AIPMT 2009

1. Trichloroacetaldehyde, CCl₃CHO reacts with chlorobenzene in presence of sulphuric acid and produces:-

(2)
$$CI \longrightarrow CI$$
 $CI \longrightarrow CI$
 CH_2CI

$$(4) Cl \longrightarrow Cl \longrightarrow Cl \longrightarrow Cl$$

AIPMT 2010

2. In the following reaction

$$C_6H_5CH_2Br \xrightarrow{\hspace*{1cm} 1. \hspace*{0.1cm} Mg, Ether} X,$$

The product 'X' is :-

- (1) C₆H₅CH₉OH
- (2) $C_6H_5CH_3$
- $(3) C_6H_5CH_9CH_9C_6H_5$
- (4) C₆H₅CH₉OCH₉C₆H₅

- **3.** Following compounds are given :
 - (a) CH₂CH₂OH

(b) CH₃COCH₃

(d) CH₃OH

Which of the above compound(s), on being warmed with iodine solution and NaOH, will give iodoform?

- (1) Only (a)
- (2) (a), (b) and (c)
- (3) (a) and (b)
- (4) (a), (c) and (d)

AIPMT Mains 2012

- **4.** Which of the following compounds will give a yellow precipitate with iodine and alkali?
 - (1) Acetamide
- (2) Propan-1-ol
- (3) Acetophenone
- (4) Methyl acetate

NEET(UG) 2019 (ODISHA)

5. The hydrolysis reaction that takes place at the slowest rate, among the following is:-

- (2) $H_3C-CH_2-CI \xrightarrow{\text{aq. NaOH}} H_3C-CH_2-OH$
- (3) $H_2C=CH-CH_2Cl\xrightarrow{aq.} N_aOH \rightarrow H_2C=CH-CH_2OH$

(4)
$$\sim$$
 CH₂Cl $\stackrel{\text{aq. NaOH}}{\longrightarrow}$ CH₂OH

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- **6.** The correct sequence of bond enthalpy of 'C-X' bond is
 - (1) $CH_3-F < CH_3-Cl < CH_3-Br < CH_3-I$
 - (2) $CH_3-F > CH_3-Cl > CH_3-Br > CH_3-I$
 - (3) $CH_3-F < CH_3-Cl > CH_3-Br > CH_3-I$
 - (4) $CH_3-Cl > CH_3-F > CH_3-Br > CH_3-I$

Que.	1	2	3	4	5	6
Ans.	1	2	2	3	1	2