108 as described above and is stored in the electrolysis cell until the NaClO solution is needed to produce the next batch of the aqueous hypochlorous acid solution. Then, the NaClO solution is transferred to the product tank, diluted, and weak acid is added to produce more of the aqueous hypochlorous acid solution. In at least some embodiments, there may be limit (for example, 12 or 24 hours) to the length of time that the NaClO solution can remain in the electrolysis cell before reverting, at least in part, to an NaCl solution via the reaction below. In at lease some embodiments, the NaClO/NaCl solution in the electrolysis cell is re-charged every 24 hours as described above, or may be re-charged every 12 to 120 hours, or may be re-charged before producing next batch hypochlorous product.

$$2NaClO \rightarrow 2NaCl + O_2$$

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The above specification provides a description of the manufacture and use of the invention. Since many embodiments of the invention can be made without departing from the spirit and scope of the invention, the invention also resides in the claims hereinafter appended.

Appl. No.: Not Yet Assigned Docket No. CCLL-1-002.0

Title: METHODS AND SYSTEMS FOR PRODUCTION OF AN AQUEOUS HYPOCHLOROUS ACID SOLUTION

Inventor: Liyu Li Sheet 1 of 2

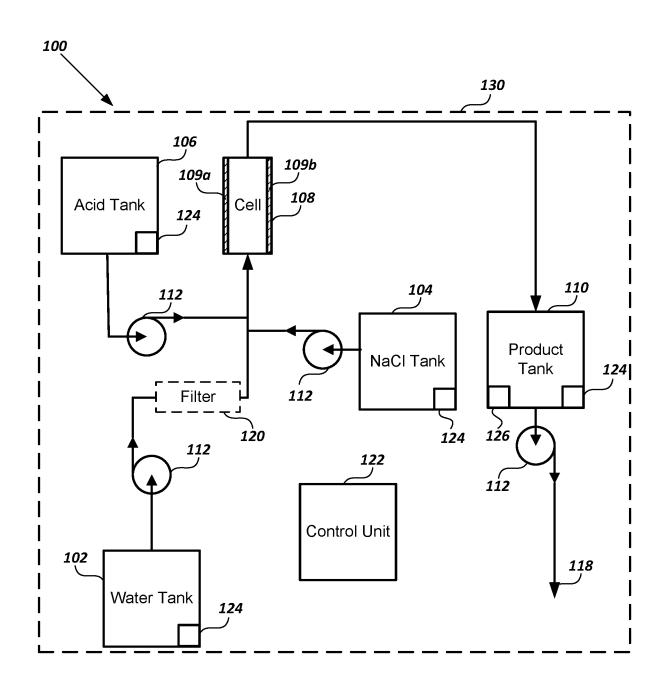


Fig. 1

Appl. No.: Not Yet Assigned Docket No. CCLL-1-002.0

Title: METHODS AND SYSTEMS FOR PRODUCTION OF AN AQUEOUS HYPOCHLOROUS ACID SOLUTION

Inventor: Liyu Li Sheet 2 of 2

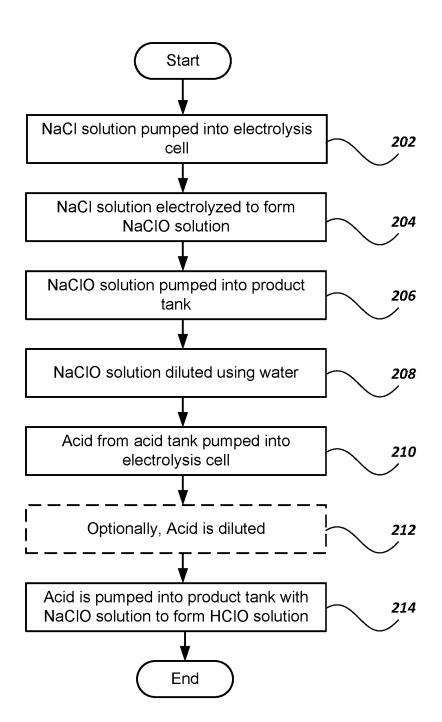


Fig. 2

Electronic Patent A	Electronic Patent Application Fee Transmittal						
Application Number:							
Filing Date:							
Title of Invention:		THODS AND SYSTE. POCHLOROUS ACID		JCTION OF AN AQU	JEOUS		
First Named Inventor/Applicant Name: Liyu Li							
Filer:	Bruce Black/Tiffany Kelly						
Attorney Docket Number:	CCLL-1-002.0						
Filed as Small Entity							
Filing Fees for Track I Prioritized Examination - Nonpr	rovis	ional Applicatio	n under 35 U	SC 111(a)			
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Basic Filing:							
UTILITY FILING FEE (ELECTRONIC FILING)		4011	1	80	80		
UTILITY SEARCH FEE		2111	1	350	350		
UTILITY EXAMINATION FEE		2311	1	400	400		
REQUEST FOR PRIORITIZED EXAMINATION		2817	1	2100	2100		
Pages:							
Claims:							
CLAIMS IN EXCESS OF 20		2202	2	50	100		
Miscellaneous-Filing:	'.		,				

Fee Code	Quantity	Amount	Sub-Total in USD(\$)			
1504	1	0	0			
2830	1	70	70			
Petition:						
Patent-Appeals-and-Interference:						
Post-Allowance-and-Post-Issuance:						
Extension-of-Time:						
Miscellaneous:						
