



Modelling of LH2 transfer operations with engineering tools

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Outline

- Introduction to DISCHA engineering tool
- Tool features
- Recent Validation
 - Emptying of an LH2 tank due to boil-off
 - Self pressurization of LH2 tanks
- Tank to tank transfer simulations
 - Filling of an LH2 truck from a stationary tank at 2 and 5 bars

Introduction



- DISCHA tool for
 - Physical properties at single phase and two-phase conditions
 - Discharge calculations
 - Tank to tank transfer calculations
- DISCHA development / validation in previous EC projects
 - NET-Tools
 - [NET-Tools_Venetsanos](#)
 - PRESLHY
 - [PRESLHY_Venetsanos.mp4](#)
 - [PRESLHY_Venetsanos.pdf](#)
 - HyTunnel-CS
 - [HyTunnel-CS_D4.4.pdf](#)



DISCHA features



- GUI (Python), Main code (Fortran)

Physical Properties and Discharge Tool

File Physical Properties Tanks Discharge Lines Steady Release Transient Release Ambient conditions Options

Stagnation physical properties

Substance = Normal H2

Select state definition mode

- Pressure, Temperature, vapor quality
- Pressure, Enthalpy
- Pressure, Entropy
- Pressure, Density
- Density, Enthalpy
- Density, Entropy
- Density, Temperature
- T, P, x from file
- P, G, h_tot

Pressure (MPa)	20.0	Saturation temperature (K)	0.000000
Temperature (K)	298.15	Saturation pressure (MPa)	0.000000
Vapor quality (-)	1.0	Spinodal temperature (K)	0.000000
Density (kg/m3)	14.482045816958072	Vaporization enthalpy (kJ/kg)	0.000000
Enthalpy (kJ/kg)	102.03636692038039	Compressibility factor (-)	1.123107
Entropy (kJ/kg/K)	-21.94833617669984	Volumetric thermal Expansion coefficient (1/K)	0.003066
Total Enthalpy (kJ/kg)	102.036367	Internal energy (kJ/kg)	-1278.983984
Mass flux (kg/m2/s)	0.000000	Specific heat under const pressure (kJ/kg/K)	14.710106
Void (-)	1.0	Specific heat under const volume (kJ/kg/K)	10.340361
		Sound speed (m/s)	1.489242e+03
		Joule-Thomson coefficient (K/MPa)	-4.029625e-01
		Velocity (m/s)	0.000000

Set saturation temperature Compute properties: Reset State Figures

Set saturation pressure

State Charts

4 In

DISCHA features

- Substances
 - Normal H₂, Para-H₂, CH₄, CO₂, H₂O, NH₃, O₂, N₂, He
- EoS
 - Helmholtz free energy, SRK, PR, RKMC, Abel-Noble, Ideal gas
- Two-phase modelling
 - Ideal mixture of liquid and vapor phase
 - HEM
 - HRM, DEM mainly for H₂O
- Various input modes for thermodynamic state definition
 - Pressure + temperature + vapor quality
 - Pressure + (enthalpy or entropy or density or internal energy)
 - Density + (internal energy or entropy)
 - ...



DISCHA features



- Discharge / tank to tank transfer calculations
 - Arbitrary network of tanks connected by transfer lines.
 - A transfer line either connects 2 tanks or connects a tank with the ambient environment
 - No direct connection between different lines (branches).



DISCHA features

- Tank modelling
 - Single zone
 - Liquid and vapor phases share the same (sat) temperature
 - Transient mass and energy equations for the entire tank volume.
 - Multizone
 - Two distinct volumes (liquid below, vapor above) separated by one interface
 - Liquid phase subcooled or saturated
 - Vapor phase superheated or saturated
 - Liquid-Vapor Interface saturated
 - Various models for evaporation/condensation through the interface
 - Transient mass and energy equations for the vapor phase
 - Transient mass and energy equations for the liquid phase
 - Wall heat transfer
 - Time-dependent energy equation within tank walls.
 - Different material layers within tank walls

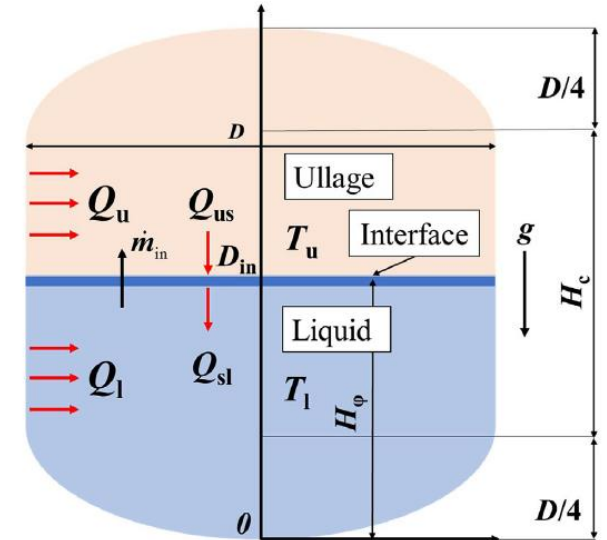


Tank conservation equations



Single zone

$$V \frac{d\rho}{dt} = \dot{m}_{in} - \dot{m}_{out} \quad V \frac{d\rho u}{dt} = \dot{m}_{in} h_{tot,in} - \dot{m}_{out} h_{tot,out} + \dot{q}A$$



Multizone zone (Wang et al. 2022)

$$\frac{d\rho_L V_L}{dt} = \dot{m}_{L,in} - \dot{m}_{L,out} - \dot{m}_{LV} \quad \frac{d\rho_L V_L u_L}{dt} = \left[\dot{m}_{in} h_{tot,in} - \dot{m}_{out} h_{tot,out} + \dot{q}A \right]_L + \dot{q}_{SL} A_S - \dot{m}_{LV} h_{sat,L}$$

$$\frac{d\rho_V V_V}{dt} = \dot{m}_{V,in} - \dot{m}_{V,out} + \dot{m}_{LV} \quad \frac{d\rho_V V_V u_V}{dt} = \left[\dot{m}_{in} h_{tot,in} - \dot{m}_{out} h_{tot,out} + \dot{q}A \right]_V - \dot{q}_{VS} A_S + \dot{m}_{LV} h_{sat,V}$$

$$V_L + V_V = V \quad \text{Volume constraint}$$

Wang H.R. et al., Modeling and thermodynamic analysis of thermal performance in self-pressurized liquid hydrogen tanks, IJHE, 47 (2022)



Evaporation/Condensation model

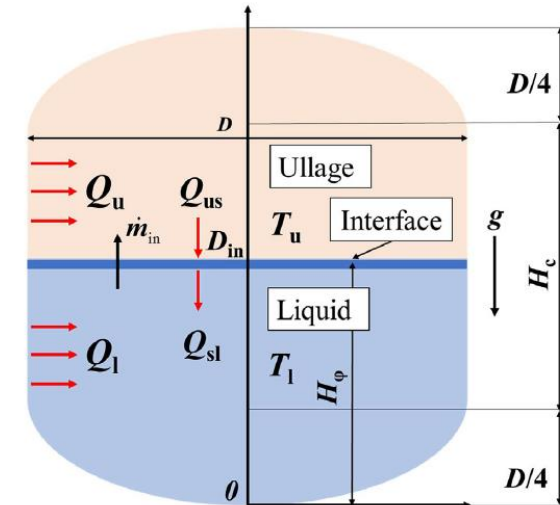
$$\dot{q}_{VS} - \dot{q}_{SL} = \dot{m}_{LV} (h_{sat,V} - h_{sat,L})$$

$$\dot{q}_{VS} = K_V a_{VS} (T_V - T_{Sat}) \quad \dot{q}_{SL} = K_L a_{SL} (T_{Sat} - T_L)$$

$$Nu_{VS} \equiv \frac{a_{VS} D}{\lambda_{VS}} = 0.27 Ra_{VS}^{1/4} \quad \text{Mc Adams}$$

$$Nu_{SL} \equiv \frac{a_{SL} D}{\lambda_{SL}} = 2.5 \left\{ \ln \left\{ 1.0 + \frac{2.5}{0.527 Ra_{SL}^{0.2}} \left(1.0 + \left(\frac{1.9}{Pr_{SL}} \right)^{0.9} \right)^{2/9} \right\} \right\}^{-1} \quad \text{Nellis \& Klein}$$

$$K_V = K_L = 0.1 \quad \text{Account for non-equilibrium}$$



Wang H.R. et al., Modeling and thermodynamic analysis of thermal performance in self-pressurized liquid hydrogen tanks, IJHE, 47 (2022)



