Safety Data Sheet GHS-Compliant

May be used to comply with OSHA's Hazard Communication Standard 29 CFR 1910.1200. Standard must be consulted for specific requirements.



REAGENT CHEMICAL & RESEARCH, INC. 115 US Hwy 202 Ringoes, NJ 08551

PRODUCT IDENTITY	
Hydrochloric Acid, 20° or 22° Baume	Safety Data Sheet Revision Date - June 1, 2015
Section 1 - Identification	
Product Name	CAS #
Hydrochloric Acid	7647-01-0
Synonym	Chemical Formula
Muriatic Acid	HCl
Chemical Name	Chemical Family
Hydrochloric Acid Solution	Inorganic Acid
Product Use	
Acidification, pH Adjustment	
Manufacturer/Supplier Name	Address
Reagent Chemical & Research, Inc.	115 US Hwy 202 Ringoes, NJ 08551
General Information	Country
1-908-284-2800	United States

Transportation Emergency Number

1-800-424-9300

Section 2 - Hazards Identification

GHS Classification:

Emergency Telephone

1-409-899-3400

HEALTH	PHYSICAL
Acute Toxicity, Oral - Hazard Category 1	Corrosive to Metals - Hazard Category 1
Serious Eye Damage - Hazard Category 1	
Skin Corrosion - Hazard Category 1	

CHEMTREC

Sensitisation, Respiratory - Hazard Category 1

Acute Toxicity, Inhalation - Hazard Category 1

GHS Label Elements:

SYMBOLS: corrosion, health hazard, aspiration toxicity







Signal Word: DANGER

Section 2 - Hazards Identification (continued)

	Label	
GHS		

Hazard Statements	Precautionary Statements
Causas assessed which houses to see domestic	De not buoothe mist/www.
Causes severe skin burns & eye damage	Do not breathe mist/vapors
Fatal if swallowed (oral)	Avoid skin contact
Fatal if inhaled (mist, vapor)	Keep container tightly closed
May cause allergic or asthmatic symptoms	Wear respiratory protection, protective gloves
or breathing difficulties if inhaled	and eye/face protection
May be fatal if swallowed & enters airway	Use only in a well-ventilated area
Causes serious eye damage	Store container tightly closed in cool/well
May be corrosive to metals	ventilated area
	Wash thoroughly after handling

Section 3 - Composition / Information on Ingredients

Component Description		Percent		CAS #	
Hydrogen Chloride		26.00 - 3	37.00	7647-0	1-0
Water	63.00 - 74.00		7732-18-5		
EXPOSURE LIMITS/REGULATORY	INFORMATION				
Substance	PEL	TLV	STEL	AWT	CEILING
Hydrogen Chloride	C-7 mg/m3	C-2 ppm	50 ppm	N/D	5 ppm
Water	N/D	N/D	N/D	N/D	N/D
N/D - Not Determined	C = C6	eiling Level			

If a known exposure occurs or is suspected, immediately initiate the recommended

Section 4 - First Aid Measures

General

procedures below. Simultaneously contact a physician, or the nearest Poison Control

Center. Inform the person contacted of the type and extent of exposure, describe the

victim's symptoms and follow the advice given. For additional information, call day or

night, Reagent Chemical (409) 899-3400 or Chemtrec (800) 424-9300.

Inhalation

Remove from contaminated atmosphere. If breathing has ceased, clear the victim's

airway and start mouth-to-mouth artificial respiration, which may be supplemented

by the use of a bag-mask respirator, or a manually-triggered, oxygen supply capable

of delivering 1 liter/second or more. If the victim is breathing, oxygen may be

administered from a demand-type or continuous-flow inhalator, preferably with a

physician's advice. Contact a physician immediately.

Section 4 - First Aid Measures (continued)

Eve Contact

Immediately flush the eyes with large quantities of running water for 15 minutes.

Hold the eyelids apart during the flushing to ensure rinsing of the entire surface of the eyes and lids with water. DO NOT attempt to neutralize with chemical agents.

Obtain medical attention as soon as possible. Oils or ointments should not be used.

