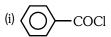
AIPMT 2007

1. Consider the following compounds:



(ii)
$$O_2N$$
—COC

The correct order of reactivity towards hydrolysis is:-

- (1) (i) > (ii) > (iii) > (iv)
- (2) (iv) > (ii) > (i) > (iii)
- (3) (ii) > (iv) > (i) > (iii)
- (4) (ii) > (iv) > (iii) > (i)
- $\begin{tabular}{lll} \bf 2. & Which one of the following on treatment with 50% aq. NaOH yields the corresponding alcohol and 100%
 - acid
 - $(1) C_6 H_5 CHO$
 - (2) CH₃CH₂CH₂CHO
 - (3) CH₃COCH₃
 - (4) CH₃CHO
- **3.** The product formed in aldol reaction is :-
 - (1) a β -hydroxy aldehyde or ketone
 - (2) an α -hydroxy aldehyde or ketone
 - (3) an α , β -unsaturated ester
 - (4) a β-hydroxy acid

AIPMT 2008

4. Acetophenone when reacted with a base, C_2H_5ONa , yields a stable compound which has the structure :-

$$(1) \bigcirc \begin{matrix} CH_3 & CH_3 \\ I & I \\ C-C \\ OH & OH \end{matrix} \bigcirc$$

$$(2) \bigcirc \bigcap_{OH} \bigcap_{OH} \bigcap_{OH} \bigcap$$

(3)
$$\bigcirc C = CH - C$$

$$CH_3 \qquad O$$

- **5.** A strong base can abstract an α hydrogen from
 - (1) Ketone
- (2) Alkane
- (3) Alkene
- (4) Amine

AIPMT 2009

6. H₂COH.CH₂OH on heating with periodic acid gives:-

(1)
$$2 + C = O$$

- (2) 2CO₂
- (3) 2HCOOH
- (4) CHO
- **7.** Consider the following reaction,

ethanol
$$\xrightarrow{PBr_3}$$
 $X \xrightarrow{alc. KOH}$ Y

$$\xrightarrow{\text{(i) } H_2SO_4 \text{ room temperature}} Z ;$$

the product Z is :-

- (1) CH₃CH₂OH
- (2) $CH_{2} = CH_{2}$
- (3) CH₂CH₂ O CH₂ CH₃
- $(4) CH_3 CH_9 O SO_3H$

8. Propionic acid with Br_2/P yields a dibromo product. Its structure would be :-

(4)
$$CH_2Br - CH_2 - COBr$$

9. Consider the following reaction :

Phenol
$$\xrightarrow{Zn \text{ dust}} X \xrightarrow{CH_3Cl} Y \xrightarrow{KMnO_4} Z$$

$$\xrightarrow{AlCl_3} Y \xrightarrow{KMnO_4} Z$$

the product Z is :-

- (1) Benzene
- (2) Toluene
- (3) Benzaldehyde
- (4) Benzoic acid

 $\begin{tabular}{ll} \textbf{10.} & In a set of reactions, ethyl benzene yielded a \\ & product D \end{tabular}$

$$\underbrace{ CH_2CH_3}_{KOH} \underbrace{KMO_4}_{FeCl_3} \underbrace{Br_2}_{FeCl_3} \underbrace{CC_2H_5OH}_{H^+} D$$

'D' would be :-

(3)
$$Br$$
 $CH_2COOC_2H_5$

AIPMT 2010

- **11.** Which of the following reactions will not result in the formation of carbon-carbon bonds?
 - (1) Friedel-Crafts acylation
 - (2) Reimer-Tieman reaction
 - (3) Cannizaro reaction
 - (4) Wurtz reaction
- **12.** When glycerol is treated with excess of HI, it produces:-
 - (1) allyl iodide
 - (2) propene
 - (3) glyceryl triiodide
 - (4) 2-iodopropane
- **13.** Match the compounds given in List-I with their characteristic reactions given in List-II. Select the correct option.

List-I (Compounds)

- (a) CH₃CH₂CH₂CH₃NH₃
- (b) CH₃C≡CH
- (c) CH₃CH₃COOCH₃
- (d) CH₃CH(OH)CH₃

List-II (Reactions)

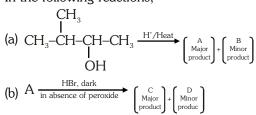
- (i) Alkaline hydrolysis
- (ii) With KOH (alcohol) and CHCl₃ produces bad smell
- (iii) Gives white ppt. with ammonical AgNO₃
- (iv) With Lucas reagent cloudiness appears after 5 minutes

Options:

- (a) (b) (c) (d)
- (1) (iii) (ii) (i) (iv)
- (2) (ii) (iii) (i) (iv)
- (3) (iv) (ii) (iii) (i)
- (4) (ii) (i) (iv) (iii)

AIPMT Pre. 2011

14. In the following reactions,



the major products (A) and (C) are respectively :-

the major products (A) and (C) are respectively :-
$$\begin{array}{cccc} CH_3 & CH_3 \\ & & | \\ (1) CH_2 = C - CH_2 - CH_3 & and CH_2 - CH - CH_2 - CH_3 \\ & & | \\ Br & & \\ \end{array}$$

$$\begin{array}{cccc} CH_3 & CH_3 \\ \mid & \mid \\ (4) \ CH_2 = C - CH_2 - CH_3 & \text{and} \ CH_3 - C - CH_2 - CH_3 \\ \mid & \mid \\ Br \end{array}$$

15. In a set of reactions m-bromobenzoic acid gave a product D. Identify the product D

COOH
$$A \xrightarrow{SOCl_2} B \xrightarrow{NH_3} C \xrightarrow{NaOH} D$$

SO₂NH₂

$$(2) \quad Br$$

$$COOH$$

$$CONH$$

(3)
$$\underset{\text{Br}}{\bigcap}$$
 $\underset{\text{NH}_2}{\bigcap}$ $\underset{\text{(4)}}{\bigcap}$ $\underset{\text{Br}}{\bigcap}$ $\underset{\text{CONH}_2}{\bigcap}$

- **16.** Clemmensen reduction of a ketone is carried out in the presence of which of the following?
 - (1) Glycol with KOH
 - (2) Zn-Hg with HCl
 - (3) LiAlH₄
 - (4) H₂ and Pt as catalyst

AIPMT Mains 2011

- 17. An organic compound 'A' on treatment with NH_3 gives 'B' which on heating gives 'C'. 'C' when treated with Br_2 in the presence of KOH produces ethylamine. Compound 'A' is :-
 - (1) CH₃CH₂COOH
 - (2) CH₃COOH
 - (3) CH₃CH₂CH₂COOH
 - (4) CH₃-CHCOOH | CH₃
- **18.** Match the compounds given in List-I with List-II and select the suitable option using the code given below.

List-II List-II

- (a) Benzaldehyde (i) Phenolphthalein
- (b) Phthalic anhydride (ii) Benzoin condensation
- (c) Phenyl benzoate (iii) Oil of wintergreen
- (d) Methyl salicylate (iv) Fries rearrangement

Code:

(a)	(b)	(c)	(d)		
(1) (ii)	(i)	(iv)	(iii)		
(2) (iv)	(i)	(iii)	(ii)		
(3) (iv)	(ii)	(iii)	(i)		
(4) (ii)	(iii)	(iv)	(i)		

AIPMT Pre. 2012

19. In the following sequence of reactions

$$CH_3\text{--Br} \xrightarrow{KCN} A \xrightarrow{H_3O^+} B \xrightarrow{LiAlH_4} C,$$

the end product (C) is:

- (1) Acetaldehyde
- (2) Ethyl alcohol
- (3) Acetone
- (4) Methane

20. Predict the products in the given reaction.

(1)
$$\bigcirc$$
 CH₂OH \bigcirc COO- K⁶

(3)
$$CH_2OH$$
 $CH_2COO^- K^{\oplus}$

- **21.** CH_3CHO and $C_6H_5CH_2CHO$ can be distinguished chemically by :
 - (1) Tollen's reagent test
 - (2) Fehling solution test
 - (3) Benedict test
 - (4) Iodoform test

AIPMT Mains 2012

22. Consider the following reaction:

The product 'A' is:

- $(1) C_6 H_5 COCH_3$
- $(2) C_6H_5Cl$
- $(3) C_6 H_5 CHO$
- $(4) C_6 H_5 OH$

NEET UG 2013

23. Reaction by which Benzaldehyde cannot be prepared:-

COOH (1) + Zn/Hg and conc. HCl

(2) + $\operatorname{CrO_2Cl_2}$ in $\operatorname{CS_2}$ followed by $\operatorname{H_3O^{\oplus}}$

(3) + H_2 in presence of Pd+BaSO₄

+ CO+HCl in presence of anhydrous AlCl₃

AIPMT 2014

- **24.** Among the following sets of reactants which one produces anisole?
 - (1) CH₃CHO; RMgX
 - (2) C_6H_5OH ; NaOH; CH_3I
 - (3) C₆H₅OH; neutral FeCl₃
 - (4) $C_6H_5 CH_3$; CH_3COCI ; $AlCl_3$
- **25.** Which of the following will not be soluble in sodium hydrogen carbonate?
 - (1) 2, 4, 6-trinitrophenol
 - (2) Benzoic acid
 - (3) o-Nitrophenol
 - (4) Benzenesulphonic acid