DISCHA features

- Discharge/transfer line modelling
 - Conservation equations
 - Steady state momentum and energy balance
 - Line resistance, area change, wall heat transfer both for single phase and two-phase conditions
 - Transient internal energy equation within pipe walls.
 - Choked flow
 - Calculated using general Possible-Impossible-Flow (PIF) algorithm
 - Discretization along discharge line is necessary (refine grid near pipe exit !!!)
 - Mach = 1 at exit is an output result not a BC
 - Fictitious nozzle
 - 7 available models



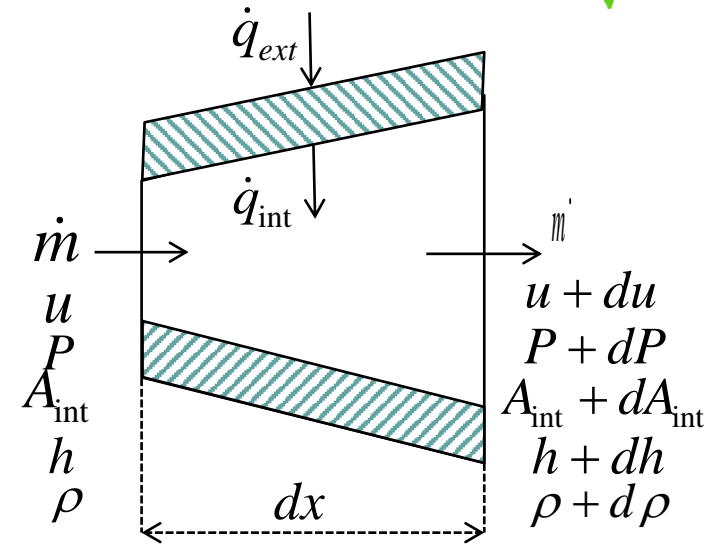
Line modeling

Pipe steady state equations:

$$\rho u A_{\text{int}} = \dot{m} = ct$$

$$\rho u du = -dP - \frac{\rho u^2 dx}{2} \left(\frac{f_D}{D_{\text{int}}} + \frac{K}{L} \right)$$

$$\rho u d \left(h + \frac{u^2}{2} \right) = \dot{q}_{\text{int}} \frac{4dx}{D_{\text{int}}}$$



Friction

Extra resistances (fittings)

Wall heat transfer
Omitted in this work

DISCHA validation

Emptying of an LH2 tank due to boil-off

Emptying of an LH2 tank due to boil-off



■ Experiments

- From Lawrence Livermore National Laboratories, see also Machalek et al., ICHS-9, 2021.

■ System

— Tank

- Vertical cylindrical tank 2 m diameter, 3.97 m height, 12.47 m³ volume.
- 11.1 mm inner steel + 50.8 mm MLI vacuum + 8.3 mm outer steel walls
- 80% initial LH2 fill

— Line

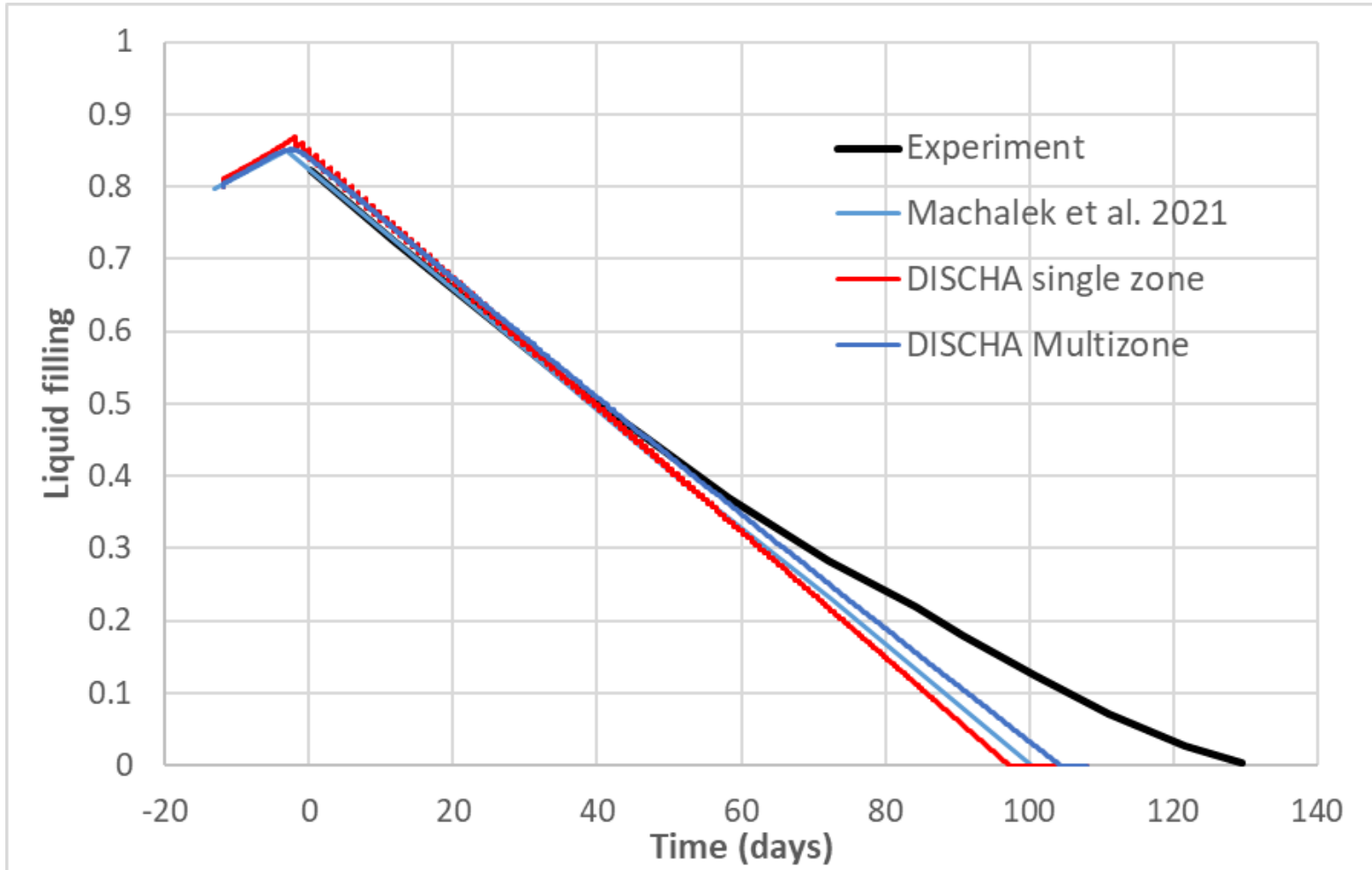
- 5 mm PRV, opens at 3.1 bar and closes at 2.9 bar

