

## INSTRUCTIONS FOR USING CONVERSION FACTOR TABLES

The tables that follow include conversion factors that are useful to the RCT. They are useful in making a single conversion from one unit to another by using the guide arrows at the top of the page in accordance with the direction of the conversion. However, when using the tables to develop equivalent fractions for use in unit analysis equations, a better understanding of how to read the conversion factors given in the table is required.

The conversions in the table have been arranged by section in the order of fundamental units, followed by derived units:

- Length
- Mass
- Time
- Area
- Volume
- Density
- Radiological
- Energy
- Fission
- Miscellaneous (Temperature, etc.)

The easiest way to read a conversion from the table is done as follows. Reading *left to right*, "one (1) of the units in the left column is equal to the number in the center column of the unit in the right column." For example, look at the first conversion listed under **Length**. This conversion would be read from left to right as "1 angstrom is equal to E-8 centimeters," or

$$1\text{\AA} = 10^{-8} \text{ centimeters} \Rightarrow \frac{1\text{\AA}}{10^{-8} \text{ centimeters}}$$

Another conversion would be read from left to right as "1 millimeter (mm) is equal to 1E-1 centimeters," or  $1\text{ mm} = 0.1\text{ cm}$ . This method can be applied to any of the conversions listed in these tables when reading *left to right*.

If reading *right to left*, the conversion should be read as "one (1) of the unit in the right column is equal to the inverse of (1 over) the number in the center column of the unit in the left column." For example, using the conversion shown previously, the conversion reading right to left would be "1 inch is equal to the inverse of 3.937E-5 (1/3.937E-5) micrometers," or

$$1\text{inch} = \frac{1}{3.937E-5\mu\text{m}} = 2.54E4\mu\text{m}$$

Multiply # of to obtain # of	by	to obtain # of Divide # of
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**Length**

angstroms ( $\text{\AA}$ )	$10^{-8}$	Cm
$\text{\AA}$	$10^{-10}$	M
micrometer ( $\mu\text{m}$ )	$10^{-3}$	Mm
$\mu\text{m}$	$10^{-4}$	Cm
$\mu\text{m}$	$10^{-6}$	M
$\mu\text{m}$	$3.937 \times 10^{-5}$	in.
mm	$10^{-1}$	Cm
cm	0.3937	in.
cm	$3.2808 \times 10^{-2}$	Ft
cm	$10^{-2}$	M
m	39.370	in.
m	3.2808	Ft
m	1.0936	Yd
m	$10^{-3}$	Km
m	$6.2137 \times 10^{-4}$	Miles
km	0.62137	Miles
mils	$10^{-3}$	in.
mils	$2.540 \times 10^{-3}$	Cm
in.	$10^3$	Mils
in.	2.5400	Cm
ft	30.480	Cm
rods	5.500	Yd
miles	5280	Ft
miles	1760	Yd
miles	1.6094	Km

Multiply # of to obtain # of	by	to obtain # of Divide # of
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**Mass**

mg	$10^{-3}$	G
mg	$3.527 \times 10^{-5}$	oz avdp
mg	$1.543 \times 10^{-2}$	Grains
g	$3.527 \times 10^{-2}$	oz avdp
g	$10^{-3}$	Kg
g	980.7	Dynes
g	$2.205 \times 10^{-3}$	Lb
kg	2.205	Lb
kg	0.0685	Slugs
kg	$9.807 \times 10^5$	Dynes
lb	$4.448 \times 10^5$	Dynes
lb	453.592	G
lb	0.4536	Kg
lb	16	oz avdp
lb	0.0311	Slugs
dynes	$1.020 \times 10^{-3}$	G
dynes	$2.248 \times 10^{-6}$	Lb
u (unified-- <sup>12</sup> C scale)	$1.66043 \times 10^{-27}$	Kg
amu (physical-- <sup>16</sup> O scale)	$1.65980 \times 10^{-27}$	Kg
oz	28.35	G
oz	$6.25 \times 10^{-2}$	Lb

Note: Mass to energy conversions under miscellaneous

<b>Multiply # of to obtain # of</b>	<b>by</b>	<b>to obtain # of Divide # of</b>
	<b>by</b>	

**Time**

days	86,400	Sec
days	1440	Min
days	24	Hours
years	$3.15576 \times 10^7$	Sec
years	525,960	Min
years	8766	Hr
years	365.25	Days

**Area**

barns	$10^{-24}$	$\text{cm}^2$
circular mils	$7.854 \times 10^{-7}$	$\text{in.}^2$
$\text{cm}^2$	$10^{24}$	Barns
$\text{cm}^2$	0.1550	$\text{in.}^2$
$\text{cm}^2$	$1.076 \times 10^{-3}$	$\text{ft}^2$
$\text{cm}^2$	$10^{-4}$	$\text{m}^2$
$\text{ft}^2$	929.0	$\text{cm}^2$
$\text{ft}^2$	144	$\text{in.}^2$
$\text{ft}^2$	$9.290 \times 10^{-2}$	$\text{m}^2$
$\text{in.}^2$	6.452	$\text{cm}^2$
$\text{in.}^2$	$6.944 \times 10^{-3}$	$\text{ft}^2$
$\text{in.}^2$	$6.452 \times 10^{-4}$	$\text{m}^2$
$\text{m}^2$	1550	$\text{in.}^2$
$\text{m}^2$	10.76	$\text{ft}^2$
$\text{m}^2$	1.196	$\text{yd}^2$
$\text{m}^2$	$3.861 \times 10^{-7}$	sq mi