Tot	al in USD	(\$)	3100			
	1504 2830	1504 1 2830 1	1504 1 0			

Electronic Ack	Electronic Acknowledgement Receipt					
EFS ID:	41038699					
Application Number:	17089537					
International Application Number:						
Confirmation Number:	1042					
Title of Invention:	METHODS AND SYSTEMS FOR PRODUCTION OF AN AQUEOUS HYPOCHLOROUS ACID SOLUTION					
First Named Inventor/Applicant Name:	Liyu Li					
Customer Number:	122997					
Filer:	Bruce Black/Tiffany Kelly					
Filer Authorized By:	Bruce Black					
Attorney Docket Number:	CCLL-1-002.0					
Receipt Date:	04-NOV-2020					
Filing Date:						
Time Stamp:	18:38:16					
Application Type:	Utility under 35 USC 111(a)					

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1	Miscellaneous Incoming Letter	CCLL-1-002Cert.pdf	096dceaaa3ff3ea2157934224346bb9cbc77 dc06	no	1
Warnings:				•	
Information:					
			386349		
2	Transmittal of New Application	CCLL-1-002AppTrans.pdf	d263f33d0968be40d978f9e4500de1237af3 9f88	no	2
Warnings:		•		•	
Information:					
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3	Transmittal Letter	CCLL-1-002FeeTransmittal.pdf	4063fa53ac960c09dd7dff81340039a1bcf9 7bef	no	1
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4	TrackOne Request	CCLL-1-002Track1.pdf	6568f5931bb5b729b1a21357c6cd63c827c 69f46	no	2
Warnings:		•			
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			1822175		
5	Application Data Sheet	CCLL-1-002ADS.pdf	d7d52ce0f738b0dcf8aef5718b04d4651de6 5619	no	8
Warnings:					
Information:					
			1361629		
6	Oath or Declaration filed	CCLL-1-002 Signed Dec.pdf	6fdb9c2adff717d4e6c0fb46182029526dd4 3568	no	2
Warnings:					
Information:					

7		CCL_002_APP_EFS.pdf	126554 08a6575b3e13412e611df9138b00b281aed a85e6	yes	17
	Multip	part Description/PDF files in .	zip description		
	Document Des	scription	Start	E	nd
	Abstrac	17		17	
	Claims	13		16	
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8	Drawings-only black and white line drawings	CCLL_002_Drawings_EFS.pdf	524665ee8e932043b801924a3958c332a2a 30ce0	no	2
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Information:					
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9	Fee Worksheet (SB06)	fee-info.pdf	211f08ea417cde2e367d1459ca9cf8707c35 6bc7	no	2
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New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

Attorney Docket No.: CCLL-1-002.0

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Utility Patent Application Transmittal (2 pages)

Fee Transmittal (1 page) Specification (17 pages) Drawings (2 sheets) Declaration (2 pages)

Application Data Sheet (8 pages)

Certification and Request for Prioritized Examination (2 pages)