Material	Inner steel	MLI	Outer steel
Conductivity	3	2e-4	15
Specific heat	25	0.1	450
Density	8050	0.1	8050

Machalek et al., Influence of non-equilibrium conditions on liquid hydrogen storage tank behavior, 9th Int. Conf. on Hydrogen Safety, 21-23 Sept. 2021, Edinburgh, UK



Emptying of an LH2 tank due to boil-off



DISCHA validation

Self pressurization of LH2 tanks

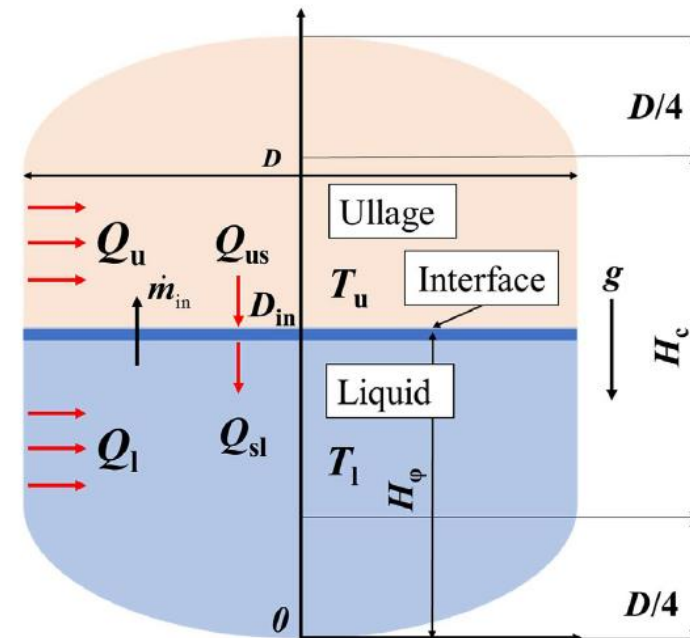
Self-pressurization of LH2 tanks

■ Experiments

- NASA multipurpose hydrogen test bed (MHTB) tank experiments see Hastings et al., (2003) and related modeling by Wang et al. (2022).

■ System

- Tank volume 18.09 m³,
- tank diameter $D = 3.05$ m
- Cylindrical height $H_c = 1.525$ m
- No venting



Hastings et al., Spray bar zero-gravity vent system for on orbit liquid hydrogen storage. NASA TM-12926 (2003)

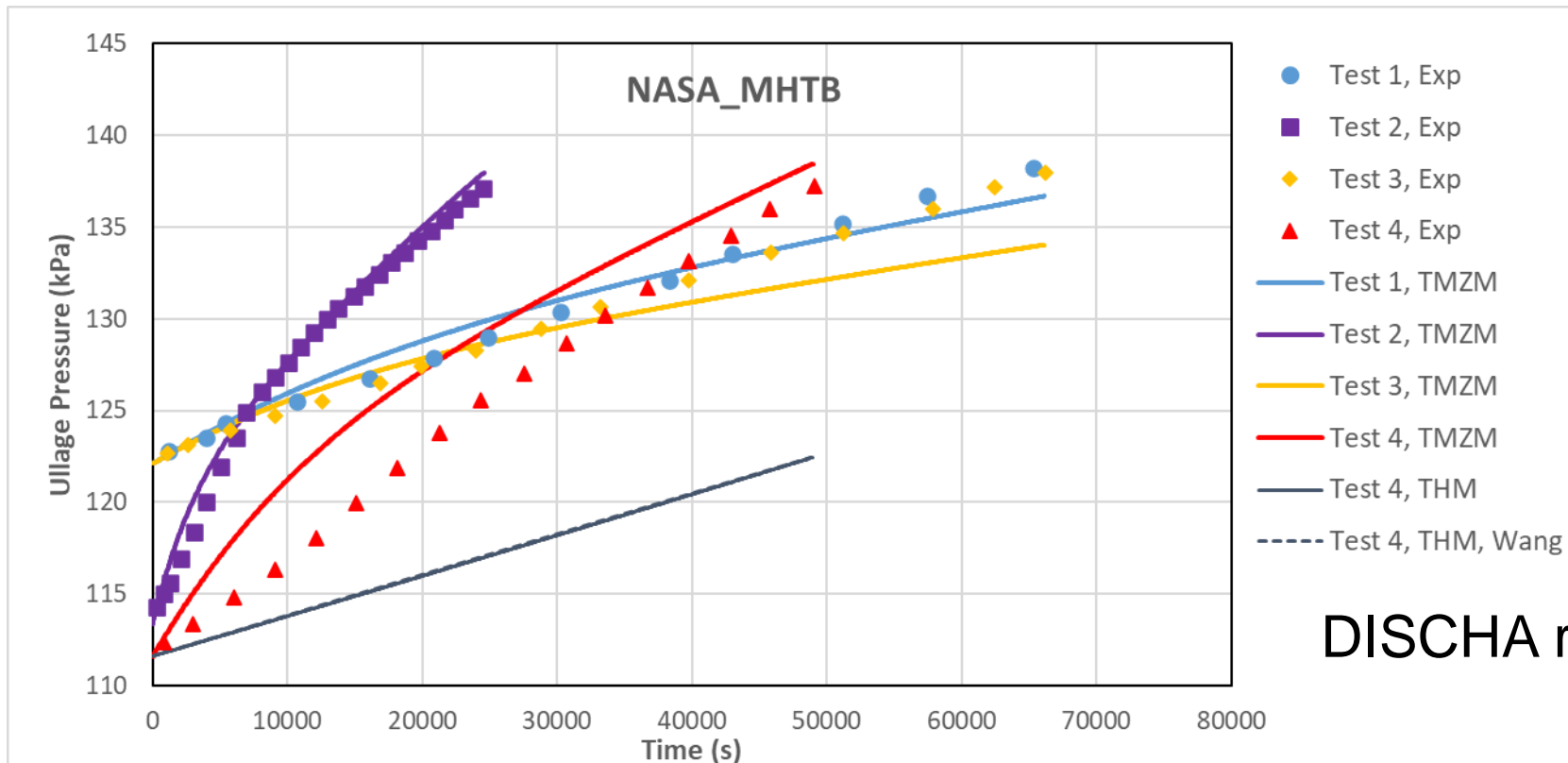
Wang H.R. et al., Modeling and thermodynamic analysis of thermal performance in self-pressurized liquid hydrogen tanks, IJHE, 47 (2022)



Self-pressurization of LH2 tanks

Table 1 – Experimental results of MHTB tank [23].

Number	ϕ	Initial pressure (kPa)	Terminated pressure (kPa)	Holding time (s)	Q_{tot} (W)
#1	25%	122.1	137.8	66,100	18.8
#2	90%	113.3	137.1	24,560	71.3
#3	50%	122.1	137.8	66,115	18.7
#4	50%	111.6	137.2	48,970	51.8