Multiply # of to obtain # of	by by	to obtain # of Divide # of
<b>Volume</b>		
cm <sup>3</sup> (cc)	0.99997	ml
cm <sup>3</sup>	$6.1023 \times 10^{-2}$	in. <sup>3</sup>
cm <sup>3</sup>	$10^{-6}$	m <sup>3</sup>
cm <sup>3</sup>	$9.9997 \times 10^{-4}$	Liters
cm <sup>3</sup>	$3.5314 \times 10^{-5}$	ft <sup>3</sup>
m <sup>3</sup>	35.314	ft <sup>3</sup>
m <sup>3</sup>	$2.642 \times 10^2$	Gal
m <sup>3</sup>	$9.9997 \times 10^2$	Liters
in. <sup>3</sup>	16.387	cm <sup>3</sup>
in. <sup>3</sup>	$5.787 \times 10^{-4}$	ft <sup>3</sup>
in. <sup>3</sup>	$1.639 \times 10^{-2}$	Liters
in. <sup>3</sup>	$4.329 \times 10^{-3}$	Gal
ft <sup>3</sup>	$2.832 \times 10^{-2}$	m <sup>3</sup>
ft <sup>3</sup>	7.481	Gal
ft <sup>3</sup>	28.32	Liters
ft <sup>3</sup>	1728	in. <sup>3</sup>
gal (U.S.)	231.0	in. <sup>3</sup>
gal	0.13368	ft <sup>3</sup>
liters	33.8147	fluid oz
liters	1.05671	Quarts
liters	0.26418	Gal
gm moles (gas)	22.4	liters (s.t.p.)

Multiply # of to obtain # of	by by	to obtain # of Divide # of
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**Density**

cm <sup>3</sup> /g	$1.602 \times 10^{-2}$	ft <sup>3</sup> /lb
ft <sup>3</sup> /lb	62.43	cm <sup>3</sup> /g
g/cm <sup>3</sup>	62.43	lb/ft <sup>3</sup>
lb/ft <sup>3</sup>	$1.602 \times 10^{-2}$	g/cm <sup>3</sup>
lb/in. <sup>3</sup>	27.68	g/cm <sup>3</sup>
lb/gal	0.1198	g/cm <sup>3</sup>

**Radiological Units**

becquerel	$2.703 \times 10^{-11}$	Curies
curies	$3.700 \times 10^{10}$	dis/sec
curies	$2.220 \times 10^{12}$	dis/min
curies	$10^3$	Millicuries
curies	$10^6$	Microcuries
curies	$10^{12}$	Picocuries
curies	$10^{-3}$	Kilocuries
curies	$3.700 \times 10^{10}$	Becquerel
dis/min	$4.505 \times 10^{-10}$	Millicuries
dis/min	$4.505 \times 10^{-7}$	Microcuries
dis/sec	$2.703 \times 10^{-8}$	Millicuries
dis/sec	$2.703 \times 10^{-5}$	Microcuries
kilocuries	$10^3$	Curies
microcuries	$3.700 \times 10^4$	dis/sec
microcuries	$2.220 \times 10^6$	dis/min
millicuries	$3.700 \times 10^7$	dis/sec
millicuries	$2.220 \times 10^9$	dis/min
R	$2.58 \times 10^{-4}$	C/kg of air
R	1	esu/cm <sup>3</sup> of air (s.t.p.)
R	$2.082 \times 10^9$	ion prs/cm <sup>3</sup> of air (s.t.p.)

Multiply # of to obtain # of	by by	to obtain # of Divide # of
<b><u>Radiological Units (continued)</u></b>		
R	$1.610 \times 10^{12}$	ion prs/g of air
R (33.7 eV/ion pr.)	$7.02 \times 10^4$	MeV/cm <sup>3</sup> of air (s.t.p.)
R (33.7 eV/ion pr.)	$5.43 \times 10^7$	MeV/g of air
R (33.7 eV/ion pr.)	86.9	ergs/g of air
R (33.7 eV/ion pr.)	$2.08 \times 10^{-6}$	g-cal/g of air
R (33.7 eV/ion pr.)	.98	ergs/g of soft tissue
rads	0.01	Gray
rads	0.01	J/kg
rads	100	ergs/g
rads	$8.071 \times 10^4$	MeV/cm <sup>3</sup> or air (s.t.p.)
rads	$6.242 \times 10^7$	MeV/g
rads	$10^{-5}$	watt-sec/g
rads (33.7 eV/ion pr.)	$2.39 \times 10^9$	ion prs/cm <sup>3</sup> of air (s.t.p.)
gray	100	Rad
rem	0.01	Sievert
sievert	100	Rem
$\mu\text{Ci}/\text{m}^3$ ( $\mu\text{Ci}/\text{ml}$ )	$2.22 \times 10^{12}$	dpm/m <sup>3</sup>
$\mu\text{Ci}/\text{cm}^3$	$2.22 \times 10^9$	dpm/liter
dpm/m <sup>3</sup>	0.4505	pCi/m <sup>3</sup>