Continue the flushing for an additional $15\ \text{minutes}$ if the physician is not available. Skin Contact

Immediately remove contaminated clothing under a safety shower. Flush all

affected areas with large amounts of water for 15 minutes. DO NOT attempt to

neutralize with chemical agents. Obtain medical advice.

Ingestion

DO NOT induce vomiting. Immediately give large quantities of water or milk, if

available. If vomiting does occur, give fluids again. Never give anything by mouth

to an unconscious person. Call a physician or the nearest Poison Control Center.

Medical Conditions Generally Aggravated by Exposure

Hydrogen Chloride will aggravate breathing disorders

Note to Physician

Attending Physician should treat exposed patients symptomatically

Section 5 - Fire Fighting Measures

Extinguishing Method

Not Applicable, use water to dilute spills and to flush them away from ignition sources.

Unusual Fire and Explosion Hazards

Non-flammable, but Hydrochloric Acid reacts with metals.

Special Firefighting Procedures

Non-flammable, but Hydrochloric Acid reacts with all metals, except gold and

platinum, with rapid evolution of Hydrogen which is flammable and explosive in air.

Firefighters exposed to Hydrochloric Acid vapors should wear Scott Air-Pak, or

equivalent. Hydrogen Chloride vapors are extremely irritating to the respiratory

tract and may cause breathing difficulty.

Section 6 - Accidental Release Measures

Steps to be Taken in Case Material is Released or Spilled

Spills or discharges into the environment involving large quantities of Hydrochloric

Acid should be controlled and cleaned-up according to a pre-determined, affirmative

written Spill Prevention and Control Program. For assistance in developing a SPCP

contact your nearest Reagent Sales Office. Refer to Section 15 for spill/release

reporting information.

Spills should be handled immediately by neutralization and dilution of the spilled

product by the use of Soda Ash (Sodium Carbonate), Lime (Calcium Hydroxide), or

Limestone (Calcium Carbonate) with large amounts of water. For an interior (inside

a closed space) spill be aware that the use of Soda Ash, Lime and Limestone will

evolve heat and carbon dioxide and that ample ventilation must be provided.

Section 6 - Accidental Release Measures (continued)

Waste Disposal

Under Federal RCRA, it is the responsibility of the user of products to determine,

at the time of disposal, whether the product falls under RCRA as a hazardous waste.

This is because product uses, transformations, mixtures, etc. may render the

resulting end-product hazardous.

Container Disposal

Containers should be cleaned of residual product before disposal. Empty containers

should be disposed of in accordance with all applicable laws and regulations.

Section 7 - Handling and Storage

Handling

Chemical goggles and full face shield must be worn at all times by personnel

exposed to or handling Hydrochloric Acid. The use of a NIOSH approved cartridge

respirator or a Scott Air-Pak should be used by all personnel exposed.

Storage

Store containers in a cool, dry location away from direct sunlight, sources of

intense heat, or where freezing may occur. Store material in acid-proof container.

Keep container tightly closed when not in use. Keep container away from incompatible

materials. All loading, unloading, and storage equipment must be inspected prior to

any transfer operations are initiated.

General Comments

Impervious clothing, gloves, footwear and head gear must be worn at all times

by personnel exposed to or handling Hydrochloric Acid.

Precautions to be Taken in Handling and Storage

Make sure all personnel involved in housekeeping and spill clean-up follow good

Industrial Hygiene practices and wear proper protective equipment.

Section 8 - Exposure Controls / Personal Protection

EXPOSURE LIMITS					
Substance	PEL	TLV	STEL	TWA	CEILING
Hydrogen Chloride	C-7 mg/m3	C-5 ppm	50 ppm	N/D	5 ppm
Water	N/D	N/D	N/D	N/D	N/D
N/D - No Data Availa	ole C:	= Ceiling Le	vel		
Respiratory Protection					
Maintain airborne contamin	nate levels belo	ow listed gu:	idelines. Use	with adequat	te
ventilation. Use a mechan	nical fan or ven	nt area to so	crubber. Use N	NIOSH approve	ed
respiratory protection if	exposure limits	s are exceede	ed.		
Ventilation Local Exhaust		Special			
If PEL exceeded		Vent fume	es to appropria	te scrubber	
Mechanical (General)		Other			

Wear neoprene rubber gloves to minimize skin contact. Additional protection may be

necessary to prevent skin contact including use of apron, face shield, boots or full

body protection. A safety shower should be located in the work area.