$$\frac{\dot{q}_V}{\dot{q}_L} = 0.5$$

DISCHA results



Tank to tank transfer

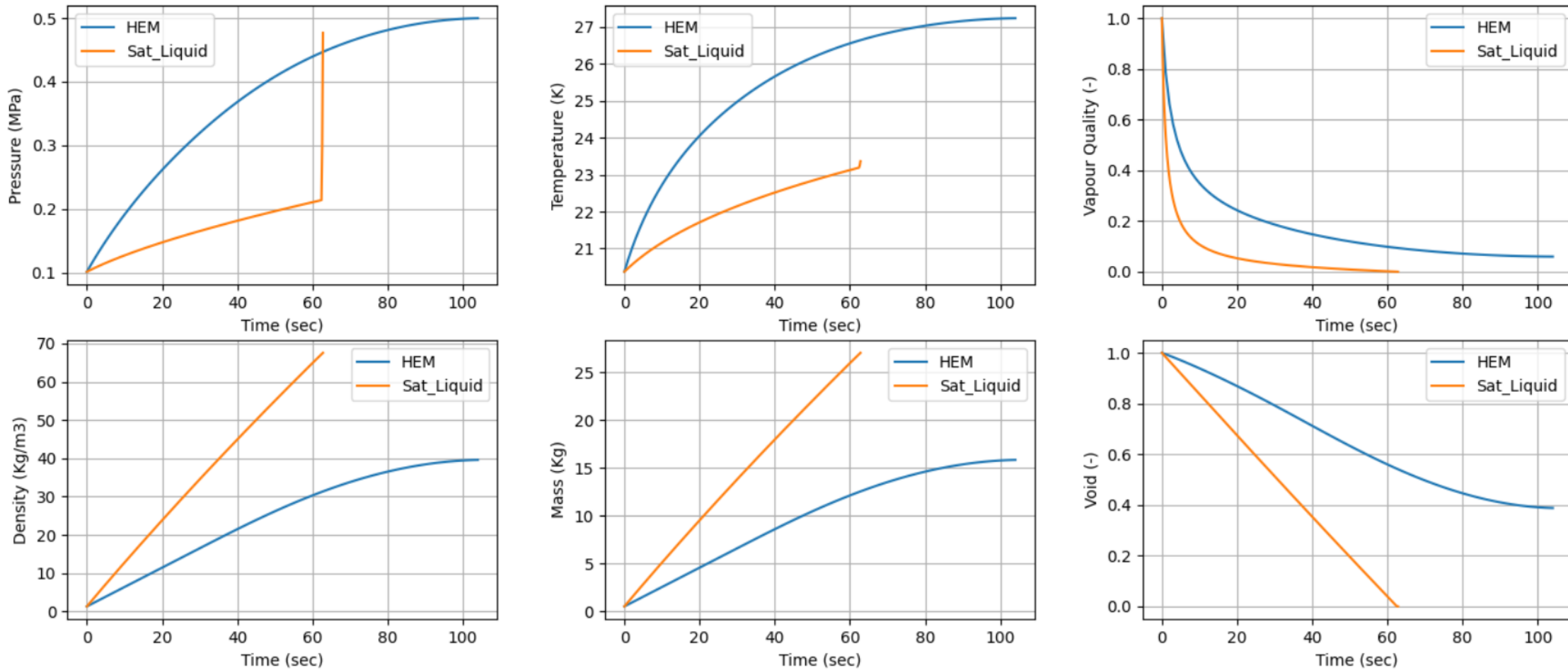
Filling of an LH2 truck from a stationary tank

Tank to tank transfer

- System Components:
 - Supply tank 12 m³ (2 or 5 bars, sat LH₂)
 - Transfer line (30 m, 2.54 cm) + (5 cm, 10mm) nozzle
 - Receiving tank 0.4 m³ (1 atm, sat GH₂) without vent
 - All components are considered adiabatically isolated.
- Models:
 - Constant pressure for supply tank (2 or 5 bars abs.)
 - HEM or Sat_Liquid for transfer line
 - Single zone or Multizone for receiving tank