**Energy**

Btu	$1.0548 \times 10^3$	joules (absolute)
Btu	0.25198	kg-cal
Btu	$1.0548 \times 10^{10}$	Ergs
Btu	$2.930 \times 10^{-4}$	kW-hr
Btu/lb	0.556	g-cal/g
eV	$1.6021 \times 10^{-12}$	Ergs

Multiply # of to obtain # of	by by	to obtain # of Divide # of
<b><u>Energy (continued)</u></b>		
eV	$1.6021 \times 10^{-19}$	joules (abs)
eV	$10^{-3}$	keV
eV	$10^{-6}$	MeV
ergs	$10^{-7}$	joules (abs)
ergs	$6.2418 \times 10^5$	MeV
ergs	$6.2418 \times 10^{11}$	eV
ergs	1.0	dyne-cm
ergs	$9.480 \times 10^{-11}$	Btu
ergs	$7.375 \times 10^{-8}$	ft-lb
ergs	$2.390 \times 10^{-8}$	g-cal
ergs	$1.020 \times 10^{-3}$	g-cm
gm-calories	$3.968 \times 10^{-3}$	Btu
gm-calories	$4.186 \times 10^7$	Ergs
joules (abs)	$10^7$	Ergs
joules (abs)	0.7376	ft-lb
joules (abs)	$9.480 \times 10^{-4}$	Btu
g-cal/g	1.8	Btu/lb
kg-cal	3.968	Btu
kg-cal	$3.087 \times 10^3$	ft-lb
ft-lb	1.356	joules (abs)
ft-lb	$3.239 \times 10^{-4}$	kg-cal
kW-hr	$2.247 \times 10^{19}$	MeV
kW-hr	$3.60 \times 10^{13}$	Ergs
MeV	$1.6021 \times 10^{-6}$	Ergs

Note: Mass to energy conversions under miscellaneous

Multiply # of to obtain # of	by by	to obtain # of Divide # of
<b>Fission</b>		
Btu	$1.28 \times 10^{-8}$	grams $^{235}\text{U}$ fissioned <sup>b</sup>
Btu	$1.53 \times 10^{-8}$	grams $^{235}\text{U}$ destroyed <sup>b,c</sup>
Btu	$3.29 \times 10^{13}$	Fissions
fission of 1 g $^{235}\text{U}$	1	megawatt-days
fissions	$8.9058 \times 10^{-18}$	kilowatt-hours
fissions <sup>b</sup>	$3.204 \times 10^{-4}$	Ergs
kilowatt-hours	$2.7865 \times 10^{17}$	$^{235}\text{U}$ fission neutrons
kilowatts per kilogram $^{235}\text{U}$	$2.43 \times 10^{10}$	average thermal neutron flu× in fuel <sup>b,d</sup>
megawatt-days per ton U	$1.174 \times 10^{-4}$	% U atoms fissioned <sup>e</sup>
megawatts per ton U	$2.68 \times 10^{10}/E^f$	average thermal neutron flu× in fuel <sup>b</sup>
neutrons per kilobarn	$1 \times 10^{21}$	neutrons/cm <sup>2</sup>
watts	$3.121 \times 10^{10}$	fissions/sec

<sup>b</sup> At 200 MeV/fission.

<sup>c</sup> Thermal neutron spectrum ( $\alpha = 0.193$ ).

<sup>d</sup>  $\delta$ (fission = 500 barns).

<sup>e</sup> At 200 MeV fission, in  $^{235}\text{U}$ - $^{238}\text{U}$  mixture of low  $^{235}\text{U}$  content.

<sup>f</sup> E = enrichment in grams  $^{235}\text{U}$ /gram total. No other fissionable isotope present.

**Multiply # of  
to obtain # of**      **by**      **to obtain # of  
Divide # of**

**Miscellaneous**

radians	57.296	Degrees
eV	$1.78258 \times 10^{-33}$	Grams
eV	$1.07356 \times 10^{-9}$	U
erg	$1.11265 \times 10^{-21}$	Grams
proton masses	938.256	MeV
neutron masses	939.550	MeV
electron masses	511.006	keV
u (amu on $^{12}\text{C}$ scale)	931.478	MeV

**Temperature**

$$^{\circ}\text{C} = \frac{(^{\circ}\text{F} - 32)}{1.8}$$

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) \left( \frac{5}{9} \right)$$

$$^{\circ}\text{F} = 1.8(^{\circ}\text{C}) + 32$$

$$^{\circ}\text{F} = \left( \frac{9}{5} \right) (^{\circ}\text{C}) + 32$$

$$^{\circ}\text{K} = ^{\circ}\text{C} + 273.16$$

**Wavelength to Energy Conversion**

$$\text{keV} = 12.40/\text{\AA}$$

$$\text{eV} = 1.240 \times 10^{-6}/\text{m}$$