AIPMT 2015

- **26.** An organic compound 'X' having molecular formula $C_5H_{10}O$ yields phenyl hydrazone and gives negative response to the Iodoform test and Tollen's test. It produces n-pentane on reduction. 'X' could be :-
 - (1) 2-pentanone (2) 3-pentanone
 - (3) n-amyl alcohol (4) pentanal

RE-AIPMT 2015

- **27.** Reaction of phenol with chloroform in presence of dilute sodium hydroxide finally introduces which one of the following functional group?
 - (1) -CHCl₂
- (2) -CHO
- (3) -CH₂Cl
- (4) -COOH

NEET-II 2016

28. The **correct** structure of the product A formed in the reaction

$$\begin{array}{c} O \\ \hline \\ H_2(gas, \ 1 \ atmpsphere) \\ \hline Pd/carbon, \ ethanol \\ \end{array} \begin{array}{c} A \quad is :- \end{array}$$

- (1) OH
- (2) OH
- (3) OH
- (4)

NEET(UG) 2017

- **29.** The heating of phenyl–methyl ethers with HI produces
 - (1) iodobenzene
- (2) phenol
- (3) benzene
- (4) ethyl chlorides

30. Identify A and predict the type of reaction

$$\begin{array}{c}
OCH_3 \\
\hline
NaNH_2
\end{array}
A$$

- $\begin{array}{cccc} & & & & \\ & & & \\ \text{OCH}_3 & & \\ & & \text{NH}_2 & \\ & & \\ & & & \\ & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$
- OCH_3 Br and cine substitution reaction
- OCH_3 and cine substituion reaction
- $\begin{array}{c} \text{OCH}_3\\ \text{(4)} & \\ \text{NH}_2 \end{array} \text{ and substitution reaction}$
- **31.** Of the following, which is the product formed when cyclohexanone undergoes aldol condensation followed by heating?

32. Consider the reactions :-

$$\begin{array}{c} X \xrightarrow{Cu/573K} A \xrightarrow{[Ag(NH_3)_2] OH/\Delta} & \text{Silver mirror} \\ (C_2H_6O) & & O \\ \hline OH/\Delta & Y \\ \hline Z & NH_2-NH-C-NH_2 \end{array}$$

Identify A, X, Y and Z

- (1) A-Methoxymethane, X-Ethanol, Y-Ethanoic acid, Z-Semicarbazide.
- (2) A-Ethanal, X-Ethanol,
 Y-But-2-enal, Z-Semicarbazone
- (3) A-Ethanol, X-Acetaldehyde, Y-Butanone, Z-Hydrazone
- (4) A-Methoxymethane, X-Ethanoic acid, Y-Acetate ion, Z-hydrazine

NEET(UG) 2018

- **33.** Carboxylic acid have higher boiling points than aldehydes, ketones and even alcohols of comparable molecular mass. It is due to their
 - (1) formation of intramolecular H-bonding
 - (2) formation of carboxylate ion
 - (3) more extensive association of carboxylic acid via van der Waals force of attraction
 - (4) formation of intermolecular H-bonding.
- **34.** Compound A, $C_8H_{10}O$, is found to react with NaOI (produced by reacting Y with NaOH) and yields a yellow precipitate with characteristic smell.

A and Y are respectively

(1)
$$H_3C$$
 CH_2 -OH and I_2

(2)
$$\sim$$
 CH₂-CH₂-OH and I₂

(3)
$$CH$$
- CH_3 and I_2

(4)
$$CH_3$$
 OH and I_2

- **35.** The compound A on treatment with Na gives B, and with PCl₅ gives C. B and C react together to give diethyl ether. A, B and C are in the order
 - (1) C_2H_5OH , C_2H_6 , C_2H_5Cl
 - (2) C_2H_5OH , C_2H_5Cl , C_2H_5ONa
 - (3) C_2H_5Cl , C_2H_6 , C_2H_5OH

CH₂CH₂CH₃

- (4) C_2H_5OH , C_2H_5ONa , C_2H_5Cl
- **36.** Identify the major products P, Q and R in the following sequence of reaction:

(1)
$$\bigcap$$
 , \bigcap ,

Q

R

NEET(UG) 2019

37. The structure of intermediate A in the following reaction is :-

$$CH$$
 CH_3
 O_2
 O_2
 O_3
 O_4
 O_4
 O_3
 O_4
 O_4
 O_4
 O_5
 O_7
 O_8
 O_8

$$(1) \begin{tabular}{c} CH_3\\ CH\\ CH_3\\ CH_3\\ \end{tabular}$$

NEET(UG) 2019 (ODISHA)