5 bars, single zone

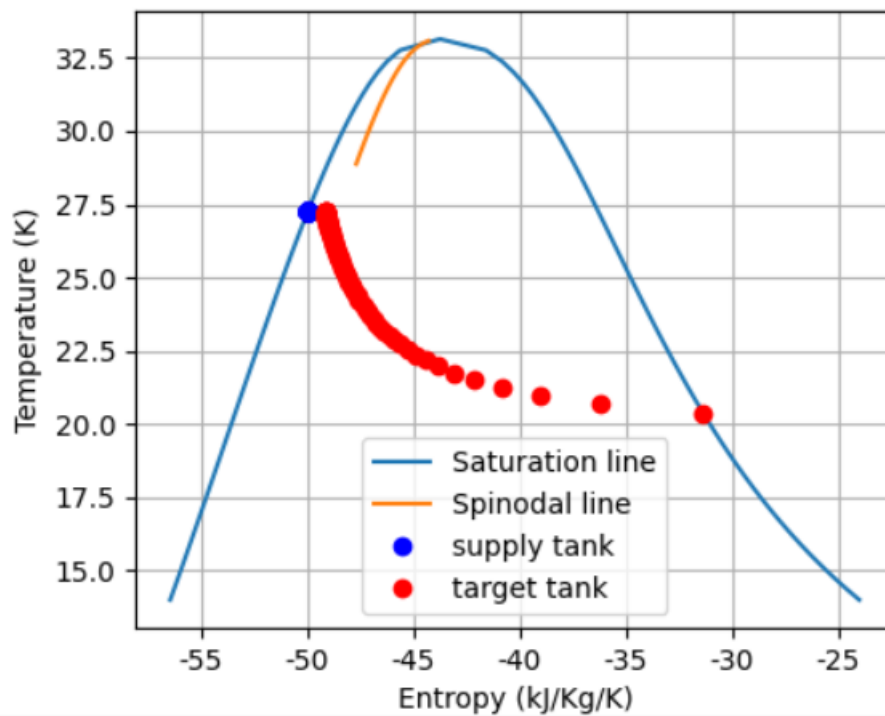
Predicted conditions in receiving tank



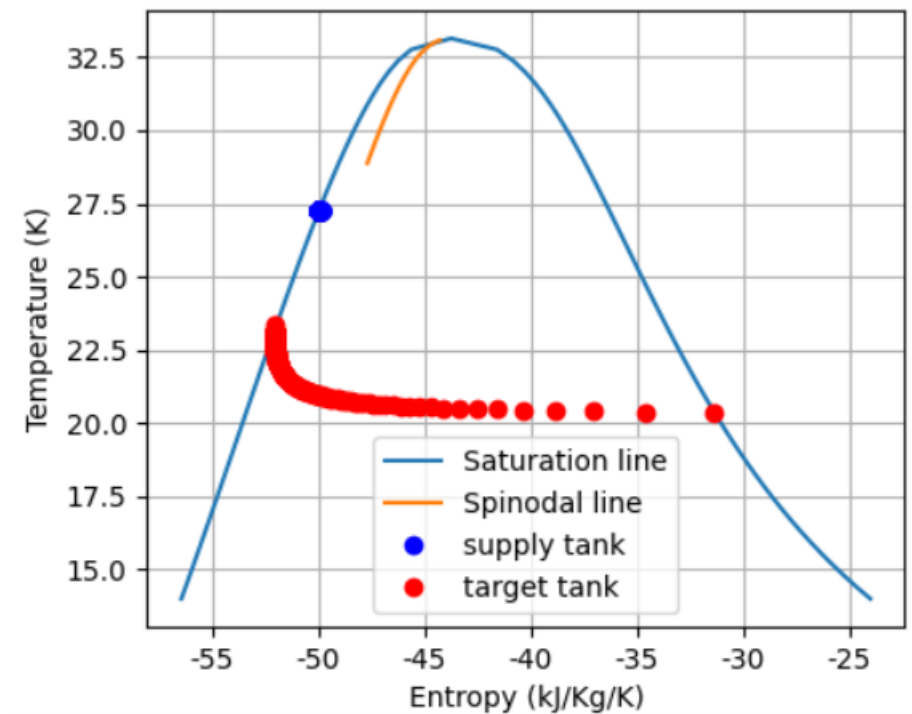
5 bars, single zone

Filling path on the TS chart

HEM

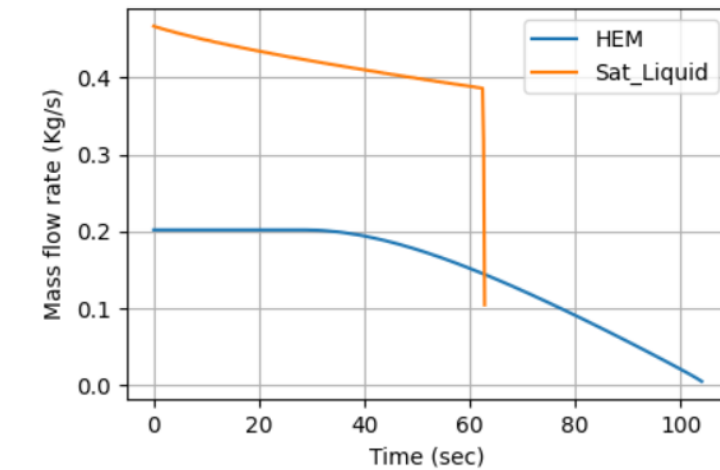
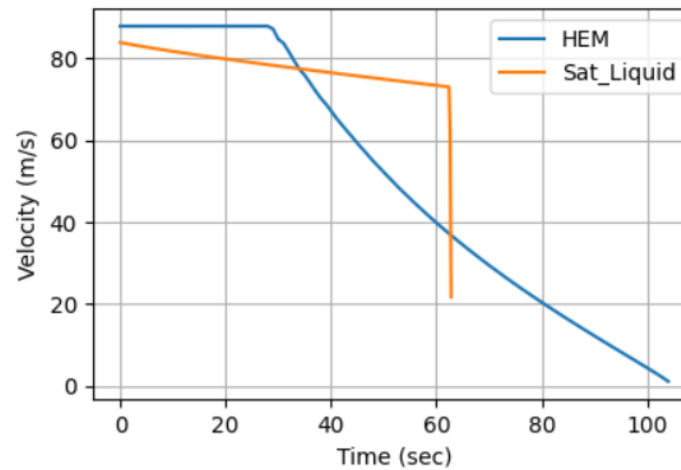
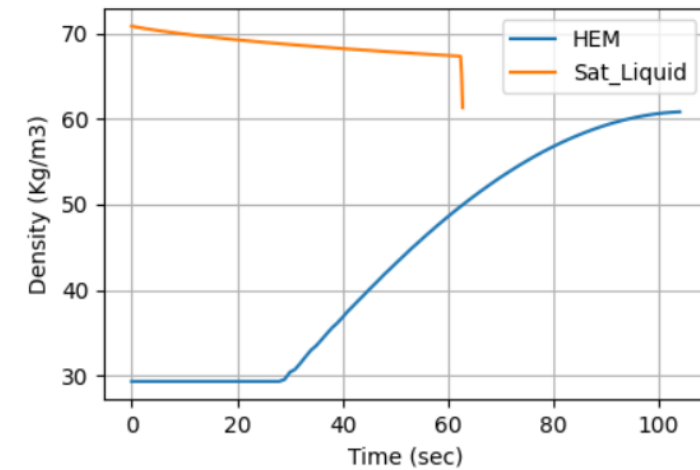
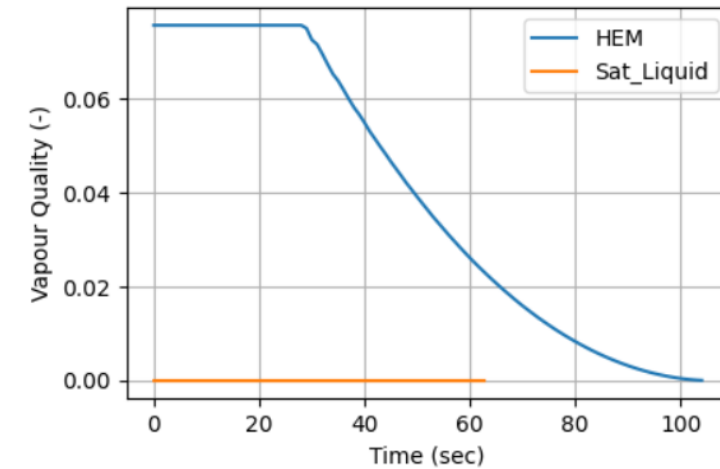
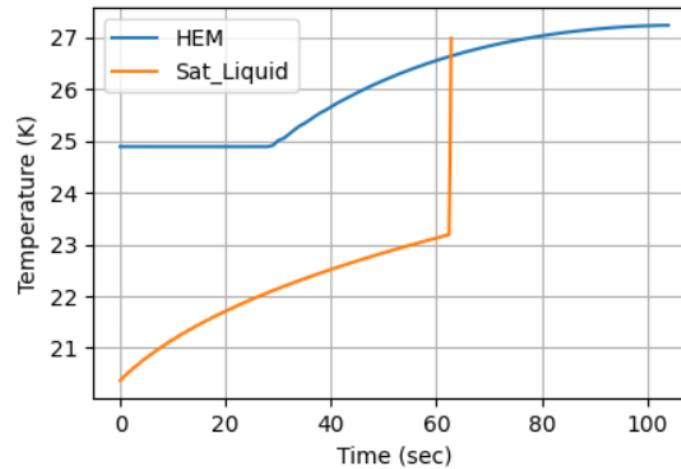
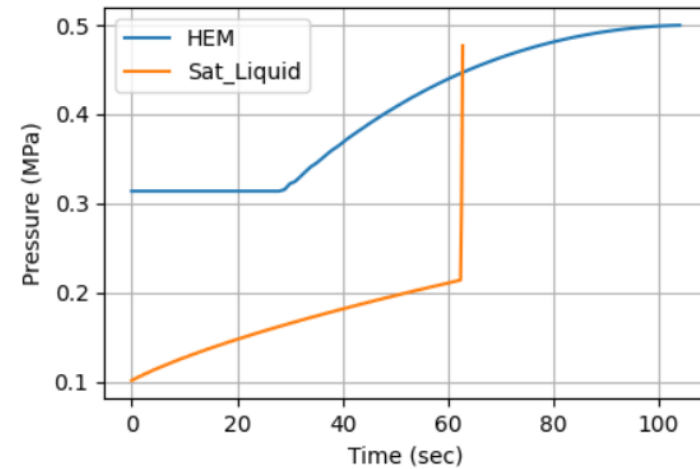