Eye Protection

Splash goggles or safety glasses. Face shields are recommended. Eye-wash stations

should be available where eye contact can occur.

Section 8 - Exposure Controls / Personal Protection (continued)

Use body protection appropriate for task. An apron or other impermeable body protection is suggested. Full body chemical protection is recommended for

emergency response procedures.

Section 9 - Physical and	Chemical Properties
--------------------------	---------------------

Boiling Point		Specific Gravity (H2O = 1)	
Boiling Point		Specific Gravity (H2O = 1)	
	230 F		1.13 - 1.19
Vapor Pressure (mm Hg)		Freezing Point	
	50 - 60 mm		12 F to -63 F
Vapor Density (AIR = 1)		Density	
	No Data Available		9.48 - 9.61
рН		Odor Threshold	
	< 1		0.25 - 10 ppm
Flash Point		Evaporation Rate	
	Not Flammable		No Data Available
Flammability		Flammability Limits	
	Not Flammable		Not Flammable
Auto Ignition Temperature		Partition Coefficient	
	Not Flammable		No Data Available
Viscosity (at 15 C)		Decomposition Temperature	
	2.3 mPa.s		No Data Available

Solubility in Water

miscible

Appearance and Odor

Clear/Slightly yellow with a sharp pungent odor

Section 10 - Stability and Reactivity

Stability	Unstable		Conditions to Avoid		
			Hydrochloric Acid is extremely reactive. Avoid contact with		
	Stable				
		X	metal surfaces and oxidizing agents.		

Incompatibility (Materials to Avoid) Hydrochloric Acid is chemically stable when properly contained and handled. It is a strong mineral acid and reacts with many metals and metal oxides and hydroxides to form the equivalent metal chloride. It reacts with zeolites and other silicious compounds to form Hydrosilicic Acid; it reacts with carbonates to form Carbon Dioxide and Water. It is oxidized by Oxygen or electrolysis to form Chlorine, a lethal, poisonous gas. It reacts with alkaline compounds to form a neutral salt. It is a hydrolyzing agent for carbohydrates, esters and other compounds. It's reaction with most metals will produce Hydrogen, an explosive gas. Violent reactions will result when Hydrochloric Acid Reacts with acetic anhydride, 2-aminoethanol, ammonium hydroxide, calcium phosphide, chlorosulfonic acid, ethylene diamine, ethylene imine, oleum (fuming sulfuric acid), perchloric acid, beta propiolactone, propylene oxide, sodium hydroxide, sulfuric acid, uranium This listing is not all-inclusive. phosphide and vinyl acetate. Hazardous Decomposition or By-products

Extreme heat may cause the product to decompose, producing toxic fumes which may

include chlorine compounds.

Hazardous	May Occur		Conditions to Avoid
Polymerization			Extreme heat and contact with incompatible materials
	Will Not Occur	Х	Page 5 of 9

Section 11 - Toxicol Route(s) of Entry:	Inhalation?	Skin?	Ingestion?
rtouto(o) or Linkly.	Yes	Yes	Yes
Health Hazards (Acute and	d Chronic)		
		d in a solution as Hyd	rochloric Acid, is a
corrosive substa	ince and can cause s	evere and painful burn	s on contact with any
part of the body	or if taken intern	ally. The mucous memb	ranes of the eyes and the
upper respirator	ry tract are especia	lly susceptible to the	irritating effects of high
atmospheric conc	entrations of Hydro	gen Chloride. The gas	or vapor is so
penetrating and	pungent that when h	igh concentrations do	occur, those exposed
	ely leave the contam		
Carcinogenicity:	NTP?	IARC Monographs?	OSHA Regulated?
	No Data Avai	lable No Data Avail	able No Data Availabl
Signs and Symptoms of Ex			L1
		cause severe burns at	the contact points
	ally Aggravated by Exposure	matitis and breathing	digardara
Toxicology	Inhalatio		aisolueis.
Hydrogen Chlorid		LCLo - 1300 ppm/30 mi	n
ilydrogen emforio	ic iramari	1300 ррш/30 ш1	11
	Rat L	C ₅₀ - 4701 ppm/30 min	
	Oral (ral		
		900 mg/kg	
		ic Effects	
	Inhal	ation: 100 ppm/24 hrs	(Chromosome damage)
	Oral:	100 ppm (Chromosom	e damage)
	Paren	tal: 20 mg (Cytogeni	c effects)