38. The reaction that **does not** give benzoic acid as the major product is :-

(1)
$$CH_2OH - K_2Cr_2O_7$$

(2)
$$COCH_3$$
 $(i) NaOCl$ $(ii) H_3O^+$

$$(4) \bigcirc CH_2OH \longrightarrow KMnO_4/H^+ \longrightarrow$$

CA0219

- **39.** When vapours of a secondary alcohol is passed over heated copper at 573 K, the product formed is:-
 - (1) a carboxylic acid (2) an aldehyde
 - (3) a ketone (4) an alkene

CC0220

40. The major products C and D formed in the following reactions respectively are:-

$$H_3C-CH_2-CH_2-O-C(CH_3)_3 \xrightarrow{excess HI} C+D$$

- (1) $H_3C-CH_2-CH_2-I$ and $I-C(CH_3)_3$
- (2) H₃C-CH₂-CH₂-OH and I-C(CH₃)₃
- (3) H₃C-CH₂-CH₂-I and HO-C(CH₃)₃
- (4) H₃C-CH₂-CH₂-OH and HO-C(CH₃)₃

NEET(UG) 2020

- **41.** Reaction between benzaldehyde and acetophenone in presence of dilute NaOH is known as:
 - (1) Cross Aldol condensation
 - (2) Aldol condensation
 - (3) Cannizzaro's reaction
 - (4) Cross Cannizzaro's reaction

NEET(UG) 2020 (COVID-19)

42. Which of the following acid will form an (a) Anhydride on heating and (b) Acid imide on strong heating with ammonia?

43. Identify compound (A) in the following reaction :

$$A \xrightarrow{H_2/Pd/BaSO_4} CHC$$

- (1) Benzoyl chloride
- (2) Toluene
- (3) Acetophenone
- (4) Benzoic acid

NEET(UG) 2021

44. What is the IUPAC name of the organic compound formed in the following chemical reaction?

Acetone
$$\xrightarrow{\text{(i) } C_2H_5MgBr, dry Ether}$$
 $\xrightarrow{\text{(ii) } H_2O, H^+}$ Product

- (1) 2-methyl propan-2-ol
- (2) pentan-2-ol
- (3) pentan-3-ol
- (4) 2-methyl butan-2-ol

45. The product formed in the following chemical reaction is

$$CH_2-C-OCH_3 \xrightarrow{NaBH_4} ?$$

$$(1) \begin{array}{c} OH & H \\ \downarrow \\ CH_2-C-OCH_3 \\ \downarrow \\ CH_3 \end{array}$$

(2)
$$CH_2$$
- CH_2 - OH

$$(3) \begin{array}{c} OH & H \\ CH_2-C-CH_3 \\ CH_3 \end{array}$$

$$(4) \begin{array}{c} OH & OH \\ CH_2-C-OCH_3 \\ CH_3 \end{array}$$

46. Match List-I with List-II.

List-I

(i) Hell-Volhard-

List-II

Zelilnsky reaction

- (b) O \parallel $R-C-CH_3+NaOX\longrightarrow$
- Gattermann-Koch
 Reaction
- (c) R-CH₂-OH + R'COOH
- (iii) Haloform

reaction

(d) R-CH₂-COOH

(iv) Esterification

$$\xrightarrow{\text{(i) } X_2/\text{RedP}}$$

$$\xrightarrow{\text{(ii) } H_2\text{O}}$$

Choose the **correct** answer from the options given below.

- (1) (a)-(iv), (b)-(i), (c)-(ii), (d)-(iii)
- (2) (a)-(iii), (b)-(ii), (c)-(i), (d)-(iv)
- (3) (a)-(i), (b)-(iv), (c)-(iii), (d)-(ii)
- (4) (a)-(ii), (b)-(iii), (c)-(iv), (d)-(i)

47. The intermediate compound 'X' in the following chemical reaction is :

$$CH_{3} + CrO_{2}Cl_{2} \xrightarrow{CS_{2}} X \xrightarrow{H_{3}O^{+}} CH$$

$$(1) \qquad CH(OCrOHCl_{2})_{2}$$

$$(2) \qquad CH(Cl)$$

$$(3) \qquad CH \stackrel{Cl}{Cl}$$

$$(4) \qquad CH \stackrel{Cl}{H}$$

Que.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Ans.	3	1	1	3	1	1	1	1	4	1	3	4	2	2	3
Que.	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30
Ans.	2	1	1	2	1	4	3	1	2	3	2	2	4	2	4
Que.	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45
Ans.	1	2	4	3	4	4	2	3	3	1	1	1	1	4	4
Que.	46	47													
Ans.	4	1													