Sat_Liquid



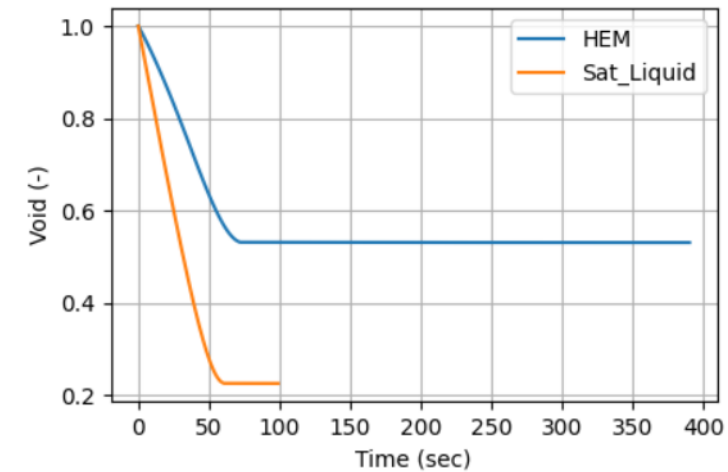
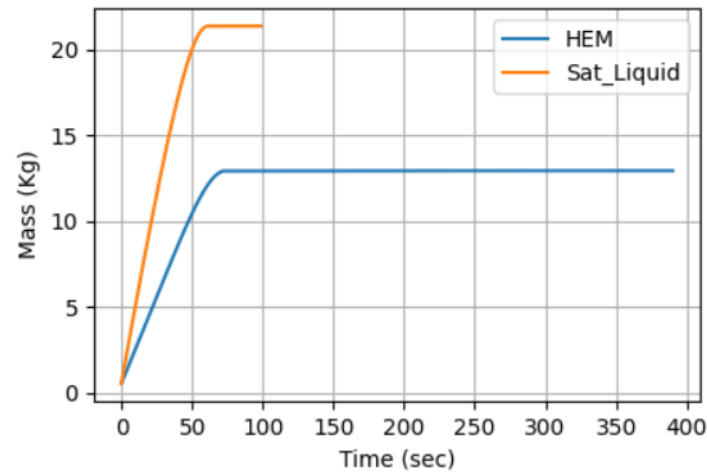
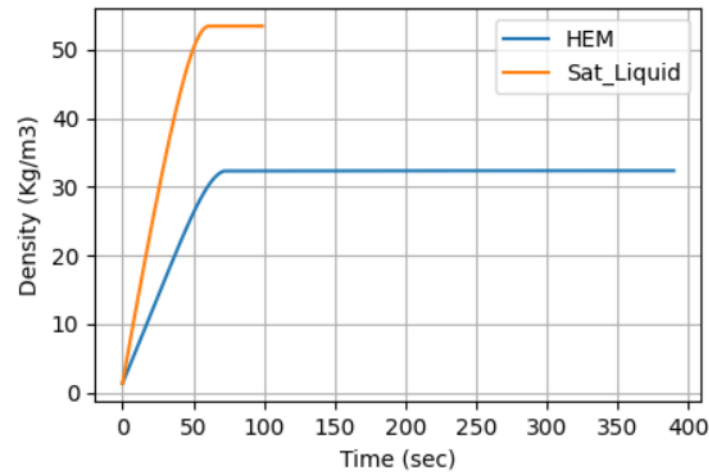
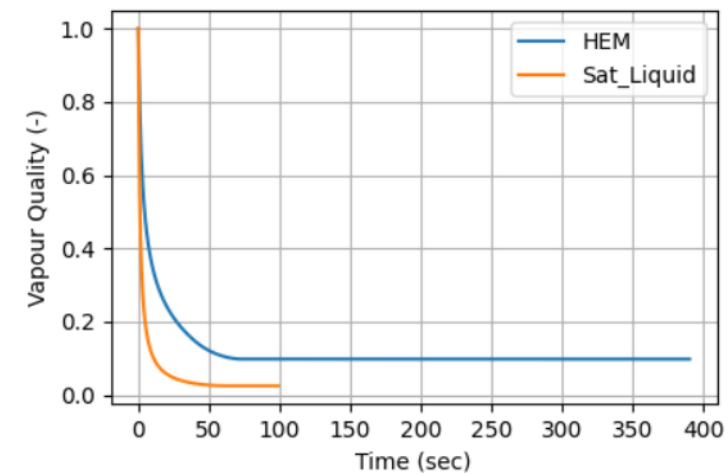
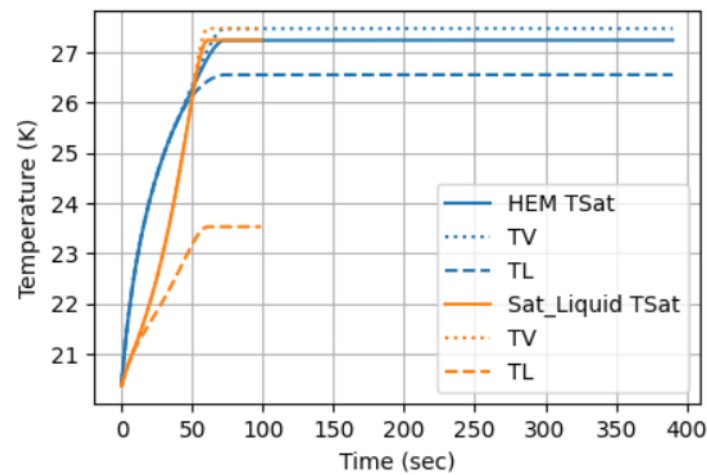
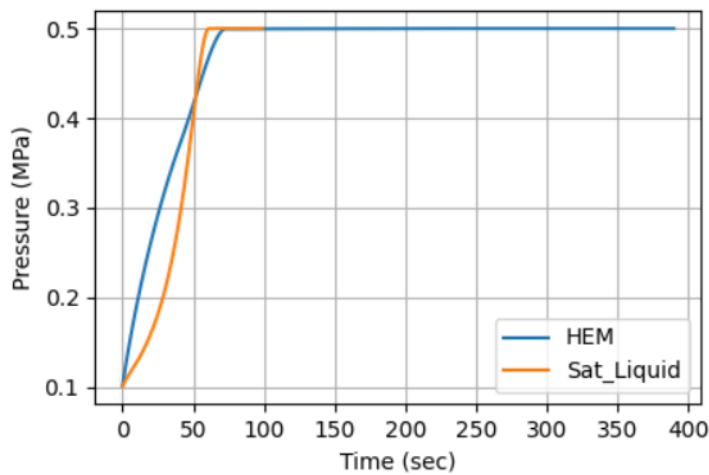
5 bars, single zone