Section 12 - Ecological Information

Ecological Toxicity

Animals exposed to hydrochloric acid solution will experience tissue damage, burns and may be killed. Plants contaminated with hydrochloric acid solutions of low pH may be adversely effected or destroyed. High concentrations have been shown to be detrimental to aquatic life. A release into a body of water will kill fish and other aquatic life. Other Ecological Information Hydrochloric acid is stable and found naturally in the environment. All work practices

should be aimed at eliminating environmental contamination.

Chemical Fate Information

Hydrochloric acid is naturally occurring in the environment.

Other Regulatory Information

No other regulatory information is available on this product.

Section 13 - Disposal Considerations

As sold, this product, when discarded or disposed of, is a hazardous waste according to Federal regulations (40 CFR 261). It is listed as Hazardous Waste Number D002, listed due to its corrosivity. The transportation, treatment and disposal of this waste material must be conducted in compliance with 40 CFR 262, 263, 264, 268 and 270.

Disposal can occur only in properly permitted facilities. Refer to state and local statutes for any additional requirements, as they may differ from Federal laws.

Section 13 - Disposal Considerations (continued) Under Federal RCRA, it is the responsibility of the user of products to determine, at the time of disposal, whether the product falls under RCRA as a hazardous waste. This is because product uses, transformations, mixtures, etc. may render the resulting end-product hazardous. Container Disposal Containers should be cleaned of residual product before disposal. Empty containers should be disposed of in accordance with all applicable laws and regulations. **Section 14 - Transport Information** Regulated Material Hydrochloric Acid is defined as hazardous by the US DOT and Transport Canada North American Emergency Response Guide Book ID # 1789 Guide #157 2008 & 2012 Revision DOMESTIC SHIPPING INFORMATION Proper Shipping Name Hazard Classification Hydrochloric Acid Corrosive UN/NA Identification Hazard Class UN 1789 Class 8 **DOT Labels Required** Packaging Group Corrosive ΙI INTERNATIONAL SHIPPING INFORMATION Proper Shipping Name Hazard Classification Hydrochloric Acid Corrosive UN/NA Identification Hazard Class UN 1789 Class 8 Labels Required Packaging Group Corrosive ΤT Section 15 - Regulatory Information U.S. Federal Regulations Comprehensive Environmental Response and Liability Act of 1980 (CERCLA): Chemical Name: Hydrochloric Acid CAS # 7647-01-0 RQ - 5000 lbs Toxic Substances Control Act (TSCA): All components of this product are included on the TSCA inventory OSHA Hazard Communication Standard Classification: Corrosive as defined by the OSHA Hazard Communication Standard. Clean Water Act (CWA): Chemical Name: Hydrochloric Acid CAS # 7647-01-0 Listed as Hazardous No chemical components listed as Priority pollutants or Toxic pollutants Clean Air Act (CAA): Hydrochloric acid, CAS 7647-01-0, is listed as a hazardous air pollutant (HAP) US Environmental Protection Agency Risk Management Plan (RMP) Regulated: No, Hydrochloric acid solution under 37% is not regulated Superfund Amendments and Reauthorization Act (SARA) Title III Information:

Hydrochloric Acid CAS # 7647-01-0

Hydrochloric Acid

SARA Section 302:

SARA Section 313:

CAS # 7647-01-0

TPO 5000 lb EPCRA RO

Section 15 - Regulatory Information (continued)	
National Sanitation Foundation Limits (ANSI/NSF	Standard 60):
Maximum Drinking Water Use Concentrat	ion - 40 mg/l
Scale and Corrosion Control at Maximu	m 40 mg/l
State Regulations California Safe Drinking Water Act (Prop 65) Li	sting:
No ingredients listed in this section	
California Right to Know Act:	
Chemical Name: Hydrochloric Acid	CAS # 7647-01-0
New Jersey Right to Know Act:	
Chemical Name: Hydrochloric Acid	CAS # 7647-01-0
Chemical Name: Water	CAS # 7732-18-5
Massachusetts Right to Know Act Substance List	(MSL)::
Chemical Name: Hydrochloric Acid	CAS # 7647-01-0
Pennsylvania Right to Know Act Hazardous Substa	nce List:
Chemical Name: Water	CAS # 7732-18-5
Chemical Name: Hydrochloric Acid	CAS # 7647-01-0
International Regulations Canadian Domestic Substance List (DSL) Inventor	y Listing:
Chemical Name: Hydrochloric Acid	CAS # 7647-01-0
Canadian Ingredient Disclosure List	
Chemical Name: Hydrochloric Acid	CAS # 7647-01-0
Canadian Workplace Hazardous Materials Informat	ion System (WHMIS):
Class E: Corrosive material	
This product has been classified acco	rding to the hazard criteria of the CPR
and the MSDS contains all of the i	nformation required by the CPR
European Inventory of Existing Chemicals (EINEC	S):
Chemical Name: Hydrochloric Acid	EINECS # 2315957
EU Labeling in Accordance with EC Directives:	
Hazard Symbols: C	
EU Risk (R) and Safety (S) Phrases:	
R23/24/25: Toxic by inhalation, in co	ntact with skin and if swallowed
R37/38: Irritating to respiratory sys	
R41: Risk of serious damage to eyes	
S36/37: Wear suitable protective clot	hing and gloves
	el unwell, seek medical advice immediately
S53: Avoid exposure - obtain special	
S61: Avoid release to the environment	
DOI: WADIR LETERBE CO CHE EHALLOHMENC	. Note: to bately data silect

Section 15 - Regulatory Information (continued)

Japanese Minister of International Trade and Industry (MITI) Inventory Listing:

Chemical Name: Hydrochloric Acid SECTION STRUCTURE # 1-324

Australian Inventory of Chemical Substances (AICS) Listing:

Chemical Name: Hydrochloric Acid CAS # 7647-01-0

US Census Bureau - Foreign Trade Identification

Chemical Name: Hydrochloric Acid HTS & Schedule B # 2806.10.0000

Section 16 - Other Information

Created By	MSDS Revision Date
Product Safety - 6/1/98	June 1, 2015
MSDS Revision Number	Revision Indicator
Revision # 010	Hazard Statement Alignment
MSDS Contact	·

Robert Dritschel 908-284-2800

Does Product Contain, or is Manufactured with, CFC's?

No

National Fire Protection Association (NFPA) Ratings:

Health - 3 Flammability - 0 Instability - 0 Other Hazard Information - ACID Hazardous Material Identification System (HMIS):

Health - 3 Flammability - 0 Physical Hazard - 0 Protective Equipment - X

North American Emergency Response Guide Book

ID # 1789 Guide #157 2008 & 2012 Revision

Disclaimer of Liability

The data contained herein is furnished gratuitously and independent of any sale It is supplied only for your investigation and possible independent verification. While the data is believed to be correct, Reagent Chemical and Research Inc. makes no representation as to the accuracy of any data contained herein. In no event shall Reagent Chemical and Research, Inc. be responsible for any damages of any nature whatsoever directly or indirectly resulting from the publication, use or reliance upon any data contained herein. Data Sheets are available for other Reagent Chemical and Research, Inc. products. You are urged to obtain data sheets for all products you buy, process, use or distribute and you are encouraged to advise anyone working with or exposed to such products of the information contained in the applicable data sheets. The data in this document is provided without any representation or warranty, express or implied, regarding its accuracy or correctness. No warranty either expressed or implied of merchant ability or fitness or of any nature is made with respect to any product referred to herein. Reagent Chemical and Research Inc. does not assume responsibility and expressly disclaims liability for loss, damage or expense arising out of or in any way connected with the handling, storage, use or disposal of the products referred herein.