Predicted conditions at transfer line exit



5 bars, Multizone

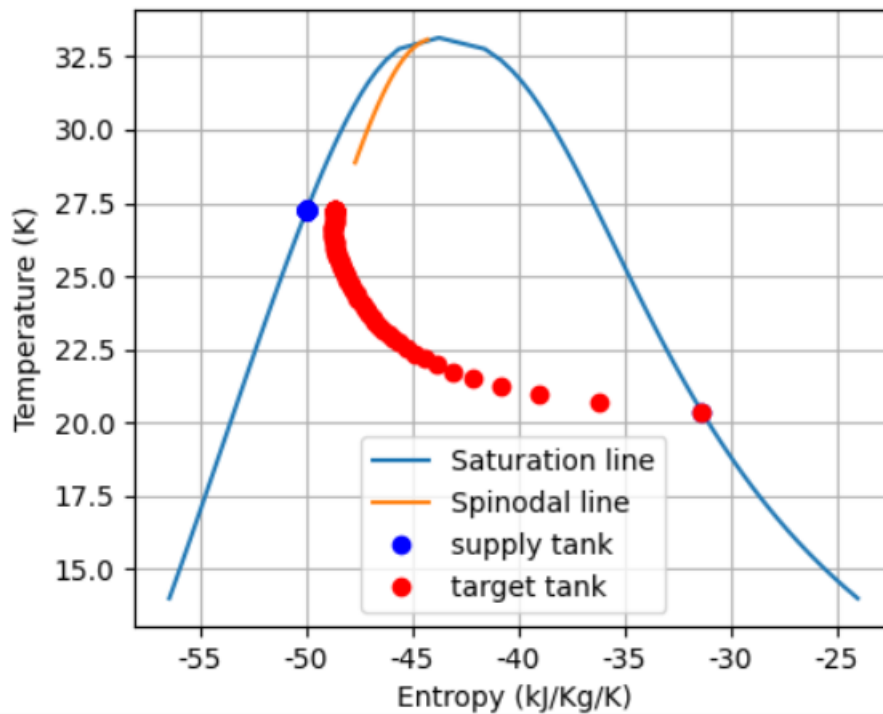
Predicted conditions in receiving tank



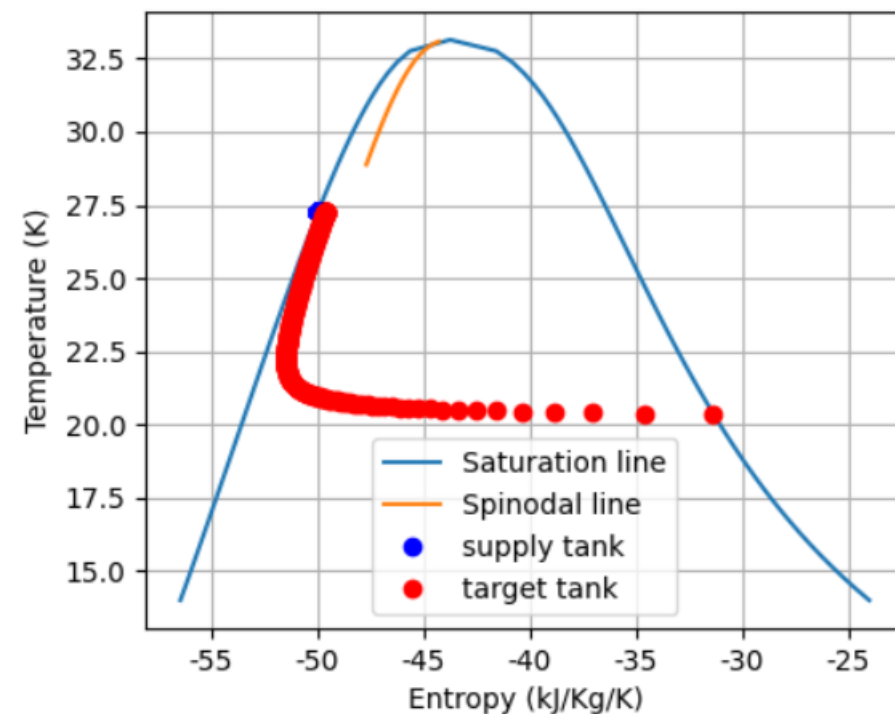
5 bars, Multizone

Filling path on the TS chart

HEM

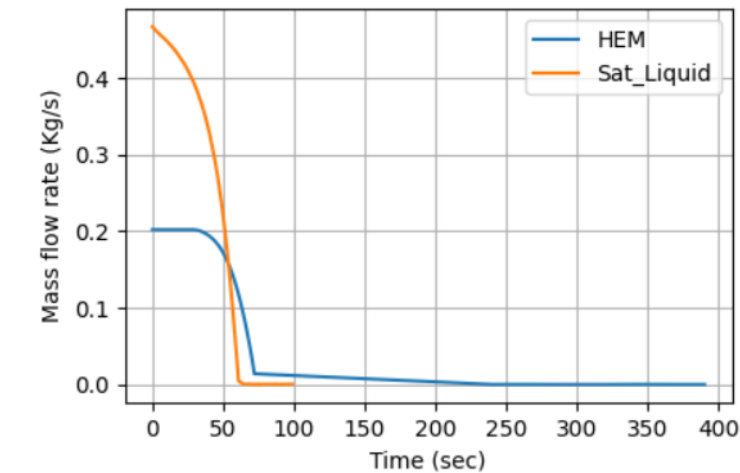
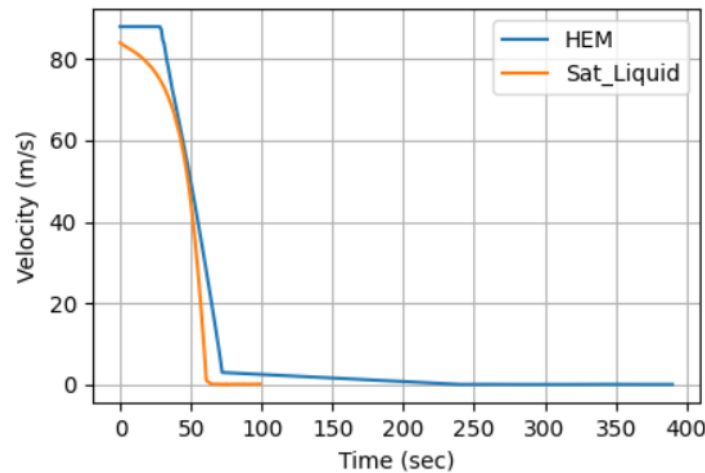
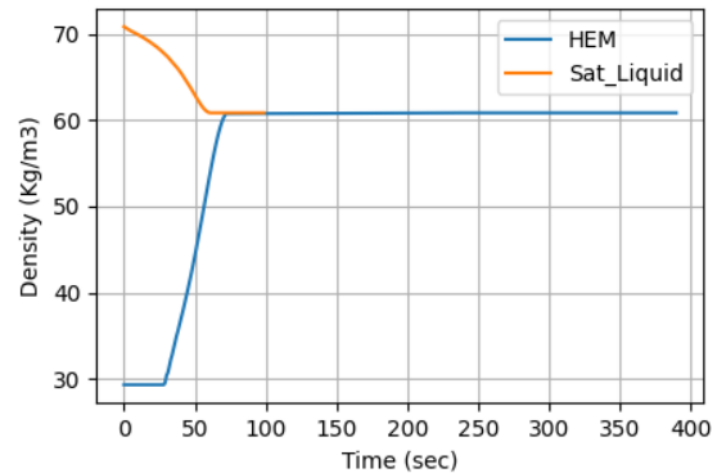
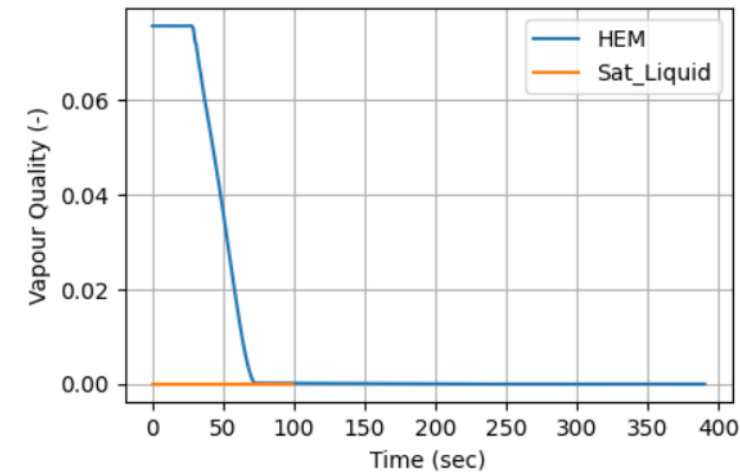
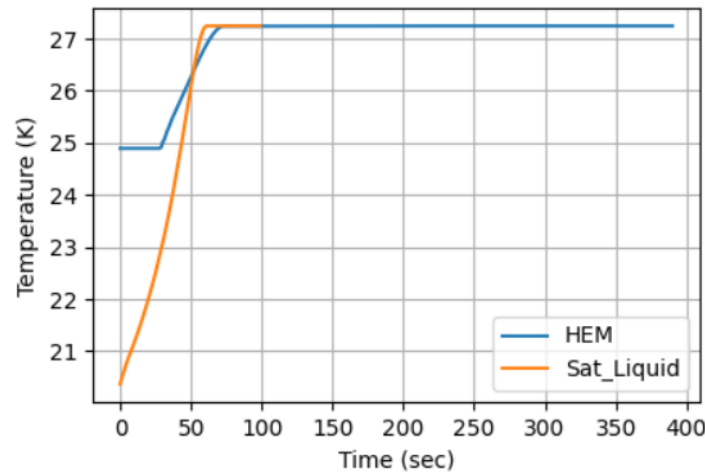
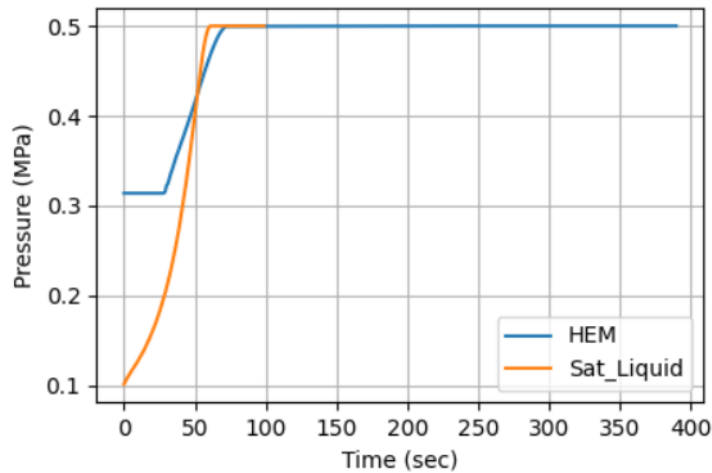


Sat_Liquid



5 bars, Multizone

Predicted conditions at transfer line exit



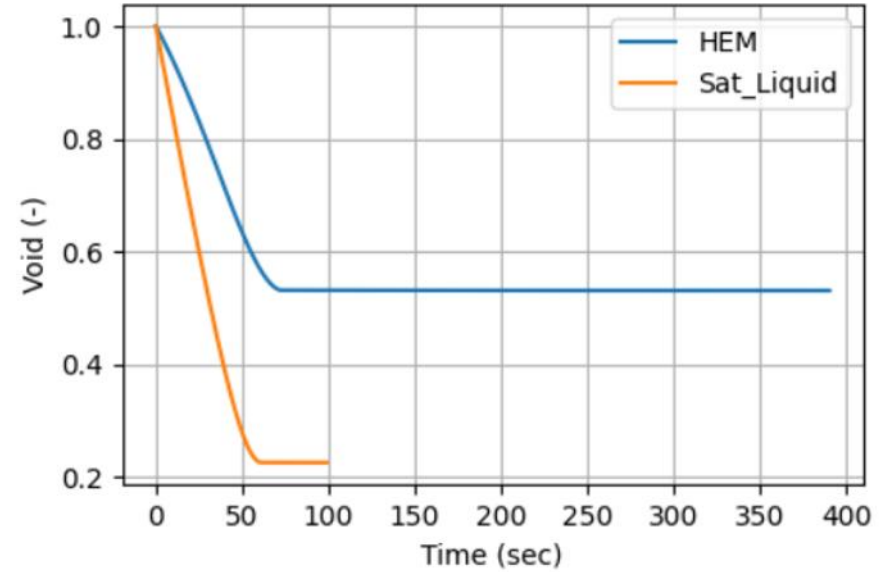
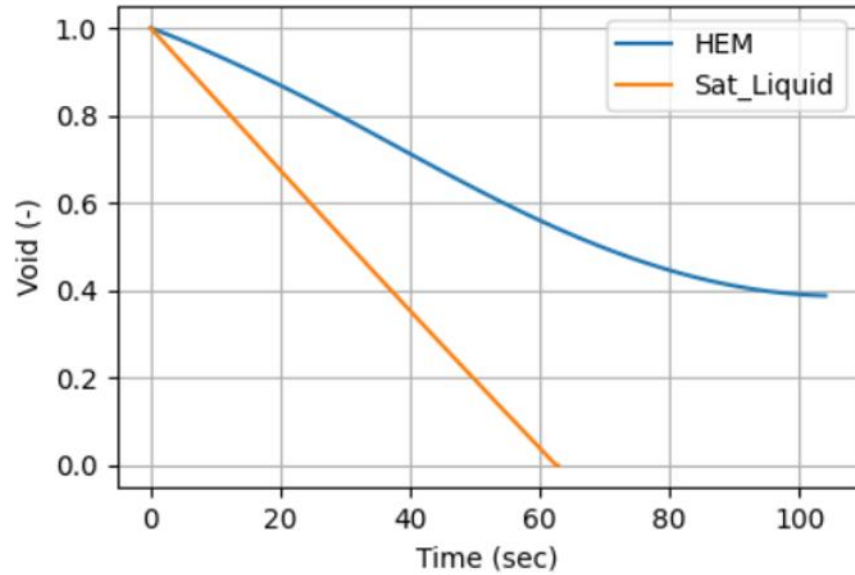
Predicted conditions in receiving tank



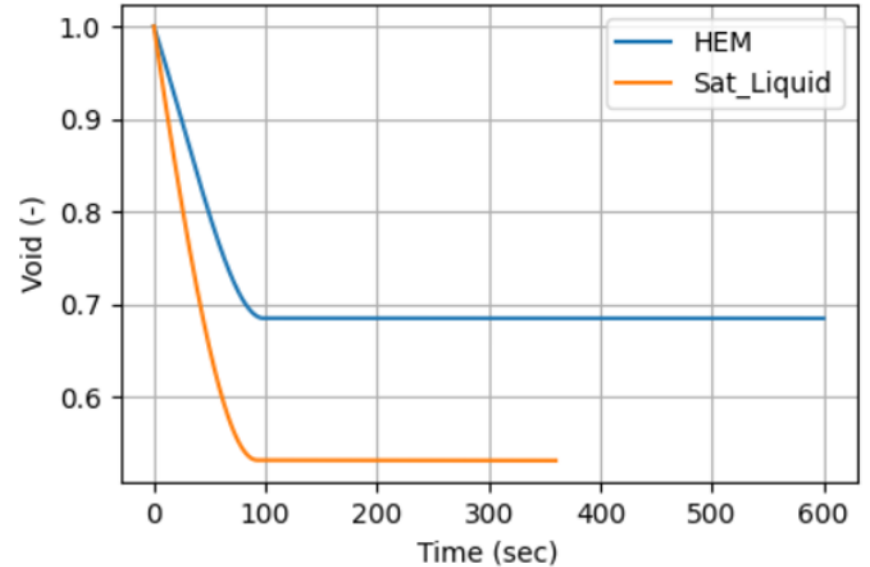
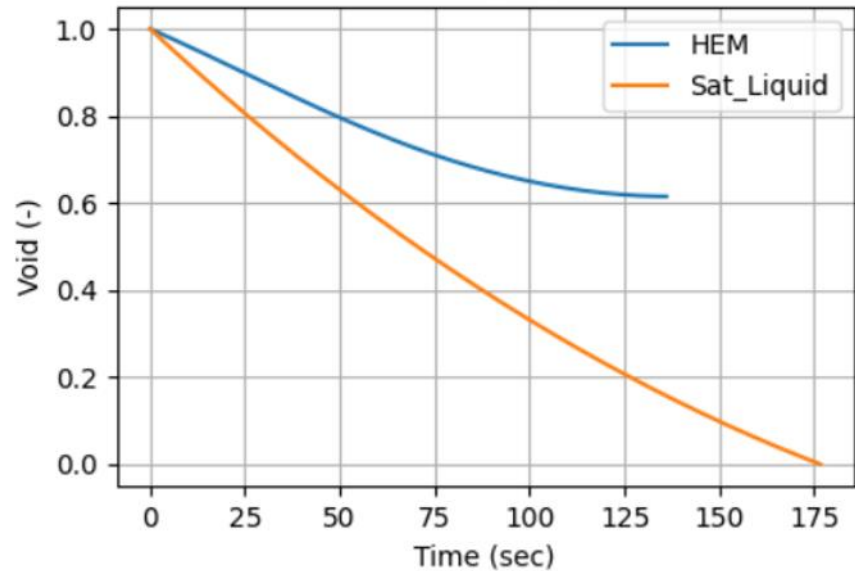
Single zone

Multi-zone

5 bars



2 bars



Conclusions



- DISCHA validation
 - Reasonable agreement against LLNL LH2 boiloff tests and NASA MHTB pressurization tests
- Tank to tank transfer simulations
 - Non-vented pressure filling can be blocked (never reach 100% fill). Happens always with HEM.
 - Multizone model gives lower tank fill compared to single zone
 - Saturated liquid in transfer line gives higher fill compared to HEM
 - Higher pressure gives generally larger fills



Acknowledgments



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